WATER RESOURCES STUDY

Metropolitan Spokane Region

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APPENDIX B
Geology and Ground Cater

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LIST OF REPORTS AND APPENDICES

REPORTS

Summary Report

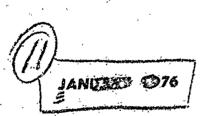
Technical Report

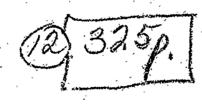
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•	GEOLOGY AND GROUND
С	Water Use
D	Wastewater Generation and Treatment
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F	Demographic and Economic Characteristics
G	Planning Criteria
H (Volume 1)	Plan Formulation and Evaluation
H (Volume 2)	Plan Formulation and Evaluation
I	Institutional Analysis
J	Water Quality Simulation Model

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METROPOLITAN SPOKANE REGION WATER RESOURCES STUDY.

APPENDIX P GEOLOGYAND GROUNDWAITER





Dopartment of the Army Corps of Engineers, Scattle District Konnedy-Tudor Consulting Engineers

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ACKNOWLEDGEMENTS

The Netropolitan Spokane Region Water Resources study was accomplished by the Seattle District, U.S. Army Corps of Ingineers assisted by Kennedy-Tudor Consulting Engineers under sponsorship of the Spokane Regional Planning Inference. Technical guidance was provided by the Spokane River Basin Coordinating Committee, with general guidance from the study's citizens ammittee. Major cooperating agencies include Spokane City and County, and the Washington State Department of Ecology. The study was coordinated with appropriate Federal and State agencies and with the general public within the metropolitan Spokane area.

The summary report was prepared by the Seattle District Corps of Engineers. The technical report and appendices were prepared for the Seattle District, Corps of Engineers by Kennedy-Tudor Consulting Engineers.

PREFACE

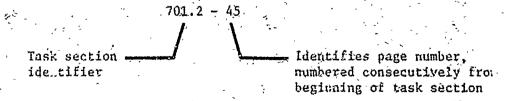
With the enactment of the Federal Water Pollution Control Act Amendment of 1972 (Public Law 92-500), new national goals have been established for the elimination of pollution discharges into our streams and lakes. This appendix is a part of the report prepared to assist local government in satisfying State and Federal Regultements relating to Public Law 92-500. The aggestions contained in this sport are for implementation by local interests with available assistance from other local, State and Federal agencies. The study suggests a regional wastewater management plan for the metropolitan Spokane urban area and provides major input to Washington State Department of Ecology Section 302e plans for the Spokane River Basin in Washington State. Also included in the study are planning suggestions for urban runoff and flood control, and the protection of the area's water supply resources.

As listed on the inside front cover, documentation for this study consists of a Summary Report and a Technical Report with supporting Appendices A through J.

The Tachnical Report summarizes Appendices A through 1, which contain 58 individual task section reports prepared during the study. These task sections are listed by title in Attachment I of the Technical Report. Generally, the numbering of appendix tack sections reflects the following system:

Study Task Sections	Type of Study Activity
308,8	Data Gollection
40018	Data Evaluation and Projection
500! \$	Identification of Unnet Needs
600Ts	Development of Alternative Plans
700's	Evaluation Comperison and Selection of Plans
800 ts	Institutional Arrangements

Pages within each appendix are numbered by task section, as illus rated below:



APPENDIX 8 - SEOLOGY AND GROUNDWATER

TASK SECTION

TITLE

Pagi

OGeology, Soils and Groundwater

303-1 to 303-95

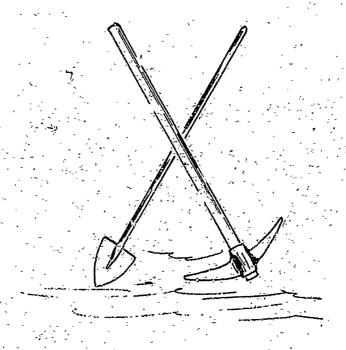
405 Groundwater Quality Dates and

405-1 to 405-90

608.1 The Effect of Surface Applied Waters on Groundwater Quantity in Spokane Valley .

608.1-1 to 608.1-42

A detailed index for each task section precedes the respective section text.



SECTION 303

Geology, soils and Groundwater



WATER RESOURCES STUDY

METROPOLITAN SPOKANE REGION

SECTION 303

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GEOLOGY AND GROUNDWATER

Prepared by Shannon & Wilson, Inc. in cooperation with Kennedy-Tudor Consulting Engineers

1 December 1974



Department of the Army, Seattle District Corps of Engineers Kennedy-Tudor Consulting Engineers

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 $[\]star$ All plates are large drawings bound at the end of this section.

I. INTRODUCTION

Purpose and Scope

The purpose of this report and the accompanying geologic and groundwater maps is to provide information on the geology, soils and groundwater resources of the study area.

The study area extends throughout the Washington State portion of the Spokane River Basin and consists of approximately 2,300 square miles of mountains, valleys and plateaus in the east-central portion of the State. Parts of Spokane, Stevens, Lincoln, Pend Oreille, and Whitman Counties are included. Particular attention is focused on the urban planning area centered on the City of Spokane, an area which occupies about 11 percent of the total study area.

A primary objective of this task is the preparation of general geology and groundwater maps at 1:250,000 scale (approximately 1 inch equal to 4 miles) to show in generalized form the gross geologic and groundwater features of the entire study area. This level of detail is appropriate for the portions of the study area lying outside the urban planning area. The needs of the study call for a more detailed analysis and presentation of information descriptive of the urban area. To achieve this, mapping of near-surface geology, soil characteristics and groundwater features at a scale of 1:24,000 (approximately 1 inch equal to 2000 feet) is presented for this area. The maps are supplemented by text and appendices. The text describes the history and present day features of the area in terms of geology, soils and groundwater. Spe-

cial attention is given to those features which are significant to water and wastewater management.

Method of Study

Research within the study area was based upon a review of available geology, soils and groundwater data gathered in the past by a number of independent investigators. The map sets represent a compilation of these separate efforts, supplemented by imited field reconnaissance. Aerial photographs, used as stereo-pairs, provided supplemental data for the delineation of soil and rock boundaries in unmapped and poorly defined areas. Subsurface explorations, detailed field mapping or testing of subsurface materials were not performed as part of this study.

Preparation of Maps. The information compiled is presented in a series of maps, Plates 303-1 through 35. Of these, two cover the entire study area, showing respectively general geology and groundwater resources. The remaining plates contain detailed coverage of the Spokane urban planning area. Four series of plates are used to present information on each of eight mapping subareas forming the urban area, and one plate presents a legend of terms used on the engineering geology maps.

Background mapping of the urban planning area was prepared at the 1:24,000 scale from the following U.S. Geological Survey quadrangle maps.

1. -- 1:24,000 Spokane N.E. 2. Spokane N.W. -- 1:24,000 Spokane S.E. -- 1:24,000 3. Spokane S.W. -- 1:24,000 4. Airway Heights -- 1:24,000 5. -- 1:62,500 6. Greenacres Mt. Spokane **-- 1:62,500** 7. -- 1:62,500 8. Deer Park 9. Clayton 1:62,500

The area immediately west of the Spokane River between the Hangman Creek confluence and the Little Spokane River confluence is not included in the large scale mapping. This stretch of river follows a natural geological boundary. The area west of the river is in the Columbia Plateau region and is adequately covered under the small-scale study area geology mapping.

The mapping of engineering geology features, Plates 303-3 through 303-11, shows the near-surface distributions and types of earth materials based on their composition and texture. Due to the nature of the source materials, the maps include reference to genetic or bedrock geology in addition to the exposed surface materials, and permit interpretation for purposes associated with engineering planning.

Two sets of maps, Plates 303-12 to 19 and Plates 303-20 to 27 respectively, are presented which show the relative permeability of soils at surface depths to three feet and near-surface depths three to five feet. The purpose of these maps is to illustrate the relative ability of the soils to permit the passage of water and, in conjunction with the other map sets, to assist in the evaluation of potential problems concerning drain fields, waste disposal, groundwater contamination, runoff, or other related concerns.

Plates 303-28 through 35 define the subterranean aquifers in the

Spokane Valley plains, the terraced valley below Spokane Falls, the Hillyard Trough, and the lower valley of the Little Spokane River. Contours of the average groundwater surface elevation are also shown in these areas. Groundwater sources in the peripheral portions of the urban area are mostly minor and, because of map scale limitations, cannot be shown on the groundwater map. These areas are, however, described in the groundwater section of this text.

Sources of Information

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Principal sources which were used in compiling geological and groundwater data are described below. Refer to the List of References.

General Geology of Study Area. When available, the principal source of information is the series of U.S. Geological Survey maps.

These are supplemented by General Soils Maps prepared by the U.S. Soil Conservation Service, covering Spokane, Stevens, Lincoln and Pend Oreille Counties. U.S. Soil Conservation Service (SCS) mapping is the primary source for areas where geological maps are not available. Soil maps, as exemplified by S.C.S. maps, and geological maps, as prepared by U.S.G.S., are made for different purposes and often have different interpretations for similar features. Where these differences occur in the reference maps, an interpretation is made which best fits the needs of this study.

The published reports and maps of previous geological investigations used as reference material included a reconnaissance map by Griggs (1966) and maps and reports by Becraft and Weiss (1963), Cline (1969) and Flint (1936). Reports of many previous investigations of the Pleistocene

glacial history, the Latah and Palouse Formations, clay resource studies, and private well and test logs, are used to develop and confirm map data.

Engineering Geology. Large-scale (1:24,000) mapping of near-surface soil classification and drainage characteristics (permeability) in the urban area are based primarily on information from previous geologic and soils mapping. Modifications are made as necessary to suit the engineering criteria of the project maps. These data are supplemented by air-photo interpretation and limited field reconnaissance.

Previous geologic mapping in the area included the 1:125,000 scale reconnaissance geologic map of the west half of the Spokane quadrangle by Griggs (1966); a map by Cline (1969), in which a portion of the Griggs map was modified to separate certain deposits on the basis of lithology rather than age; and a 1:62,500 scale geologic map of the Greenacres quadrangle by Weis (1968). U.S. Soil Conservation Service maps of Spokane County by Donaldson and Giese (1968) provided data on soil classification and provided assistance in delineation of soil boundaries.

Where possible, formation contacts were drawn on stereo paired air photos. These were checked and completed in the field, where soils classifications were also verified on the basis of available exposures. The boundaries were then transferred to the base map via mylar tracings of the photo maps. In areas that had been previously mapped by others, checks were again made on the boundaries. Since the previous investigations were intended to show somewhat different features than the project maps, some modifications and interpretations were necessary.

In areas where drill log, test pit or other pertinent information

had been previously gathered, further checks were made. In inaccessible mountainous areas, reliance was chiefly upon air photos and previous investigations. No test pits or borings were made during this study.

Information from previous studies on soil permeability as related to grain size, together with information from Spokane County soils maps, provided basic data for the drainage maps.

Groundwater. Information on the groundwater potential of the study area is known in part by published information, but is primarily derived from a general knowledge of regional groundwater conditions and the water bearing characteristics of the various soils and rocks of the study area. Specific sources of historical information are referenced in the text. Water level records from selected wells were used to plot groundwater contours of the aquifers on the maps.

Limitations

As soil and rock formations vary considerably over both vertical and horizontal distances, boundaries indicated on the accompanying maps delineate the predominant soil or rock present in a given area. There may, of course, be several different types of soil or rock in lesser quantities within each boundary. Also, the origin of the materials within each boundary may vary locally from that shown. Furthermore, horizontal changes in soil types are often gradual; therefore, boundaries are drawn at what is considered to be midway between one soil type and another.

Because the reliability of existing available data varies and in some aspects is entirely lacking, the results presented herein reflect con-

siderable experienced judgment. The maps and accompanying discussions are not intended to preclude the need for detailed explorations and study in specific areas, but only to provide a sound basis for overall management planning.

Professional Services

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This entire section on geology and groundwater of the study area was prepared by geologists and groundwater specialists on the staff of Shannon & Wilson, Inc., Geotechnical Consultants.

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II. REGIONAL GEOLOGY AND GROUNDWATER

Geologic Setting

Topography. The Spokane River Basin as a complete hydrologic unit is a roughly elliptically shaped drainage area lying within northern Idaho and northeastern Washington. The total area of the basin is approximately 6,600 square miles. About 2,400 square miles of this lies within the State of Washington, and constitutes the study area. The study area's topography is dominated by the broad east-west trending Spokane Valley in Washington which becomes the Rathdrum Prairie in Idaho. The valley is flanked on the north and south by mountains, plateaus and tributary valleys. These gravelly valley plains slope gently west from mountain valleys in Idaho to the basalt ledges at Spokane Falls. Down valley from Spokane Falls, the valley plains continue only as remnant terraces along the sides of the entrenched course of the Spokane River, which follows a 60-mile canyoned course to its confluence with the Columbia River. Along its course, the Spokane River is fed by a number of tributaries, notably the Little Spokane River, Hangman (Latah) Creek and Chamokane Creek. Numerous small streams, some of them intermittent, feed into these tributaries from the highlands and plateaus.

The topography of the basin varies, from predominantly mountainous in the northern and eastern portions to rolling plains south of the Spokane River and west of Hangman Creek. The plains are a portion of the over 100,000 square mile Columbia Plateau. This plateau is formed of many horizontal lava flows of basalt and is thinly covered in places with low

hills of wind-blown silt (loess) and, in other places with glacial and flood outwash sand and gravel. North of the Spokane River Valley, remnants of the Columbia Basalt flows have formed Five-Mile, Orchard, Pleasant, Peone and Manito Prairies and Orchard and Green Bluffs. Each of these has a veneer deposit of reworked loess covering the basalt capping. Remnants of old granite mountains protrude through the basalt flows and loess south and west of the Spokane Valley.

The region generally north of the Spokane River is composed of the Okanogan bedrock highlands on the west and the Selkirk bedrock highlands to the east. The two highlands areas are separated by the Deer Park Basin and the Little Spokane River valley. The highlands are subdued, mature north-south trending mountain ranges modified by glaciation. Maximum relief in the highland provinces is on the order of 4,500 feet, with local relief of about 500 to 700 feet. Elevation extremes range from 1,289 feet M.S.L. at Lake Roosevelt in the Columbia River Gorge to 5,878 feet M.S.L. on Mount Spokane.

One striking difference between the Columbia Plateau and the Okanogan-Selkirk Highlands is the character of the valleys. The basalt plateau south of the Spokane River has a gentle southward slope but drops abruptly northward at the Plateau edge causing the valleys tributary to the Spokane River to be short in length and high in gradient. These valleys are generally narrow, having been carved by limited seasonal runoff and rimrock spring flow. The topographic characteristics at the plateau edge reveal sharp breakoff slopes dissected by steep, narrow valleys. The valleys are separated by broad, relatively flat plateau segments.

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Northward across the Spokane River in the granitic and metamorphic rocks of the Okanogan-Selkirk Highlands, the valleys are generally broad, rather flat bottomed, and are formed between relatively narrow, north-south trending mountain ranges. The reasons for the differences in the topography north and south of the Spokane River are twofold. The Columbia Plateau basalt flows reached their northward extent against the highlands, and also the southward extent of the Pleistocene glaciation was near the present Spokane River Gurge. Therefore, glaciation helped shape the valleys north of the river, but had little effect on the plateau area.

Differences in the topography of these two areas are reflected in differences in drainage and groundwater conditions. Surface drainage near the plateau edge is locally poor where natural drainage channels are not well established. Ponding of water is evident in small irregular basalt basins and in shallow loess pockets. However, in areas where loess cover is significant and where natural channels lead surface water to the tributary valleys, drainage is good. The drainage in the Okanogan-Selkirk highlands is generally well established along south and southwest flowing tributaries to the Spokane River. Some ponding occurs, however, in some valleys where glacial outwash or till deposits dammed up surface drainage and formed small swamps or lakes.

Other physiographic sub-divisions of the study area include the Little Spokane-Deer Park Basin, the Hangman Creek valley and the Chamokane Creek valley. Numerous natural lakes dot the basin, with Newman, Liberty, Eloika and Diamond Lakes being among the largest. These lakes are mostly

land-locked and are dammed by glacial lake, outwash, or till deposits.

Other small lakes and ponds scattered over the plateau area are generally basins in the plateau basalt and represent an impoundment of water from local surface runoff.

Geologic History. The following brief summary of geologic history is intended to provide insight into the events that have produced the present geology and groundwater situation.

The oldest rocks in the basin are Pre-Cambrian age metamorphic formations which are probably related to the Belt Series of the northern Rocky Mountains. These rock formations are exposed in the northern and central highlands as well as occasional "islands" surrounded by the basalt flows on the plateaus. Many varieties of metamorphic rocks are known to occur, including phyllite, quartzite, schist and gneiss.

Large bodies of granitic type rocks of Cretaceous age have intruded into the older metamorphic rocks of the highland areas often forming large batholiths or plutons. The contact zones between the metamorphic and granitic rocks are "host" areas for many small mineral deposits known in the region. Uranium is the chief mineral presently being mined, although, in the past, copper, molybdenum, lead, zinc and tungsten have been mined in small quantities (Becraft and Weis, 1963).

The metamorphic and igneous intrusive rock masses form the mountains in the northern and eastern portions of the study area. A regional drainage course through the Spokane area during early Tertiary times (50-60 million years ago) eroded deep canyons into these rock masses. In the ancient Spokane Valley, the canyon bottom was carved to an elevation

of approximately 800 feet above present sea level elevation. The same metamorphic and intrusive igneous rock masses also underlie the rest of the basin where they are covered with sedimentary rocks, unconsolidated soils and volcanic rock flow.

Two types of Tertiary volcanic rocks occur in the study area.

Of these, basalt, with its associated breccea and tuff deposits, is

by far the most abundant. The other, andesite, is found at scattered

locations in the western portion of the basin.

During the Miocene Epoch of the Tertiary Period, successive lava flows from huge earth fissures spread over the metamorphic rocks. Accumulation of this Miocene Columbia River basalt blocked the westward flow of rivers and streams and created large lakes in the canyons. Clay, silt and sand, eroded from weathered igneous and metamorphic mountain areas to the east, and airborne volcanic ash from the west, were deposited into these lakes. Near the north and east edges of the Columbia Plateau these deposits interbedded with some of the successive lava flows. This deposition occurred throughout much of the middle Tertiary period, and the sediments varied from fine clay deposited during quiet periods to sand and gravel transported to the lakes by floods or landslides. Leaf fossils now found in these sediments indicate deciduous forests grew along the lake shores. The sediments thus formed have consolidated into soft sedimentary rock and are now known as the Latah Formation. High "rimrock" bluffs west of Spokane and remnant mesas, such as at Five-Mile and Orchard Prairies, are typical of the interbedded basalt and Latah deposits.

After the cessation of the lava outpourings, the ancestral Spokane River carved a gorge around the north edge of the great lava field. By the beginning of the Pleistocene Ice Age, a broad valley had developed at about 1600 feet altitude. This pre-glacial drainage course apparently passed through the Spokane area either by way of a bedrock canyon in the Hillyard Trough, or what are referred to subsequently in this report as the "North Central Gap" and the "Shadle Park Gap." During the Quaternary Period, which included the Pleistocene and recent Epochs, a variety of deposits occurred. By mode of origin, they may be classified as eolian, morainal, glaciofluvial, glaciolacustrine, and alluvial.

The eolian (windblown) deposits consist of clay, silt and fine sand. Known as the Palouse Formation loess, these deposits occur as dunes and rounded hills covering most of the plateau areas. The loess soil is valuable to the economy of the area as crop land, although careful farming practices must be observed because of the high susceptibility to erosion. Among portions of the northern margin of the plateau and on the Spokane area mesas, the loess is mixed with gravel and occasional cobbles and boulders, having been reworked by stream action and deposition into glacial lakes. The thickness of loess deposits varies considerably from a few inches to almost 100 feet.

Pleistocene glaciation extended into the Spokane River basin several times, although most of the glacial features now in evidence are the result of the last ice advance, some 10,00 to 15,000 years ago. The number and extent of glacial advances and retreats has not been definitely established, as each advance and the flood water from each retreating

glacier have destroyed evidence of the preceding glaciation. However, it is generally accepted that several floods occurred from melting glaciers and the sudden outbreak of glacial lakes to the north and east of the area. The ice lobes, extending southward from the Cordilleran ice sheet, entered the Columbia River Gorge, the Colville-Chamokane Valley, the Little Spokane River Valley and the Purcell Trench-Spokane River Valley. Most recent investigations (Weis and Richmond, 1965) have concluded that ice reached further south than the plateau edge only in the Columbia River Gorge. As a result, glacial till and icemodified topography is mostly in evidence north of the Spokane River.

Clacial lobes north of the Spokane River, primarily in the Newport-Colbert (Little Spokane River valley) and the Chamokane Creek areas, left poorly sorted, morainal deposits of silt, sand and gravel till. Most of these moraines are poorly exposed and are generally thin. Some glacially polished bedrock is also evident in these areas. In the Spokane valley, the effects of the continental glaciation of the Pleistocene epoch consisted mostly of the deposition of gravel, sand and silt, laid down by floods and melt waters from glaciers which stopped short of the valley.

Features of the glaciofluvial deposits in the Spokane River valley include widespread occurrence of sand, up to elevations as high as 2,200 feet, deposited during times when there were glacial obstructions of the regional drainage farther down stream, a preponderance of coarse gravels following along the central part of the valley, and an overall buildup of the glaciofluvial deposits to a thickness of at least 300 to 400 feet in the central part of the valley all the way to the mouth of the

Spokane River. At the depositional maximum, the glaciofluvial deposits underlaid a wide gravelly plain with an altitude of about 2,000 feet, at Spokane.

In the area affected by glaciation, mountainous terrain was modified by the removal of residual soils, talus, and highly fractured surface rock. Higher mountainous areas and other highland areas that escaped glaciation, such as Mt. Spokane, and Browns Mt., were exposed to long periods of surface weathering, largely as a result of physical disintegration from the freeze-thaw process. This has produced a thin mantle of residual soil and rock rubble over portions of these areas. A similar mantle has developed on more level basalt surfaces. In the southern and eastern portions of the study area, chemical alteration has caused deep decomposition of some metamorphic rocks.

Several times during the periods of glacial advance and retreat, lakes were formed when ice blocked main and tributary river valleys.

Outwash silt, sand and gravel deposited in these lakes are now in evidence as stratified terrace and valley fills in tributary entrants and on the floors of topographic basins. Two large areas which show evidence of glacial lake sedimentation are the Deer Park Basin and an area north and east of Wellpinit on the Spokane Indian Reservation. Terraces along the lower Spokane River give evidence of other periods of glacial damming and deposition at a lower elevation. Subsequent erosion of these lake deposits has washed away all but a few terraced remnants along the valley flanks.

At one or more times during periods when lakes filled the Spokane River Valley, the lakes spilled excess water out over

the Columbia Plateau to the south. Large quantities of overflow water washed away some of the loess cover and eroded channels into the basalt plateau. During different stages of the flooding, layers of boulders, cobbles, gravel and sand were deposited over some areas of the eroded surface. In quieter times, small ponds and swamps filled with silt and clay. Today, this area, known as the "channelled scablands", reveals bare rock scablands, gravel bars, thin cobble, gravel and sand veneers, silt and clay deposits, and islands of loess to give evidence of its varied history.

Present geologic evidence indicates that several catastrophic floods occurred through portions of the study area during late Pleistocene times. These floods had a significant influence on the present day topography and geologic features. These floods, collectively called the Missoula Flood, occurred an estimated 22,000 years ago (Richmond 1965), and was the largest fresh water flood ever documented in geologic records. The causes and results of the flood have been well documented by Baker (1973), Bretz (1923, 1969), Bretz, Smith, Neff (1956), and many others.

During late Pleistocene time, a glacial lobe, at least 2,000 feet thick, blocked the mouth of the Clark Fork River in northwestern

Montana, ponding an estimated 500 cubic miles of water. The lake, Glacial

Lake Missoula, filled and overtopped the ice dam, causing a rapid failure

of the dam and a quick release of water. The water could only take one

course, down a valley in northern Idaho, which now comprises part of the Purcell

Trench, and out through the Spokane River Valley. It has been estimated that the

flood reached a peak flow of 752 million cubic feet per second. The entire

flood is believed to have lasted only a week or two (Baker, 1973, p. 65). Constrictions in the lower Spokane River Gorge backed the flood waters up until they overflowed southwestward over the Columbia Plateau towards the Quincy and Pasco Basins. The onrush of water scoured out the valleys along its course and eroded a wide swath of loess from the surface of the Columbia Plateau southwest of Spokane. As the flood waters spread out, slowed down, or were locally ponded, the immense quantities of materials eroded along its path were deposited in the valleys and on portions of the plateau. Much of the gravel and sand now filling the Spokane Valley and in filled channels from Spokane to Pasco Basin are believed to have been deposited by the Missoula Flood.

The present course of the Spokane River follows the path coded across the surface of the glaciofluvial deposits during and after the last years of the Pleistocene epoch. The channel crosses bedrock spurs at Post Falls, Spokane Falls, Nine Mile and Long Lake Dams. It remains fixed in these bedrock notches even though deeper parts of the alluvial fill permit groundwater to pass around them. Up valley from Spokane Falls, the post-glacial downcutting by the river has been minimal, the last of the glaciofluvial channels being visible on the prairie surface westward to beyond the state line. Downstream from Spokane Falls, the river has cut a gorge several hundred feet into the glaciofluvial deposits. This action has resulted in notched terraces on either side of the canyon which display the depositional surfaces of the glaciofluvial sediments.

In the last 10,000 years, since the end of the Pleistocene Ice

Age, the soils and rocks exposed at the surface have undergone further change.

In the valley areas, these changes include: erosion and redeposition of earlier deposits by normal stream and river activity; the formation of irregular sand dunes by prevailing southwesterly winds in the outwash areas northeast of Hillyard; and also the deposition of lake sediments and the formation of peat bogs in the Saltese Flats, and near Newman and Liberty Lakes.

Recent alluvial deposits are generally small in extent and are limited for the purposes of this study to recent clay, sand and gravel deposits in river and stream valleys. Deposition of colluvium has continued since Pleistocene times, in the form of unconsolidated soils, landslides and talus. Examples of these gravity deposits are found at the base of cliffs formed by the basalt flows that rim portions of the Spokane Valley and Hillyard Trough.

Description of Mapped Geologic Units

Two exhibits have been prepared to record the geologic features of the study area, mapped at 1:250,000 scale. Plate 303-1 shows the gross geologic features of the region. Plate 303-2 shows the features of subsurface geological structure which are significant for the purpose of analyzing the region's groundwater resources.

Map symbols are identified in the legend shown on Plates 303-1 and 303-2. For certain materials, the symbol system utilizes a capital letter first to indicate geologic origin followed by a lower case letter to indicate physical properties. In the text which follows, the legend used on the appropriate map is listed in parentheses.

Bedrock

are predominantly granitic and include granodiorite and quartz monzonite rock types. The symbol (GR) also indicates areas where smaller intrusions of aplite and alaskite occur. Granite is the predominant rock of the Okanogan Highlands and also occurs as "islands" on the Columbia Plateau where it is surrounded by basalt flows. The granitic rock is generally massive to locally blocky, medium gray in color, and is fine to coarse grained. It is surficially weathered where exposed, except along glacial flood scoured valley walls where it appears fresh and hard. Deep alteration and decomposition often occurs in contact zones between metamorphic rocks and intrusive rocks.

Metamorphic Rock (ME): Metamorphic rocks of the study area include phyllite, quartzite and carbonate rocks in the western portions of the study area and schist and gneiss in the eastern portions. The rocks are generally intensely metamorphosed sedimentary shales, sandstones and limestones. Original thickness of the sedimentary deposits is unknown. The metamorphic rocks have been extensively intruded by granitic rocks and have been further modified. The metamorphic rocks occur in eastern Washington in an area about 30 miles wide and 50 miles long (Weis 1968). The metamorphic rocks in the study area are surficially weathered except where scoured by glacial flood waters. Deep weathering or alteration occurs in some protecter areas and near contact zones with granitic intrusives.

Columbia River Basalt (BA): Basalt of the Columbia River Group is the predominant rock of the southern half of the study area,

although there are scattered remnants of basalt in the northern half. The basalt is composed of many separate flows sometimes separated by weathered zones, or as in the Spokane area, siltstone interveds. Individual flows range in thickness from 10 feet to 200 feet and total thickness ranges upward to 1,500 feet in the study area and to 5,000 feet elsewhere on the plateau. The rock is highly jointed and blocky and talus slopes usually form at the base of most basalt cliffs. Columnar jointing occurs in some areas. Pillow lava structure is exposed at various locations, indicating lava deposition into shallow lakes. The basalt is generally dark gray to black, dense and fine grained. The upper few feet of each flow is commonly vesicular.

Latah Formation Siltstone: (ST) In the ctudy area, the Latah Formation occurs generally east of Nine-Mile Falls, where it lies beneath the Spokane Valley gravels, the rimrock basalt, and is interbedded with the basalt flows along the flanks of the Spokane River Valley. Thickness of the formation varies considerably. The siltstone underlying the basalt may be as much as 1,500 feet thick and the interbedded deposits are generally less than 100 feet thick. The Latah Formation is primarily siltstone, but contains some clay, sand, or gravel beds and lenses. The formation is overconsolidated due to the overlying weight of the basalt flows. Colors of the siltstone range from gray to gray-brown, but upon exposure to weathering, turn to distinctive buff, white or light yellow. Many beds of the Latah Formation are fossiliferous and contain abundant imprints of plants, diatoms and sponge spicules.

Surface exposures of the Latah Formation are too small to be

mapped at 1: 250,000 scale on Plate 303-1. They are of significance to considerations of slope and local groundwater conditions in the Spokane urban area, and are mapped at 1: 24,000 and discussed in a later section of this report.

Aeolian and Glacial Deposits

Palouse Formation Loess (Wm-1, Wm-2): In the study area,
Palouse Formation silt loess covers most of the Columbia Plateau south of
the Spokane River and the small remnant basalt flows north of the river.
The formation varies greatly in thickness due to its dune-like deposition,
erosion, and the fact that it was deposited on an irregular plateau surface.
Thickness varies from a few inches to nearly 100 feet. Soils of the Palouse
Formation consist primarily of tan to brown silts which contain varying
amounts of clay or fine sand. North of the Spokane River, the Palouse formation silt has generally been mixed or reworked with sand and gravel.
When glacial lakes covered this area, sands and gravel from the highlands
were deposited in the lakes, mixing with the submerged silt. In addition,
gravel, cobbles and boulders trapped in floating blocks of glacial ice
were deposited in the lakes as the ice melted. These deposits of reworked
Palouse Formation loess are designated Wm-2 on Plate 303-1.

Outwash and Flood Deposits (G): These were carried by the large meltwater and flood releases during periods of glacial recession.

The Spokane River Valley, Hillyard Trough, Little Spokane River Valley, and Chamokane Valleys received large quantities of these deposits. Thicknesses of these deposits are unknown, but estimates derived from seismic and drill

hole information indicate thicknesses exceeding 400 feet in the Spokane Valley and 700 feet in the Hillyard Trough. Flood deposits, the primary fill of the Spokane Valley, consist of relatively clean, rounded gravel with varying amounts of sand, cobbles and boulders. Outwash deposits, in Hillyard Trough, Little Spokane Valley and Chamokane Creek are generally finer and are composed of stratified silt, sand and gravel with only occasional cobbles or boulders. Outwash deposits in the Spokane River gorge west of Long Lake Dam primarily consist of silt and sand.

Glacial Till Deposits G(t): Deposits of glacial till are limited to portions of the northern half of the study area. These deposits were left by glacial lobe advances down the Spokane River Valley, the Little Spokane River Valley and a large part of the northwest portion of the study area as far south as the Spokane River Gorge. The till deposits are generally in low terminal and lateral morainal hills, and relatively flat ground moraines. The till cover is thin and rarely exceeds 30 to 40 feet in thickness. The morainal material consists of unsorted, unstratified clay, silt, sand, gravel, cobbles and boulders.

Glacial Lake Deposits (L): Glacial lake deposits occur in areas along the sides of the Spokane River Valley and Gorge, in canyons tributary to the Spokane River, and in the Deer Park and Wellpinit areas. Thickness of these deposits vary from thin edges at shorelines to as much as 300 feet. The glacial lake deposits generally consist of stratified silt and sand, but clay beds or gravel lenses do occur in most deposits.

Recent Deposits

Alluvial Deposits (A): Alluvium is deposited along most of the tributary streams of the Spokane River, although the Latah Creek, Little Spokane River and Chamokane Creek valleys are the only ones with significant accumulations. The deposits are generally thin, with the greatest thickness recorded being 38 feet in the Little Spokane Valley (Cline 1969). Alluvial deposits along the Spokane River upstream of Nine-Mile Falls are generally restricted within the river banks and to "Flood stage" gravel bars. These features are too small to map on Plate 303-1 or on the engineering geology maps at 1: 24,000 scale. The alluvium accumulations consist of silt and sand with lenses of gravel. Some clay and organic silt is also found in the meandering reaches of the lower Little Spokane River.

Residual Deposits (R): Residual materials are those formed in place by disintegration or decomposition of rock and which have not been moved beyond the local area. Residual soils and rock rubble has accumulated over much of the Okanogan-Selkirk Highland area where bedrock has been exposed. Thickness of these deposits is generally thin and probably averages less than 10 feet. However, local conditions of rock type, topography, jointing, and protection from exposure vary the degree and depth of weathering.

Residual materials commonly consist of fine to coarse sand and rock rubble from gravel to boulder sizes. Some rocks such as basalt and some metamorphic rocks have, under certain conditions, weathered to clay sized particles and occur locally within the plateau and highland areas.

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Colluvium: Although not in deposits large enough to map on Plate 303-1, colluvium occurs in many places of the study area.

Colluvium occurs at the base of the rimrock basalt cliffs and on the lower slopes of many highland areas. Colluvium consists of unconsolidated rock materials that accumulate by gravity or slopewash at the base of cliffs or slopes. The deposits at the base of the rimrock cliffs are primarily basalt talus and blocks, but in the Spokane region, are mixed with weathered silt from the Latah Formation interbeds. The colluvium in the highland areas generally ranges from silt to boulders and is often mixed with pre-existing glacial, alluvial or other materials. Thickness of the colluvium deposits varies upwards from a few inches to nearly 100 feet.

Groundwater Geology. Groundwater occurs in varying amounts throughout the Spokane River Basin. The largest amounts occur in the glacial flood gravels of the Spokane Valley between the Washington-Idaho state line and downtown Spokane. Lesser amounts are found in the Little Spokane River-Deer Park Basin and the Columbia Plateau. The least amounts of groundwater may be expected in the Spokane River Gorge west of Nine Mile Falls and north of the river in the bedrock areas of the Okanogan-Selkirk Highlands. The following general descriptions explain the relationship of groundwater occurrence to the geologic units shown on the groundwater geology map (Plate 303-2).

Pre-Tertiary Rocks (pTr): The granitic and metamorphic rocks of the Okanogan-Selkirk Highland province and of the Columbia Plateau "islands" are essentially impermeable. However, the weathered top (uppermost 10 to 20 feet) and joint cracks in the upper part are slightly

permeable and, in places, afford small yields of water to wells and springs. This type of perched groundwater forms the source of many small water supplies in the uplands underlain by these rocks.

Tertiary Rocks (Tcr): The geological units identified by this symbol on Plate 303-2 include materials of volcanic origin and sedimentary deposits. The volcanic materials consist primarily of basalt, in which there are local conditions which produce a waterbearing capability. The sedimentary deposits are those of the Latah Formation.

The predominantly clay and silt composition of the Latah Formation, with its resultant low permeability, makes it generally a non-water bearing unit. In places, sand beds within the Latah Formation have a higher permeability and afford limited yields from wells, adequate for household supply.

Permeable parts of the basalt of the Columbia River Group consist of the (1) rubbly tops of some of the lava flows, (2) pillow-palagonite phases formed locally where lava disintegrated by flowage into ponds, and (3) a rare occurrence of flow breccia. Where the lava is interlayered with sedimentary materials, there is less yield from rubbly flow tops and flow breccia, because the invasion of overlying sedimentary materials decreased this permeability. Because the more pervious zones generally occur horizontally, the groundwater is mostly confined, or both perched and confined.

Quaternary Deposits (Qgd & Qd): Glacial drift materials (Qgd), consisting of till, outwash deposits and lacustrine fine-grained materials, underlie several areas of the Okanogan-Selkirk Highlands.

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The sand and gravel members of the drift are principal aquifers of the Chamokane Creek and Little Spokane-Deer Park Basin and other smaller valley areas. The aquifers afford small and moderate yields to many wells and springs. Many local "Pockets" of these glacial deposits in the mountain areas afford storage for groundwater and provide water for houses, camp sites, stock, wildlife and streams.

Gravelly glacial outwash (Map Symbol Qg) is the main aquifer of the valley areas including the Spokane Valley. In the Spokane River Gorge below Nine-Mile Falls, the gravel outwash is intermixed with other finer grained glacial outwash and lacustrine sand and silt deposits; in some of these down-valley areas the principal gravel outwash strata lie beneath terraces and above the level of the main water table.

Groundwater Occurrence and Use

Throughout the Spokane River Basin, groundwater is encountered in varying amounts. The occurrence is most prolific in the Rathdrum Prairie of Idaho, from the Spokane Valley glacial outwash deposits which extend downstream to the City of Spokane. The groundwater resources of the Spokane Valley Aquifer are described in a subsequent part of this section.

In addition to the principal aquifer, there are other geological formations within the study area which give rise to the occurrence of groundwater in usable quantities. These are outlined in the following text, and may be related to the information shown on Plate 303-2, "Ground-

water Regions of the Study Area."

Okanogan-Selkirk Highlands: This region, generally north of the Spokane Valley, is characterized by the occurrence of pre-Tertiary bedrock formations, the oldest in the basin. The region encloses major tributary valleys of the Spokane, including the Little Spokane-Deer Park Basin and the Chamokane Valley, which have been influenced by Pleistocene glaciation.

The water supply of the upland areas is derived principally from springs and shallow wells in "pockets" of glacial deposits and alluvium or from near surface fractured zones in the bedrock. Residences and the sites of recreation, agriculture and forest management are centered about these scattered sources of groundwater, mostly in the valley areas. The groundwater supply is sufficient only for a sparse population and for grazing and wildlife purposes. The smaller upland valleys are similarly limited. In favorable places, the larger lowland valleys have sufficient groundwater for towns and small communities. Large withdrawals, in amounts greater than about 10 million gallons per day, could exceed the capacity of the available groundwater sources unless procedures, such as artificial recharge by stored surface water, were followed. In general, future large withdrawals are feasible only in the larger valleys such as those of the Little Spokane River and Chamokane Creek, and those withdrawals will need to be wisely located and administered.

Intermountain valleys have small to moderate supplies of groundwater, primarily from irrigation, industrial and public supply wells. These wells yield in the range of 50 to 500 gpm from sand and gravel members within the glacial drift and glacial outwash deposits. In the Chamokane Creek and Little Spokane River valleys, most households can readily obtain domestic water from nearby springs or by drilling wells 50 to 200 feet deep. Larger withdrawals of groundwater are obtainable in a few places where greater thicknesses of saturated sand or gravel occur and the sites are not too high above the valley water table.

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Groundwater is contained in minor aquifers in the blocked tributary valleys containing Newman, Liberty and Hauser Lakes, the Saltese Flats and the upper Little Spokane River. The meager groundwater information which is available in these areas indicates that the gravel and sand aquifers are lenses in finer grained material; the natural water table, like the lakes, stand at higher levels than the water table in the valley aquifer, but the water levels in pumped wells draw down unlike those of pumped wells in the valley. Hence, they appear to be effectively isolated from the principal valley aquifer and, while providing limited supplies for domestic use, are not considered major sources of groundwater.

Just before the glaciofluvial episode, these tributary valleys must have been relatively narrow, with the bed of the downstream portions graded to the 1,650-foot elevation of the pre-glacial valley of the Spokane River. The filling episodes of the glacial outwash not only blocked the lower parts of these side valleys with the progressively thicker coarse deposits in the main valley, but must have caused fine grained materials to be deposited in the slack water that occupied them. The fine grained deposits in the side valleys built up to elevation 2,100

to 2,200 and lapped up onto the bedrock sides of the pre-glacial valleys. The fine grained fills across the lower parts of the valleys and, later, the insulating seals deposited within the lakes, largely isolated the side lakes to a runoff-evaporation balance within their own drainage basins.

Spokane River Valley and Gorge. From the Washington/Idaho border, the course of the Spokane River is underlain by glacial outwash deposits. Upstream of Nine Mile Falls, these deposits contain significant amounts of usable groundwater; downstream, in the Spokane River Gorge, occurrence of groundwater is relatively sparse. That part of the aquifer which is upstream from Nine Mile Falls is the source of municipal, irrigation and industrial water supplies of the Spokane urban area. This major aquifer which is the most important groundwater body in the study area, is discussed separately and in detail in Part IV of this report. A brief description is included here together with a description of the downstream or Spokane Gorge groundwater area to complete the study area overview.

The Spokane Valley aquifer, as the upstream part is designated, includes four surface physiographic units, the Spokane Valley plains, the terraced valley below Spokane Falls, the Hillyard Trough and the lower valley of the Little Spokane River. The primary flow path through the aquifer turns northward from the vicinity of Spokane Falls and passes through the Hillyard Trough to the lower reaches of the Little Spokane.

The flow paths west of the Hillyard Trough are uncertain. Two gravel filled gaps are believed to exist linking the Hillyard Trough and the branch of the aquifer underlying the river valley, but their existence is not well documented. Groundwater in small amounts can be seen flowing out at river level in the "Downriver Springs" just west of the North Central Gap, in the SW 1/4 Sec. 12, T. 25 N, R. 42 E, and may be indicative of a westward

movement of groundwater. A similar situation is present in the northern ("Shadle Park") gap, but the surface of the gap is higher and is almost 200 feet above the known elevation of the water table in the Hillyard Trough.

A comprehensive subsurface investigation in these gaps would be required to determine if there is a groundwater supply in these areas, presently without wells, which could augment supplies to portions of west and northwest Spokane.

Throughout the gravelly deposits of the Spokane Valley aquifer, investigation of the specific capacities of wells has shown that well yields of 400 gallons per minute per foot of drawdown are equaled or exceeded throughout the aquifer from westernmost Idaho to the basalt at Spokane. (Frink, 1962).

In the Spokane River Gorge below Nine Mile Dam, groundwater supplies are developed from a variety of sources in addition to the glacial flood outwash materials. These other sources include rimrock springs, wells in the weathered top of granitic bedrock and alluvium. The rimrock springs yield small amounts of water from the base of the rimrock basalt, wherein the groundwater is perched upon the underlying, interlayered Latah silts and clays. Some of the glacial outwash contains sand members that overlie clay beds and afford small supplies of perched groundwater to wells located on the valley terraces. Aside from places where groundwater in coarse-grained outwash materials can be tapped by wells, the river gorge is generally an area where groundwater supplies may be lacking at any particular site.

The southern side of the gorge, much of which is underlain by slumped parts of the Latah Formation, may be a particularly difficult area in which to obtain a water supply from groundwater.

In the Spokane River Gorge, a great variation is observed in the yields of wells, from 2,000 gpm in some riverside wells in glacial

outwash to little or no yield from some wells drilled into the Latah Formation or granite bedrock beneath the slopes of the gorge. The indication is that, should increased water supplies be required, large amounts of water may need to be withdrawn in favorable places and transmitted extensively. Large amounts of water should be available to additional users in the future by use of such distribution systems; without this distribution of water, large areas on the slopes of the gorge may be restricted to sparse, poorly-watered habitation.

Columbia Plateau: Except for a few areas which have perched water that occurs in the outwash gravel of some of the channeled scablands and that which occurs in some surficial parts of the pre-Tertiary rock outcrops, the basalt is the only source of groundwater. Wells of 100 to 300 foot depth are commonly used for household and stock supply. Common yields of wells in the basalt range from 2,000 gpm in large deep irrigation or public supply wells to 1 or 2 gpm from 100 to 300 foot deep domestic wells. The water-yielding capability of the basalt is greater where the basalt section is thicker--consisting of a greater number of lava flows--away from the plateau's northern edge and away from the edges of the inlying granitic and metamorphic rock. Because the aquifers in the basalt consist mainly of the rubbly tops of the lava flows, the yield of wells is commonly in proportion to the number of aquifers penetrated; hence, it is in general proportional to the depth of the well (Newcomb, 1959). However, the completion of a largeyielding well that draws from several horizons of different hydrostatic heads may be impractical.

The nearly horizontal layers of basalt in the Columbia Plateau province make it difficult for water on the surface to seep through the lava flows and reach the permeable horizons below; consequently, natural recharge of the groundwater through the basalt may be small and water pumped from these horizons may not be adequately replaced by natural recharge. Significant recharge from the main Spokane Valley aquifer is prevented by low-permeability Latah silt interbeds and basalts that form the aquifer boundaries. For these reasons, the groundwater in the basalt cannot be considered as a source of water supply much beyond that now extracted. Artificial recharge of the horizontal basalt aquifers can be accomplished in a practical manner only by the intra-well injection of clean purified water. While artificial recharge may provide a future method for some cheap storage of water, it is unlikely to be a source of large supplies of water. Future needs for large amounts of water in the plateau area may not be feasible from groundwater withdrawals, though the groundwater supply is adequate for many small dispersed withdrawals from household and stock wells.

Near the north and west edge of the 1 lateau, wells may encounter much interbedded clay of the Latah Formation. This area, extending as much as 4 or 5 miles south and west of the Spokane River canyon, is an area of low well yields. In the southern part of the area, near Cheney, Rockford and Tekoa, yields of up to 500 gpm are obtained from wells 700 to 1,000 feet deep.

III. GEOLOGY AND SURFACE SOILS OF THE SPOKANE URBAN AREA

Introduction

Information on the engineering geologic and surface soil characteristics within the Spokane urban area are presented on three sets of 1:24,000 scale maps, plates 303-4 through 27. The text which follows is intended to assist in interpreting the information displayed on the maps.

Engineering Geology (refer to Plates 303-4 to 11)

Map Symbols: Map symbols used to depict soils and rock types on engineering geologic maps are not yet standardized. A set of symbols used for the maps of this study has been agreed upon by the agencies involved.

Symbols for bedrock shown on the engineering geologic maps consist of two capital letters. Unconsolidated soils are designated by a capital letter which indicates the geologic source or origin of the soil, followed by one or more lower case letters that indicate soil type. In mixed soils, where more than one soil type symbol follows the origin symbol, the first one indicates the predominant soil, followed by symbols of soils in decreasing order of quantity present. Special depositional categories are designated by lower case letters in parentheses. The following symbols are used:

Geologic Origin	Symbol	Soil Type	Symbol
Colluvial	C	Clay	c
Glacial	G	P ea t	p

Geologic Origin	Symbol	Soil Type	Symbol
Alluvial	A	Silt	m
Eolian	W	Sand	8
Lacustrine	L	Gravel	g
Residual	R	Boulders	g b
Manmade fill	F	Rock Rubble	r
Special Depositional			
Category	Symbol	Bedrock	Symbol
		Basalt	BA
		Metamorphic	ME
Talus	(ta)	Granitic	GR
T111	(t)	Siltstone	ST

In areas where one soil overlies another type of soil or rock at shallow depth, dual symbols are used. For example, the symbol $\frac{Rs}{ME}$ denotes a shallow mantle of residual sand overlying metamorphic rock. Dual symbols are only used where surficial soils are believed to be less than 5 feet deep. However, due to subsurface irregularities, it should be assumed that localized areas may have overburden depths as great as 15 feet. Because of the variation in soil depths, no depth symbol is shown.

Each soil or rock unit shown on the maps is bounded by a dashed line. Since a gradual transition usually occurs between adjacent soil types, the lines are arbitrary, but are as accurate as map scale and technique allow. Where soil units abut rock masses, and in some lake or stream deposits, the boundaries are better defined and are more accurately mapped.

Within soil boundaries, there may be small areas of different materials, as well as like materials of different origin. Mapping these areas on a smaller scale with more detail would be required for special

land use planning.

Following are more detailed descriptions of the bedrock materials and soil types found in the urban planning area. This expands descriptions which have been introduced in more general terms under study area-wide geology.

Characteristics of Bedrock

Metamorphics. The metamorphic rock exposed in the Spokane area is predominantly schist and gneiss formed from pre-existing sedimentary rocks. These rocks are intensely metamorphosed and exhibit layering and granitization. Most metamorphic rocks are believed to be related to the Pre-Cambrian Belt Series rocks of the Rocky Mountains. Physical characteristics of the metamorphic rocks vary widely over short horizontal and vertical distances. Some are highly altered by hydrothermal processes and by deep weathering. In other areas, the rock is hard and fresh. Lack of good surface exposures precludes accurate delineation of altered and fresh rock areas on the geologic maps.

In place, metamorphic rock generally provides good to excellent foundation siting. However, metamorphic rock types in the study area vary from hard massive gneiss and quartzite to highly fractured, thinly foliated phyllite and schist. Accordingly, foundation and excavation requirements must be determined on the characteristics of each individual area prior to any planned construction.

Granite. Igneous rocks that intruded the metamorphic rocks are found in some portions of the urbanizing area north of the Spokane River.

These rocks are medium to coarse grained, hard, and generally massive. Boundaries with the metamorphic rocks are often indistinct.

Weathering of the granitic rock occurs in areas of low slopes and the less exposed sides of mountain ridges. The weathering is less intense than it is in the metamorphic rocks and does not exceed 5 feet below the exposed rock surface is most areas.

Areas of granite outcropping generally are excellent for foundation siting. The rock is generally massive, but local jointing causes some blockiness. Excavation usually requires blasting below any surface residual deposits. Deeply altered zones near contact zones with metamorhpic or other igneous rock bodies may require special investigation prior to planning construction in these areas.

Basalt. Basalt is the most conspicuous rock in the Spokane area. It is in evidence at the falls near downtown Spokane, in the rimtock bluffs south and west of the city and in many local outcrops throughout the area.

Columbia River Basalt is dark gray, dense and fine grained.

Basalt is inherently jointed, causing it to disintegrate and to break into blocky fragments when exposed to freezing-thawing conditions.

The basalt in the Spokane region varies from slightly to highly vesicular and from moderately hard to very hard. Residual weathering of the rock is generally slight, although near the contacts with some Latah siltstone interbeds, some weathering and alteration has occurred.

Thickness of individual basalt flows in the Spokane area ranges from 50 to 150 feet. The overall thickness of the successive flows ex-

ceeds 5,000 feet in some areas, but is in the order of 400 feet within the Spokane urban area.

The basalt forms an excellent foundation stratum and supports many of the larger buildings in Spokane. The surface of the basalt is highly irregular, however, and comprehensive explorations are necessary to establish the surface of the rock, prior to planning construction in such areas.

The basalt is, for the most part, impervious, although highly fractured zones occur locally, particularly near flow contacts, which permit the passage of subsurface water. Water is also encountered at the bedrock surface contact in some areas. However, the location of water-bearing zones within and on top of the basalt is often indeterminate without extensive explorations and available quantities are generally small. Seldom are they a source of groundwater supply when other sources are more readily available.

The basalt rock generally requires blasting to excavate, although large backhoes, rippers and jack-hammers are successful in highly fractured near-surface zones and for excavation "dental" work. Excavation slopes in moderately fractured basalt will stand near-vertically, but a safety area is generally required at the base to catch spalling blocks.

Latah Formation. Siltstone of the Latah Formation is rarely exposed at the ground surface, but is an important subsurface feature in the Spokane area. The siltstone ranges from soft to hard, depending upon its proximity to the surface. It ranges in color from yellow to gray brown

and often contains leaf fossils. The Latah is practically impervious, and is considered a groundwater barrier. Springs are common on the surface of the Latah where interbeds daylight on hillside slopes.

The Latah Formation is predominantly composed of silt, but has some zones of clay, sand or gravel. The Latah beds may be as thick as 500 feet below the glacial outwash gravels in the Spokane Valley (Newcomb et al., 1953) and in places rest directly upon the metamorphic "basement" rocks. Latah silts, interbedded with Columbia River basalt flows, also exist over a wide area adjacent to the Spokane River Valley, but the interbeds are generally less than 100 feet thick. Pardee and Bryan (1926) state that the Latah Formation may be as much as 1,400 to 1,500 feet thick in the vicinity of Hangman (Latah) Creek.

Near the surface of many beds, the Latah siltstone has softened to the consistency of a stiff soil. At depth it is highly overconsolidated. The Latah, in general, is considered an excellent foundation stratum, although local groundwater conditions and the degree of natural softening may dictate special foundation considerations in certain areas. Excavations into the Latah can generally be made with conventional power excavation equipment.

There have been very few man-made cuts in the Latah and very little direct evidence of recent landslides or failures in the natural slopes where the Latah is exposed at the surface. Where earth movements have been observed, they generally have involved the soils overlying the Latah, or were confined to the softer, weathered near-surface silts. In either case, groundwater was also involved. There is also limited evidence

of ancient movements, such as fissures and offsets, in some areas.

Without direct data or experience, man-made modifications to slopes underlain by the Latah silts should be preceded by careful study.

Characteristics of Soils

Unconsolidated Glacial Deposits. The dominant soils in the Spokane Valley area are sands and gravels deposited by glacial activity. The gravel and sand in the urban section of the Spokane Valley is poorly sorted, fine to coarse, and contains cobbles and boulders. These soils were deposited from glacial flood waters, and are relatively free of silt and fine sand, except in the upper 3 to 5 feet where the finer materials generally fill the voids of the coarser sands and gravels. The permeability of these soils is high, allowing for high groundwater yield. In the Spokane Valley area, the coarse sands and gravels are believed to extend to a depth of about 400 feet below the surface at the Washington-Idaho state line, to an elevation of 1,600 feet. This estimate, however, is based upon interpretive seismic data (Newcomb et al., 1953), since to date (1973) no boring has ever penetrated through the stratum.

Glacial outwash deposits adjacent to Fivemile Prairie and in the Hillyard Trough are finer grained than those in the Spokane Valley. These deposits consist mainly of stratified sand with minor amounts of gravel and silt, and, in places, occasional boulders. Stratification and sorting of these materials, which exhibit some crossbedding, are moderate to well developed. The Hillyard Trough deposits are known to exceed 780 feet in depth (Rieber and Turner, 1963) and to become thinner north of Mead (Cline, 1969) in the Little Spokane River Valley.

Other glacially derived deposits were formed when sand and gravel-laden glacial outwash streams deposited their loads in a lake that formed behind a glacial lobe at Long Lake (Richmond et al., 1965). These deposits are classified as lacustrine sand (Ls) on the geologic maps and are located north of Mead and on the uplands flanking the Spokane and Little Spokane Valleys. These deposits are generally stratified, well sorted and flat lying and are finer grained than those in the main valley. Though chiefly sand, these deposits may contain occasional gravel or silt interbeds. Occasional ice rafted erratic boulders up to 10 feet in diameter are found in the lake sands. These deposits range in thickness up to 300 feet (Cline, 1959) and vary widely in permeability.

Other glacially derived deposits in the study area are relatively minor. Glacial moraine materials (till), consisting of unsorted clay to boulder size materials, are presently found mixed with glacial outwash, colluvial and eolian soils along the lower flanks of the Spokane Valley. According to Weis and Richmond (1965), a glacial lobe reached the vicinity of east Spokane. Because the mixing of materials from different sources often makes identification difficult, glacial till may be included in map units designated as colluvium.

The sands and gravels in the main valley are excellent foundation soils, while the finer sands in the northern portion of the urbanizing area and Hillyard Trough are excellent for most one and two story structures. Since large variations in the relative density of the finer sands occur, study of individual sites is advisable prior to constructing large structures. Foundations for such structures may have to penetrate several feet beneath the surface.

Most of the unconsolidated glacial deposits can be excavated with conventional excavating equipment, although large boulders may require special handling. Excavation slopes steeper than about I vertical on I horizontal generally require shoring. The finer sands will exhibit apparent stability when excavated at near-vertical slopes, but such stability is temporary and unpredictable.

Eolian Deposits. Windblown silt and sand deposits are prevalent in the Spokane region. Silt occurs as loess in the Palouse Formation, and mixed with glacial sand deposits, and dune sands also occur in the area near Mead. The silts of the Palouse Formation are identified on the geologic maps by the symbol Wm-1. The mixed loessial and glacial soils are designated Wm-2 and the dune sands by Ws.

Loess of the Palouse Formation is fine grained, tan to brown silt with occasional clay and fine sand. The silt is generally indistinctly stratified and has poor permeability. Loess is characterized by many fine, usually vertical root tubules and root remnants. Thickness of the loess deposits varies greatly as a result of the highly irregular surface of the rock that generally underlies it and because of the dune—like character of the windblown deposits. Thickness of the loess in the Spokane area is on the order of 3 to 20 feet, although a test hole on Orchard Prairie (Cline, 1969) encountered a thickness of 73 feet. South of the urbanizing area, the thickness of the Palouse Formation sometimes

exceeds 100 feet (Scheid et al., 1954).

Palouse silts are a marginal foundation soil and, while adequite for light structures, heavy loads often require special foundation treatment or deeper excavation. Permeability of the silts is poor, although the presence of vertical root tubules in some areas may provide a high degree of vertical permeability. Shallow excavations in the silt will generally remain stable at near vertical slopes, while deep unsupported slopes require flattening. Local groundwater conditions such as springs usually require special slope treatment.

Deposits of loess in the uplands surrounding Spokane Valley and Hillyard Trough occasionally contain glacial sand and gravel which indicates a mixing with more recent glacial lake or glacial till deposits. The properties of the mixed soils are difficult to generalize because the degree of mixing and relative proportions of each component will significantly affect the engineering characteristics. Drainage, however, is considered poor.

Windblown fine sand has formed dunes near Mead in the northwestern area. This sand is blown from glacial outwash and lake deposits and the maximum depth of accumulation is approximately 50 feet. All windblown deposits in the Spokane area are highly susceptible to surface erosion from both wind and water.

Like the silts, the windblown sands have marginal foundation properties. All excavations require flattening of unsupported slopes. Permeability of the dune sands is generally variable depending upon the proportion of silt mixed with the sand.

Colluvial Deposits. Colluvium consists of accumulations of unconsolidated mixtures of soil and rock and is generally found on the lower slopes and at the base of cliffs. The colluvial soils vary widely and may include slopewash silt, sand and gravel, similar residually formed materials, and talus from rock cliffs. These materials may often overlap at their contacts with glacial outwash, till or with the underlying formation.

A colluvial feature peculiar to the Spokane region is the occurrence of basalt "haystacks" downslope from the basalt rimrock cliffs. The "haystacks" are basalt blocks that have broken from the flow edges and have moved downslope by gravity. In most instances, the blocks are resting on, or are partially buried in, the underlying Latah Formation. The term "haystack" was derived from the characteristic rounded shape. Haystacks range in size from 2 feet to more than 100 feet in diameter. Griggs (1966) has mapped these features as landslides but, because they are predominantly colluvian in origin, they have been mapped as colluvium for this study.

The highly variable nature of the colluvial soils precludes generalization of engineering properties. However, since most colluvium occurs upon or at the base of slopes, sites underlain by these soils are generally the most difficult to develop. Studies of the specific site conditions in colluvial areas must be made and foundation, drainage and excavation requirements must be tailored to these conditions.

Residual Deposits. Residual deposits of the area are formed by weathering and in-place decomposition of the granitic and metamorphic

Control of the second of the s

rocks of the mountainous regions. Generally, the soils remain essentially where they were formed and consist primarily of silty sand and gravel size particles, although larger pieces of rock rubble are common. The residual soils in inaccessible mountainous areas were mapped from aerial photographs and the soils classifications were interpreted from soils maps (Donaldson and Giese, 1968). Thickness of residual soils is mostly unknown and probably varies considerably. However, examination of roadcuts and stream banks indicates that depth of residual soils ranges from 1 to 5 feet, with depths up to 15 feet locally.

Residual soils generally are suitable for foundations and, when questionable, good foundation rock is usually available at reasonable depth beneath them. Excavation of residual soils can usually be accomplished with conventional excavating equipment with heavier rippers being required where the rock-to-soil breakdown is not yet complete. Because of the high silt and clay content of the residual soils and their proximity to the underlying parent rock, the permeability of these soils is generally found to be low. Water may be encountered seasonally at the residuum-bedrock contact.

Alluvium. Soils mapped as alluvium are silts, sands and gravels that have been deposited along stream valleys, chiefly the Little Spokane River and Latah Creek. These deposits vary widely in composition and are generally stratified. Thickness of the alluvium deposits is rarely over 30 feet. Alluvial deposits often merge into glacial outwash or lacustrine materials.

Because alluvium generally occurs as a variety of soil types and often in conjunction with materials of different origin, the engineering properties vary. Generally, the finer grained alluvial soils are unsuitable, or at best marginal, foundation soils, while the coarser sands and gravels are adequate. Excavation stability is usually poor as most alluvium occurs in areas containing a high water table. Sites underlain by alluvium should be investigated in detail prior to planning construction.

Lacustrine Organic Deposits. Saltese Flat, Newman Lake and Liberty Lake are shallow basins of deposition for soils washed down from adjacent mountainous areas. Plant growth in swampy areas around the basins has accumulated and, over a period of time, has formed peat bogs and organic silt deposits. These areas are soft and spongy and characterized by high water tables.

Three types of peat have been identified at each site (Rigg, 1958). Muck is peat that has decomposed to the point where plant remains are indistinguishable and usually forms a very soft surface layer of from 1 to 5 feet deep over underlying deposits. Fibrous peat generally underlies the muck and thicknesses range up to 15 feet. Fibrous peat is composed of coarse grasslike plant remains that are incompletely decomposed. Sedimentary peat is composed of small plant remains (algae, diatoms, etc.) and the remains of aquatic animal and bacteria life that have been deposited in water. Sedimentary peat is often very soft and plastic. This type of peat is found near the bottom of the deposits at Saltese Flat and Newman Lake.

The organic deposits are not suitable foundation soils and must

be removed or foundations must penetrate through them. They are soft and compressible and surface loads such as embankments will often produce settlements of several inches that may continue over a period of several years. Excavation of the peats generally requires a dragline.

Permeability. (Refer to Plates 303-12 to 27.) Two sets of permeability maps have been prepared to illustrate the relative ability of the surface and near-surface soils to transmit water. One set, designated "Surface Soils" (Plates 303-12 through 19), denotes the relative permeability of the soils from the surface to a depth of about 3 feet, while the set designated "Near Surface Soils" (Plates 303-20 through 27) presents similar data on the soils at a depth of about 5 feet below the surface.

The engineering parameter used to describe the ability of a soil to transmit water is called the "coefficient of permeability." The one that was selected for the purpose of this study is Darcy's coefficient "k", which represents the saturated flow velocity of water through a unit area of soil under a hydraulic gradient of unity. The coefficient is expressed as a velocity with units in the metric system (centimeters per second). Permeability varies over a wide range and, within the soil spectrum, can vary from about 100 cm/sec. for coarse gravels to 10^{-10} cm/sec. for highly plastic clays. The permeability of a particular material varies according to particle size, particle shape, particle orientation, density and stratification, so that values may vary widely even for a given type of soil. Field and/or laboratory tests are required to establish realistic values for given soils and soil strata with certainty. Informa-

tion regarding such tests is lacking for the soils in the Spokane area. For the purpose of this study, permeability evaluation was based upon three very broad classifications: 1) Good; 2) Poor; and 3) Practically Impervious, corresponding to permeability coefficients of: 1) greater than 10^{-4} cm/sec.; 2) between 10^{-4} and 10^{-6} cm/sec.; and 3) less than 10^{-6} cm/sec.

In terms of the soils found within the Spokane Urban area, the permeability and soil type relationship is as follows:

Permeability Classification	k (cm/sec.)	Soil Type	Typical Geology Map Symbol
CTABSTITCACTOR	x (cm/sec.)	SOII Type	Georgy Map Symbol
Good	10 ⁻⁴	Sands and Gravels relatively free of silt and clay particles.	Gg, As, Ls, Cs/g, Gs, Rs, Ws
Poor	10 ⁻⁴ to 10 ⁻⁶	Very fine sands, silty sands and gravels, silts.	Wm, Am, Lm, Cm, As/m, Ls/m, Gs/m, Rs/m, Cs/g/m
Practically Impervious	10 ⁻⁶	Clayey silts, rock	Ba, ST, Am, Lm

The primary basis for the permeability/soil type relationship tabulated above is the U.S. Soil Conservation Service (U.S.S.C.S.) soils data for Spokane County presented by Donaldson and Giese (1968). Permeability in the U.S.S.C.S. publications represents an infiltration rate expressed in inches per hour, through a unit area of soil under a given head, usually 0.5 inch (U.S. Department of Agriculture, 1951). A limitation of the U.S.S.C.S. data for Spokane County is that they are based upon only a few tests with the majority of the permeability values being assigned on the basis of a visual classification of the soil. Further, the U.S.S.C.S.

system contains several categories, spanning a relatively narrow portion of the total permeability spectrum, which may be unrealistic if unsupported by direct testing. Therefore, only the relative magnitudes of the published data were considered and judgment was applied on the basis of the grain size characteristics presented by the U.S.S.C.S., particularly that portion passing the No. 200 mesh seive, constituting the portion of silt and clay particles.

Boundaries on the surface drainage maps generally coincide with those on the engineering geologic map. However, because different soils may have similar drainage characteristics, fewer boundaries are used. Like the soils boundaries, the drainage boundaries are arbitrarily drawn, consistent with the scale and technique used.

IV. GROUNDWATER RESOURCES OF THE SPOKANE URBAN AREA

Historical Background

From the times of pioneer settlement, springs and shallow wells were the most popular early-day means of supplying water. The first public supply at Spokane was pumped from the river at Havermale Island. The system was promoted by a private company, but was taken over by the City of Spokane in 1885, a few months after supply was initiated. Nine years later, the capacity of the system was enlarged, and the source moved upstream to Parkwater.

The City looked to groundwater sources when further expansion of supply was contemplated and installed its first groundwater pumps in dug wells at Parkwater in 1908. The main gravel aquifer of the Spokane Valley plains has provided essentially the entire supply of domestic, public service, industrial and irrigation water for the Spokane Urban area. Spokane is now the largest city in the nation to rely entirely upon groundwater sources for its public water supply.

In spite of the early reliance upon underground sources of water supply, little was known of the extent and properties of the Spokane aquifer until the early 1920's. It was then that the first studies were begun by the Washington Water Power Company to learn more about the regimen of the water available for electric power generation. The data and interpretative results of their efforts, reported by E. R. Fosdick (1931) have since been substantiated and expanded by later studies by several public agencies. Pertinent data from these later studies are given by

Weigle and Mundorff (1952) in Washington and by Nace and Fader (1950) in Idaho. The principal presentation of engineering data for well design is contained in Frink's 1962 paper for the Bureau of Reclamation.

Explanation of Maps

Features of the principal subsurface aquifer in the Spokane urban planning area are presented on a series of eight maps, Plates 303-28 through 303-35.

The groundwater maps whow the average summer elevation of the groundwater table within the principal aquifer of the Spokane Valley, the Hillyard Trough and in the lower valley of the Little Spokane River.

Water level records from selected wells were used to derive the groundwater contours of the aquifers shown on the maps.

The well records used as a data source are presented in Appendix I. Well locations are shown on the groundwater maps. The wells listed in Appendix I are those whose records afford reliable water level information over an extended period of time and are strategically located within the study area. Some of the wells shown on the maps represent a group of closely-spaced wells. In such cases, the one with the best level record has been selected. Most of the wells listed in Appendix I were constructed prior to 1952. The reason for this selection is that wells with good long-term water level information were considered to be the most valuable data sources. The data further indicate that there has been no significant change in the water table during the last 20 years.

A small part of the groundwater discharges from known and named

springs. These are described in Appendix II and are located on the groundwater maps.

Lateral boundaries of the aquifers are shown on the maps, although their location is only approximate. The elevation at which the water table intersects an impervious barrier is considered to be the aquifer's lateral boundary. In the Spokane area, the aquifer boundaries are normally sloping bedrock or Latah Formation preglacial channels.

The Spokane Valley Aquifer

Configuration of the Aquifer. The principal aquifer consists of a massive body of glacial outwash sands and gravels that underlies the Spokane Valley in Washington and the Rathdrum Prairie in Idaho. The principal aquifer branches at Spokane Falls, with one branch following the route of the lower river valley and the other passing through the Hillyard Trough to the lower Little Spokane River Valley. The general outline of the glacial materials which constitute the aquifer is indicated on Plate 303-2. Within the study area, the glacial outwash extends along the Spokane River to within 10 miles of the confluence with the Columbia River, and fills the Chamokane Valley.

The detailed description which follows refers only to that portion of the aquifer lying within the urbanizing area—between the Idaho border and the confluence of the Spokane and Little Spokane Rivers, at the head of Long Lake. Within this area, geologic and hydrologic data is relatively plentiful, by contrast with the more westerly portions of the Spokane Valley.

Through the central part of the Spokane Valley the aquifer consists of very permeable coarse gravel and sand that contains beds of boulders and cobble size as well as boulders and cobbles scattered through finer gravel. The materials are so bouldery that, prior to World War II, wells were dug and masonry curbings were handlaid more cheaply than machine drilling could be done. Even after machine drilling became the usual practice, most wells were not drilled more than a few tens of feet below the water table. Consequently, there is little drilling information on the materials in the lower part of the aquifer, below a depth of about 50 feet.

The high permeability of at least the upper part of the aquifer is known to continue west to the area of the basalt at Spokane Falls.

Northward through the Hillyard Trough, the aquifer materials are progressively finer grained; sand and even fine sand and clay beds become an increasing part of the materials below the water table toward the northern end of the Hillyard Trough. Throughout the aquifer, finer grained materials have been encountered in a number of drillings. One test well, 1/4 mile south of the Spokane City Well Field at Parkwater, is said to have been drilled 111 feet deep to an elevation of 1,842 feet, and abandoned because sand predominated below a depth of about 60 feet (Weigle and Mundorff, 1952). A 6-inch test well at the Diamond Match Company at the eastern side of Spokane (E 1/2 Sec. 14, T25N, R43E) is reported to have been drilled 90 feet deep to elevation 1,780 in saturated gravels and solely in "clay" to an unreported depth below elevation 1,780 (Newcomb,

1933). The well drilling records of Phillips et al (1962) and Crosby et al (1968, 1970, 1971) show rare clay or silt lenses were found in their wells and much of the sand and gravel encountered in the drilling contained 2 to 5 percent of materials in the silt and clay particle sizes.

The glacial outwash and flood materials which constitute the Spokane Valley aquifer extend to abut the sloping sides of the pre-glacial bedrock valley. In places, the gravel grades laterally into less permeable materials of outwash age; one place being at the Pasadena Park Elementary School, where reworked Latah materials interbedded with glaciofluvial sand and gravel near the side of Spokane Valley have been reported by Phillips, et al (1962). Thus, the lateral boundaries of the main gravel aquifer mostly consist of abruptly sloping surfaces of essentially impervious granitic and metamorphic rocks, slightly permeable Latah Formation and basalt with generally low but irregular permeability. In a few areas, the lateral boundaries include fine-grained outwash and other sedimentary deposits of outwash age which limit the width of the gravel aquifer. These boundaries between the aquifer and finer grained sedimentary deposits occur mainly across some of the larger tributary valleys, such as those containing Newman and Liberty Lakes. The dividing line between the coarse materials of the valley aquifer and the finegrained fill across these side valleys is only generally known from records of a few wells.

One significant deficiency in information on the boundaries of the aquifer occurs in the northwest part of Spokane where two gravelfilled gaps exist in the western side of the Hillyard Trough. The outcropping area of the basalt ledge, over which the river falls at Spokane, extends north nearly one mile to the vicinity of North Central High School. Between that end of the basalt and a 1/4 square mile knob of basalt, centered at Courtland and Elm Streets, a one-mile sag in the drainage divide is underlain by gravel. To the north of the basalt knob, another gravel-filled gap 1-1/2 miles long extends through Shadle Park and north to the slope which rises to Five Mile Prairie. These two gravel-filled gaps are shown on Plate 303-32, as "North Central Gap" and "Shadle Gap," respectively. While there is a lack of information in these areas, either or both of these gaps are suspected of serving as conduits for the westward movement of groundwater which has been inferred from river flow measurements described later in this report.

Plates 303-28 through 303-35 show the lateral boundaries of the aquifer, at the elevation of the water table, within the Spokane urban planning area subject to the limitations of data described above.

Knowledge of the thickness of the valley aquifer is derived mainly from geologic interpretations, drillers' logs of a few wells that have been drilled through the aquifer along the sides of the valley, two seismic cross sections (Newcomb et al 1953), and two deep wells drilled by the Washington Water Power company and the Bonneville Power Administration in the Hillyard Trough.

The seismic section across the valley just east of the Idaho

State Line indicates that the base of the glaciofluvial gravels rests on

material typical of the Latah Formation, which has low permeability. The

aquifer, some 3 miles wide, averages a depth of over 400 feet below ground

surface. The seismic information shows that the base of the aquifer is in the form of two broad "channels," the northern one being the deeper and reaching down to elevation 1,600. The average elevation of the aquifer base inferred from the seismic survey is about 1,690, approximately 300 feet below the water table for this section of the valley (Newcomb et al. 1953).

A partial check of the validity of the seismic data was provided by a test well drilled for the Bureau of Reclamation a mile to the west of the seismic section. This test well was near the state boundary in the middle of the valley and near the site of the Bureau's "Site 11" area which is in the NW 1/4 SW 1/4 Sec. 31, T26N, R46E. The test well was drilled to a depth of about 300 feet, or within about 100 feet of the aquifer base as inferred by the seismic survey, and did not reach the base of the glaciofluvial gravel aquifer. At a few places along the edges of the valley, wells have encountered the sloping top of the bedrock (granitic, basaltic or Latah sedimentary rocks) before reaching the level of the water table in the gravel aquifer, but no well is known to have been drilled completely through the aquifer in the central part of the Spokane Valley.

The seismic section across the Hillyard Trough shows the base of the valley gravel aquifer is an essentially even surface at a depth of some 300 feet, at elevation 1,700 (Newcomb et al 1953). The water table is at about 1,820 feet elevation indicating that the depth of groundwater flowing in the aquifer is approximately 120 feet through this section of the Hillyard Trough. At the north end of the Hillyard Trough

the base of the aquifer, below elevation 1,650, is believed to be uneven and be partitioned by bedrock spurs, based on observation of the ground-water discharges to springs and to the Little Spokane River at elevations ranging from 1,620 to 1,540 within a lateral width of some 2 miles.

A test well for the Washington Water Power Company was drilled in 1962 through the gravel aquifer in the Hillyard Trough. The well, 10cated in the NE 1/4 SE 1/4 Sec. 20, T26N, R43E, 1000 feet north of the seismic line, was drilled to a depth of 780 feet. Because the well was drilled to test the sub-aquifer section for gas storage possibilities, a record of the aquifer materials was kept only on a nearby water well (26/43-20J1, in Table I-1, Appendix I). However, the sub-aquifer samples start at 345 feet depth, the depth reported on the seismic section as the base of the aquifer. From 345 to 715 feet the well was drilled in clays, silt and sand. Cobbles and boulder gravel were reported from 715 to 780 feet. The description of the clay, silt and sand appears to show that this section of the drilling penetrated sedimentary materials of the Letah Formation, but the report on the drilling (Rieber and Turner, 1963) concludes that the material was of Pleistocene age. The cobble and boulder materials from 715 to 780 feet may have been a basal conglomerate of the Latah Formation, as the seismic section interprets granitic bedrock at a depth of about 740 feet. A mile north of the seismic section, a water well drilled for the Bonneville Substation near the center of Section 16 shows the base of the sand aquifer at 272 feet depth, 20 feet below the elevation of the base at the seismic section a mile to the south.

Thus summarized, data on the base of the gravel aquifer of the

Spokane Valley and the Hillyard Trough indicates a wide planar pre-glacial surface sloping gently westward to Spokane where the altitude of the base is unlocated but presumed to be near or below elevation 1,650, 190 feet below the basalt rim over which the river flows. The possible higher elevation of the base of the aquifer in the northern part of the Hillyard Trough may indicate that the base of the main course of the gravel aquifer trends west through one of the two gaps in the basalt rim south of Five Mile Prairie. Also, this higher base of the aquifer in the northern part of the Hillyard Trough suggests that the sand and gravel deposits of the northern part of the Hillyard Trough were laid down by southward flowing outwash streams tributary to the meltwaters coursing west through the Spokane Valley. The base of the aquifer may be underlain largely by the Latah Formation, but knobs, spurs and "islands" of the granite bedrock form the base of the aquifer in places.

Hydraulic Characteristics of the Aquifer. East of the study area, the aquifer extends into Idaho, where the elevation of the water table at its extreme northern end is influenced by Lake Pend Oreille. Prior to 1952, the natural level of the lake was at elevation 2,051. From this elevation, the water table had a gradient of about 2 to 2-1/2 feet per mile beneath the valley plains in the Athol, Idaho area to within about 5 miles of the State line. The gradient of the water table for the 5 miles east of the State line was 5 feet per mile (Newcomb, 1933). With the closure of Albeni Falls Dam in 1952, the normal pool level of Pend Oreille Lake was raised to about 2,059 feet, and the level of the water table in the North Pole and Athol areas adjusted to the new level.

Depths to the water table at the northern end of the aquifer in Idaho are in the order of 300 to 400 feet. At the Washington/ Idaho state line, the water table lies near elevation 1,980, about 100 feet below the surface of the valley plain. In its 20-mile course from the state line to downtown Spokane, the surface of the water table drops 1?? feet, at slopes varying locally from 4 to 10 feet per mile. The depth of the groundwater below the surface of the valley as it approaches the City of Spokane is in the order of 40 feet, and at a few locations, sand and gravel quarry pits penetrate beneath the water table. West of Greenacres the surface of the water table is at a higher elevation than the bed of the Spokane River channel, a factor which has a significant bearing on the transfer of water between the aquifer and the river.

At the basaltic rock ledge which forms Spokane Falls, the principal branch of the aquifer is that which turns north through the Hillyard Trough. Along this course, the water table slopes evenly at about 10 feet per mile from the 1,860 feet elevation which prevails immediately east of Spokane Falls, and then steepens to as much as 70 to 80 feet per mile near the spring areas in the Little Spokane Valley. As the terrain rises from Spokane Falls to the north into Hillyard Trough, the depth to groundwater increases to about 150 feet.

From the spring areas westward down the Little Spokane River, the water table drops from elevation 1,600 to elevation 1,540 at the confluence with the Spokane River below Nine Mile Dam. The groundwater in the sand and gravel underlying the floor of the Little Spokane River Val-

ley is joined, near the mouth, by groundwater contained in the other branch of the aquifer which follows the course of the Spokane River.

This second branch of the gravel aquifer begins below the rock ledge of Spokane Falls and apparently receives some groundwater that percolates through gravel-filled gaps in the largely impervious materials that form the west side of the Hillyard Trough. Beneath the Peaceful Valley area, just west of the basalt obstruction at Spokane Falls, the water table is at elevation 1,740. From there it slopes about 15 feet per mile, in general conformity with the river's gradient, to reach the 1,602 foot level of the pool behind Nine Mile Dam. The granitic constriction that caused Nine Mile Falls, and serves as a foundation for Nine Mile Dam, forms an obstruction which results in the steep descent of the water table to the 1,540 foot elevation at the mouth of the Little Spokane River referred to above. The resulting high gradient in the groundwater accelerates percolation through the gravel east of the granite knob, especially during rising stages of Nine Mile Reservoir, as described by Broom (1951).

Over a period of years, studies have attempted to determine the patterns and quantities of groundwater flow within the aquifer, and the water balance between the surface and sub-surface flows. The principal sources of information are flow records of the Spokane and Little Spokane Rivers, which indicate that these streams gain and lose increments of flow which must represent an exchange of flow within the subsurface aquifer. The exchange can be observed at numerous springs in the gaining reaches of the river valleys. The increments of flow are sub-

ject to pronounced seasonal and year-to-year variations, as a result of the interaction of the elevations of the water table and the river profile (Fosdick, 1931; Broom, 1951). The mechanism is, however, complex, and has defied several attempts at compilation (McDonald and Broom, 1951).

A comprehensive approach to the estimation of average flows was adopted by Pluhowski and Thomas (1968), who analyzed 50 years of streamflow records. They reasoned that "below the City of Spokane, the river flows on nearly impermeable basaltic bedrock, so that virtually the entire water yield of the Spokane River basin is measured at the Long Lake gage." For the upstream end of the study area, they developed an estimate of the sources in Idaho contributing to flow in the aquifer near the State line is as follows:

		Average Flow (c.f.s.)
Infiltration from Pend Oreille Lake (approximate)		50
Infiltration from Coeur d'Alene Lake and Spokane River above Post Falls Infiltration from Spokane River	250)))	370
(Post Falls to Otis Orchards) Infiltration from streams draining onto the valley floor and precipitation on the valley floor	120)	530
Infiltration from irrigation with surface water		50 1,000

The magnitude of this estimate is consistent with the conclusions of

Piper and La Rocque (1944) and the flow increments to the Spokane River observed by Fosdick (1931) and Broom (1951).

With the aid of the seismic data referred to previously, estimates of the permeability and transmissivity of the aquifer can be made, and compared with conclusions reached in the course of well pumping tests. The computations are to be found in Table 1, and indicate transmissivities of 25.4 million gallons per day per foot at State Line, and 1.9 m.g.d/ft. in the Hillyard Trough.

Determinations of transmissivity by the U. S. Geological Survey from six well-pumping tests in Spokane Valley are reported (Frnk, 1962) to have ranged from 1.3 to 13 m.g.d/ft. Pumping tests were run on many of the wells in the eleven tested areas of the Bureau of Reclamation in the eastern part of the Spokane Valley and yielded transmissivity values ranging from 6 to 13 m.g.d/ft. (John W. Frink, personal communication, 1973). Each of the wells tested by the Geological Survey and Bureau of Reclamation penetrated only part of the aquifer.

and the comments on the relative permeability of geologic units concern mainly the horizontal direction. In water-laid materials, the permeability in a vertical direction is generally less than that in the horizontal, and the numerical comparison varies greatly with the type of material and horizontal layering - from one-half the horizontal permeability in clean sand and gravel to as little as one-ten-thousandth in finely bedded fine grained materials." Though an estimate of the vertical permeability of the Spokane Valley aquifer might be

within the range of one-half to one-tenth the horizontal permeability, the figure is largely irrelevant because the known movement of water vertically to the water table is by unsaturated transfer and not by vertical saturated percolation.

Except for some areas of silty soils on some of the terraces and a general incomplete silt and sand filling of the gravel interstices in the top 5 feet of the subsurface, the gravel and sand above the water table is readily susceptible to the transfer of water downward from the surface. Air moves through the coarser gravel members during changes in the atmospheric pressure, and some depletion of capillary moisture must take place by evaporation; however, this evaporative loss from the subsurface is estimated to be a small part of the infiltrated water, probably much less than 10 per cent.

Westward from Post Falls, where the river descends 120 feet to the level of the water table at Greenacres, the river is subject to continuous infiltration losses which recharge the groundwater. The resultant river losses are especially large when the river is rising rapidly. Unrecorded visual tests of the unsaturated transfer process by which water moves down from that part of the Spokane River which is losing water to the aquifer have been made along the river's edge in the State Line vicinity. When a shaft is put down beside the river's edge, it proceeds in moist unsaturated sand and gravel; and, when a short adit section is then dug laterally to be in a position vertically beyond the river's edge, a drop by drop fall of water can be observed from the unsaturated sand at the crown of the adit.

The drop from the crown falls on the sand and gravel floor of the adit and disappears without evidence of wetness or saturation. Though no formal record of this hydrologic mechanism has been made in the Spokane Valley, by lysimeter or such measurements, the process is well known in hydrologic science. Also well known is the partial sealing effect that can build up at the surface as sediment carrying water infiltrates an unsaturated permeable granular subsurface. This phenomenon must occur in the Post Falls - Greenacres reach of the Spokane River and account for the river's ability to cross that permeable reach during periods of low flow.

Some spot tests on the moisture content of the sand and gravel material between the surface and the water table in Spokane Valley were made and results published by Phillips et al (1962), Crosby et al (1968), Crosby et al (1970) and Crosby et al (1971). These tests indicated that a moisture deficiency below field capacity existed at the time and place of sampling, including areas directly below drain fields of septic tanks. Crosby et al concluded that water does not move vertically down at the sites tested; and, consequently, would not do so from precipitation or other such surface or near surface additions of water. However, their data shows that some samples, particularly those containing some fine-grained materials do hold water at or above field capacity (Crosby et al 1971). In accordance with the rules of capillary movement of water in the unsaturated zone, it is probable that

their near-surface moisture moves down along devious subvertical paths formed by the finer grained materials which have greater capillary tension. This unsaturated transfer, by materials at field capacity, would form but a small part of the vertical column below recharge points and would be difficult to identify in the vertical drill holes used by Phillips et al (1962) and Crosby et al (1968, 1970, 1971). Some of their data show that no identifiable buildup of chemical residues occurs at the waste water disposals, a situation indicating that the downward transfer of the water and solutes along capillary-preferred routes is more likely than their conclusion that it (and the area's precipitation) is stored in the upper strata and removed by evapotranspiration during the growing season.

Currently the U. S. Geological Survey (USGS) is undertaking an extensive study of this aquifer which should provide an improved quantitative understanding of the operation of this aquifer. The USGS study is not scheduled for completion until after the completion of this study. The USGS study is described as follows, quoted from a release by USGS:

"The objective of the study is to accurately determine the hydraulic characteristics of the aquifer, including its geometry (especially in the primary recharge of discharge areas). A digital model of the flow system that will be developed to provide a basis for guiding a basic model upon which a water-quality model may be superimposed at a later date to develop an understanding of the possible effects of various water-disposal land-use practices. The model will be developed to simulate the hydraulic response of this aquifer from the state line to the terminus of the gravel aquifer near Long Lake.

Emphasis in data collection will be on: (1) Drilling and geophysical surveys to determine aquifer geometry at the Washington-Idaho border and in the areas north and west of the

City of Spokane; (2) development of water level distribution in, and pumpage from the aquifer; (3) determination of inflow from adjacent highlands; (4) determination of T and S values from pumping tests, especially in the area of the state line and northwest of Spokane; (5) determination of the hydraulic connection between surface and ground waters".

Use of Groundwater

Withdrawals from the primary aquifer are developed in other sections of this report, results of which are summarized below:

Use Category	Annual Withdrawal Acre Feet
Domestic	90,000
Industrial	17,000
Agricultural and non-agricultural irrigation	22,000
	129,000

The annual withdrawal is equal to an average rate of 178 cfs which is about one-fifth of the 1000 cfs estimated to percolate westward into the aquifer from Idaho. The domestic component of use includes a significant part that is for domestic landscape irrigation. The peak rate of summer withdrawal caused by the total of all irrigation peaks superimposed on the year around components of domestic and industrial use is of the order 400 cfs which is 40 percent of the estimate average aquifer inflow.

Of the total average groundwater withdrawal of 129,000 acre feet per year, only that part which finds its way to the City of Spokane sewer system is diverted to surface water. The estimated annual discharge of the City of Spokane sewage treatment plant, exclusive of storm

drainage and infiltration is 35 cfs or 25,000 acre feet per year. The remaining 104,000 acre feet are used (1) as domestic water supply in areas having drain field disposal of their wastes, (2) homeowner and public agency landscape irrigation and (3) agricultural use. All of these waters are returned to the surface of the aquifer. The extent to which any or all of these surface applied waters reach and recharge the aquifer is unknwon.

Interchange with Surface Waters

It has been shown above that approximately 1000 cfs of ground-water enters the study area at the interstate boundary and percolates westward through the main gravel aquifer of the Spokane Valley. It has also been shown that at present an annual average rate of 178 cfs is withdrawn from the aquifer for domestic, irrigation and industrial use of which 35 cfs is discharged to surface water and the remainder is returned to the surface of the aquifer for agricultural or landscape irrigation or through drainfield disposal of wastewaters. An unknown portion of that returned to the surface again reaches groundwater. It has long been suspected that the fate of the groundwater crossing the boundary, other than that withdrawn by man, is to be returned to surface waters by natural interchange.

In 1948 a study was undertaken by the U. S. Geological Survey in cooperation with the Bureau of Reclamation to obtain a better understanding of the interchange mechanism between the groundwater and surface waters of the Spokane and Little Spokane Rivers. The approach taken was to establish additional gaging stations on these livers to determine by

difference the gain or loss from and to groundwater. This study and results are reported by Broom (1951).

Although high flows during part of the year tend to mask the differentials which at that season are of the same order as expected gaging accuracy, the measurements of gain or loss from the Spokane and Little Spokane Rivers give the best available approximation of the interchange process and the transmissive capability of the aquifer. Except for a few springs above the level of the Little Spokane River near Dartford and a few small springs along the banks of the Spokane River, the groundwater discharges rather inconspicuously in the gaining reaches of the Spokane River and Little Spokane River. Only in certain parts of gaining reaches of the river can water be seen entering in sandboil fash on, and this groundwater discharge is evident only during lower stages of the river. The reaches of the river to which groundwater discharge takes place are those where the water table stands above the level of the river. Because the levels of both the river and the water table fluctuate as much as several tens of feet each year, the limits of the gaining reaches of the river change with the respective levels of these interacting water bodies. The water table fluctuates through an annual recharge and discharge cycle and in reponse to changes in river level as well as in response to other cyclic and noncyclic recharge and discharge events, including changes in response to long term multi-year variations in precipitation. Annual cyclic changes of 10 to 20 ft. in the level of the water table in wells have been described by Piper and

La Rocque (1944). The lowest general levels of the water talbe were reached in 1931 during a low point in the long range precipitation cycle.

Broom (1951) was able to reach the following conclusions based on measurements for the water year 1950:

				cfs	3		
		<u>Gai</u>	n by R	<u>ive</u> r	Los	s by R	iver
	River Reach	Aver	Max	Min	Aver	Max	Min
Spo	kane River						
1.	Post Falls to Greenacres		52 9		78	757	
2.	Greenacres to Trent Bridge	370	1140	39			
3.	Trent Br. to Green St.	566	1650			12	
4.	Green St. to Hangman Cr.		216		39	428	
5.	Hangman Cr. to Seven Mile	126	427			184	
6.	Seven Mile to Nine Mile	21				1028	
7.	Nine Mile to Long Lake	157				1422	
Lit	tle Spokane River						
8.	Dartford to Mouth	218	250				
	TOTALS	1458			117		

Note the large variation throughout one year, not only in magnitude, but in direction. The location having the most consistent direction of flow and quantity is the Little Spokane.

The general conclusion indicated by the average direction of flow in Bloom's investigation is confirmed by the water table contours developed

in this study. The reaches wherein the river recharges the groundwater and others where it receives discharge from the groundwater are evident from a comparison of the levels of the river and water table. For example, in the section from Greenacres to Green Street, where Broom shows the largest consistent gains by the river, the water table is seen to stand normally a few feet above the river level. However, rapid rises in the river level result in correlative rises in the groundwater near the river. These recharge waves in the water table are well shown by graphs such as Plate 6 of Piper and La Rocque (1944). The discharge from riverbank wells of the City of Spokane at Parkwater is commonly chlorinated during strong recharging of the groundwater by these episodes of rapidly rising river level.

Another aspect of the aquifer that is demonstrated by the water table contours developed in this study is the relative permeability. More permeable parts of the aquifer are indicated by flatter gradients; and less permeable parts of the aquifer are indicated by steeper gradients.

In order to make a water balance on the groundwater regime it is necessary to estimate recharges to the surface of the aquifer within the study area. Rainfall on the 90 square miles of valley floor itself creates a potential contribution of 26 cfs if 4 inches (about one fourth of the annual) are assumed to percolate its highly permeable surface. Rainfall on the 80 square miles of surrounding uplands could contribute another 15 cfs. As pointed out above, irrigation and drainfield

create potential for some fraction of 178-35 = 123 cfs to contribute.

If 15 percent of the latter is assumed to reach groundwater, the total potential surface recharge is 26 plus 15 plus 18 = 59 cfs. These highly doubtful items are seen to make up only a small part of the water balance. Thus a crude water balance would be as follows:

	Average Addition to Groundwater cfs	Average Deletions from Groundwater cfs
Entering the study area		
from Idaho	1000	
Recharge from the Spokane R.	117	
Recharge from Surface	59	
Discharges to the Spokane R.		1240
" " Little Spokane R.		218
Pumpage		178
	1176	1636

It is evident that either the Pluhowski and Thomas (1968) estimate of the inflow is too low or that the Broom (1951) estimate of river gains from groundwater is too high. A check made for other water years indicates that the years measured by Broom may be above average. On the other hand, Pluhowski and Thomas were working from methods inherently less reliable than Broom. Nevertheless, despite these inconsistencies, the general size of the groundwater stream and its interchange with the river appear to be confirmed by the water balance.

TABLE I
HYDRAULIC PROPERTIES OF THE SPOKANE VALLEY AQUIFER

	East of Idaho State Line	Hillyard Trough
Saturated section (sq. ft.) Width (ft.) Average Thickness (ft.)	3.82 X 10 ⁶ 13,400 285	2.10 X 10 ⁶ 14,000 150
Estimated Average Flow (cfs)	1,000	200
Effective porosity (estimated)	.25	.20
Velocity (ft./day)	90.5	41.1
Water Table Gradient (ft./mile)	10	26
Permeability (gals/day/sq.ft.) (cm/sec)	89,143 4.2	12,600 0.6
Transmissivity (gals/day/ft.)	25.4 X 10 ⁶	1.9 x 10 ⁶

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APPENDIX I

Representative Water Wells in the Principal Aquifer of the Spokane Valley in Washington and adjoining portions in Idaho.

AIT.	Date casing Kemarks		4-10-51 1679.2 (2)(3)	28(4) 11-28-41 1707.2	4-24-51 1710	4-10-51 1748.5	- VO)		4-10-51 2047.2	4-25-51 1946,0	4.23-51 17.42	5-8-51 1891.7	5-12-51 1856.3	2-10-21 1845.2
Water level	Depth		26.5	28 (4) //	15.9 4	34.4		7 2 2	183 4	78.9 4	74,0 4	14.1 5	12.7 5	20,4. 5
Water-bearing Zone	Depth)		93-125	40-79	32-46	34-59			198-212	50-124		78-02		
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Bia.	(£n.)		42		022/	00		40	°	912	24-12	ઝ	9;0	30
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Well	Number	7.25No.18.42E	341 3	11E1	11.81	2381	38+/ 1/28-1- 03-7:	•	100	N W W	7 11 8	; ;	, T.	

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Water	Depth		8:86	89.4	129,4	119.7	92.3	6.68	67.2	74.5	73.6	112.4	3%2
Water-bearing Zene	Depth		521-86			119-142			67-73			112-127	,
Water-be.	Alaterià l		Gravel		24 & dr.	Grave!	do	0,0	do	Sd & 9r	Grave!	sd lgr	Q.0
Casing	(F.4.)		90			150	,			8			117
Depth Dia.	(Ft.) (In.)	red	72	1.5	84	72-6	7.8	72.	24	60-30	3	30	4 %
Depth	(±)	ntin	57./	%	1+7	192	128	114	29	90 90	103	127	117
	5 3 2 3 3 3 3	T. 25 N., 12, 44 E Cantinued	Vera Irrig. Dist.	W. Shaw	15E1 Modern Flectric	Vera, Land Water	16E1 Modern Eketric Wafer Co.	Do	F. Lawhead	1901 Edgecliff Sanitar	20A1 S. Stauser	20KI L. Orstreicher	2151 Modern Electric Water Co.
V/c 11	Nomber	7.25	I3WI	14PZ	1561	15.57)3-80	17.87.1	18m1	1901	20il	20K1	2111

Bibbs at a historial distriction of the second of the seco

	Date casing Remai		0.2302 12-52	6-12-5/ 2016,5	6-11-51 20.5.0	6-71-51 2083,5	27.70	76.77 74.97-6	6-12-57 2-064	do 2066.0	5-29-51 2000	do 2040	7602 1576-9
Water level	A			80.5.	135.7			70.2	148.1	126.8	73.0	108.4	103.9
aring 20he	Depth CF#.)		165-181 165.3		135-163	196-180 176.9-		26-06	148-172	991-921		108-122	
Water-bearing zone	Material		Sd, coarse	2/69/2	Srave)	o D		1 amos	90	84, course 126-166		Sand	
Casing	depth (Ft.)		157		7.10	081		36	152	291			
Die.	(£h;)	und	\$ 4	2/2	18	9-72	81-8-6	4	72.	72.	36	<u> </u>	- \
Depth	(ft)	outio	181	105	891	180	. 61	76	921	991	<i>∞</i>	727	- 2/
<	Owner	T. 25 N., 12. 44.E. Contin	21N1 Model Water &	Light Co. Vera Irrig. Co.	Modern Electric	22RI Vera Irrig. Dist.	L. Lewis	R. Danklofs	2601 Yera Tring, Dist.	15.00 A	bj.	2981 M. Lettenmeier	- 13:31 J com
Mell	Nomber	7,25	21N1	22.HI	2.2NI	303·	2301	24.81	1077	261.)	1wsz	2981	- 2 3

Remarks									BR site 1A			BIR site 8A			BK site CA	•	•	
Alt.	casing		2015	2032	2030			2047	1.2062.1	2033	6-4-51 2059	1202 2057	2045		81202	2000	6202	12035
level	Date	.•	6-27-51	4-9-45	do			6-26-51	104.0 12-29-64 2066.1	6-22-51 2033	15-4-9	4110	8 -1 ret 2045		69-8-01	11-13-51 2060	3-27-42	5602 15-1-11
Water level	Vepth Date		72.9	8.0	8			64.9	104.0	49.0	86.2	1	75.5	79.5	106.2	126	17.9	17.7
Aring Zone	(FF.)			95-100		001-46			822-891		•		95-220 95.2		81202 10-8-64 25278	٠	·	
Water-bearing zone Debth	Materia			Sand		do			Gravel	do		٠	13240		John Mer			
Casing depth	(F.E.)			-		100			228	7.7	· ·		210	9//	214	130		_
	イカン	Je:	8	} i,	n	9		4	41-91	7)	9	20-16	৩	20-16	V		ė
Depth Dia.	(+1.)	tone	9		u L	100		88	225	2	0	& = =	220	9//	225	8	. %	<u> </u>
Owner		T. 25W, R. FIE, Continued	11. 11.	nomana	G, 100er7	G. Haw kins	T. 25 V., R. 45 E.	C. Sterfiensmier	U.S. Bur. Reclam.		人、られいとにか	Hell baum	3F1 (4.5, Bur, Reclaim,	F. Dilley	402 U.S.Bur. Rechm.	4EI O. Darrah	B. Eddy	4PI S. Bergerson
Well	Namber	7. 25		3046	240	3+14	T. 25	141		3-8	 7 12	341	3FI	32	4CP	461	4NI	4.01

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	1	Remari					BR site.									11-2-64-12035.31BIR site
	A1t.	casing		2060	2026	2033	611202	2015	20532	2019.2	2028	2060		7112	205516	2035.3
٠	Water level	Depth Date		6-14-51 2060	Ö	3-26-42	8-14-64	15-72-9	6-13-51 20532	do	do	op	6-26-51	3-19-45	92502 15-61-9	11-2-64
	Wate	Depth		87.8	623	26.7	70.9	1:29	87.6	50,5	01.0	12.1	58.1	7.191	94.9	87-230 87.9
	Water-bearing Zone	Depth (Ft.)				86-93	138-185	26-29							· :	87-230
	Water-be	Material				Grave!	-	Gravel	20	5d 4gr	Sraval	Sand			546gr	Gravel
	Gasing	depin (Ft.)			<i>o</i>		-85 -25		111				90		128	207
	i	(fn.)	pi	7/	47	30	20-16	84	9-09	36	. 98	. 🤣	৩	4	26	97-02
	Depth	(ft)	marke	137	8,0	93	195	96	111	29	85	721	9%	775	129	230
	•	Owner	T. 25 N. R. 45E., Coltinued	T. Skinn	Inland Paser Co.	W. Tenman	U.S. Bar. Reclam.	792 G. Grrier	SRI R. Rudebaugh	loci W. Lielman	10F1 G. Neyland	Jodi J. Morris	A. Winnestarfer		16C1 Inland Empire Paper Co.	1701 U.S. Bur. Reclam.
	We 11	Number	7.25	1H3	571	145	7.81	762		1007 3-83	10F1	INOI	11811	IAMI	1661	lati

ortheach solding in each case, by material and research sold excelses and an each ordinated to the contract of

	Remarks			1-18-64 2044,6 BK SITE JN:				7-16-69 2091.5 BK site ZA			•			♠	et same, con
AIF.	casing	72.5		2044.6	20/6	2056.5	1.40%	2091.5	2083		0,07		1555	1550	1545
Water level	Date	2-2 6.43	74 - 77 - 7	K-18-69	6-13-5/ 20/6	1-21 38 20%.5	3-25-42 2.041	1-7-91-2	15-8-9	.1	15-41-11		5-17-51	5-17-51	g,
	Depth	ه ا	72.8	94.2	11211	90.1	95.3	88.0	11211	· · ·	29.0		4.9	6. 6:	4.4
Water-bearing zone	Depth CFt.)		92-104 72.8	94-197		86-06	66-56	88 -217			29-80		· :		· ,•
Water-be	Materia)		ord sd	grave/		5d 4.gr	Gravel	Gravel			Gravel			<i>:</i>	
Gasing	depm (Ft.)			197			6.6	2/7	,		0%	,			
Depth Dia.	(Ft.) (In.)	200	72	16-14	હ	72-30	(n)	20-16	00		V		36	12/	1. 20
Depth	(Ft.)	ntinu	t 0,	207		96	66	230	134		80		721	<i>></i>	75
!	OWNER	T. 25 NJ, R.45E, Continued	17E1 Community Service 10+	U.S. Bar. Reckm,	R Toffore	o Wilson	T. Mac Donald	U.S. Bur, Rechm.	Zomi V. Hepton	T, 25N, P.46 E.	6FI Cenerale Inds.	T. 26 N., R 42 E	3EI J. Willis	4L1 A. Rattig	FFI W. N. P. Scher
Mell	Number	7.25	17E1	1961	1761	1 4 8 1		188) 3-84	2.0M1	T, 2	- 6F	7:2	361	471	T. I.

APPENDIX I (continued)

	Remark											•	· ·	•
A1+.	casing		1543	16.05	いいと	163¢	3:5	552	1665	829/	1627		1682.	Manda Pergerap
Water level	Date		5-17-5	5-16-57	0	9 5		C S S / 1 S - 2/- C	5-17-51 1665	15-91-9	90	•	17-11-21	
Water	Repth		.6.3	77.5	67	13.4	4.3 4.		56.8		16.3		15.8	
ing Zone	Depth (Ft.)		• -			•							٠	
Water-bearing zone	Materia)										6 ra ve 1	o O		•:
Casing	depth (Ft.)								4-2	× ×	1	45 4	90	
Depth Dia.	(Ft.) (In.)	S	12-24	<i>∞</i>	2	12	36	v	35	v	30-6	09	09	
Depth	(f+;)	linue	9/	, 00	451	50	m N	204	42	22	2/6	34	20	
	782 n	T. 26 4, 1. 42 E. Continued	541 W. Norman	Wash. Water	Fower Co.	R. Fellow	W. Allen	C. Swan	G. Mc Lr Van	H. Wilson	W. Danly	21E1 11/13 J. Ray	28D1 4.5. Gout.	
Mell	Number	T. 26.	175	1219	ואר	IN'S	303	-85	20A1	2161	2101	2.1E.1	2801	

METROPOLITAN SPOKANE REGION WATER RESOURCES STUDY.

APPENDIX P GEOLOGYAND GROUNDWATER

JANUSES 076

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(2) 325p.

Department of the Army Corps of Engineers, Scattle District

Konnody-Tudor Consulting Engineers

K KENNEDY



		Remarks		Cline, 1767	•	Cline, '969		Cline, 969		de.		Chine, 1969.		d		•	•
•	A14.	casing		1870	0681	1830	1.4.81	15.80) ;)	1760	0.3/11.	1775	1730		2	1782	7
•	Water level	Depth Date		164.2 11-14-62	1957	- 7-17-64 1830	4-17-45	3-30-46		.4. 5-7-63	74-4 6-19-51 1715.0	1063	4	. •			7-18-4.5
٠,٠		Depth De			100		27-26 22.8		0001	93-114 43.4	71 88 74				59-30 38		0114
	Water-bearing zone	Materia}) - ()		Gravel	Sand 9	_	•	Sand	SJEGr	Sand S		Sand
	Birsh	aepin (Ft.)		112			į		٠ ن ن	93		80 80	54.		90		0.
	٠	$\overline{}$		47	્		7	45	12	9		72	ઝ	. 01	7.7		4
	Depth	(ft) (fh)		7115	150	•	ţ	56	30	124) !	88	5.4.	6 33	90		4
	1	OWNER	T, 26 N/, R. 43 E.	O. Humbhries		***	Nera Jeneol Dist	H. Wood	Riville Noice Co.	Wash. Moter	Power Co.	Mary Felate	m. Deshier				8F1 E. Billberg
	Mell	Number	T, 261	101	747	<u> </u>	4 ·	401	189	303-8		7.01	8		700	÷ 0.0	8F1

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Warden markey a way of

		Remark		Chre,175							Cline, 196	Top Lalah at 295.	Chie, 190	
	A1t.	casing		0:61	6061	0161	1938	1937	6	× 6 6 7 .	1940	1942.3	1940	57.61
	level	Date		3-29-62	6061 145	0161 1461	4-17-51	4-12-51		15-11-9	4-25-55	11-24-42	5-1-64	5-21-57
	Water level	Depth		88.8		Dry .	162.3.	159.9		159.5	157 4	i tesi	173,5	7.86.1
	Water-bearing zone	Depth CF+.)			112-130						238-268		199-220	150-183
	Water-be	Material		Sd है पुर	1 op	् च	ס	•	Q Q	9	do	Jo	30	Sand
	Casing	depm (Ft.)		901	124.		777		.247	286	238	34.6	724	· ,
		(Ft.) (In.)	ρö	2-22	9	00	£ ~ //		9-8	9/-02	24-18		9-8	૭
	Depth	(Ft.)	+100	106	130	735	i i	780	247	987	2.68	556	2.24	83
	1	Owner.	T. BONL, R. ABE, Continued	Wash, Water	المعامل الم	e de la la		Chem. Carp.	00	00	J,	Bonne Wille Power	Misson Landscap. Service	J. Miller
	Well	Number	V9 :: 1.	IOKI	7 7 7	1901	1361		1091 30	£09- 3-87	15F2	1591	1811	1921
1		i												•

	•	Remarks		Cline, 1969.					Cline, 1969	Do. Found top laise im. at 360ft.	Cline, 1969.			•	•
•	A14.	casing			5161	1935.6		0202	0561	1102		2761		01.41	203.0
	Water level	· Date		175,1 4-27-65 1945	5161 17-52-21	7.5261 15-61-9		15-22-5	6-12-62	5-2-63	3-29-65 2040	7-7-42	, , , , , , , , , , , , , , , , , , ,	5-25-51 1410	12-20-41
	Water	Depth Date		175.1	. 8'091	135.0		7 8	152	176.3	2.802	C.081	: 9:29/	170	171.4
:	Water-bearing zone	l Depth (F+.)		2.28-7.43	160-180			220-248	·	512-561		186-206			
	Water-	Material	•	. 167 PS			348	90	Sd. 2.9 p	sd, fine	54 4gr	Sand	op	op	90
	Casing	4epin (Ft.)		248	140				253	270	238	180			180
	Dia.	(£n.)	Ţ.	9	72.	•	74-60	૭	21	\	. 7	84	96	8	4
	Depth Dia	(Ft.) (In.)	tinue	24.8	700		191	2.48	50 50 50	430	238	. 907	163	216	200
•	,	Owner	T. 2.6 K., R. 43E., Continued	1751 El Paso Natural	643 Co.	Wreterries compe	County Homes Estate	c. Wespect	Whitworth Water Dist. z, well z	2011 Wash. Wotor	20NI Wie Joser Power	Co. Pac. Niv Alloys		23 M DL: 11/48 Refin. 6.	27F! National Pole
	Well	Number	7, 2.6 K	1751	1 60	250	1941	1461	70 D S S S S S S S S S S S S S S S S S S	2-03.1	20N1	77	2161	73000	27F!

in contact contributions and the contribution of the contact contact and the contact of the contact of the contact of

					• • •		-		-			• • •				•-
		Remarks			Cline, 1969	Do.	2		, , ,	مدوون رو		Spok. Terrasc	Central Ave			•
	A14.	casing		1995	2025	12031	1502		2053	2046	7:02	20.20		6-19-51 2035.6		2020
•	Water level	Depth Date		12.22-5 8.271	197.0 10-9-64	5-2-64	211.9 10-9-64		209.5 7-16-64 2053		-	02 02 15-91-11	59-21-8	6-19-51		5-23-51 2020
		Depti		172.8	197.0	190	6112	•	209.5	204,8	229.7	4.82.4	522	172.9	, 	1911
	Water-bearing zone	il Depth		172-207	262-012			211-251	239-259	204-2-18	852-522	229-293	012-522		• .	··· ·
	Water-	Material		548gr	9	~	S)	Sand	Gr # 3d	167 PS	Sand	54 & cyr	0) -		St & gr.
	Casing	depin (Ft.)			310		_	رة الم	652	602	2.38	293			240	
		(Ft.) (In.)		26	30-19		70	9.	00	90	8/	2			72	54
	Depth	(Ft.)	inued	207	3		250	152	259	817	17.00	293	27.6	. 2	240	140
		OWNER	T. 26 N, R. 4-3E., Con inued	221 Milion Poke	36.77	C, Cainins	Nispokane +rrig. Dist. 8, well 2	C. Calkins	G. Chortor	Holu Gross Gem.			power Co.	Sur-Jock to his	34PI Great Karthan Ry.	35EI J. McKay
	Well	Number	T. 26 N		17) 7	1087.	28H1	28Ni1	303-	3061	1308	30R2	ŕ	JAK!	3461	35E1

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APPENDIX I (continued)

•		Remar		***************************************												
	A14.	casing		2.003	2.100.7	2100		2.130	2195	0602	2079.0	1.602	2105	2063	2013	L9c 2
	Water level	Date		3-12-28	6-19-51 2001.7	11-19-51			10-31-51	102.6, 11-5-51	0,0	11-1-21	329 4-5-42 210S	do	3-17-28	15-11-9
	Wat	Depth		102.	92.3	72.2		14.5	152,9	102.6.	116.7	100	32.9	76	99.9	9:16
	Water-bearing zone	Depth CFt.)			92-113					•.				101-26		
	Water-be	Material	•		Gress			Sand		Gravel	PS & 15			jenery		Gravel
**************************************	Casing	depin (Ft,)			= %	14.0		156		126				101		S
	D. ts.	(In)		99	72-92	9		٠,	৩	৩	50	٧3		4.	2.5	<u> </u>
	Depth	(ft)		106	<u>[3</u>	15.8		981	651	92)	661	72.61	14.8	. 96	121	123
	200		7. 26 NO. 8. 44 E.	32NI R. Frazier	Hutlen Settlement	C. Sharber	T. 261/1, 12, 45E	13N2 E. Mellick	1, 807	2 SDI A. Maurer	W. Beck	V. Pinther	W, Chff	6, Sly	Ver . 7	33Pl G. Kenney
L	Well	Number	T. 26 H	32NI	32R1	3,81	T. 26M		-90	las z	1352 .	2:111	32.31	33.	3352	3301

APPENDIX I (continued)

Cepth Dic. Casing Water-bearing zone Water level top top to Casing Cepth Depth							,		6R S.7e 10A		BR site 78.		Remarks	
a. Casing Water-bearing zone Water (Ft.) Material Depth Depth (Ft.) Material Depth (Ft.) Material Depth (Ft.) Material Depth (Ft.) Material Depth (Ft.) Depth (Ft.) Depth (Ft.) Material Depth (Ft.) Dep	16602	1.602		5802	2076	20%	3307	8202		2017			casing	A.H.
a. Casing Water-bearing zone Water depth (Ft.) Material Cepth Depth Depth (Ft.) Material Cepth Depth Depth (Ft.) Material Cepth Depth (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) (Ft.) Material Cepth (Ft.) (Ft.) (Ft.) (Ft.) Material Material Cepth (Ft.) (Ft.) (Ft.) (Ft.) Material Material Cepth (Ft.) (11-29-41 2099	3-10-28		11-14-51	6-14-51			15	29-6-21	3-17-28	57-91-1		Ì	r level
1. Casing depth (Ft.) (F	139.4	124.2.	Diğilmiyi in ildə i	110.5	7:86	123	011			94.8			Pepth	
1. Casing depth (Ft.) (F	Gravel 139-140+	•					110-123		184-227		171-228		Depth CF+.)	earing zone
ia. Casing depth (Ft) (Ft) 5 135 123 142	Gravel				•	ه ت	Grave!		Gravel		Gravel		Material	Water - b
Nner Repth Dia. ". 45 E., Continued (ft.) ". 45 E., Continued 238 20-16 ". Bur. Reckum. 238 20-16 Ahite 125 2.5 Bur. Reckum. 232 16-14 Ceeber 141 6 Siener 123 5 Imms Donn. 142 5 Siener 129 6 Sennett 111 6 C. 46 E. 6 Beck 130 30	140	-			•	142	123	135			228		aepin (Ft.)	Gasing
### (Ft.) ". 45 E., Continue ". 45 E., Continue Bur. Reclam. 232 Bur. Reclam. 232 Res ber 141 Res ber 142 Jener Co. 142 Jafer Co. 142 Sennett 1117 C. 46 E. Beck 130	1901 22-3			V	S	Ŋ	h	0	41-91	2,5	20-15	70-1	(In.)	Dia.
nner ". 45 E. Co ". 45 E. Co Bur. Rechm. Ree ber Tiener Jiener Jufer Co. Jafer Co. Sennett C. 46 E. Beck	1.01	130		11.17	129	142	123	141	232	571	238	1 inc	(ft.)	Depth
2 2 8 8 2 5 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	od.	J. Beck	T. 26 NI, R. 46 E.	3601 I. Bennett	36NI H. Segerstrom	E. Farms bom. Water Co.	M. Diener	C. Reeber	35F1 U.S. Bur Reclam.	R White	U.S. Bur. Reckm	V, R. 45 E, Co	, sumo	7.50
Well Number Number 7. 26 34.2 34.2 35.7 35.7 7. 26 01 35.7 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26 7. 26	- 32m1	35.E	7:26	2601	. 35NI		36 A1	35.71	35 F1	3+P1	34.2	7.26	Number	Well

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		Kema Ks			1777 1-29-65 2090,5 BRSIFE 11A	-							
•	A/t.	Casing	•	2085.9	2090,5 6			<i>4.7805</i>		2/00.6	2075.7	2/12.2	
	Water lavel	Date		6-14-51	59-62-1			F-28-48 2087.4		3-50-49 2-100.6	5-30-48 2075,7	5-30-48 2112.2	
	Wate	Depth		106.2	. 7.771			103,4		120.5	87.3.	33.8	
٠	Water-borning zone	Depth (++.)		106-147 106.2 6-14-51 2085.9	183-238			611-801		120-200 120.5	87-130	33-55	
		Material		Gravel	90		•	65.3401		5r.4.3d	Gravel	, P	
	Casing	(FL)		14.7	80 27	01		611	,	200	130	SS	
	Depth Dia,	(Ft.) (In.)	pur	9	20-16	AH		9	•	0	9	80	
	Depth	(Ft.)	ntin.	147	249	IDA		611	0.01x1	200	132	5.5	
	(Owner	T. 26 N. R. 46E., Continue	31 MI M. Arnold	U.S. Bur. Reelam.	·	T. 50N, R. 5W.	5MI G. Anderson	T. 50N", R. 6 W.	State Line Hillage	J. Holland	M. Holland	
	Well	Nanioor	7.26	31 MI	31112		•			171 .	12K1	1361	
- 1							3(03-92	!				

THE PARTY OF THE P

-			<u></u>	-		1						
	17211 Aly or ber		6wner	Depto	Dia. (fr.)	Casing depth (Ft.)	Water - be	Water-bearing 7 eme	Water level Depth Dale	levol Dale	Alt. Ter Casing	Remark
			IDAHO	0 - 0	ntinued	نز م						
	7.5	T. 51 N. R. 5W.	2.5W.								į	
	1361	1 11.	H. Just	200	∞	2.00	grave)	182-200		9-6-49 4112.3	2172.3	
	146.2		C. +Fr. 22	122		122	90		116.7	1,1,9-1,	21762	
	1981	1 NP Ry		181.5	12-10	5'181	op		•	8-30-49 2105.4	2/05,4	
	303-5		P. Buck	156		156	. ر م	901-11				
	i	T. 5111, R. G.W.	.6 W.									
	525	25JI C. Beck	· 	175	ی	173	á'o	67.9.6 0.811 56.49	118.0	9.6.49	2/012	,,
	Fee 1	Footnotes: (1) Numb grid The (2) 100 til	of Numbers for Wells and springs are derived from their location in the land survey of Numbers for Wells and range are followed by 3. hyphen and than the section. The letter refers to the 40-acre tracts the lettering is in successing west east across triers from A in NET NET to R in SELSET, I and O are omitted to mells in Washington is largely from Weigle and Mundor 17.1952. in I daily are from	Ship &	d spr nd ra the from	nge a: 40-ac: A in N	e derived e followe e tract; 154 NE4	from the to by 2. h the letter to R in	ir loca yphen a ring is SELLSE	tion in sur in sur f. I ar	the la n the cessin d O ar	nd survection. y west somitte
armit Sa		Na.c.	Nauce and Fadely 1950, Data on Burleau of Reclamation wells are from Unished rec	5 1950	Datal dupih	on Bur	eau of Recland sur	lamation face datu	wells is	ere frontstant	Heir pui	Vished rec
	i A	STrain	(5) Indicates a few feet of pumping drawdown when water level measured.	feet o	f pum	ping di	awdown	uhan wat	er lere	I mea	sured.	
The state of the s		the state of the state of	_									
	er en		i. A	क प्रदूष्णिय कृष्णि क्षित्र कृष्ण क्षेत्र कृष्ण क्षेत्र कृष्ण	The state of the state of	The transfer of the state of th		en e	YEAR TEREST	ERREMENER	TENNESS SERVICES	

THE PROPERTY OF THE PROPERTY O

APPENDIX II

Springs that discharge groundwater from the Principal Aquifer of the Spokane Valley.

	7	Kemarks	One of several flews from continues	one of several of southern part of Down river Surs.	Fish rear- Hatchery Spirs,	Country Club Sprs.	•	Wandenme Spes
	Principal	əs /7	None	Non.	Fish rear-	.Trr. Dom.		fish cearing Irr, Dan.
	Water-yielding	material	Sund and gravel	do,	erarel	d <i>o</i>		5.21-55 Sand & gravel
	10	Date	8-13-73	25 8-13-73	59-51-9 0089			5.21-55
	Vield	m 95	75	. 52	6800	4000	•	3 300
	Alt	(Ft.)	1700	1700	1590	1590		5991
			7.25N, R. 42E.	•	T. 26 N., R. 42 E. 11515 Wash. Dept. Same	12 Als Spokane County	12 L/s	Carrie merestre 1665
	Spring	Number	7.25	12PIs	7.26	12 Als	12 6/5	27.73
J		•			303-94			

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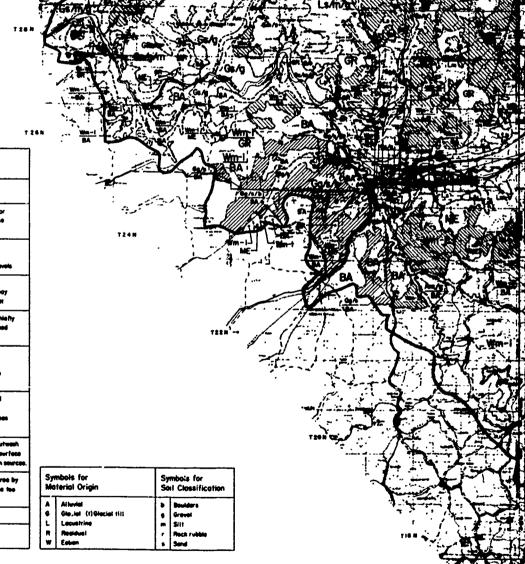
APPENDIX II (continued)

THE PROPERTY OF THE PROPERTY O

	Remarks		Waikiki Sprs. Cline, 1969	Do.				he mumbers r s.
	use Use		None	Fish raning Domi		•		er as t the letti
Water-111e lding	Water-yielding material		Gravel	do,	·			(1) Wilmhors of springs are derived in the same manner as the numbers assigned to well (Table 1), with the addition of the letter s.
19	Date		0061	1900				ved in
Yield	@pm		4000	1500		-	•	der (Table
AIT	(Ft.)	P	1600	1600			 	is an
	Owner		6015 Waikiki Synd. 1600	De	· ·			a ssigned to
Spring	Number	T. 26 N., P.43E., Confinard	6015	7818				() ₁ K::

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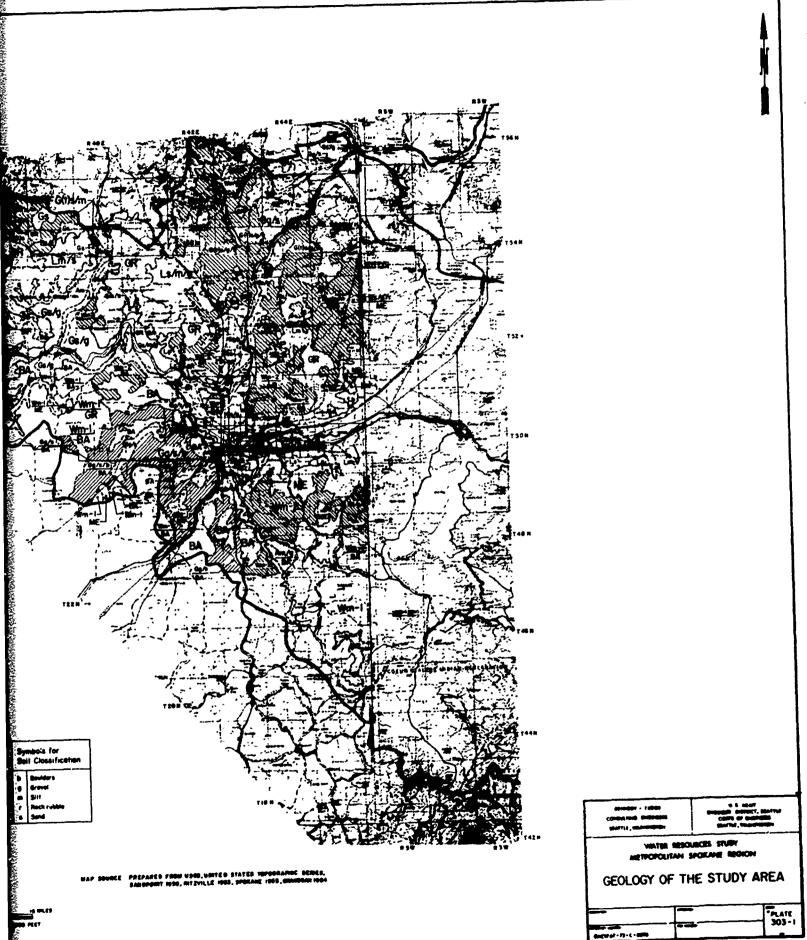
Symbol	Material	Description					
Am Lm	Alluvial Silt Lake Silt	Streem had deposits Undifferentiated set, organic sitt and peat deposits					
Wm-1	Lesss — Silf, Clay, Fine Sand	Loses of the Polouse Formation. May contain clay and for fine sand. Primarily sellan, but includes some local lake and revaried loses deposits.					
	Silt, Send, Gravel, Boulders	Primerity reworked Palouse Formation less. Often mixed with till, outwesh, leke, or collusted meterials. Generally shotlow (<20'), overlies reck or outwesh grovels.					
Gs Ls	Send	Prodominently send from glocal or locustrine origins. Often in stratified, terraced deposits. Surface areas may be reworked by wind, Occasionally contains gravel or set					
	Send, Gravel, Rubble	Unconsolidated sendy sale of the mountainous areas. Chiefly of residual and collevial origin, but accessionally is mixed with HII or autuses materials, everiles bedreck					
64	Gravel Send	Prodominantly glocal flood and outwoth deposits in velloys and on the lower slopes of flonking ridges. Generally scarce sends and fine to occrse ground with cobbins and boulders					
35	Gravel, Sand, Boulders	Thin (<20°) outwesh and fleed deposits over secured baselt surfaces, west of Spokens Highly irregular in thickness. May centain silt or clay leneas, includes areas of baselt outcraps, (Scathand deposits)					
8::	Gravel, Send, Rocky THI, Boulders	Ground ormeroi.not titl deposits. Generally mixed with outwook materials: Shallow to medicately doop (5'-30'). Near surface zones generally mixed with silt from locustrine or collen source					
84	Basell Flock	Besett love flows. Fractured, Undertein in the Systems area by sitistens of the Latch Formation. Exposures of sitistens too small to map at 1°250,000 scale.					
ME	Metemorphic Rock	Onaico, achiet, quartzita, etc.					
GR	Granitic Rock	Intrusive rachs, including granite, quartz, manzanite, andasite, etc.					

NEVISORS								
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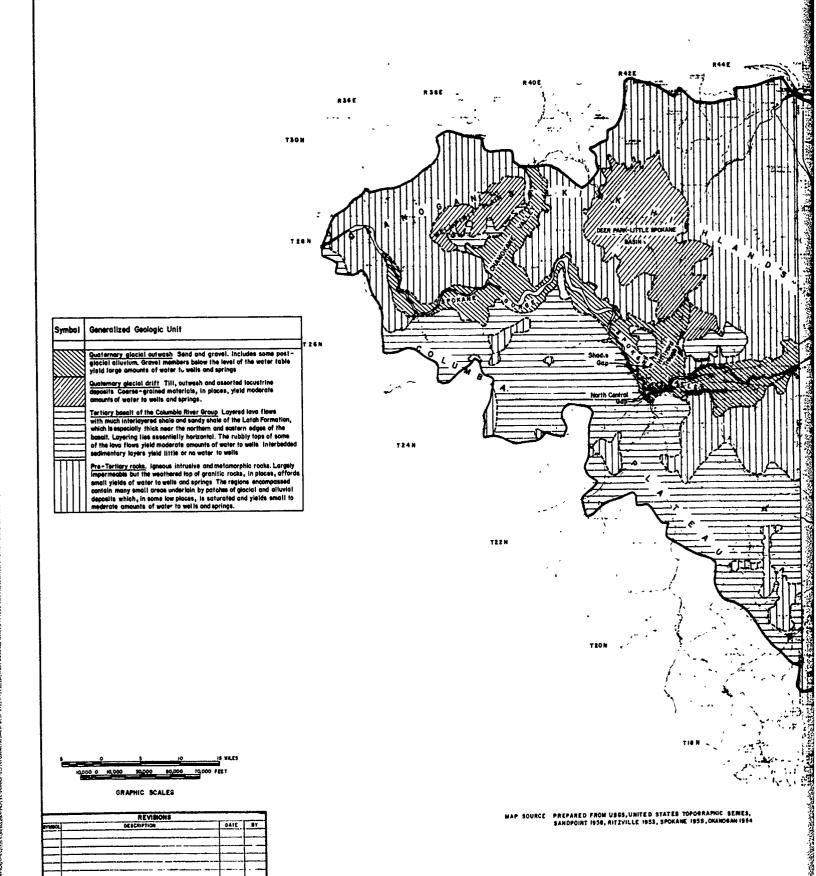
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WATER RESOURCES STUDY

GROUNDWATER REGIONS OF THE STUDY AREA

*PLATE 303-1

8

EVIAL CO	P loz	Map Unit Symbol	TASCRIPTIUM	DRAINAGF	FOUNDATION STABILITY	2
OPLANT			Lacustrine post deposits Fibrous and granular organic deposits in recent shellow lake and marsh areas; grades to organic and sale) and near outer margins of peat bods. Recent and Late Pleistoceans.	Poor because of high groundwater Permeability usually high in fibrous peats, but decreases rapidly if peat is consolidated	Not suitable for building foundations Debantment fill loads may settle several inches and settlements will continue over several years.	Does not add when dry, bu
	+		Recent and late Pleistocene Bolish sits (Pelouse Formation Loss); tan to brown, includes time sand, clay and occasional caliche zones. Pleistocene	Poor Permeability may be moderate in zones with a high	Generally adequate for light frame structures; heavy structures usually require special foundations; semi-stable particle	Usually sta failures as subjected t slopes wilk
			Provided Palouse Portation with (Lores): iscludes becant and Playsropers (Dest) Octobrolly Sind With hand grows and cobbins from residual collustation of glacial sources; mixed with Alluvial or Industries deposits in places and occasionally contains ice rafted erratic boulders.	concentration of root lubules. Whenly susceptible to erosion on slopes.		back or she
317		Am Am/s Am/c	Alluvial sitt. Located along streams in Local decressions, and mantles flood plains any contain sead and gravel, and may overlie glacial flood or la-ustrine deposits. Recent.	Foor, clayey areas are practically impervious Generally occurs in low areas. Infiltration is slow.	•	Cenerally &
	-	ia Is/a	Levestine silt Lake deposits in electably blocked valleys tributary to the Spokane and Little Spokane valleys; may contain sand and ice-rafted boulders; merges with past deposits near Herman Lake, Liberty Lake and Saltese Flats. Recent and late Pleistocene.			47941,
		Cs Cs/s	Colluval sits Principally alogowash from aeolian and residual hattrials on mountain slopes; say contain colluvial sand and residual sand and gravel. Recent.	Poor, but variable Susceptible to crosson on slopes	variable but generally adequate for light frame atructures; heavier attructures require special foundations Frost suaceptible.	Usually sta failures an aubjected t
		A0/4	Allovies cand Deposited along atrease, prancipally elong Latch Creek and Liftle Spokane River; locally grades into allowies site (Am) and gravel (Ag). Secent.	Generally poor due to high water table Cood when above water table (As), but poor in silty areas (As/m)	Generally poor because of high water table and local interlayering with allowish milts (Am)	Generally of poor, due t
		La La/g Ls/m	Lecustine sand. Clocial outwash deposited in Pleistoceme lakes, usually well sorted and stratified. May contain occasional sist or gravel.	Generally good, but affected by degree and type of stratification. May be goon in near surface some containing appreciable assuming of silt (Le/m). Infiltration rates are generally high. Finer sends (Ls) subject to erosion on unprotected alopes.	Generally of medium density and suitable for most light executives. Looke somes any dictate special foundation restants, pericularly in dune anadistis. Limiting settlements assually govern foundation design. Frost assorptibility moderate to high, being most severe in areas containing appreciable slit. Generally enaturally enables and severe in areas containing appreciable slit. Similar to above, although foundations will often penetrate through these soils and bear on underlying parent rock. Similar to be, We and Ge series.	
93		Ye .	Bolian sand. Dane deposits; wind reworked glacial outwash sand, Recent.	Cenerally good Subject to erosion on unprotected slopes		
		Gs/m Gs/m/q Gs/g/m	Clarial outwash and. Principally electal fluvial deposits: Includes zone kame deposits and vind recorded surface dumes. Generally stratified and may contain occasional fine gravel or silt bedding. Pleistocene.	Generally wood, but affected by degree and type of stratification foor where appreciable silt is present (Gs/s, Gs/s/q). Finer sands (Gs) subject to erosion on slopes.		
		Ro/g	Residual sand. Derived by weathering of metamorphic and grantic focks; generally less than 1ft thick, aithough say exceed 10 ft locality contains gravel to boolder size us'dual rock frequents and may be mixed with glacial outwesh or till, sand and gravel. Recent.	Generally poor duc to high slit content (RE/r/m, RE/m/I) and proximity of parent rock. Good in slit-free areas (RE/I,RE/q).	Similar to above, although foundations will often penetrate through these soils and bear on underlying parent rock.	
		Cs/q/m Cs/q/m Cs/m Cs/m/r Cs/1	columin) and openue). Principally alone wash from mountain aloness includes residual material and is generally mixed with resorted deads. May be mixed locally with glacial outwash or till materials, Accent.	Righly variable, but generally poor due to silt content (Cs/u/m).	Similar to Lo, Ws and Ga series,	
di,		Ag/s/# Ag/s	Allowisi graves, Deposited along portions of portions of Latah Creek and the Spotene River in bers and nerrow terraces. Generally occurs with allowisi sands (As). Recent. Glacial flood and outwass grave! Principal deposit of the Spotene	Permeability is very high, drainage poor due to high water within river and atream channels, Cood Permeability wary high, drainage at surface low to moderate	Suitable for most structures Suitable for most structures	Does not o
Can		Gg/s b Gg/s Gg/m Gg/s/m C(talb	Clacial flood and outwass gravel Principal deposit of the Spokene Valley. Rosity Flood gravels deposited following benechings of any dass in the Clark fork and Coeur d'Alene Valleys. Upper 1 to ft, is sized with silt and safel Jeposit is mostly coarse, poorly sorted gravel with some sand, cobbles and boulders. Pleistocene. Colluvial talus Gravity transported rock fragments from	where mixed with milt (Gg/m, Gg/m/m). Generally poor due to high proportion of milt. May be yord at	Generally poor, but highly variable and individual sites must	flattening are unpred
HIXTORES		C(ta)b C(ta)s/m C(ta)a C(ta)m/s C(ta)m	upslope Took Outcrops, size ranges from gravel to "house" airs blocks to Large bessit blocks are locally called "haystocks: Talus is often sized with ailt and sand and locally overlaps outwesh and till deposits. Recent	surface where silt is missing, and infiltrated water will flow through buried more pervious zones. Runoff is high on slopes	be investigated Generally ound at the base of basait cliffs in areas most difficult to investigate and develop.	vill gener treatment. conditions
AND SOIL NI		G(t)= G(t)=/b G(t)=/=	Glacial till. Sand to boulder site, unsorted moralnal deposits; Often mired with silt and outweak material.	Highly variable depending on proportion of silt, sand and gravel. Jenerally moderate to good,	Generally good	Cenerally
100		C(1e)	Colluval Indelida Seposits. Cravity transported local earth saterilals deposited under side or flow conditions. Unsorted materials may range from clay to boulder sizes Recent.	identified only at one remote location within area and sufficient data is not available to characteristics. Secause of the variable nature and generally development of area underlain by mammade fit by thorough investigation and study.		
1		Fs/q Fs/q/d Ft Fd	Namede fills. Symbol shows predominant material where known, Hanilary landfulls and mixed debrie fills are designated Fd.			
		3.4	Baselt rock flows of dense, derk gray, jointed baselt of the Columbia Biver Group; essentially flat lying; may be tens of feet thick. Tertiary.	Fractured zones and flow context somes in besalt may carry water Seegges often occurs along rock surfaces. Permeability varies from high to impervious according to the openness of fractured zones and the characteristics of the interbed or interformation context.	gcclient. Neavity loaded areas on heaslt should be explored to determine the presence of highly resicular on fractured gones immediately below the footing	Steep alope provided at with time,
, jog		ΧĽ	hetemorphic tock primerily belt series gesies, schist and quartitic. The seess is typically banded and medium to coarse testured; the schist is typically contorted and fine to medium testured. Shallow to moderately deep alteration may occur locally. Pre-Cambrian. Granitis rock. Redium to coarse grained; may be local's altered. Cretacons.	In ervious where Massive but may carry or store water in 'ractured mone, joints, or in weathord or sitered tones Firmedbility is dependent upon degree of fracturing, open-mas of joints, and type and extent of alteration.	Generally excellent mithough highly weathered or altered sites may require special investigation gard, unweathered phases excellent. Where exposed near surface. Generally	
*	,,,,,,	GR		Programme to the state of the s		
		57	Siltatone (Latah Pormation) Predominately consolidated lacuation deposits but contains some cluvins islt and fine sends tan yellow or gray in color. Relatively soft where exposed. Underlies and is interbedded with the Columbia Siver beselt flows. Contains plant feesils. Tertiary.	Practically apprisons Springs may flow along surface or through joints and flasures.	MarG, unweithered present sections to the control of the poor and softened by weathering, foundation stability may be poor to wood.	wathree
402-		and rock to planning 4	s) enginearing evaluations listed above are based judgment and incolledge of the generally defined soil stands of the generally defined soil sons of development and are in no vey intended to normal size investigations.	Symbols For Material Origin A Alluvial C Collevial Clay F Jill (mannade) d bebris G Glicial E W Bolian Special Cvalifying Symbols (t) Clacial till (te) Talus (le) Landslide Symbols For Material Clay Special Cvalifying Symbols (t) Clacial till (te) Talus (le) Landslide	rock	General I
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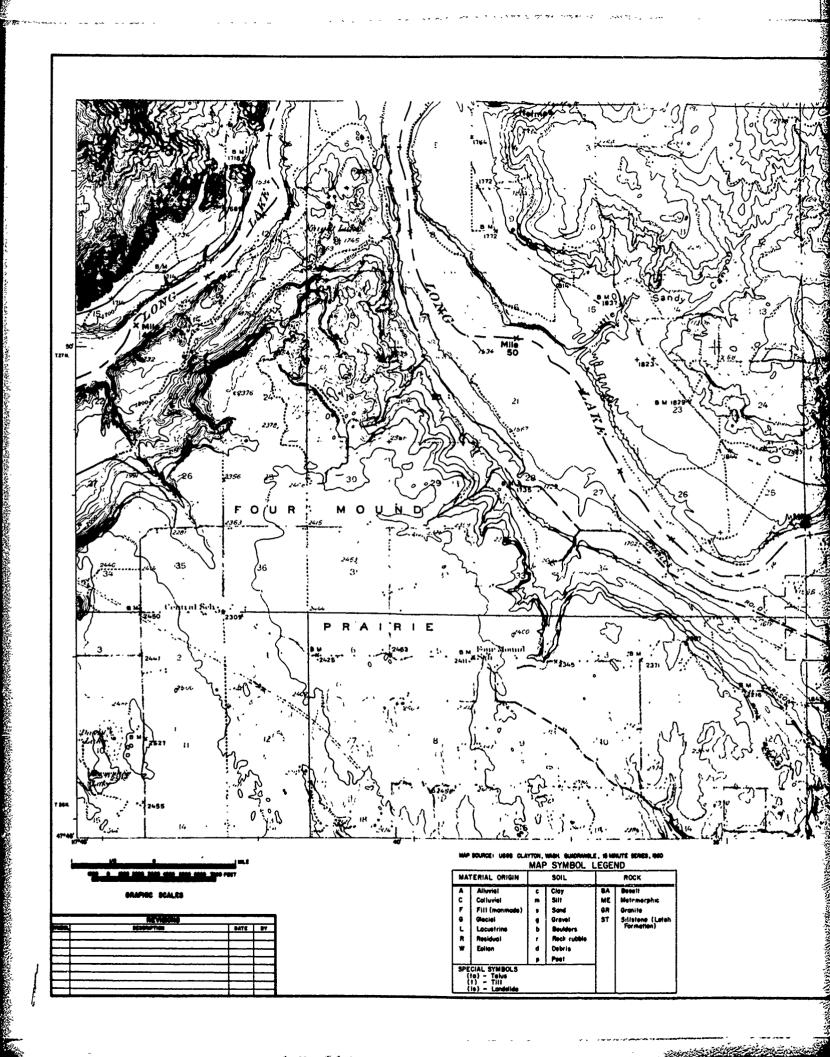
MINACE	FOUNDATION STABILITY	SLOPE STABILITY	EXCAVATION						
Permeability Lumally high in midly if peat is consolidated	Not suitable for building foundations. Embentment fill loads may sattle several inches and sattlements will continue over several years	Does not occur on natural slopes; steep slopes relatively stable when dry, but unstable when wet.	Usually requires dredging Or dragline Operation						
mate in somes with a high material susceptible to erosion on	Generally adequate for light frame structures; heavy structures usually require special foundations; semi-stable particle structures may cause large sattlements upon saturation. Frost susceptible.	Usually stable when dry Net condition may produce shallow failures and mud flows on natural slopes, particularly when subjected to freeze-thaw cycles Shallow (10 ft.) excavation slopes will usually reason stable at steep angles unless unfavorable groundwater conditions are present; high slopes should be cut back or shored.	Can generally be excavated with hand or power equipment, but mobility of equipment is poor under wet conditions						
ly impervious Generally Occurs	Not regarded as a muitable foundation soil Settlements may be excessive and irregular. Front susceptible.	Generally does not occur on slopes Excevation stability poor in wet areas. May stand on steep alopes teaporarily in dry areas.	Generally easy to excevate with hand or power equipment, but mobility of equipment is poor under wet conditions						
to erosion on slopes	Variable but generally adequate for light frame structures; heavier atructures require special foundations Prost susceptible.	Usually stable when dry Wet condition may produce shallow failures and mud flows on natural slopes, particularly when subjected to freeze-thaw.	Can generally be excavated with hand or power culpment, but mobility of equipment is poor under wet conditions						
table Lood when above water mes (As/m).	Generally poor because of high water table and local interlayering with alluvial silts (Am).	Generally does not occur on slopes Excavation stability poor, due to high water Slopes must be flattened or shored.	Can generally be excavated with hand or power equipment above vater table. Dradging, dragline, etc required below water unless site is dewestered. Presence of site (As/s) will reduce equipment mobility.						
Magree and type of stratification. Containing appreciable amounts emperies alopea. Magnee and type of stratification. Magree and type of stratification. Magnee and type of stratification. Magnee and type of stratification.	Generally of medium density and suitable for most light structures. Loose somes may dictate special foundation treatment, particularly in duce sandsties), tipicing settlements usually govern foundation design. Froat susceptibility moderate to high, being most severe in oreas containing appreciable siit.	Cenerally good, but may be marginal where underlying impervious materials and springs are present. dilfaide estes should be investigated. Excavation slopes must be fattened. Steep alopes exhibit temporary stability, but are unpredictable and should be shored.	Can generally be excavated with hand or power equipment. Can generally be excavated with hand or power equipment. Boulders (Gs/8) say require special handling.						
matent (Re/r/m, Re/m/r) and in Silt-free areas (Re/r,Re/q).	Similar to above, although foundations will often ponetrate through these soils and bear on underlying parent roct.		Solidis (1975) By Todalia Special Institution						
due to silt content (Ca/g/m).	Similar to Ls, We and Ga peries.								
mue poor due to high water within	Switable for most structures.	Does not occur on natural slopes Excavation slopes require flattening.	Can generally be excavated with power equipment. Dredging required below water. Migh permeability and provinity to rivers and atreas Often make devatering impracticable without special treatment.						
mbinage at curface low to moderate (Am).	Su _t table for most struc [,] res	Natural slopes are generally stable Encavation slopes require flattening. Steep slopes exhibit temporary stability, but are unpredictable and should be shored.	Lan unnersily be excavated with power equipment targe boulders 'Gg/b, Gg/a/b) may require special handling.						
mation of silt. May be good at the infiltrated water will flow may be amount in high on slopes.	Gunerally poor, but highly variable and indivious sites must be investigated. Generally found at the base of basalt cliffs in areas most difficult to investigate and develop.	Good to poor; individual sites must be investigated Excavations will generally encounter local seepage that may require special treatment. Steep slopes may remain stable depending upon local conditions.	Excavation difficult because of presence of large Doulders and haystacks which usually must be broken up by blasting Heavy power equipment generally required.						
paction of silt, sand and gravel.	Generally spond.	Generally good.	Con generally oe excavated with power equipment. Large boulders may require special handling						
Identified only at one remote location within the urbanizing area and sufficient date is not available to evaluate engineering characteristics.									
characteristics. Because of the variable nature and generally uncontrolled placement, development of areas underlain by manuade fills must be preceded 1, through investigation and study.									
Somes in basalt may carry water. Surfaces. Permesbility varies my to the openness of fractured the interbed or interformation	Excellent. Meavily loaded areas on baselt abould be explored to determine the presence of highly vesicular or fractured somes immediately below the footing.	Steep stopes will generally remain stable. Safety area should be	community requires blasting. Reavy power aquipment and rippers sometimes have ascess removing highly fractured basalt, but this condition cannot siways be determined beforehand.						
Feefry Or store water in Mithered or altered zones. Megree of fracturing, open,ess of alteration.	Generally excellent although highly weathered or altered sites may require special investigation.	Steep slopes will generally remain stable. Seftly area should be provided at the tee to ratch spalls and block that may loosen with time, particularly in basalt. Highly fractured and weathered rock may require slope flattening.							
Bey flow along surface or	Herd, unweathered phases excellent. Where exposed near surface and softened by weathering, foundation stability may be poor to good.	Conerally good in deep unweathered phases Poor in softer weathered romes if spring water is present.	Unweathered Latah inquires power equipment; softer phases can be escavated by machine or by hand with picts.						
Origin Symbols For Material Class b Boulders c Clay dis) d Debris g Gravel a filt f Rock-Trubble DA Basalt befrock	1Elcation	CONSULTING ENGINEES COMPS OF ENGINEES STATELE COMPS OF ENGINEES STATELE, WASHINGTON WATER RESOURCES STUDY METROPOLITAN SPOKANE REGION							
e Rock rubble 9 Sand DA Basalt befrock 22 Metamorphic bedroc Co- GR Granite befrock 111 ST Siltatone (Latah F		ENGINEERING GEOLOGY LEGEND							

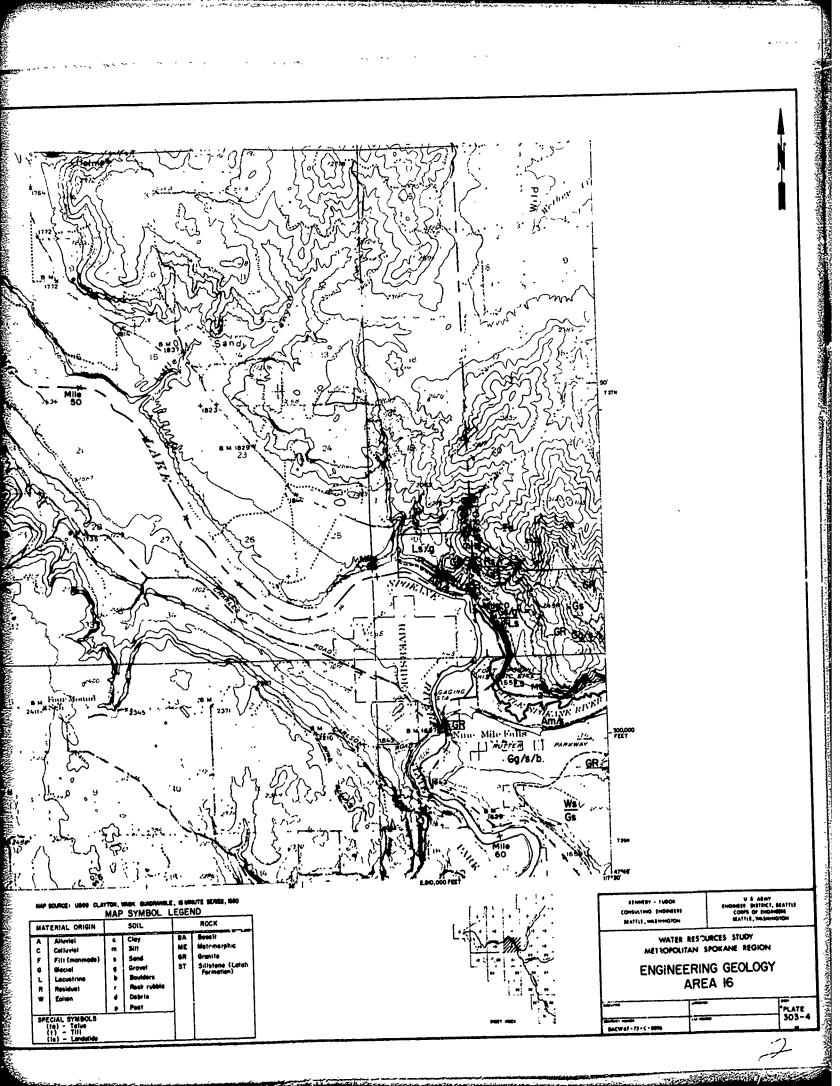
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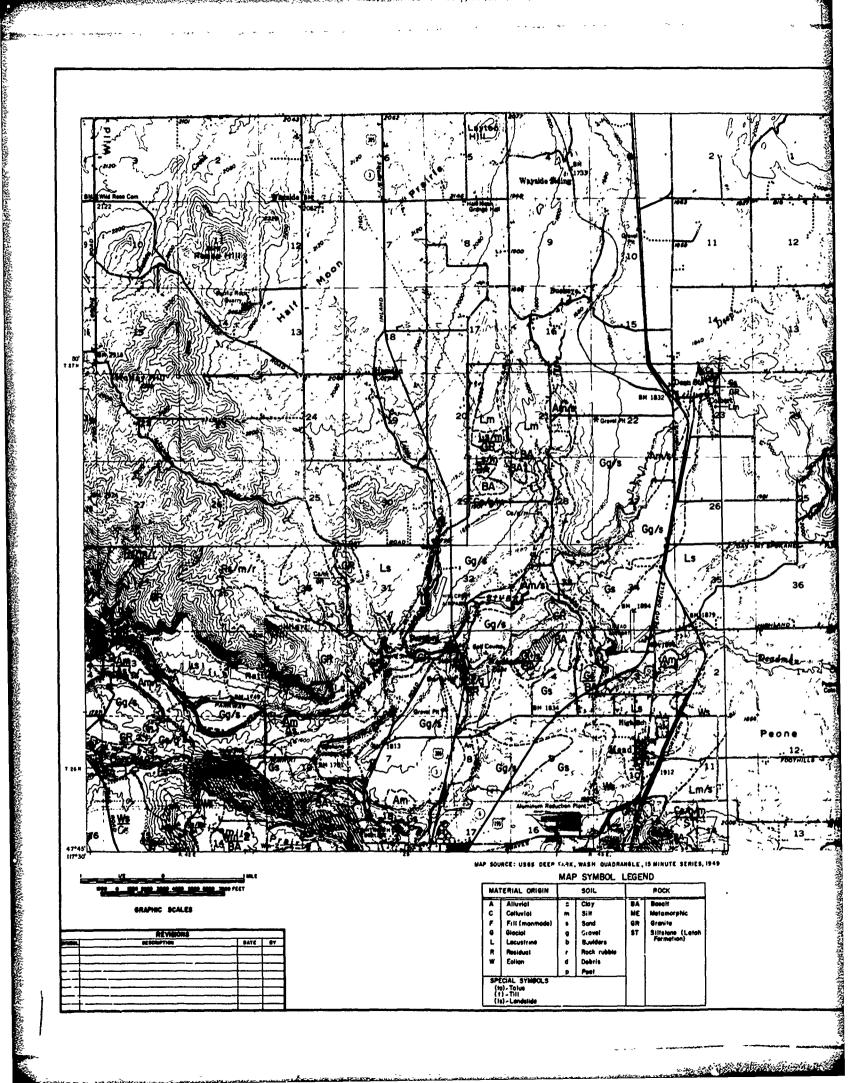
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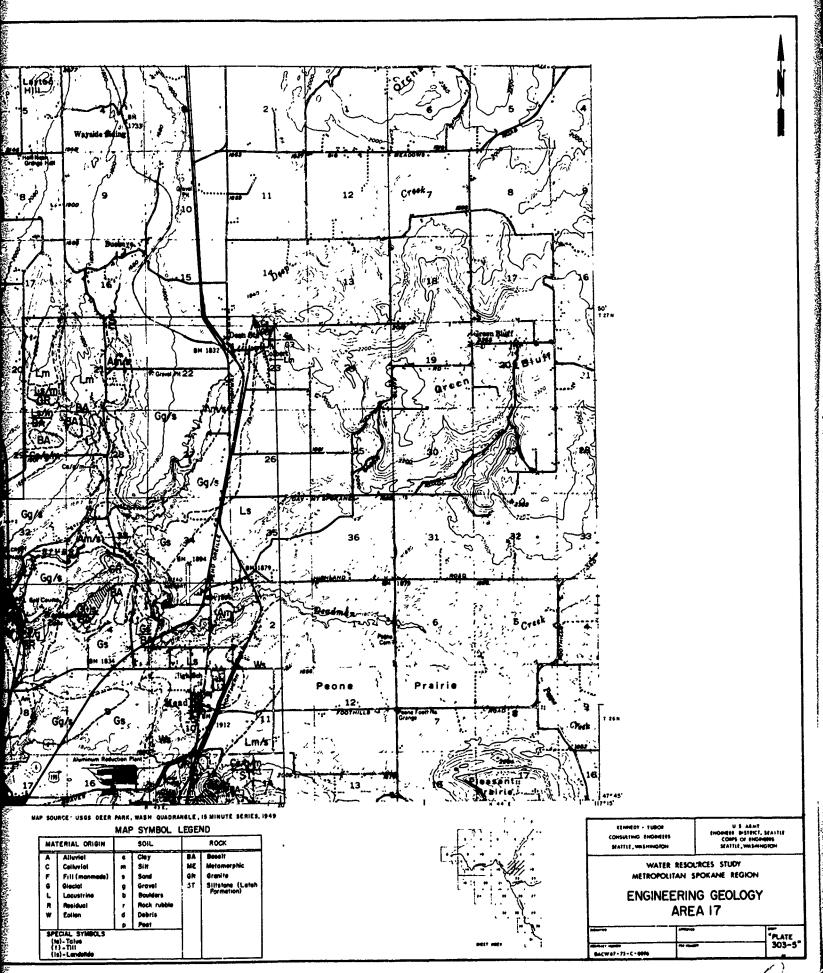
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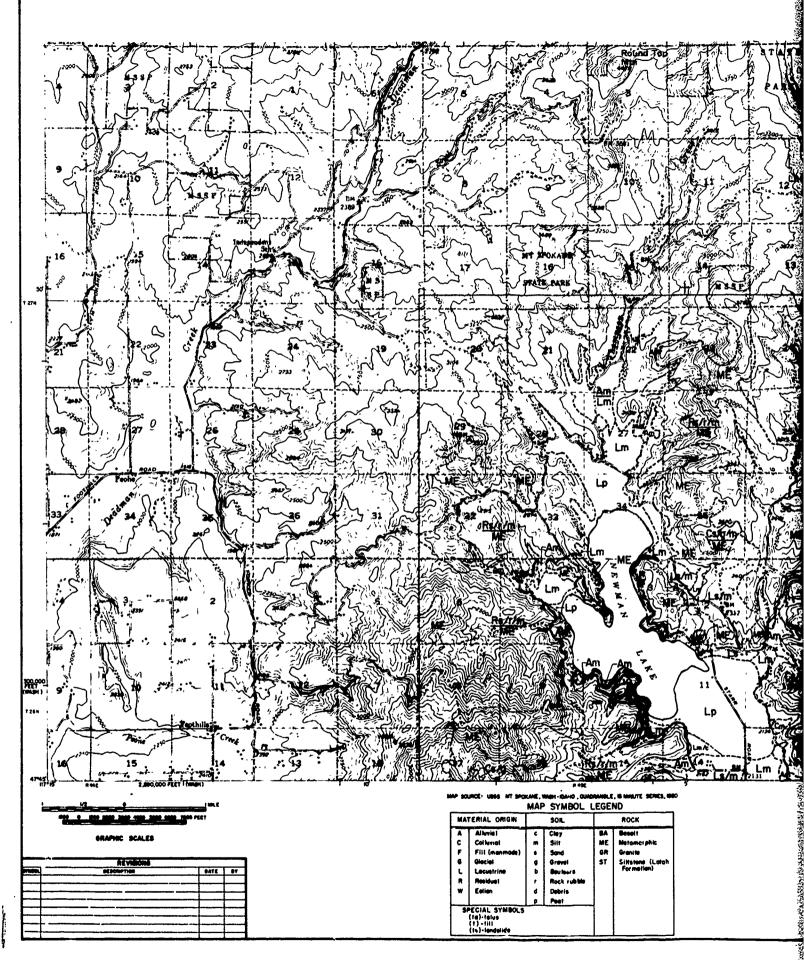


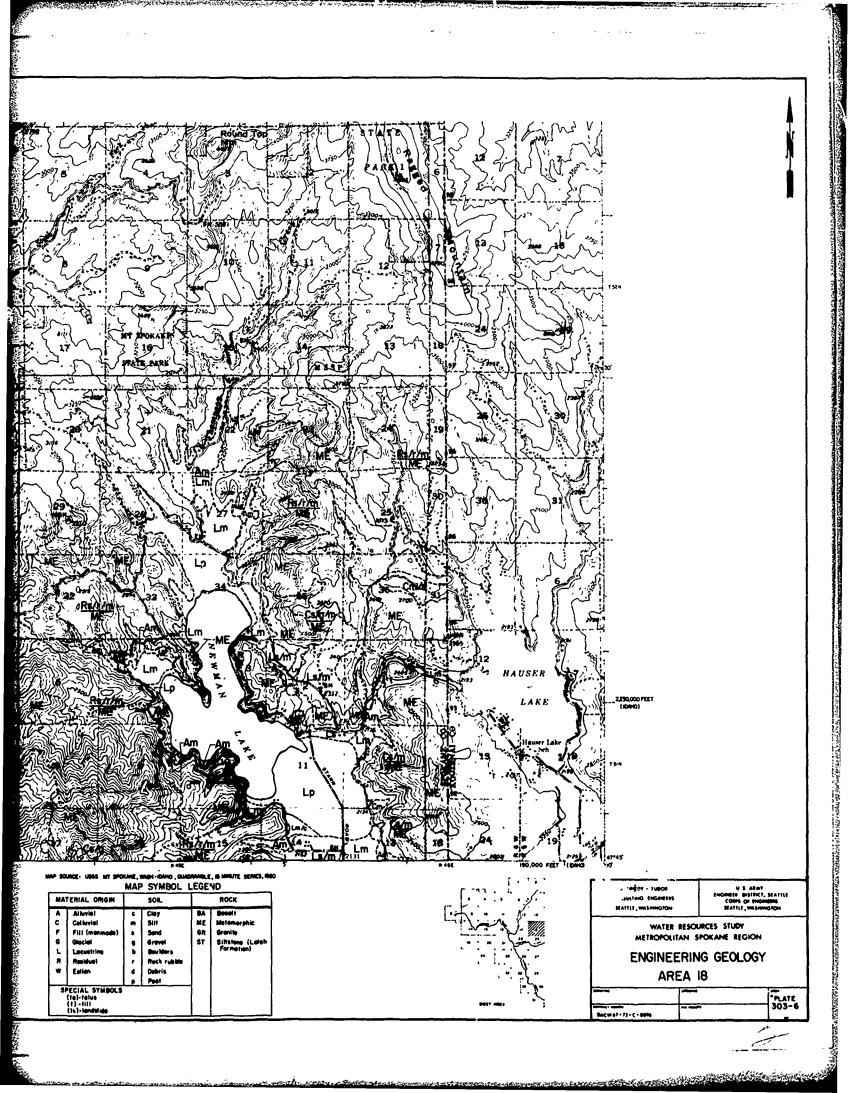


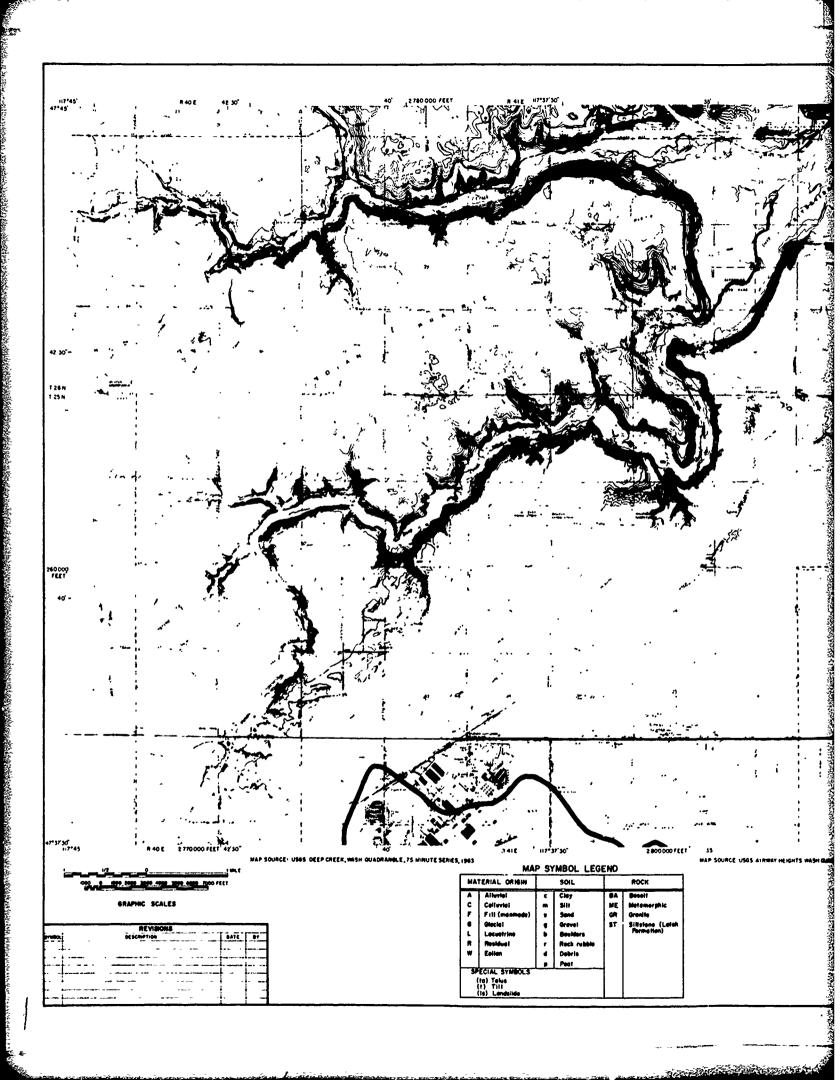


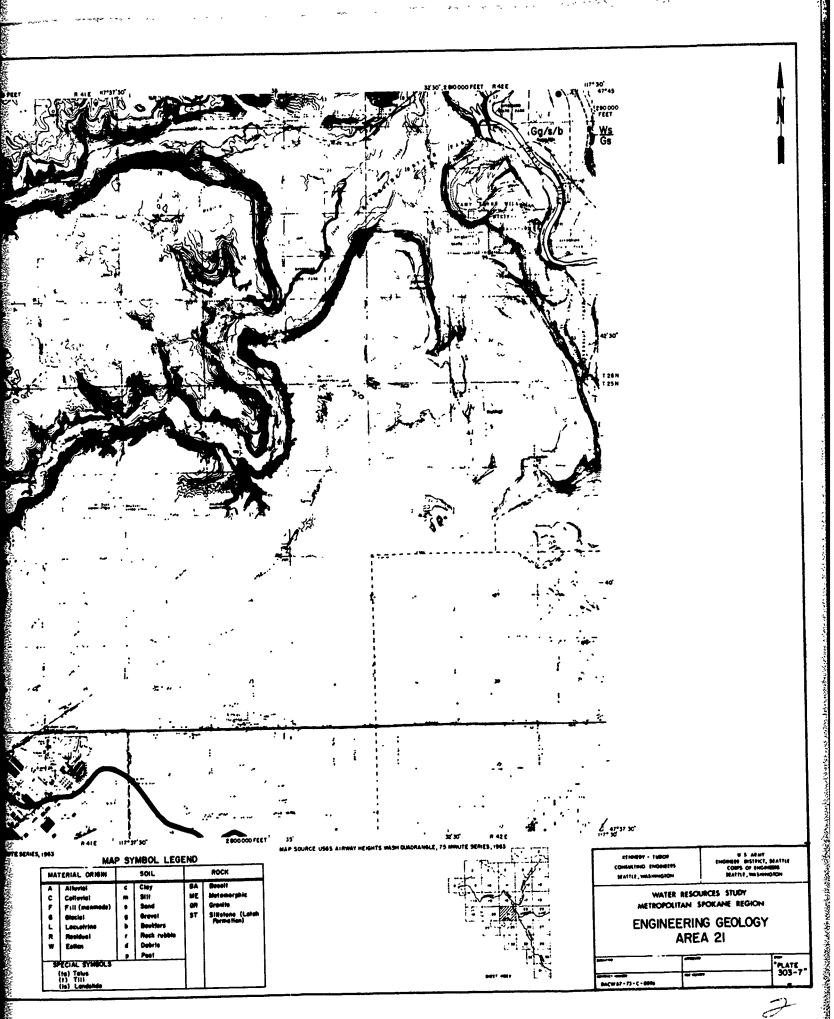


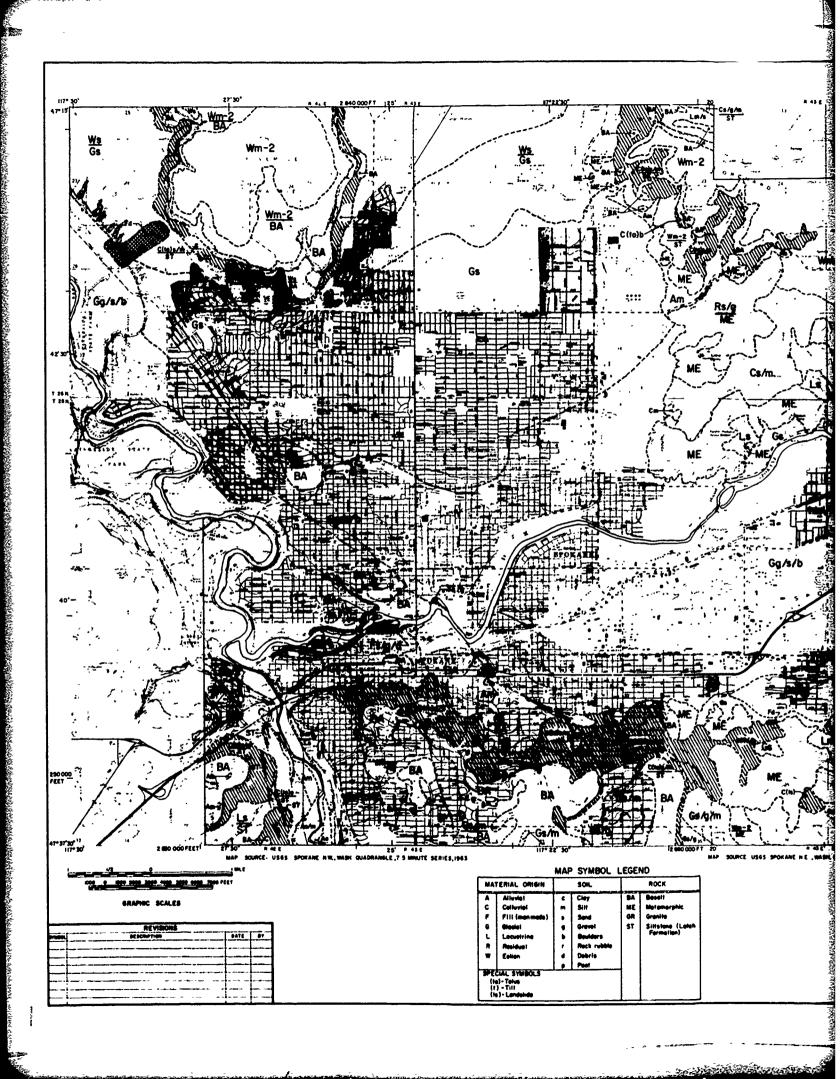
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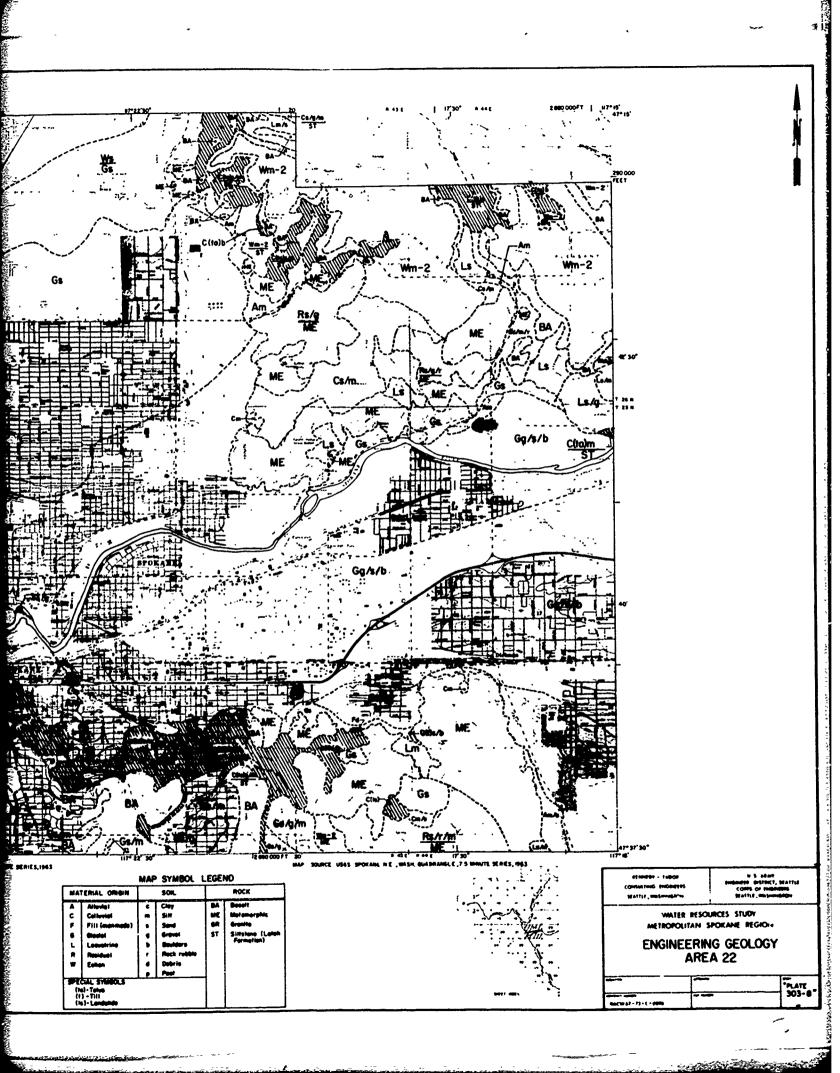


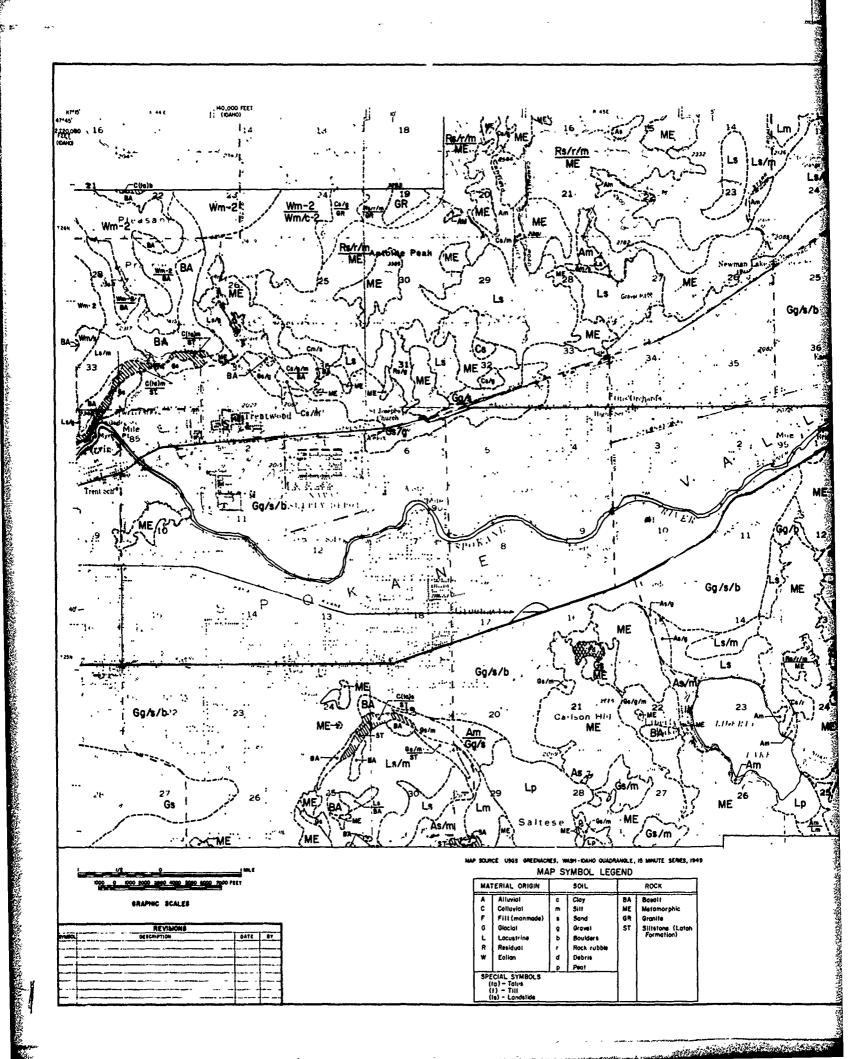


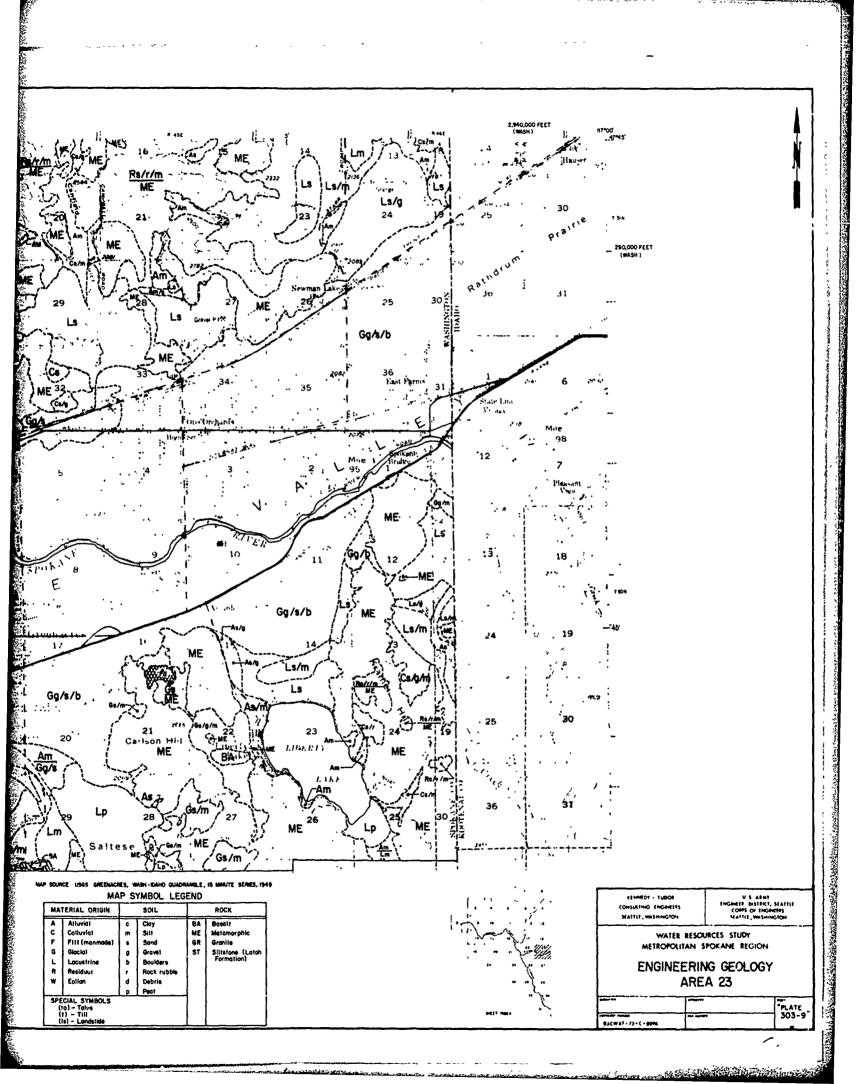


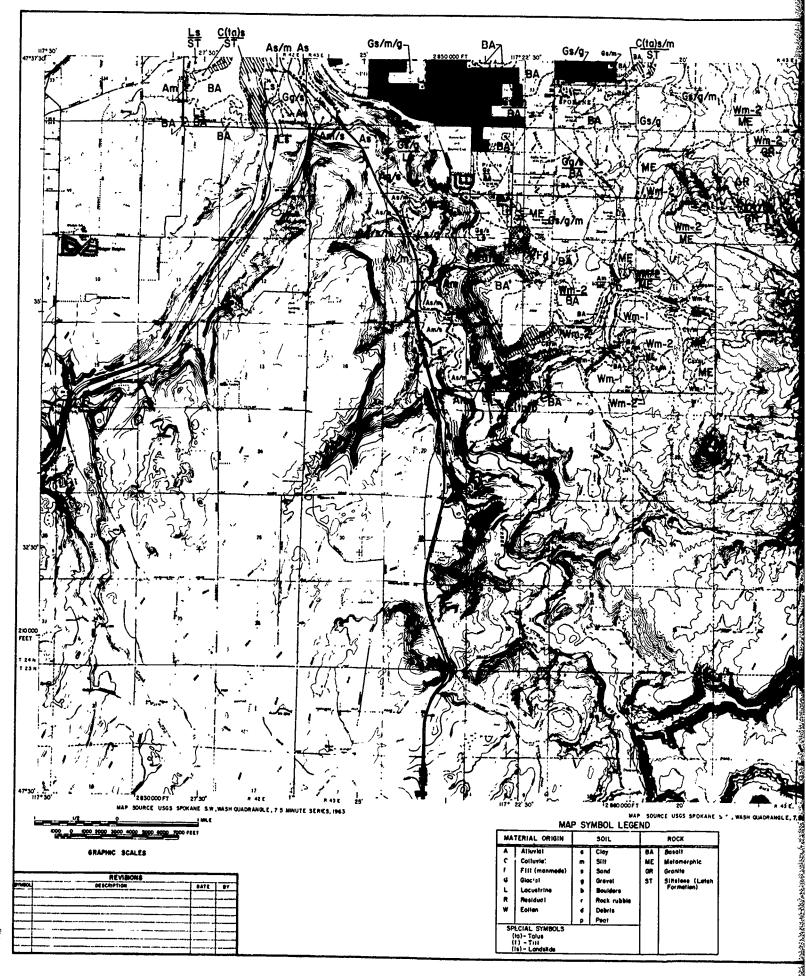


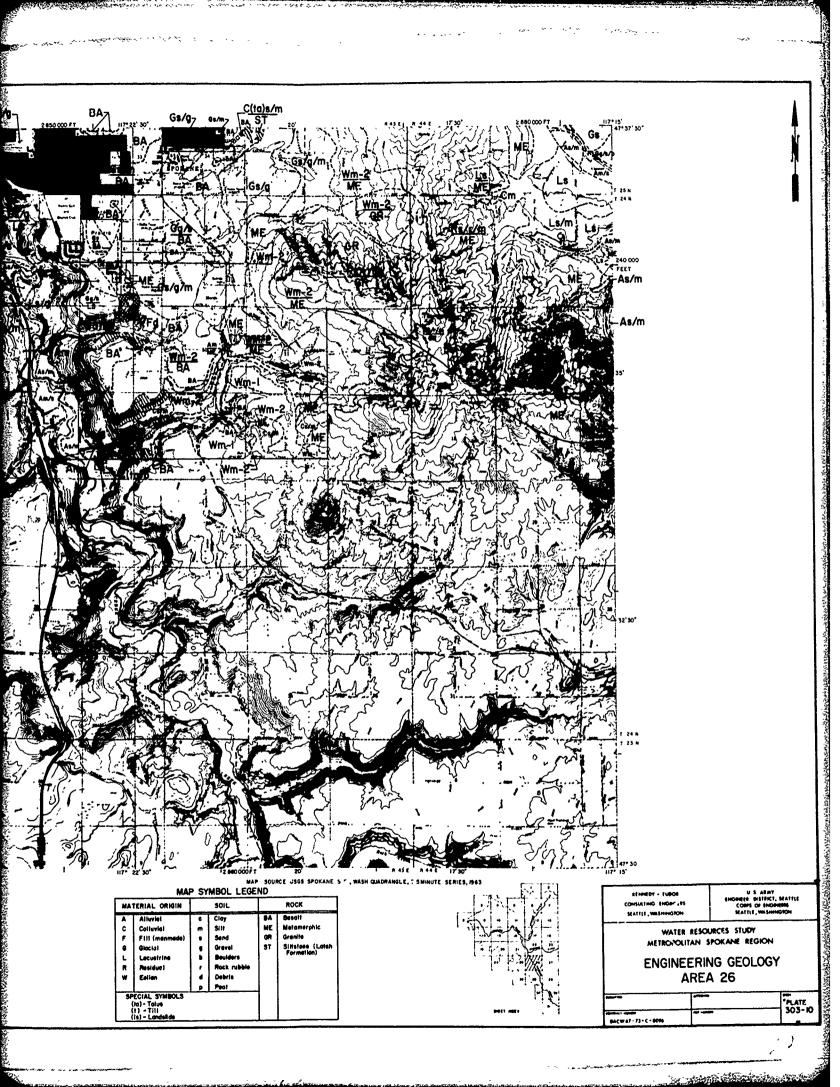






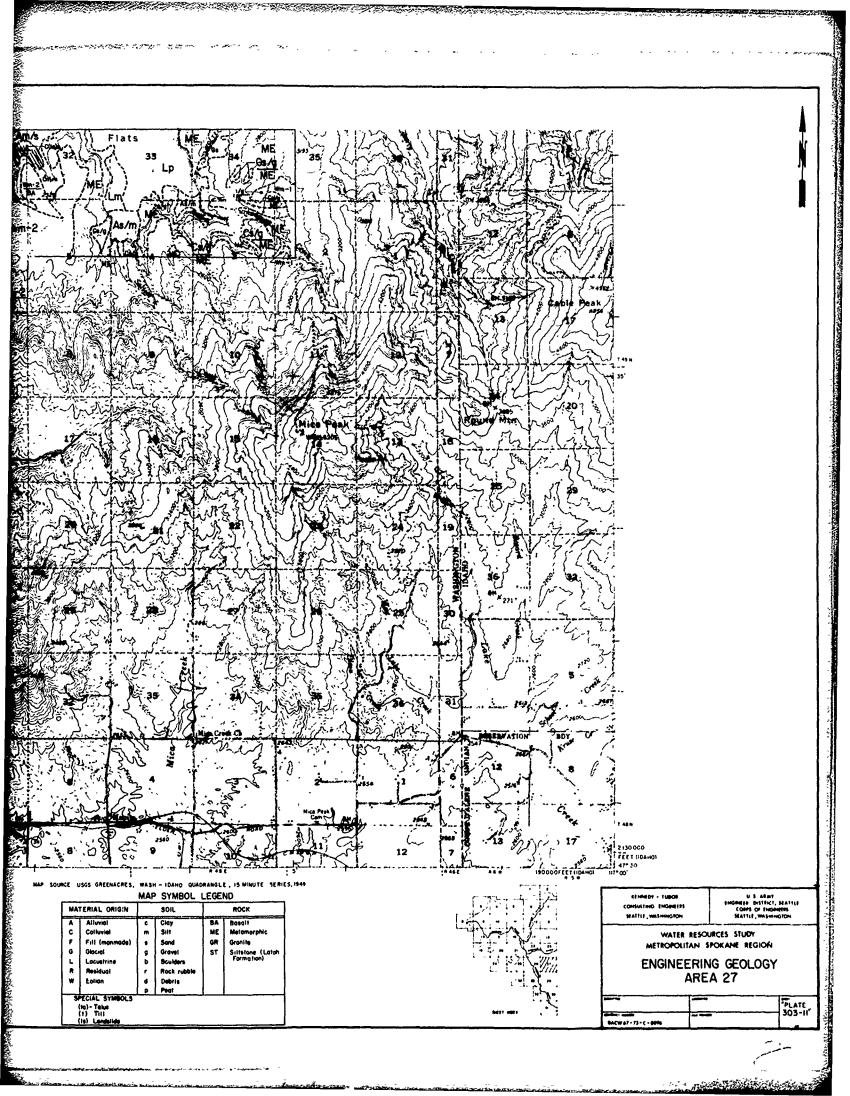


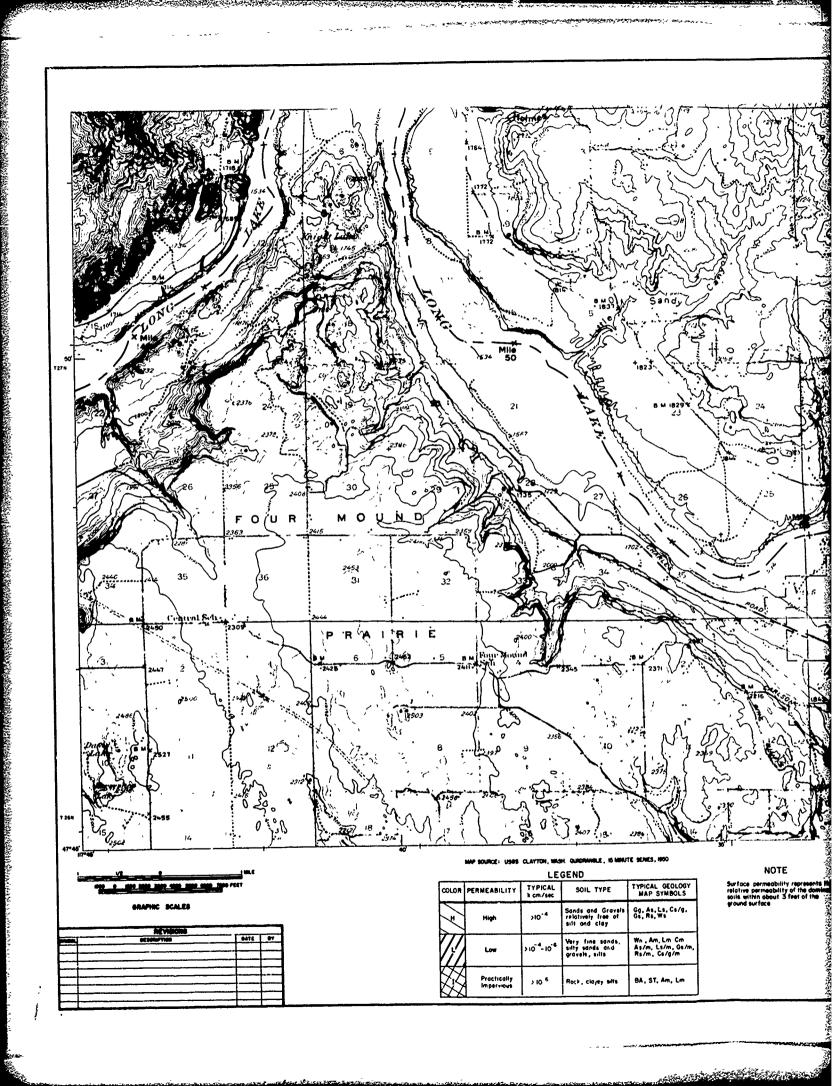


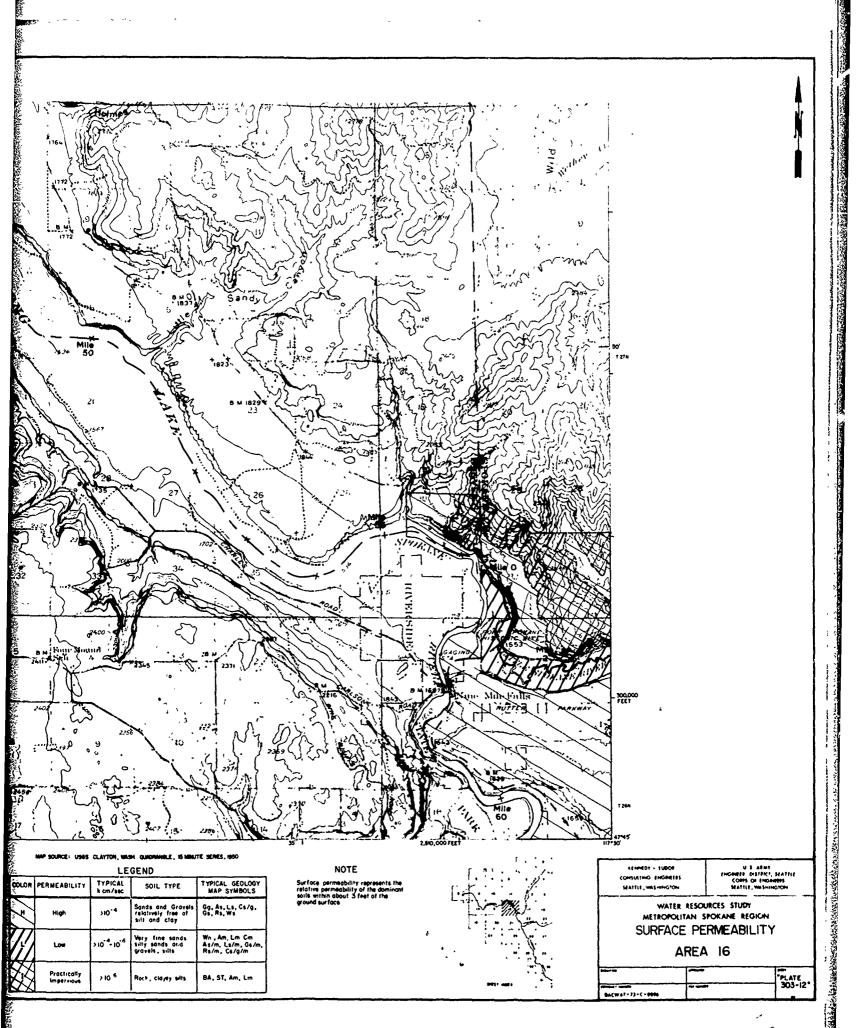


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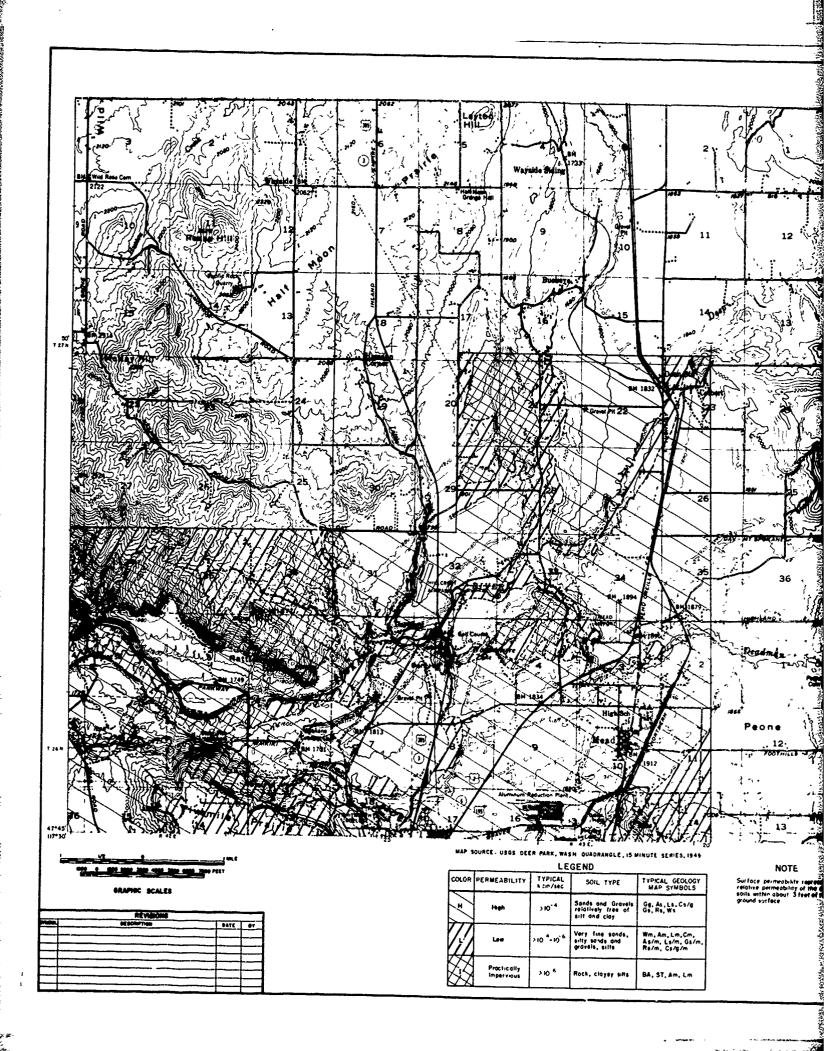
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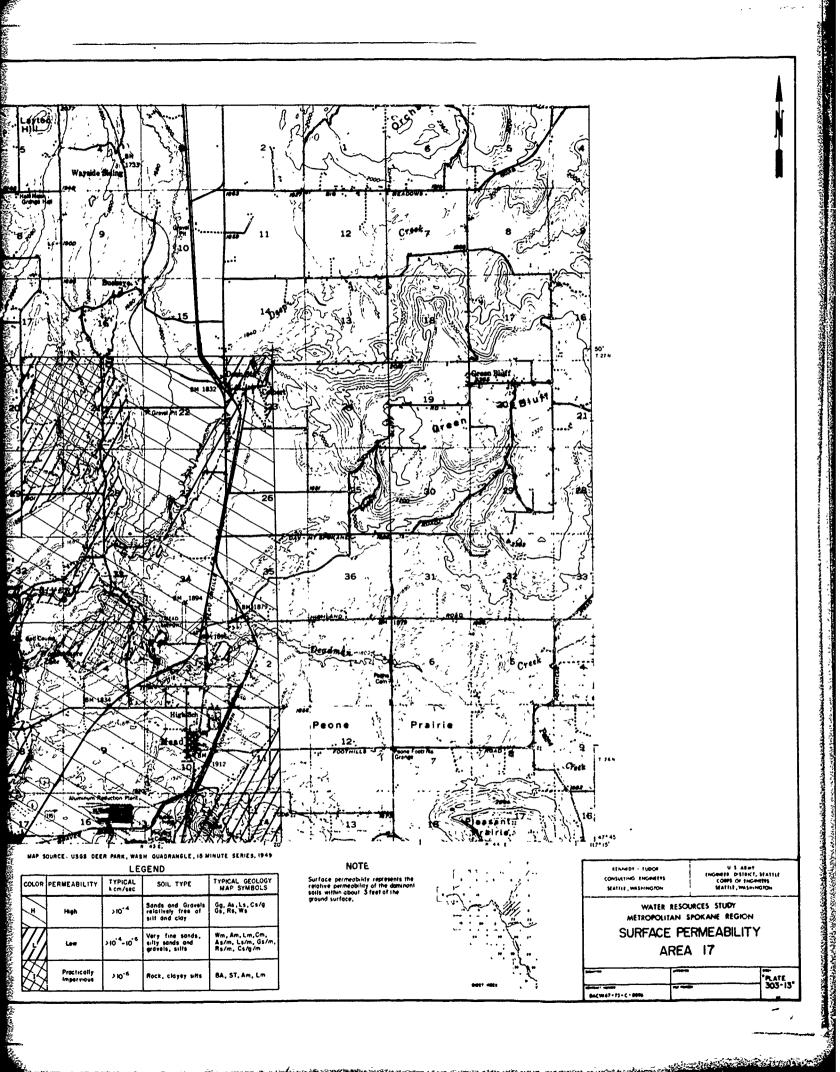


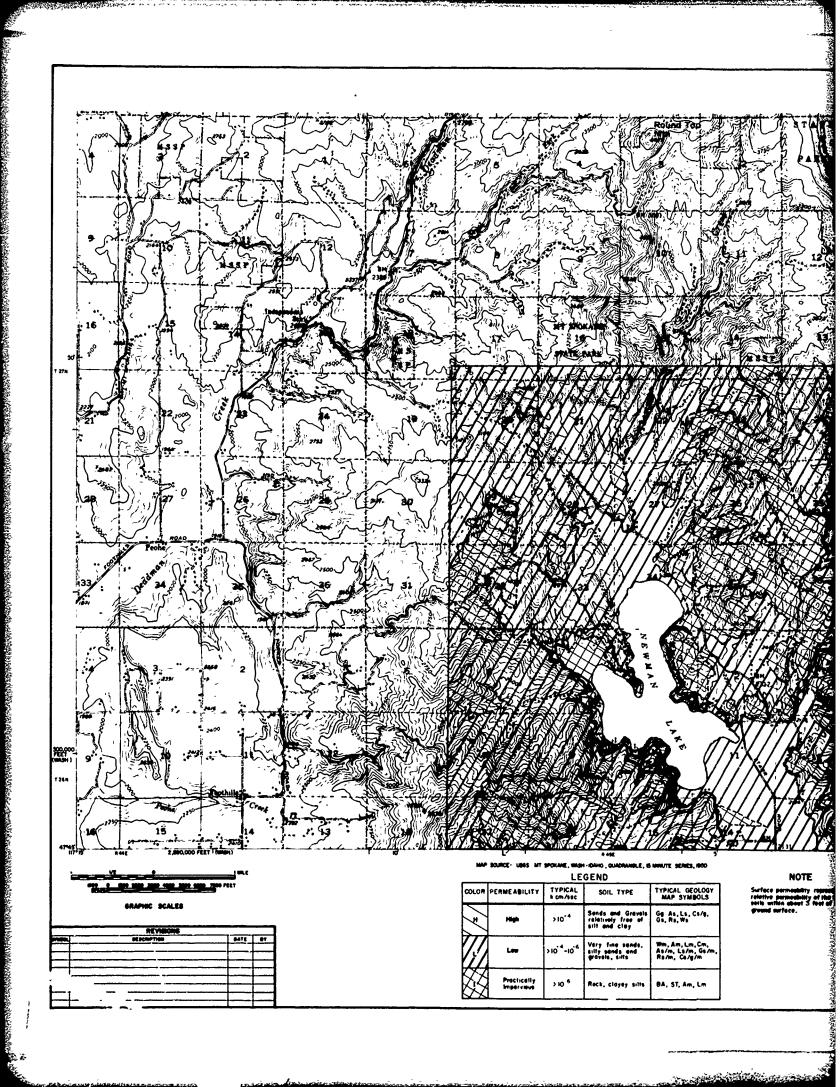


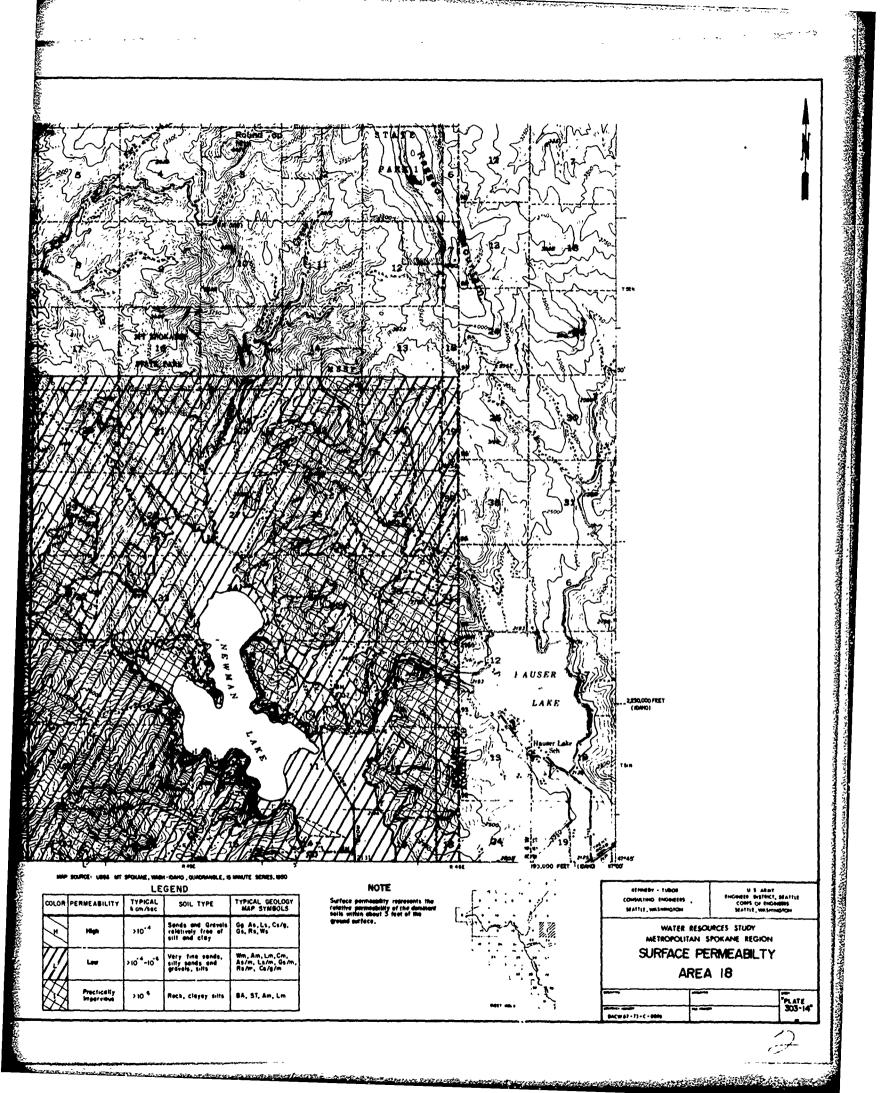


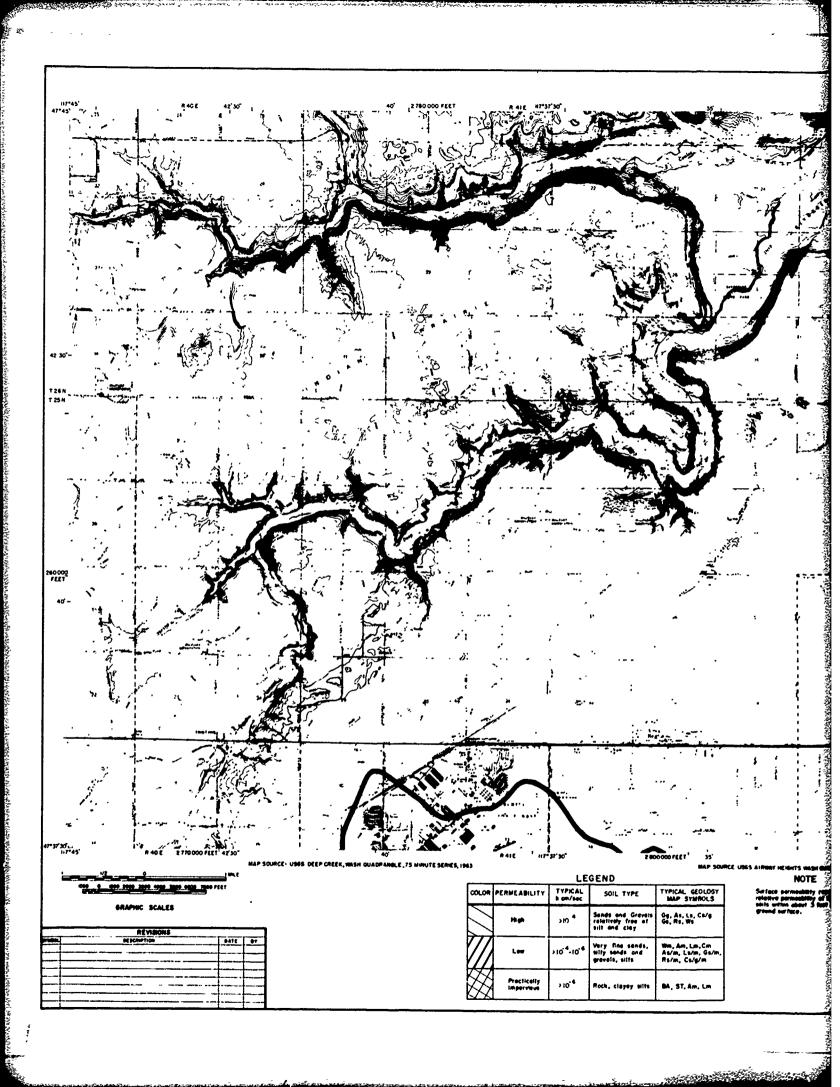
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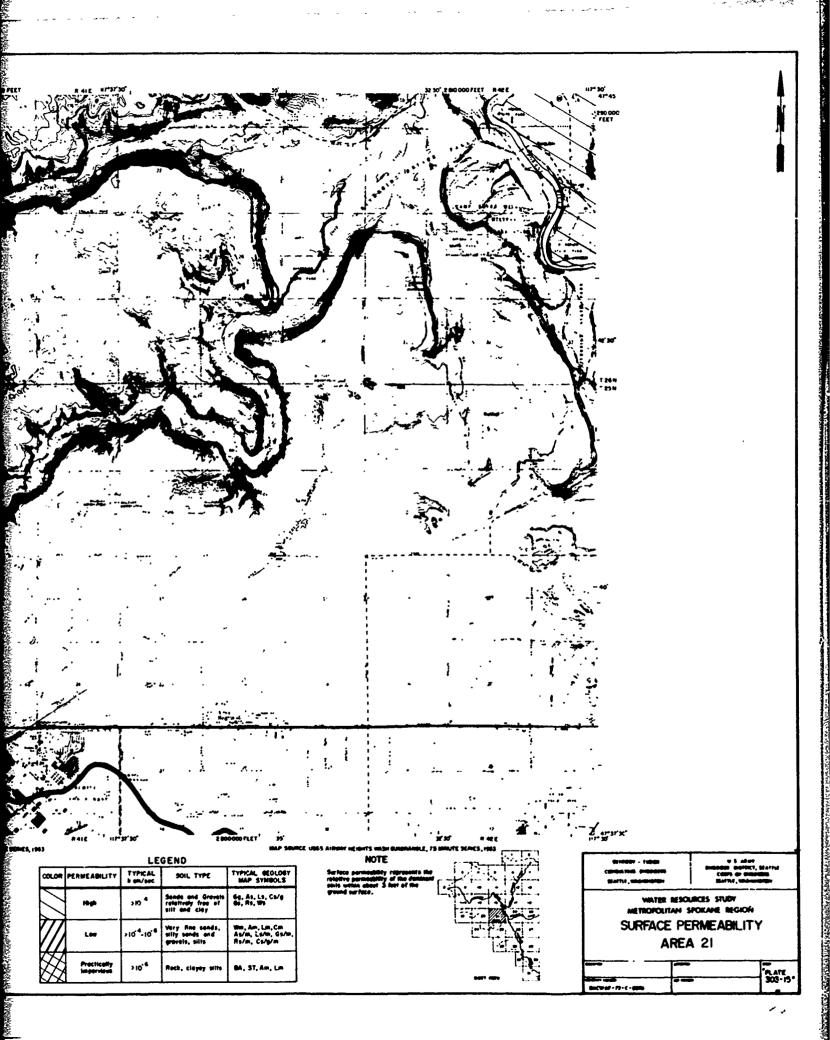


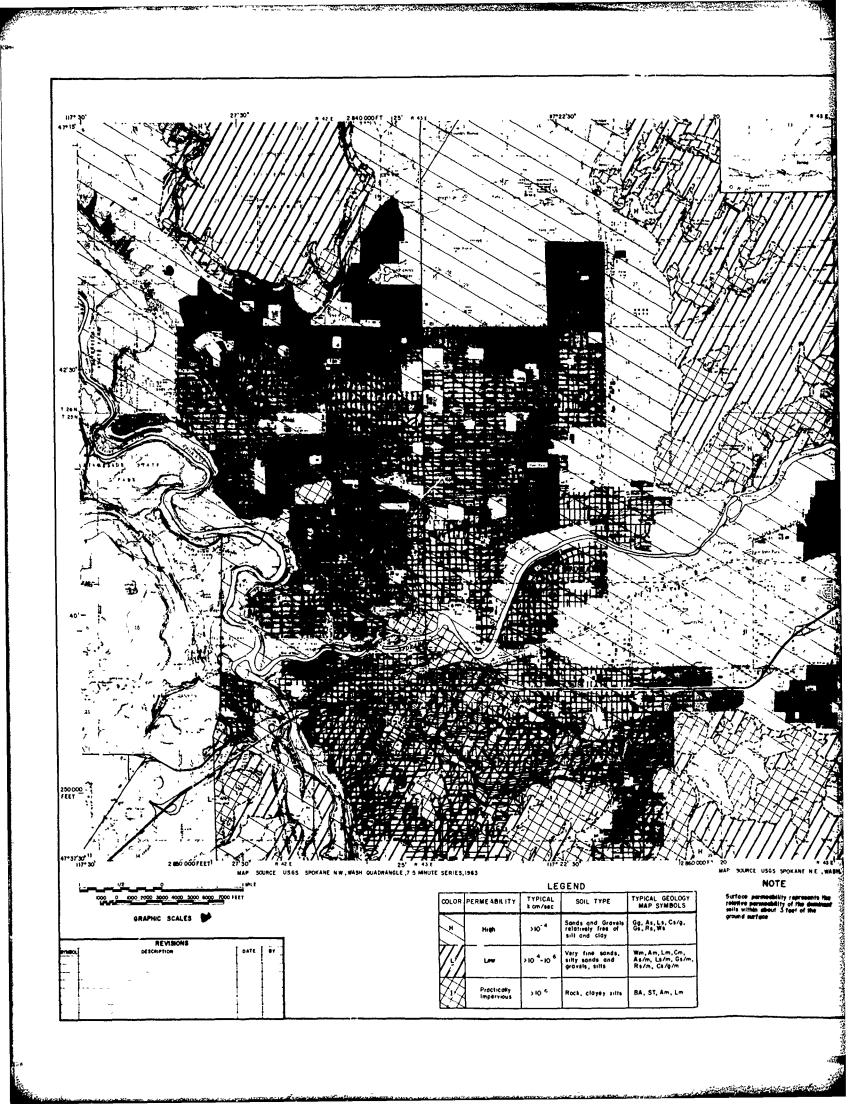


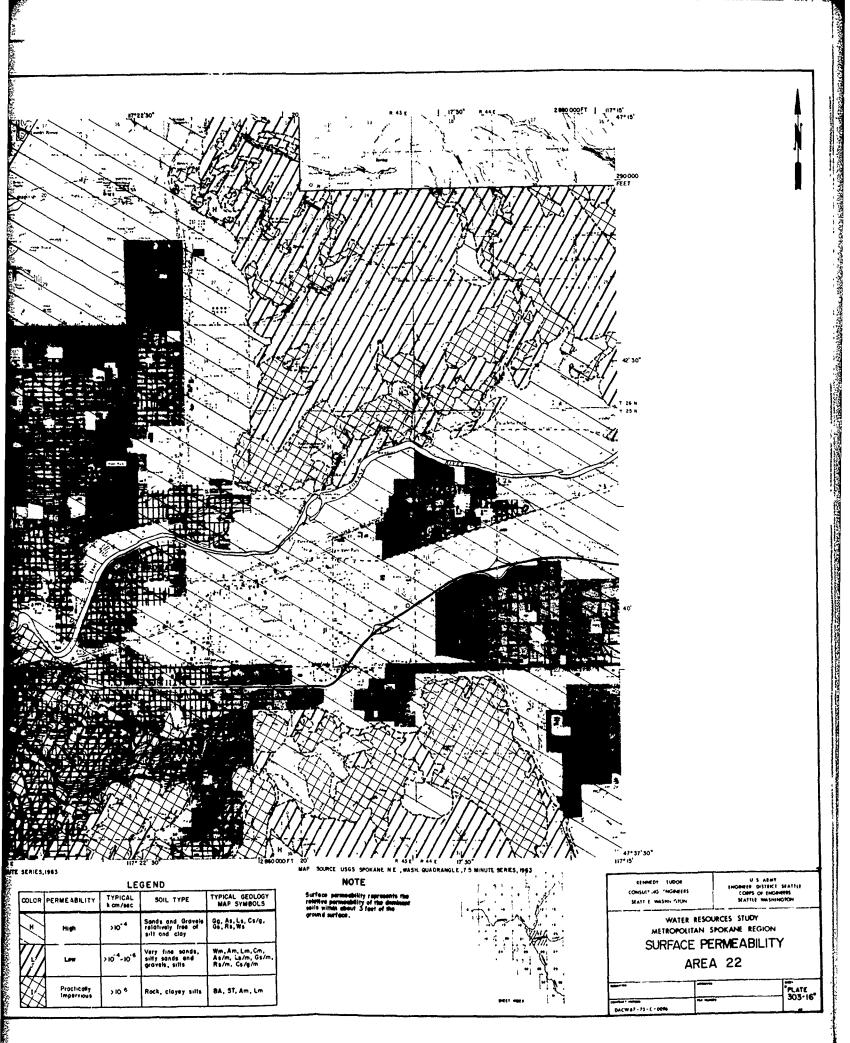




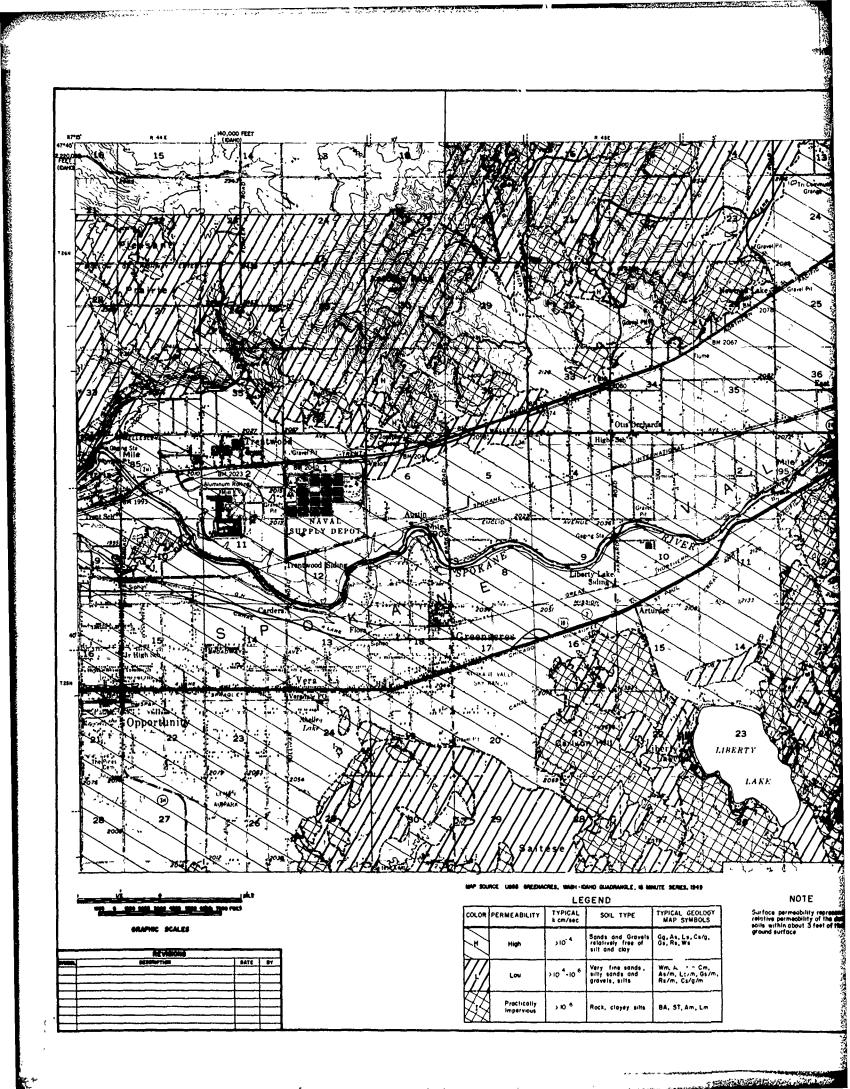


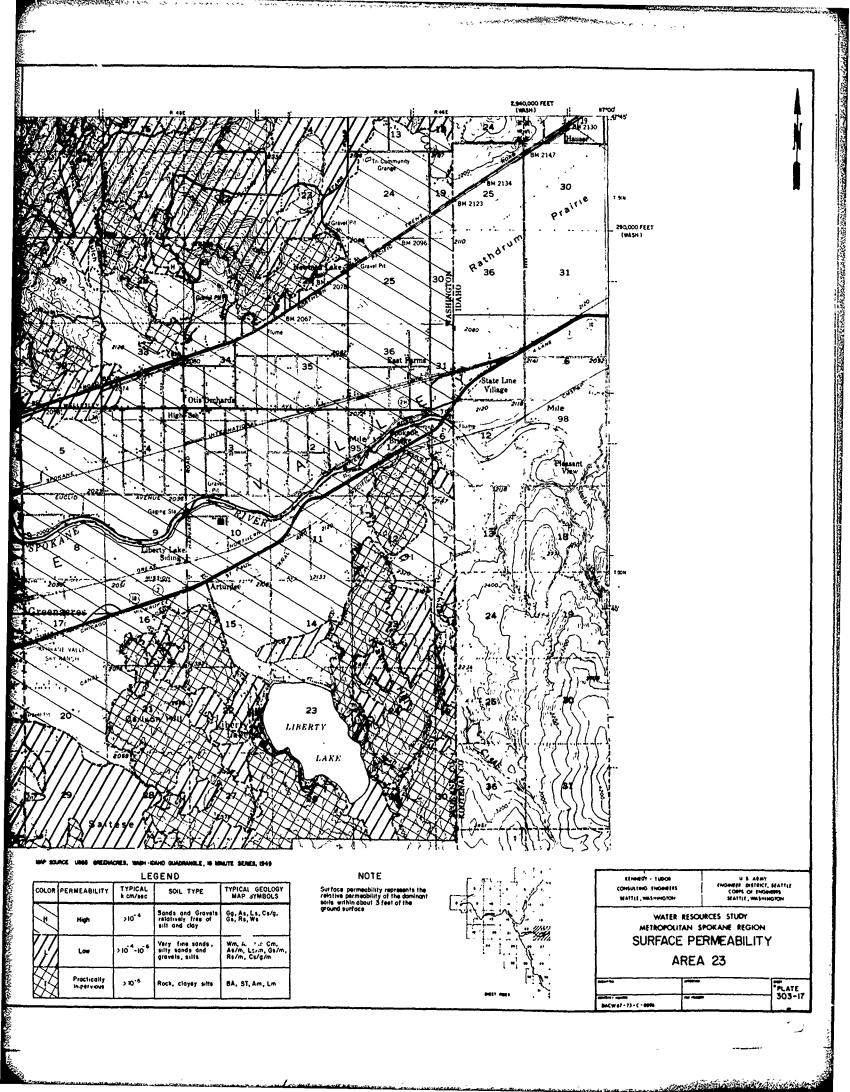


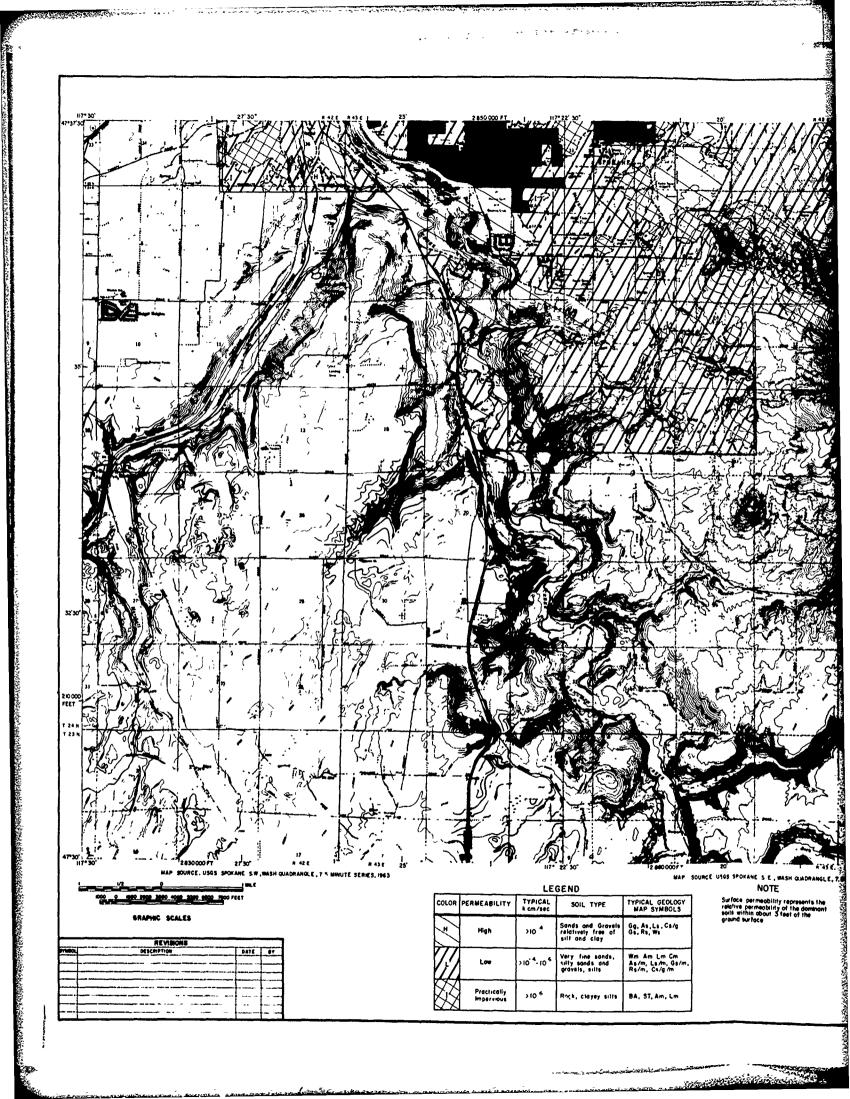


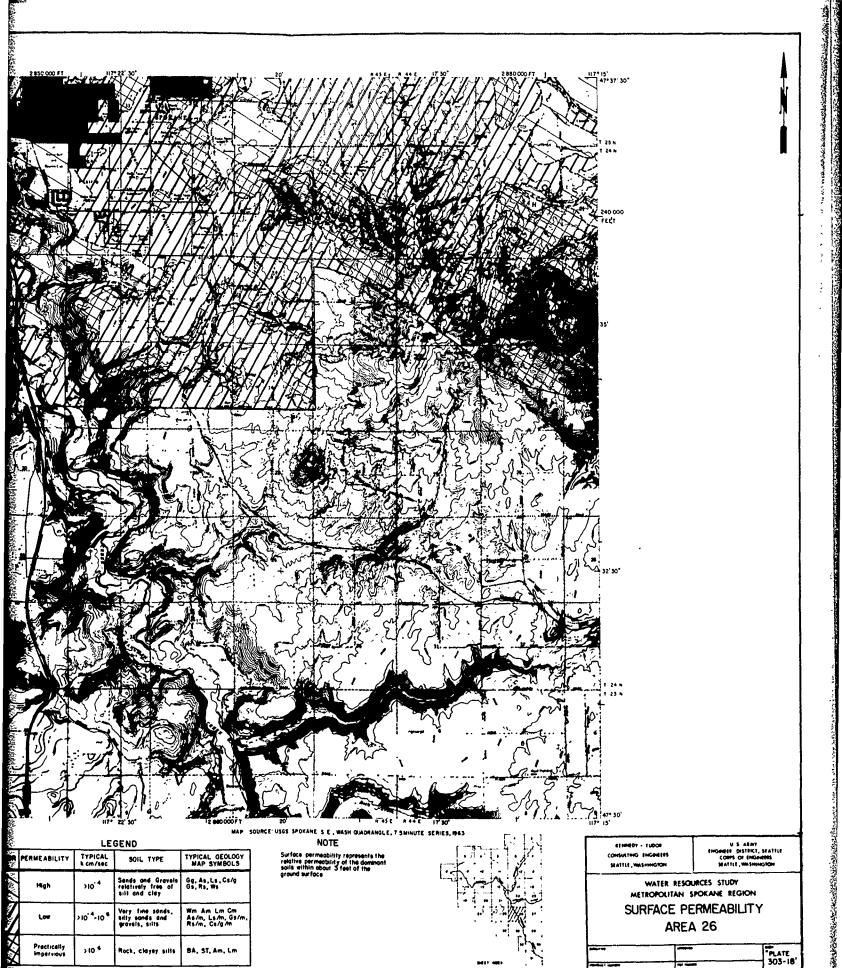


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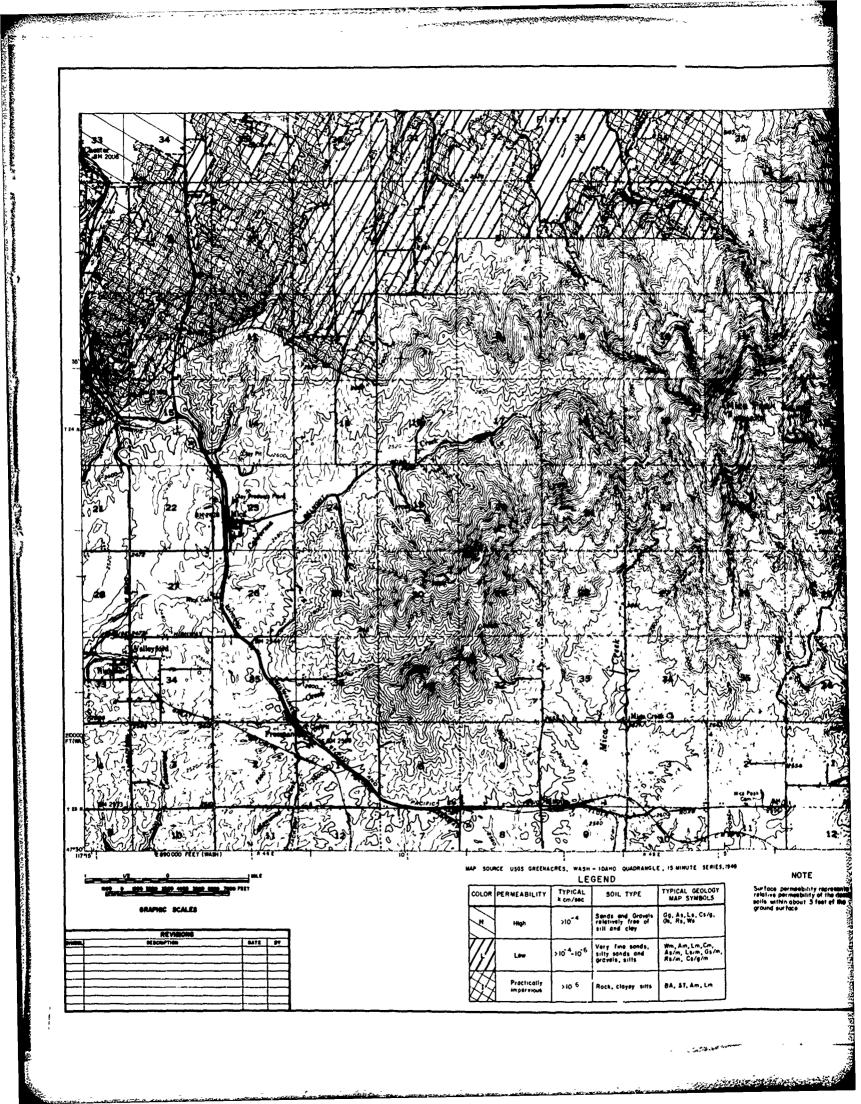


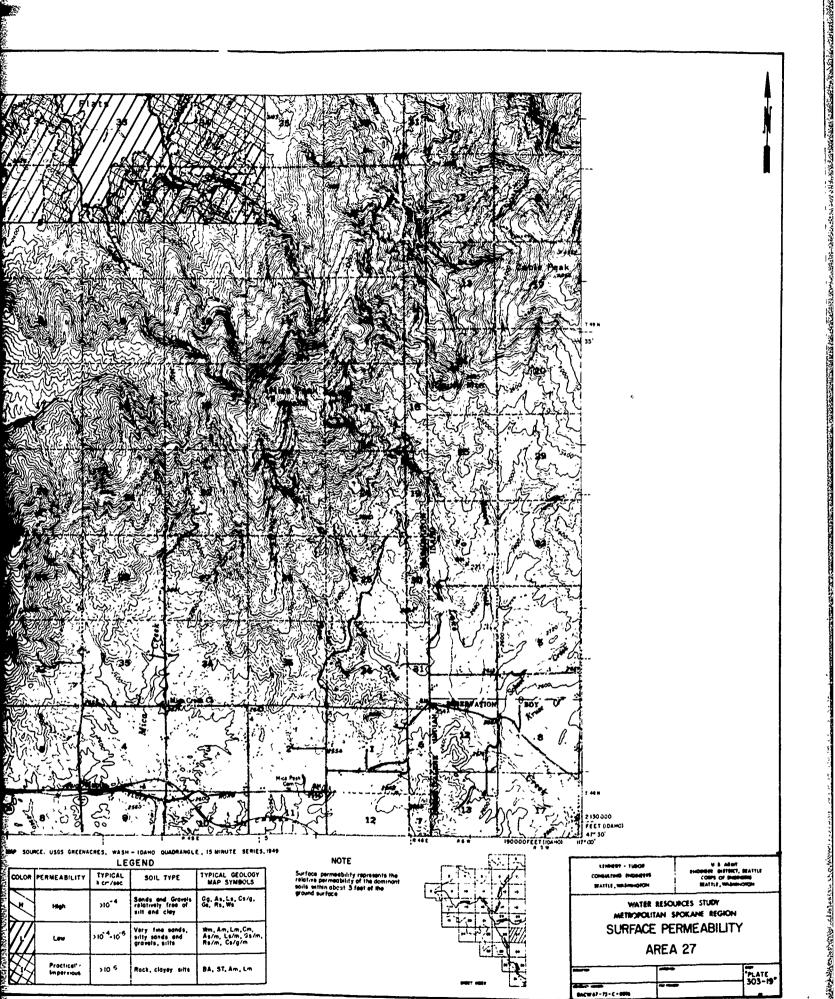


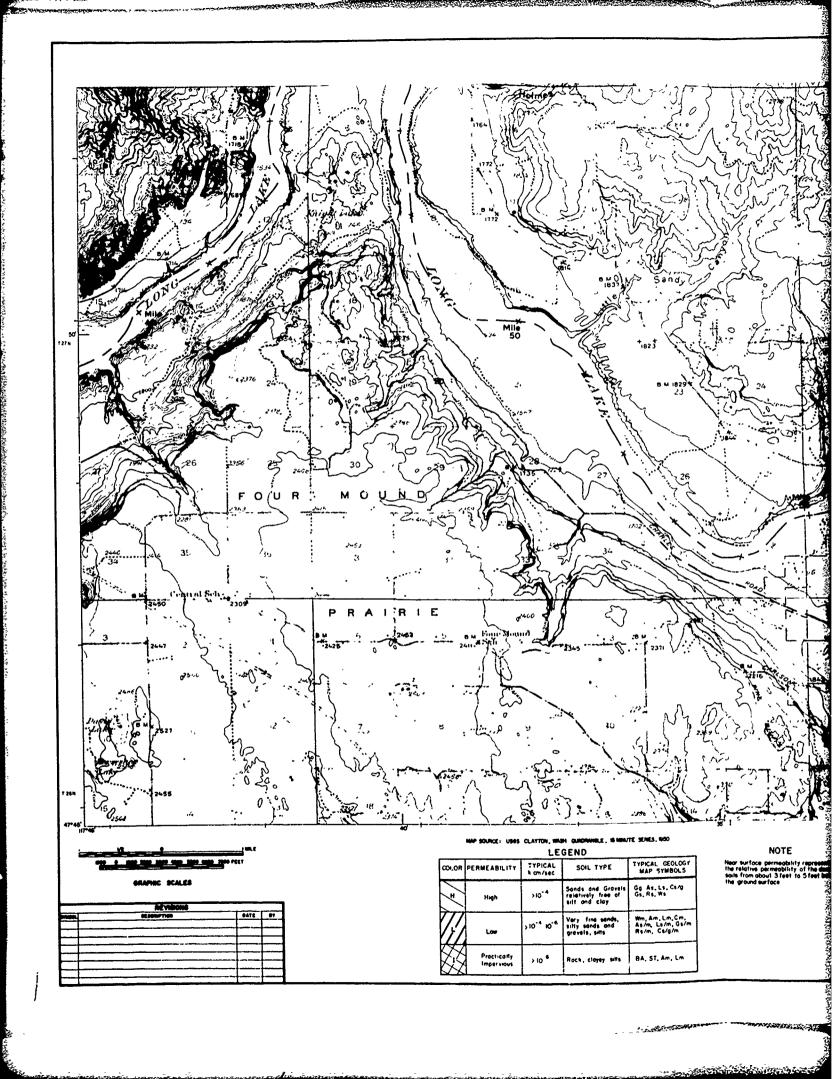
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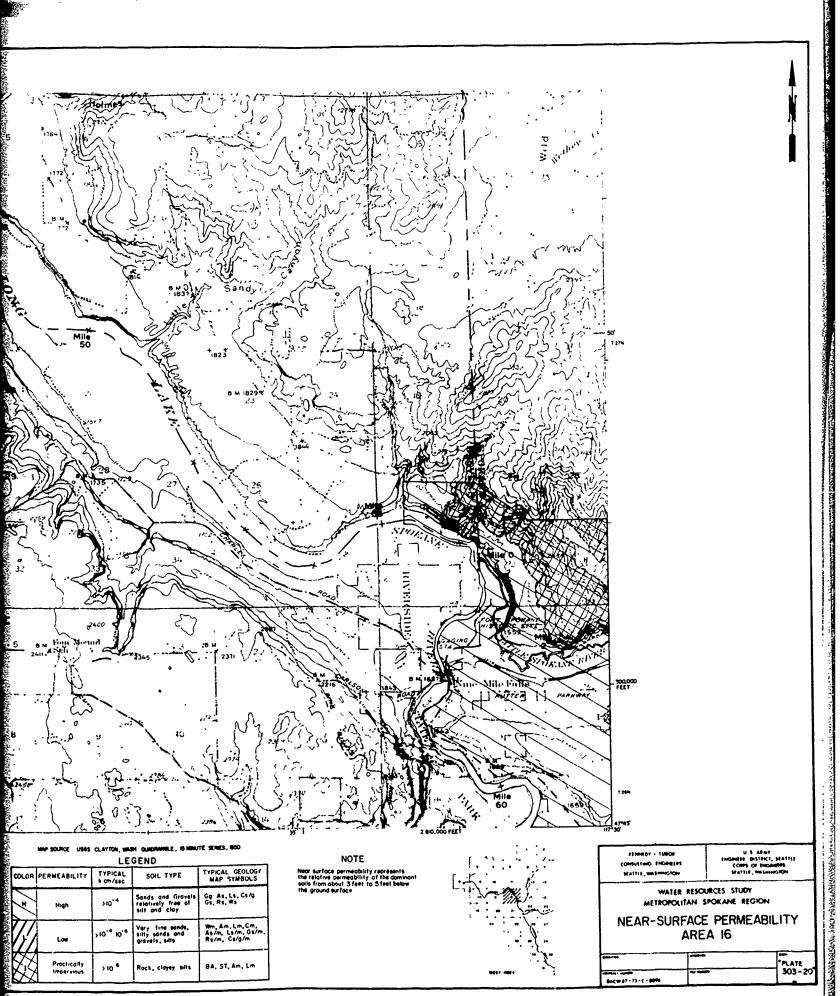
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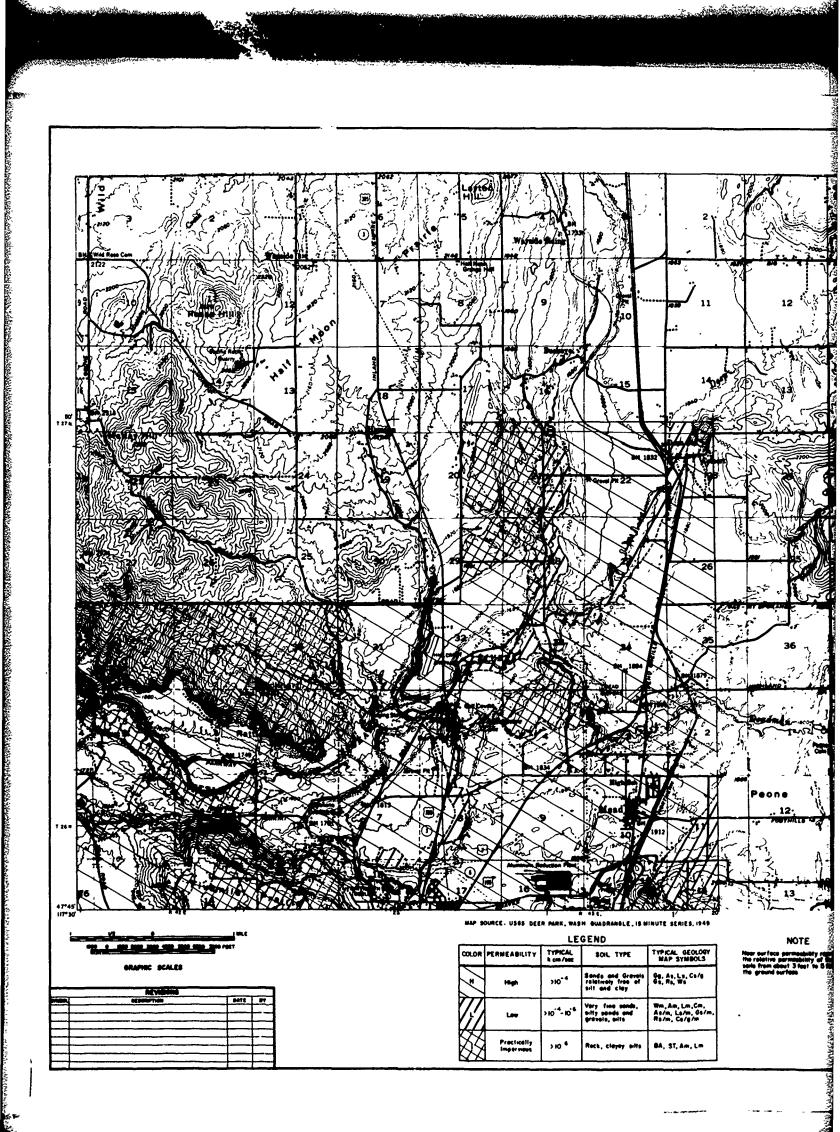
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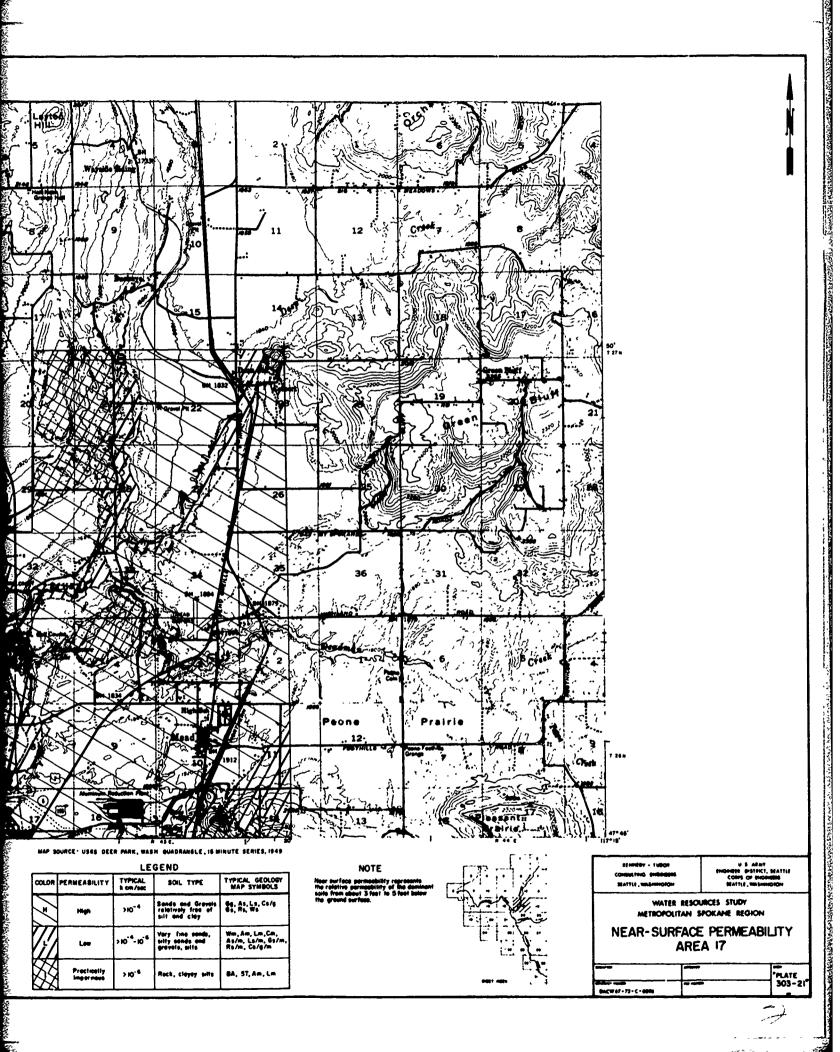


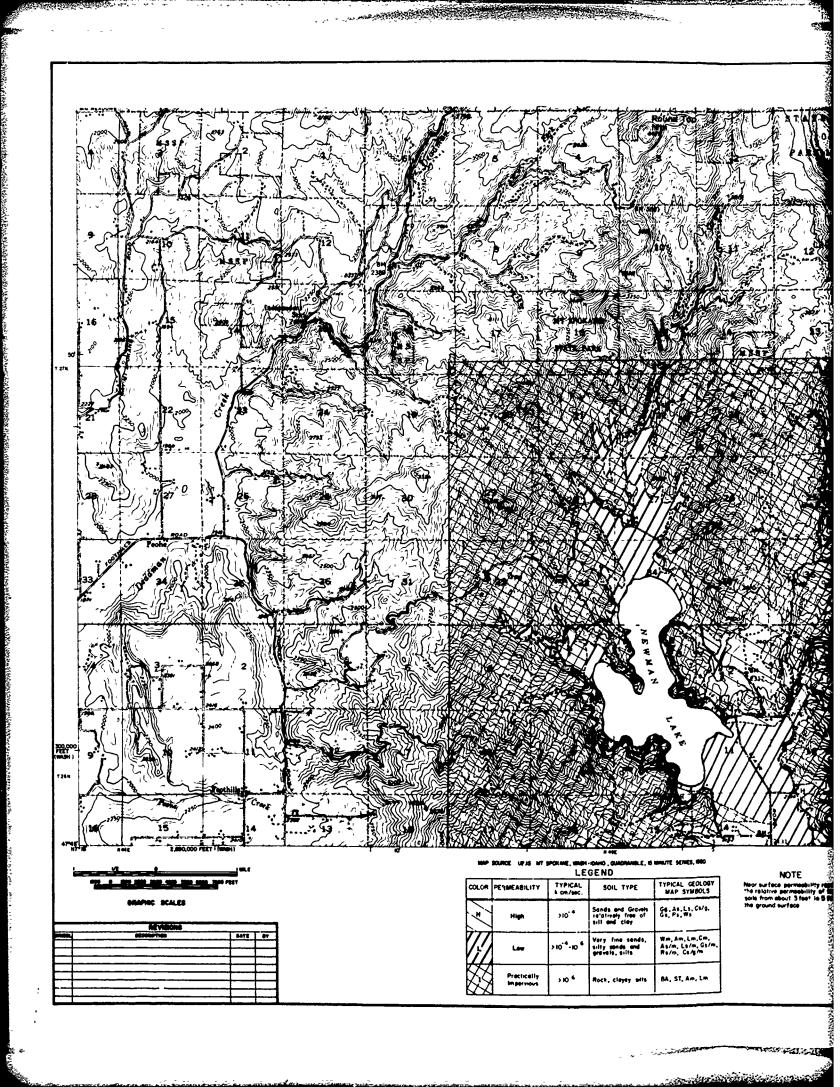


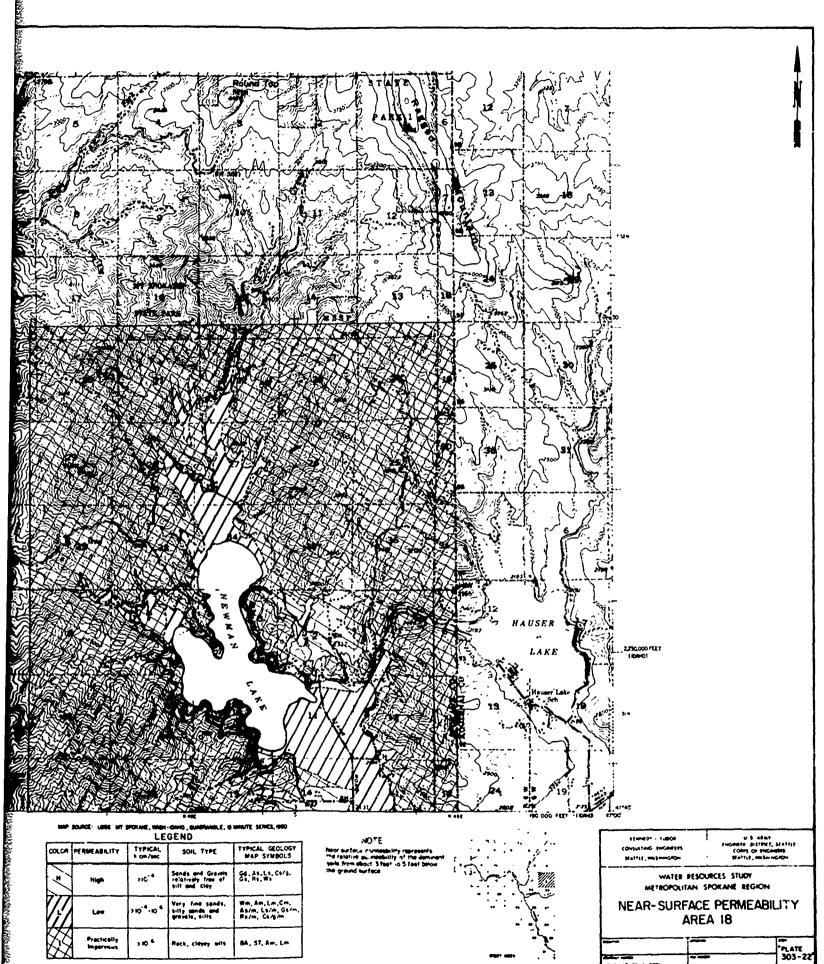






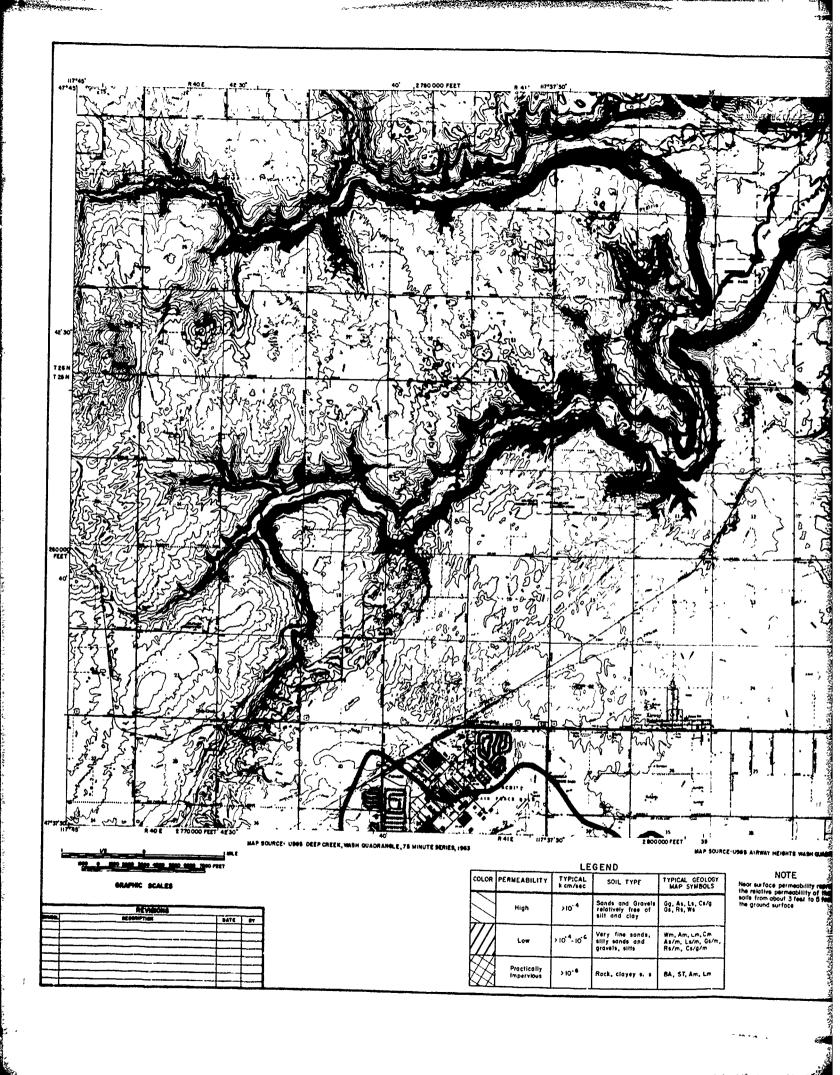


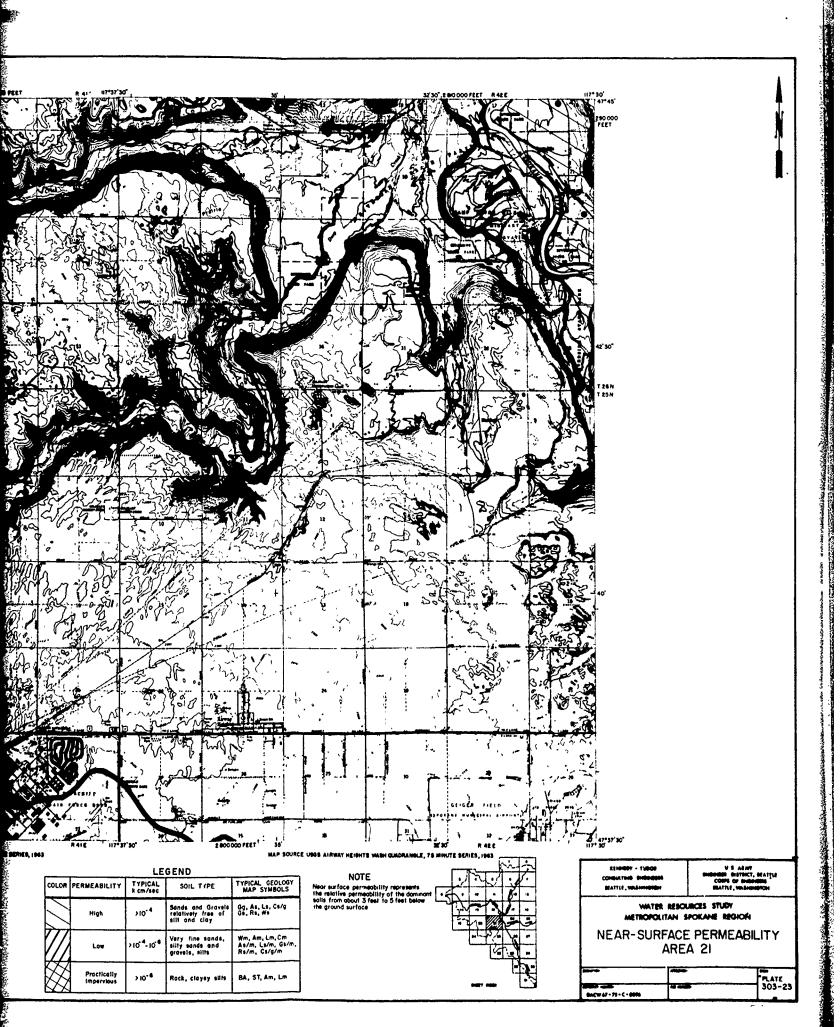


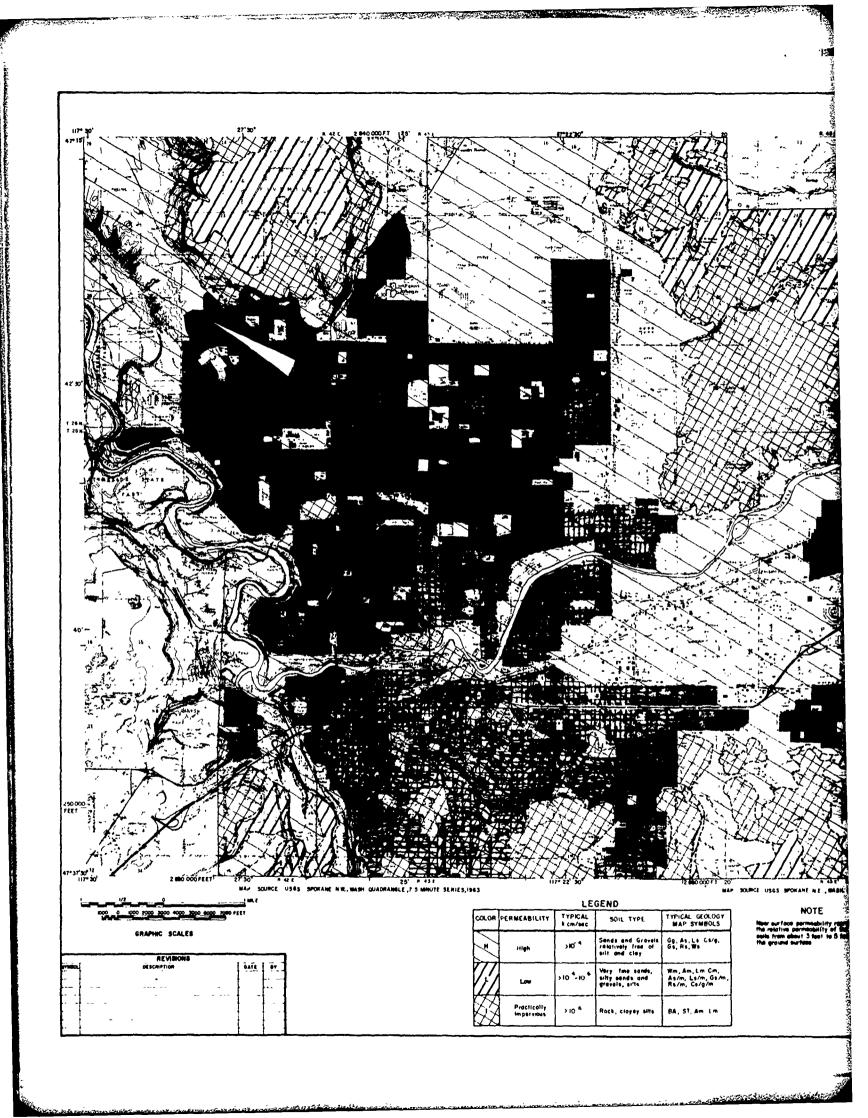


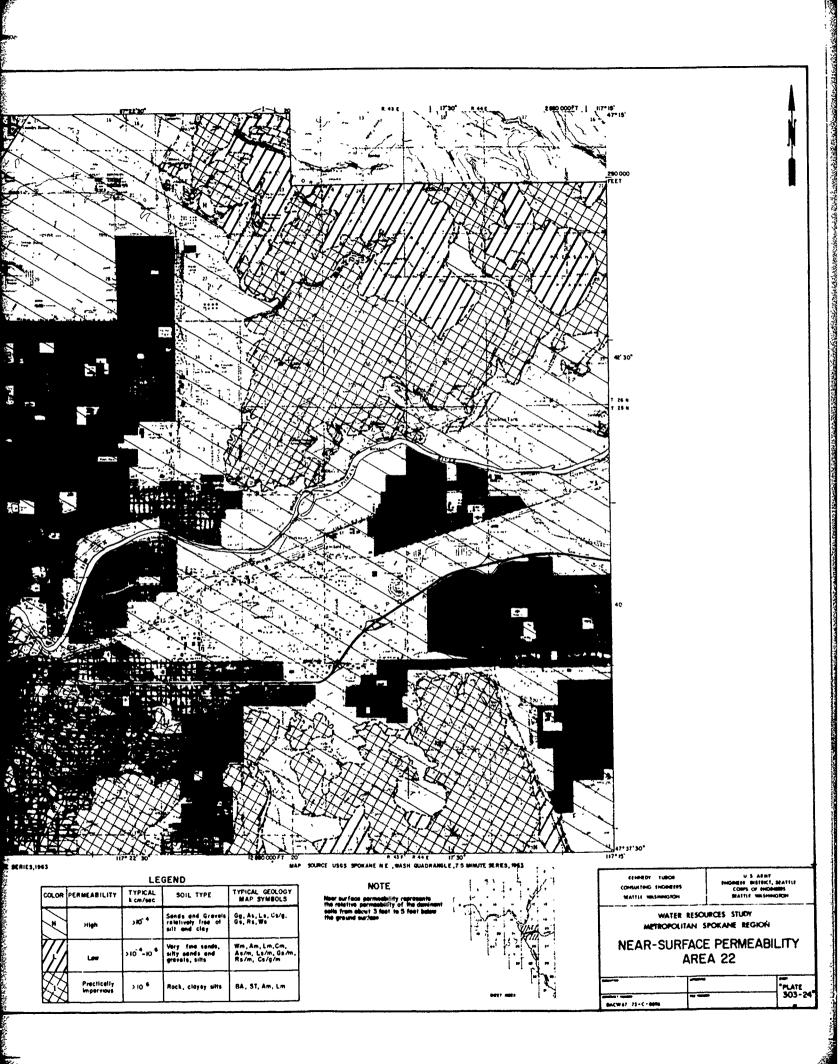
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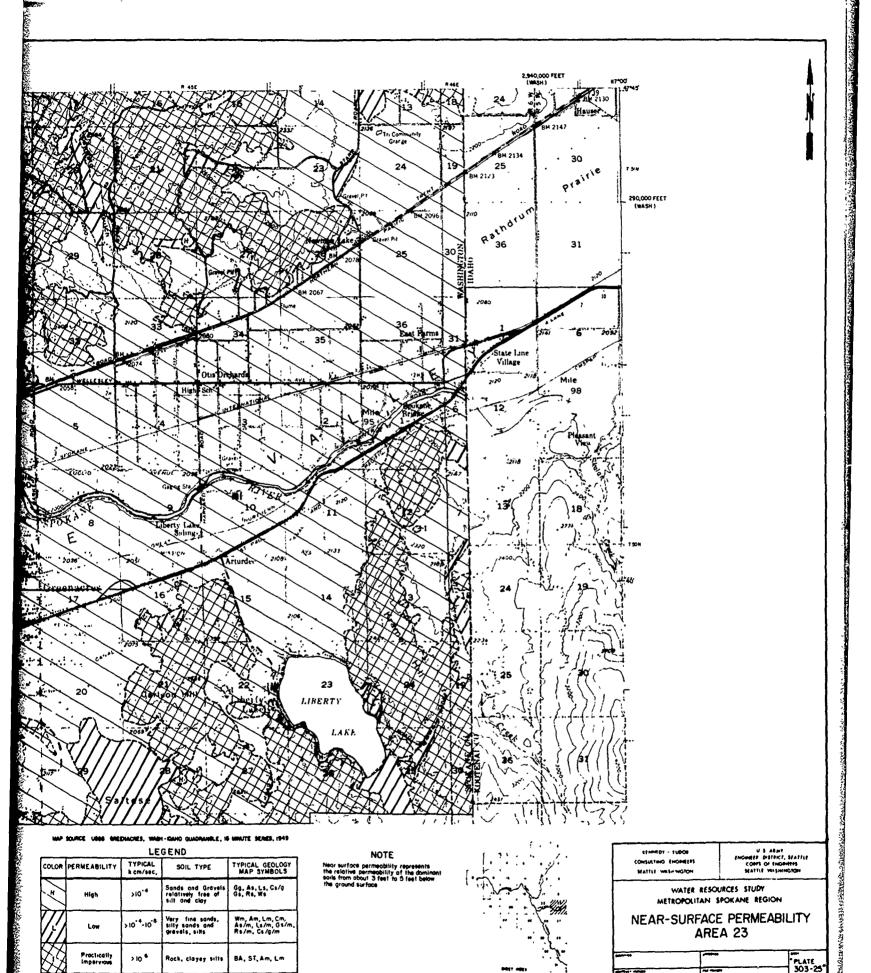




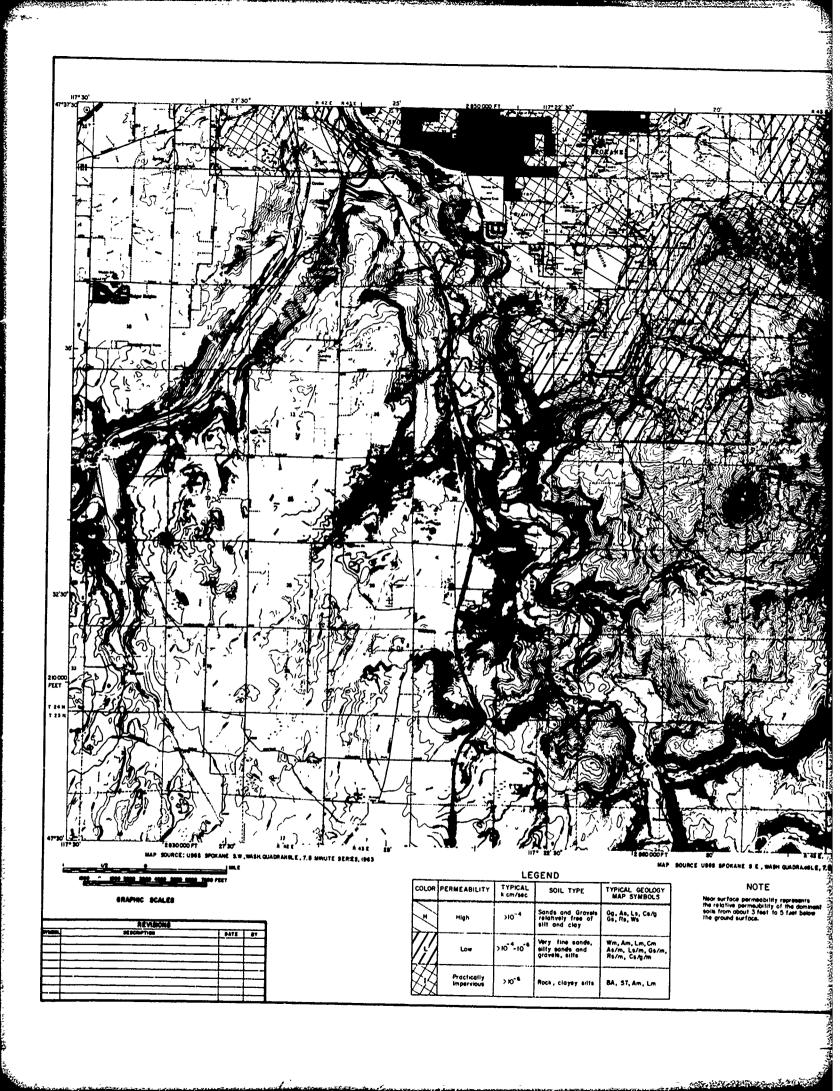


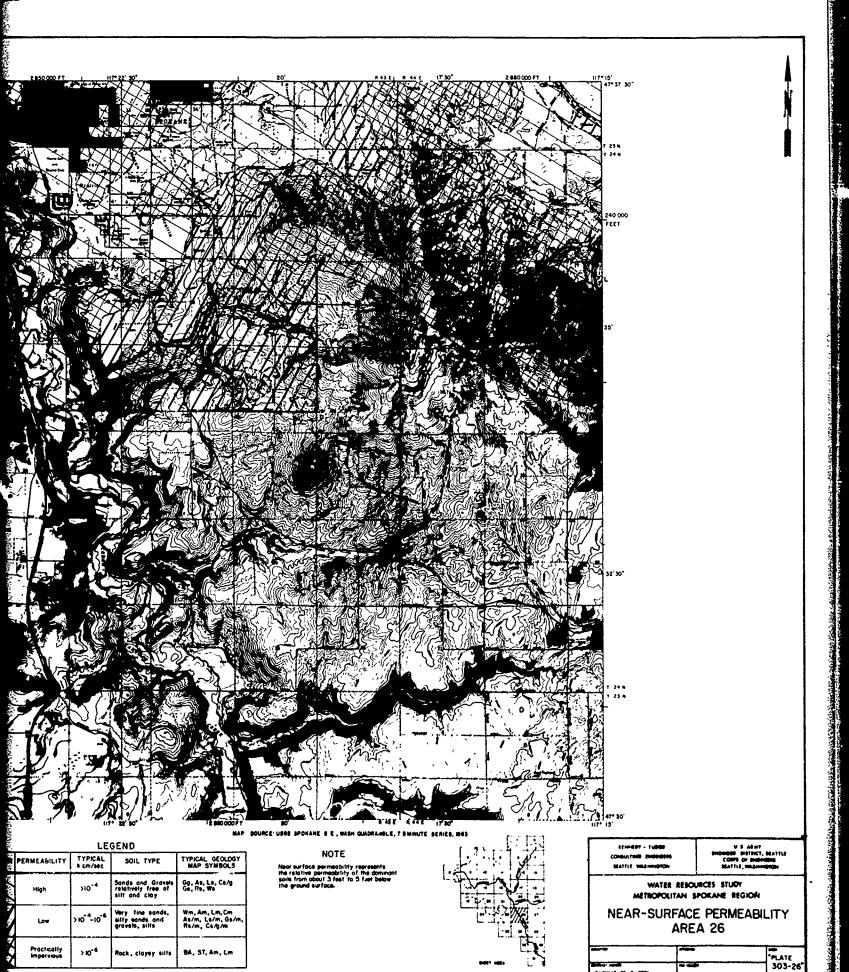


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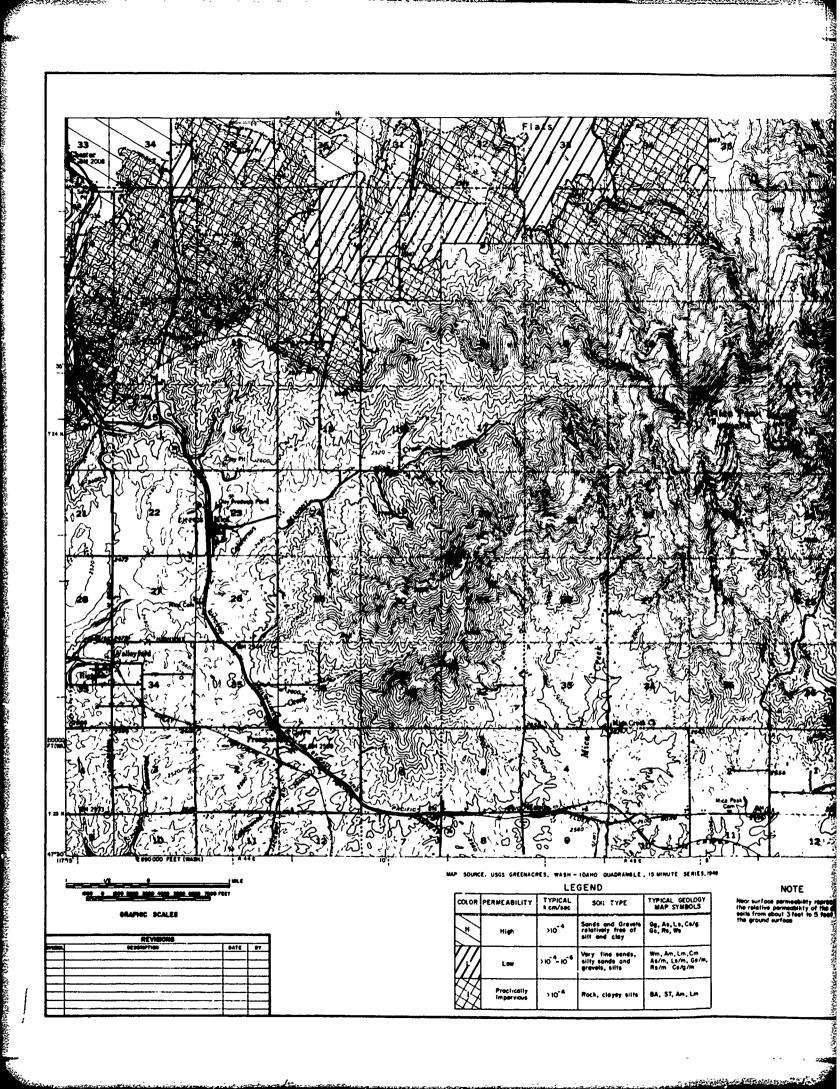
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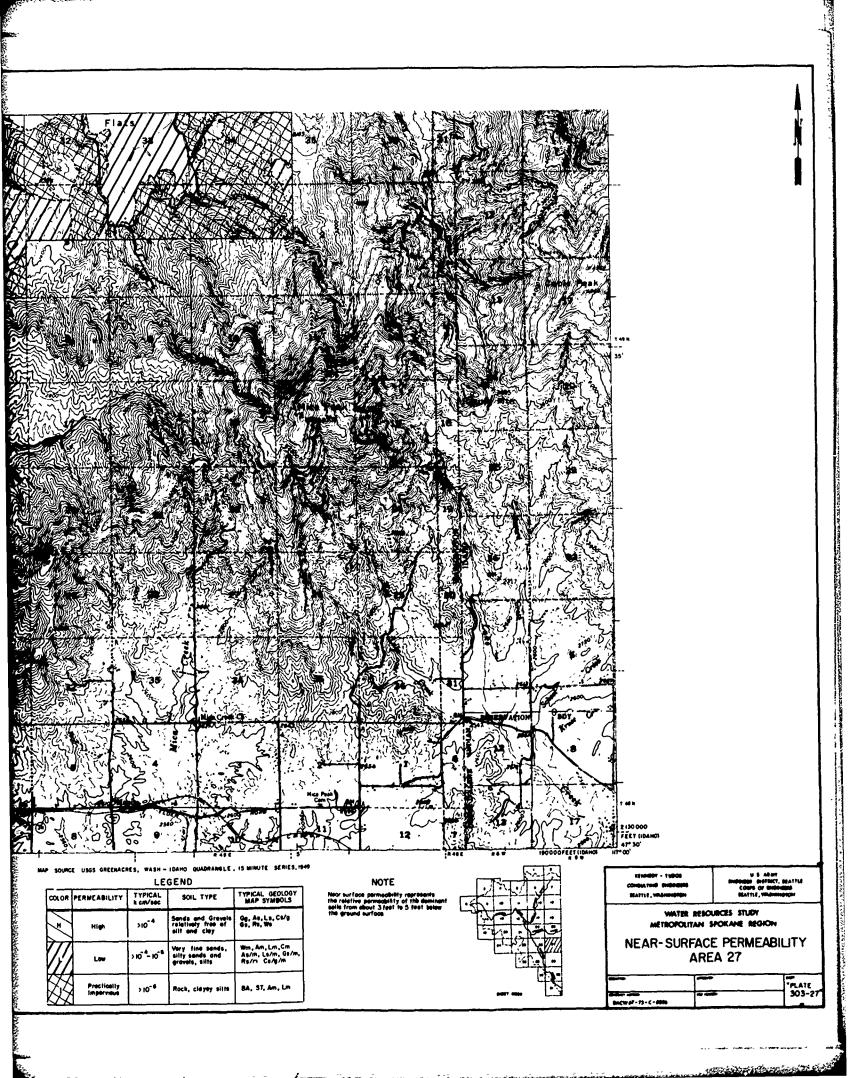


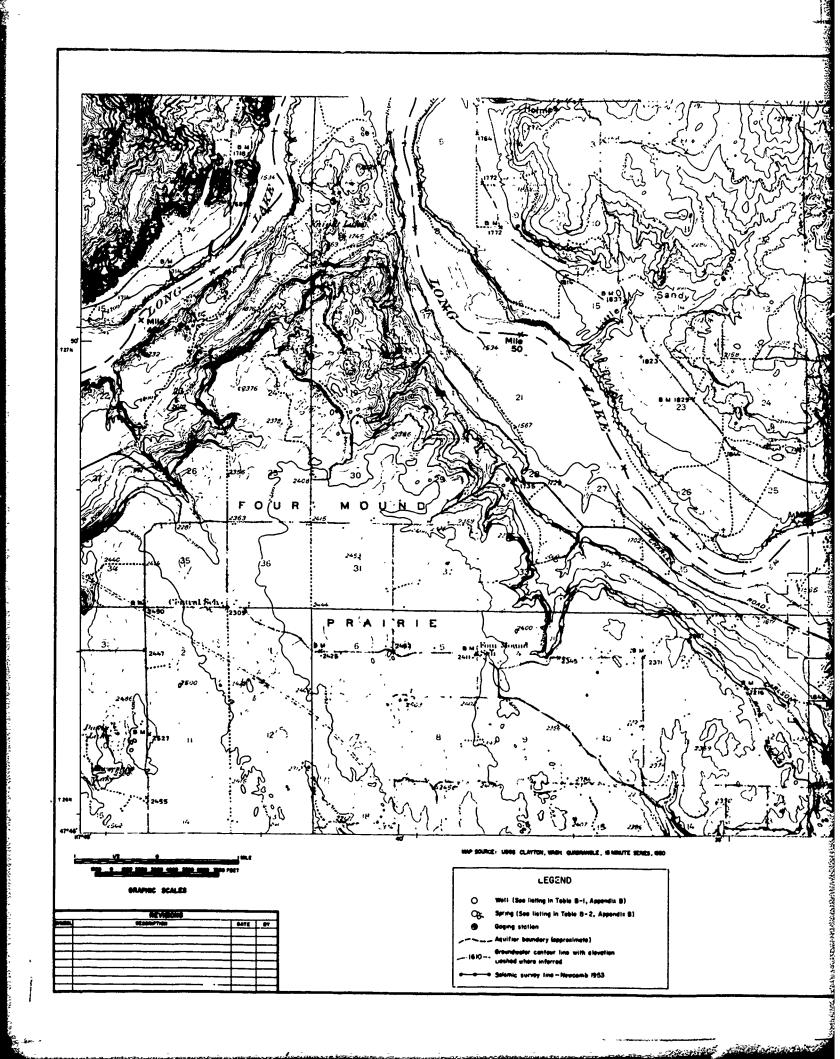


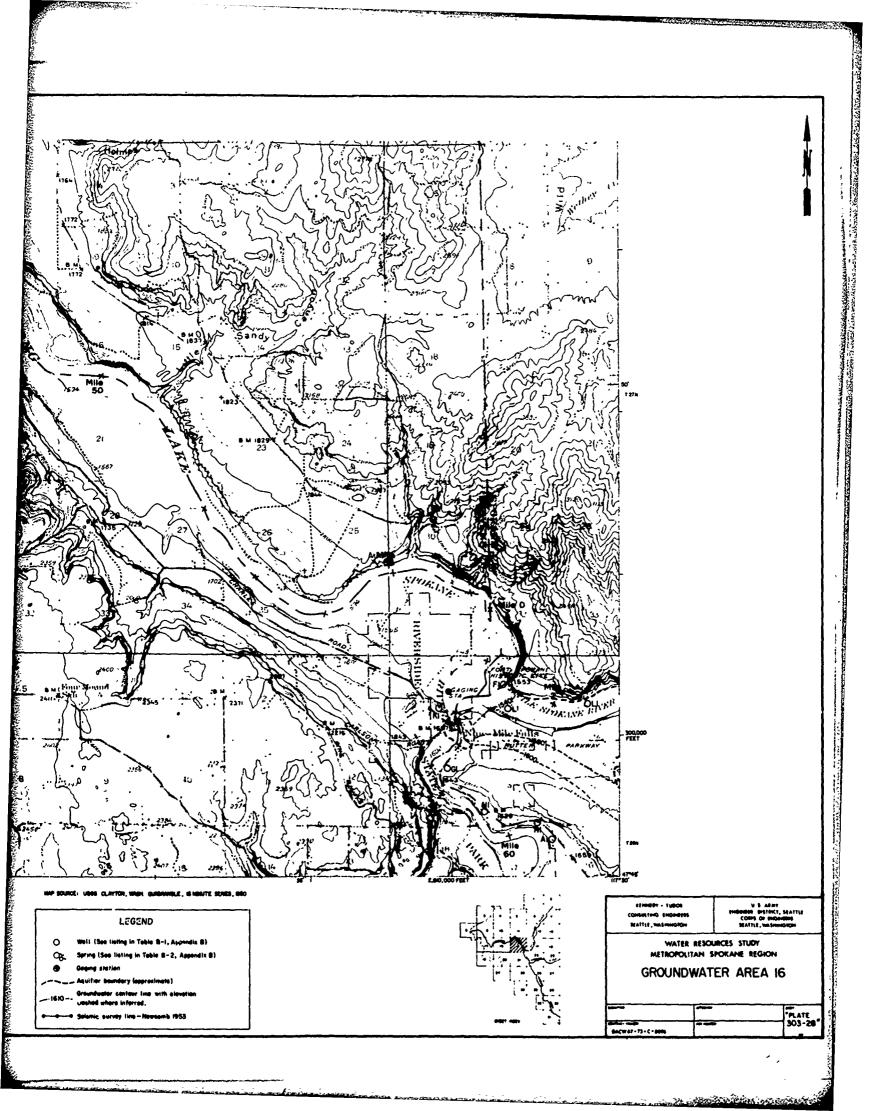
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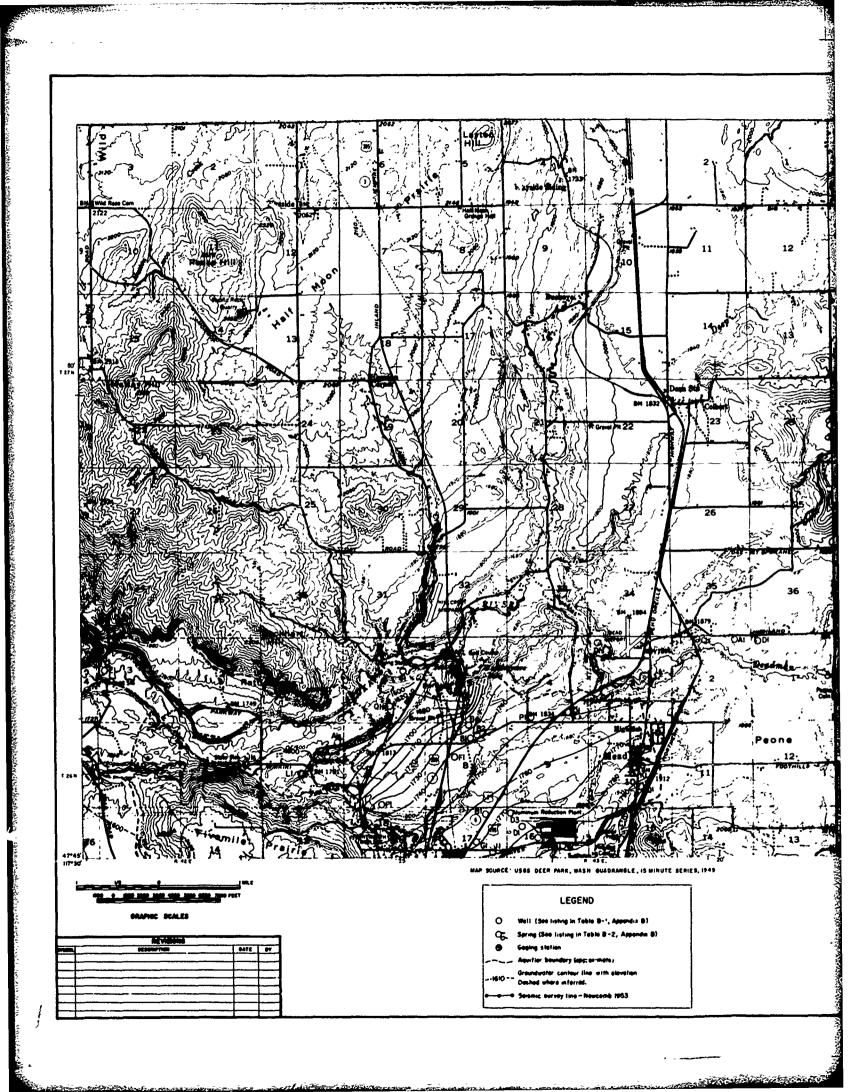
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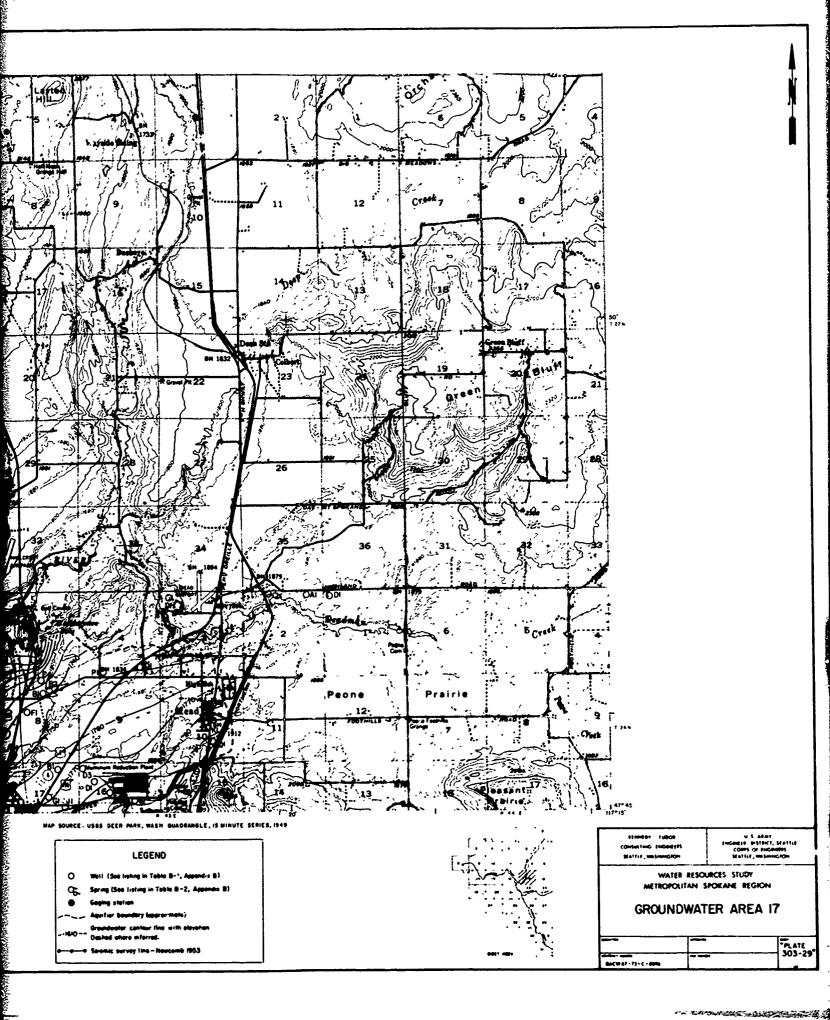


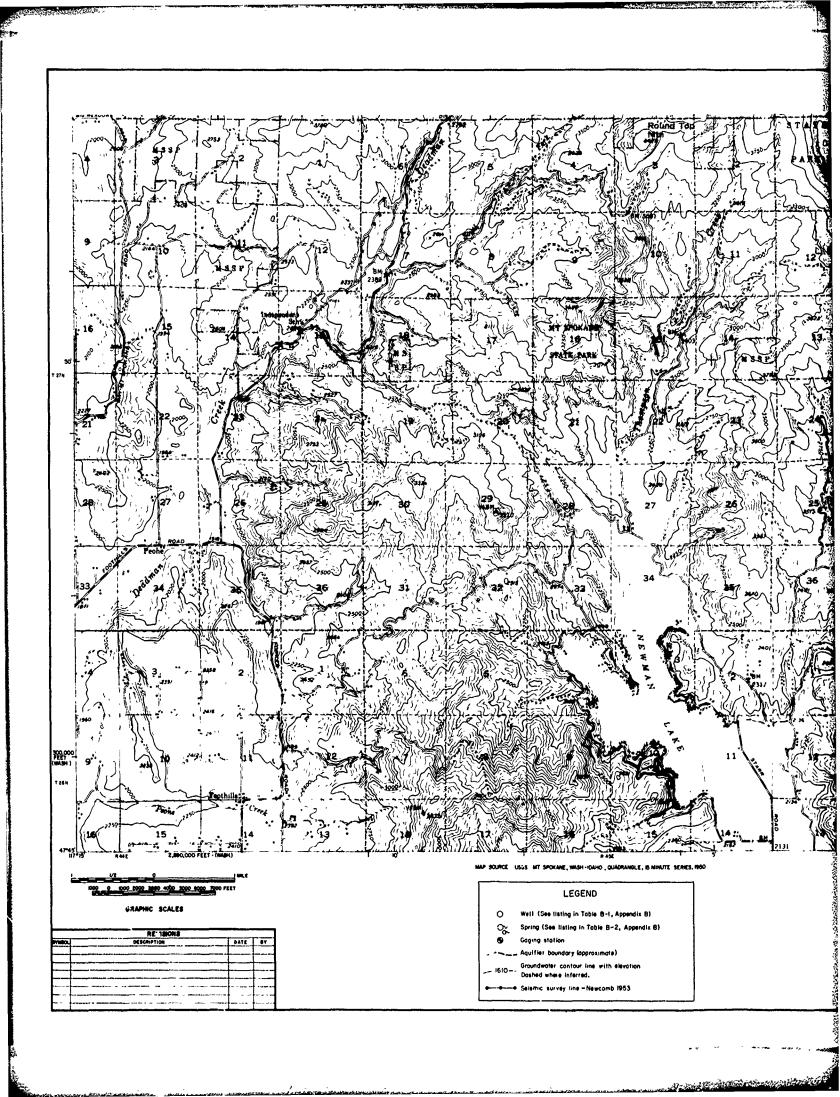


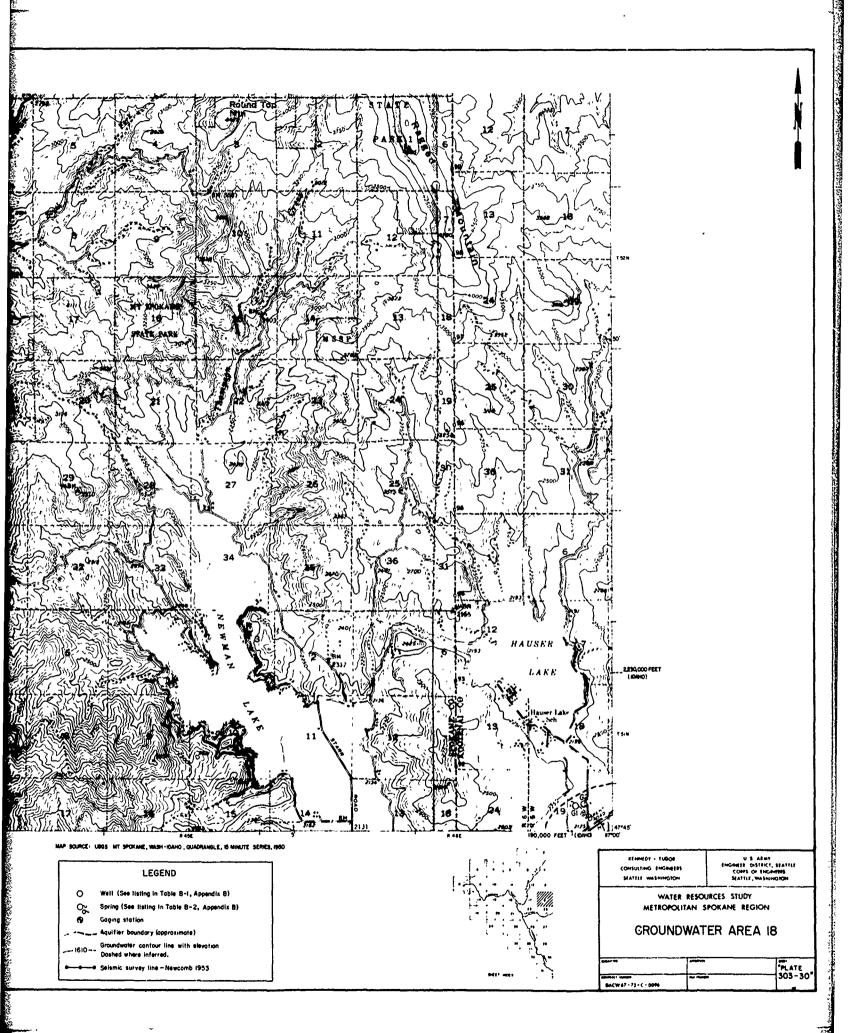


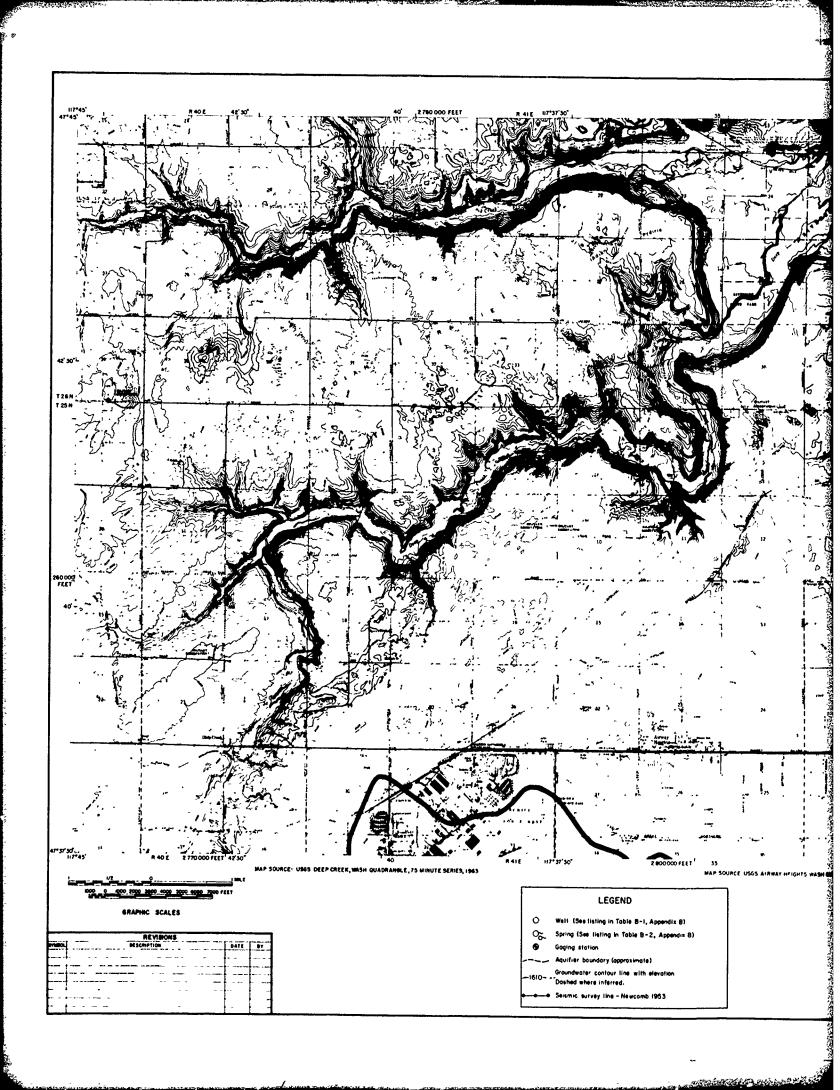


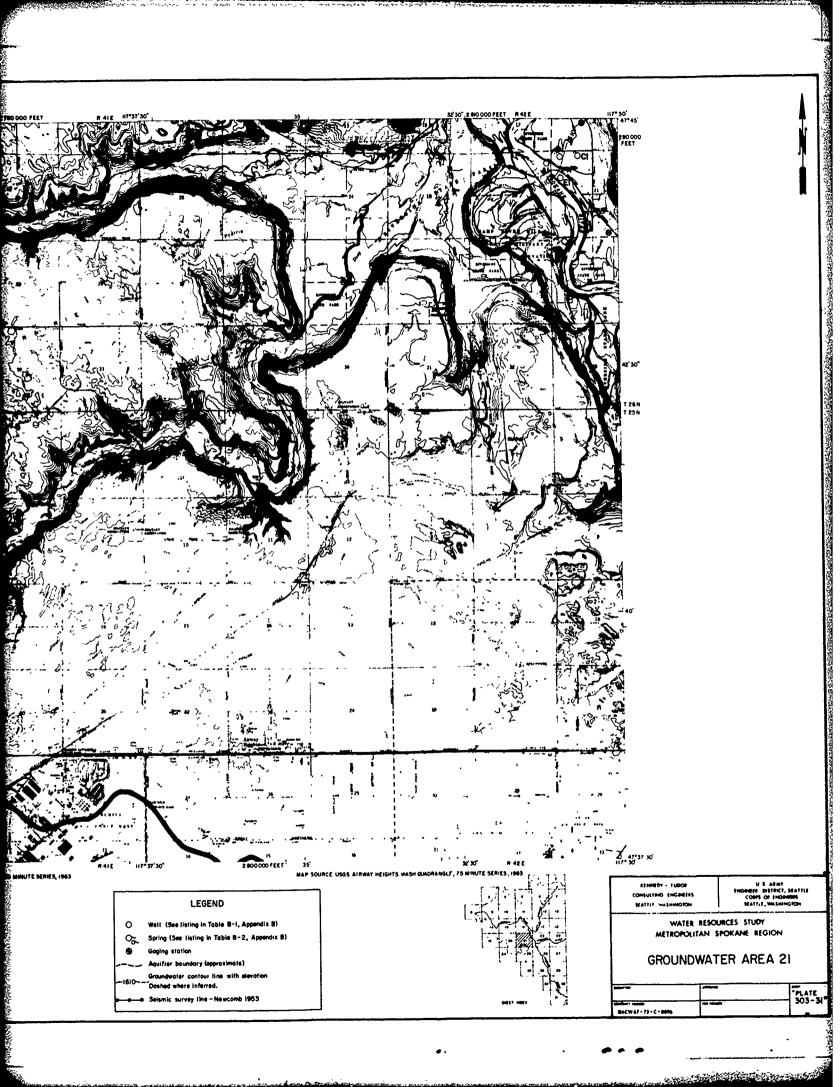


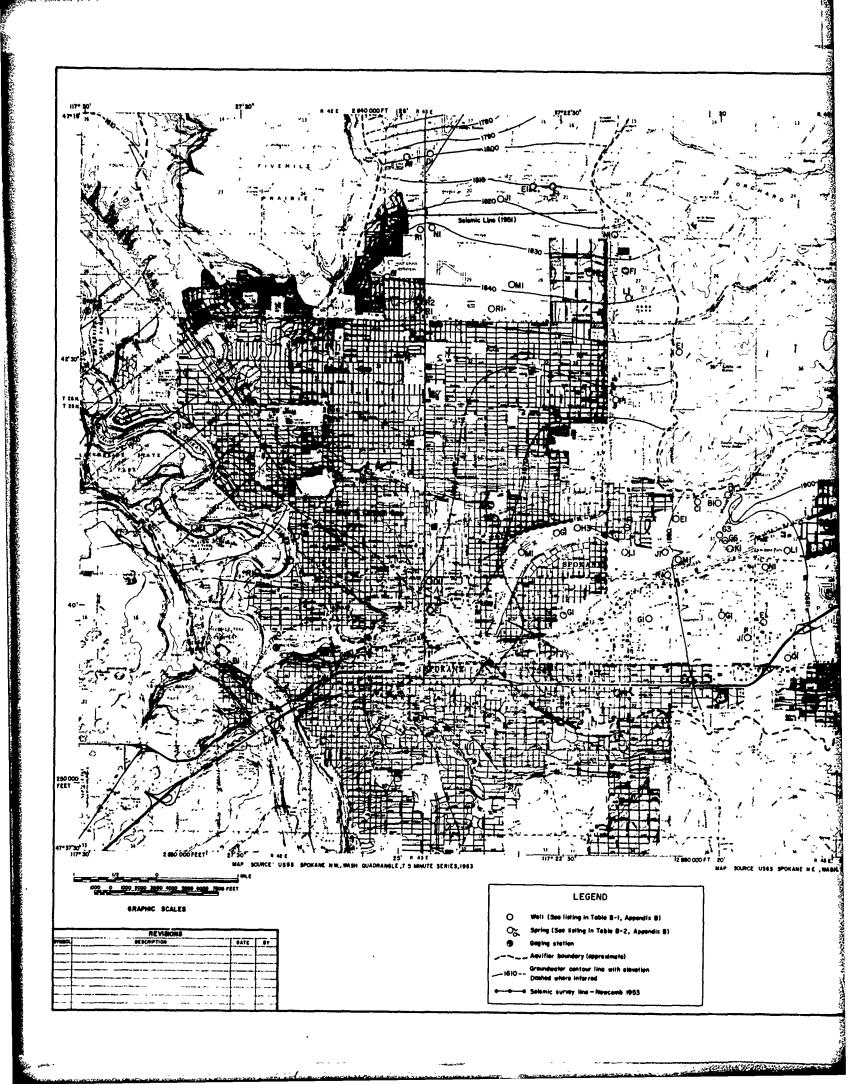


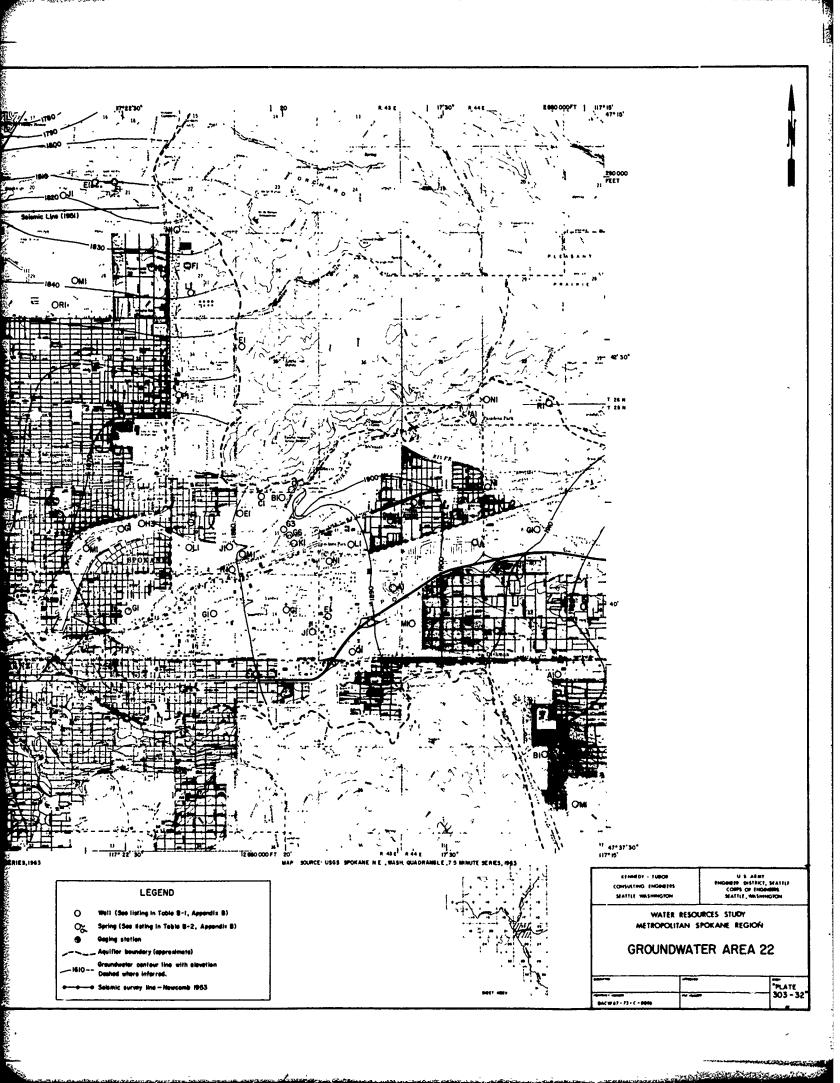


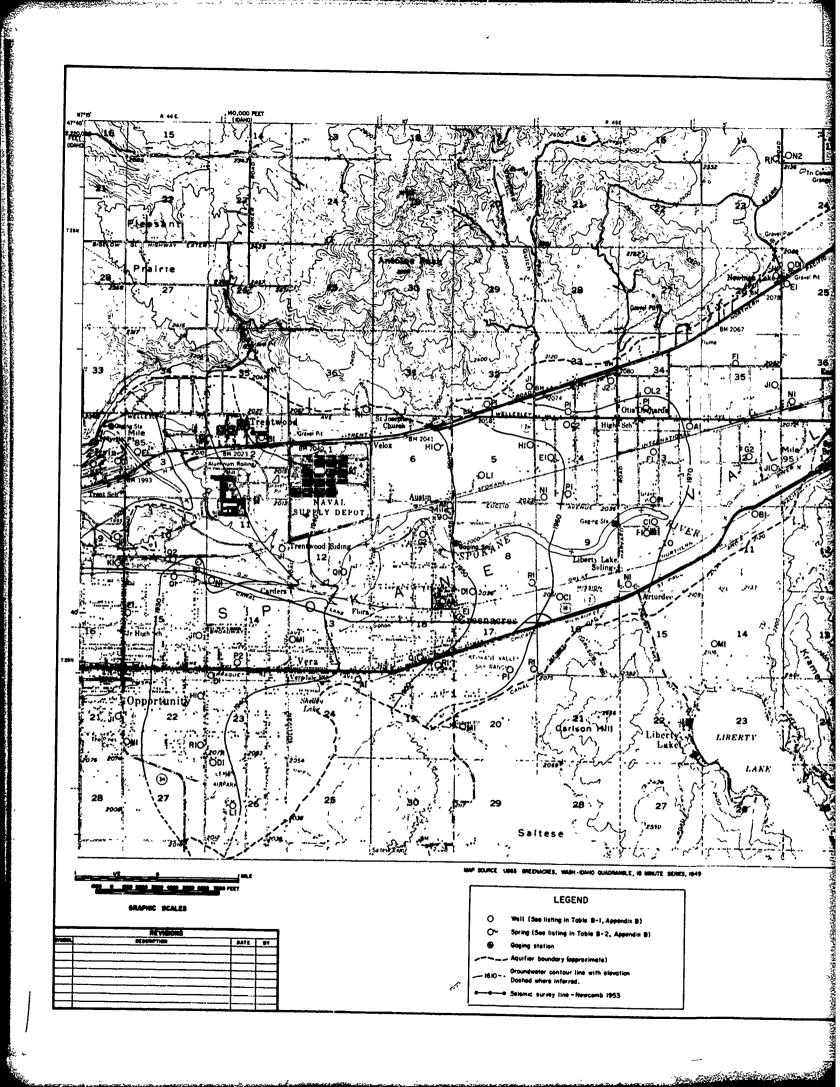


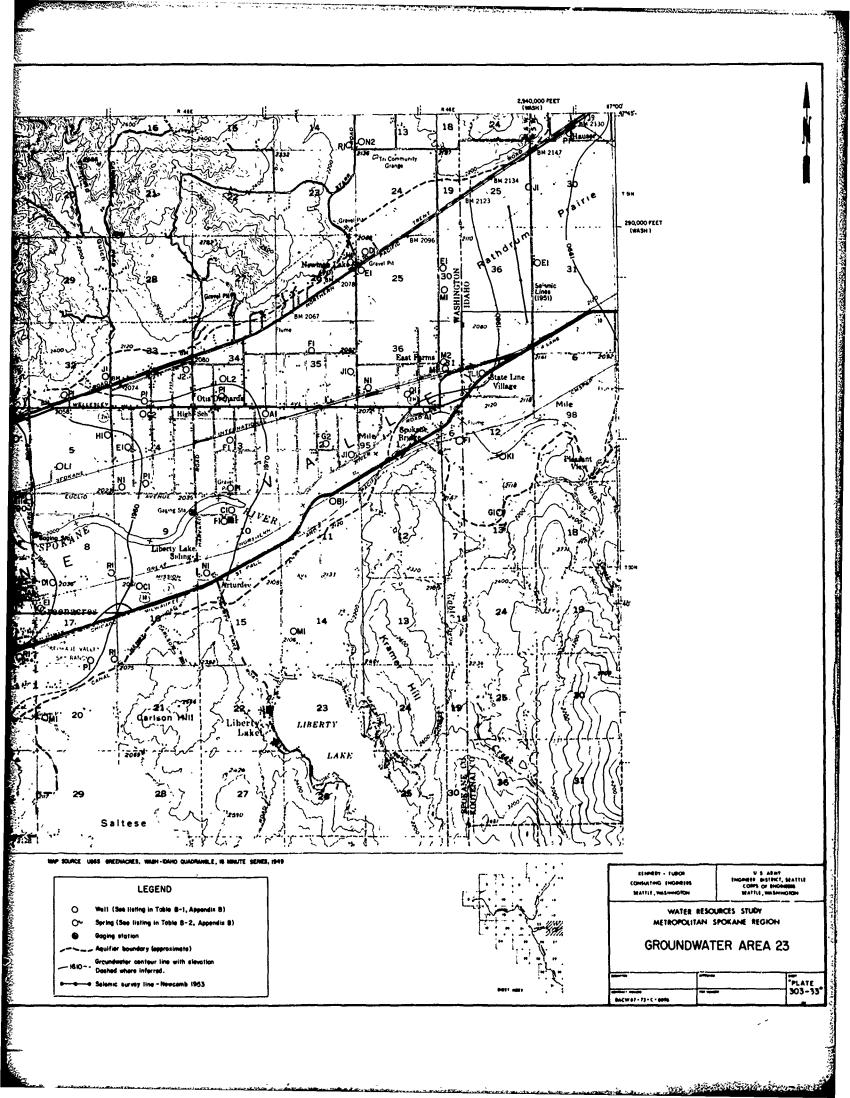


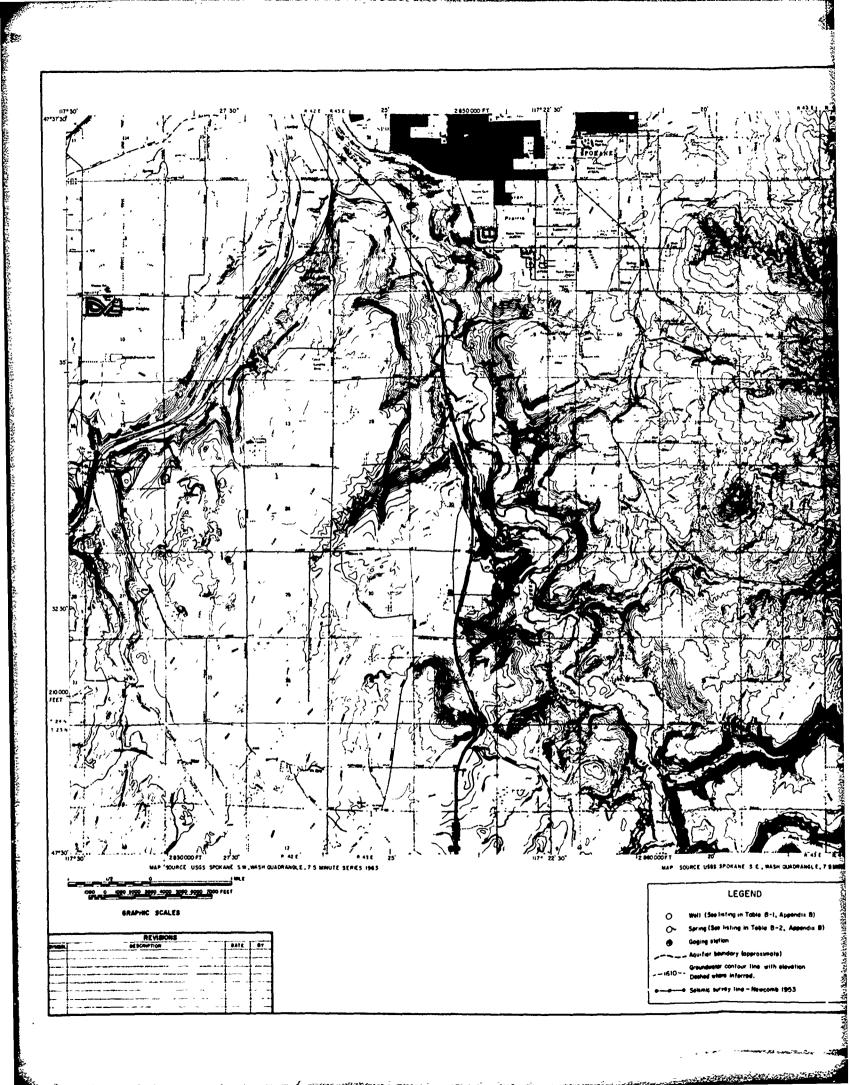


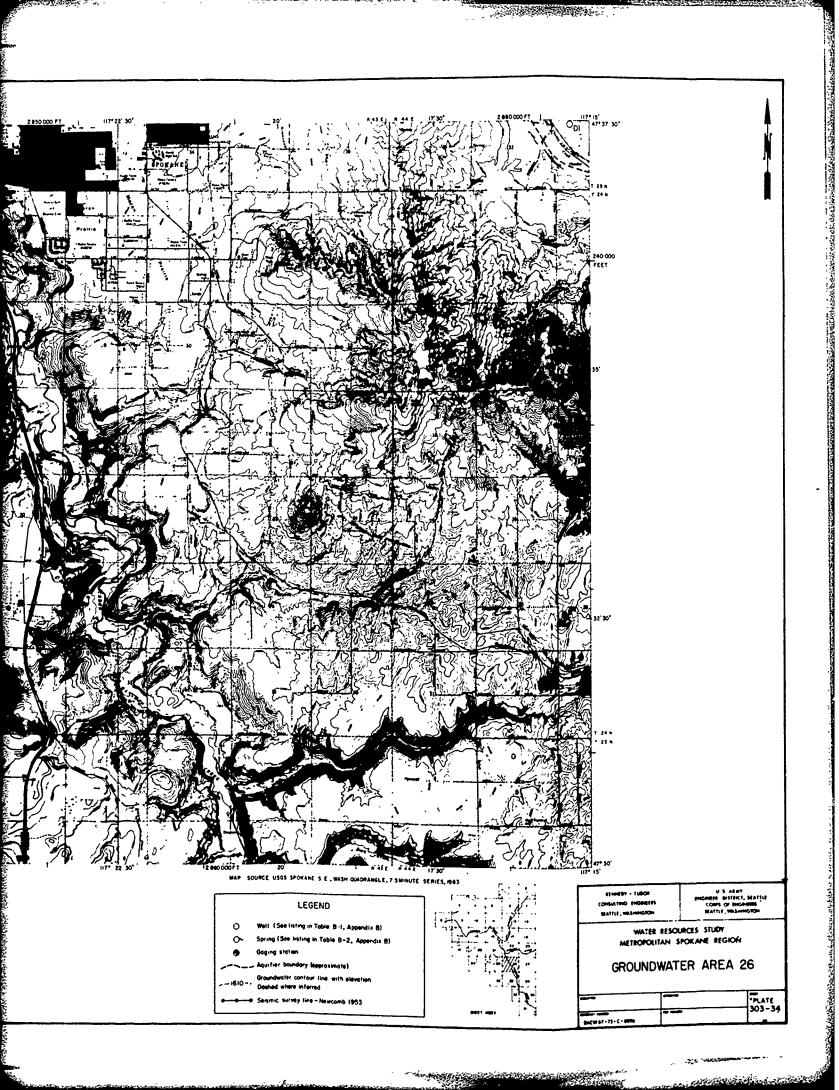


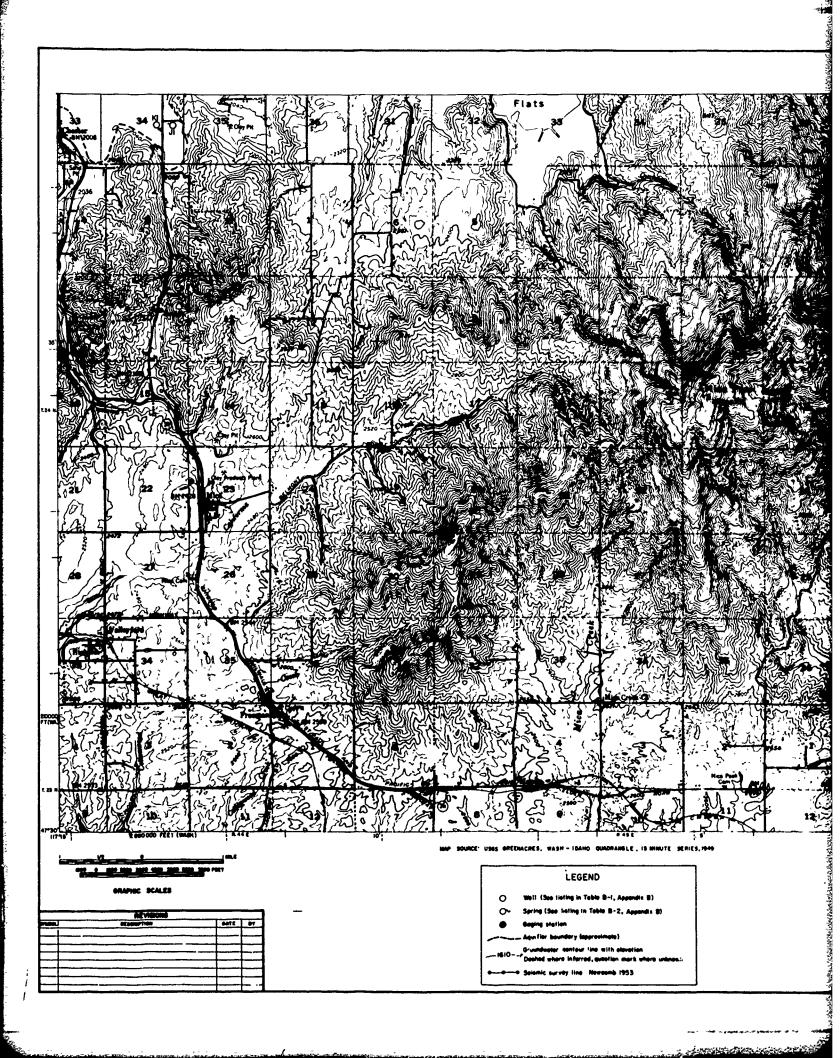


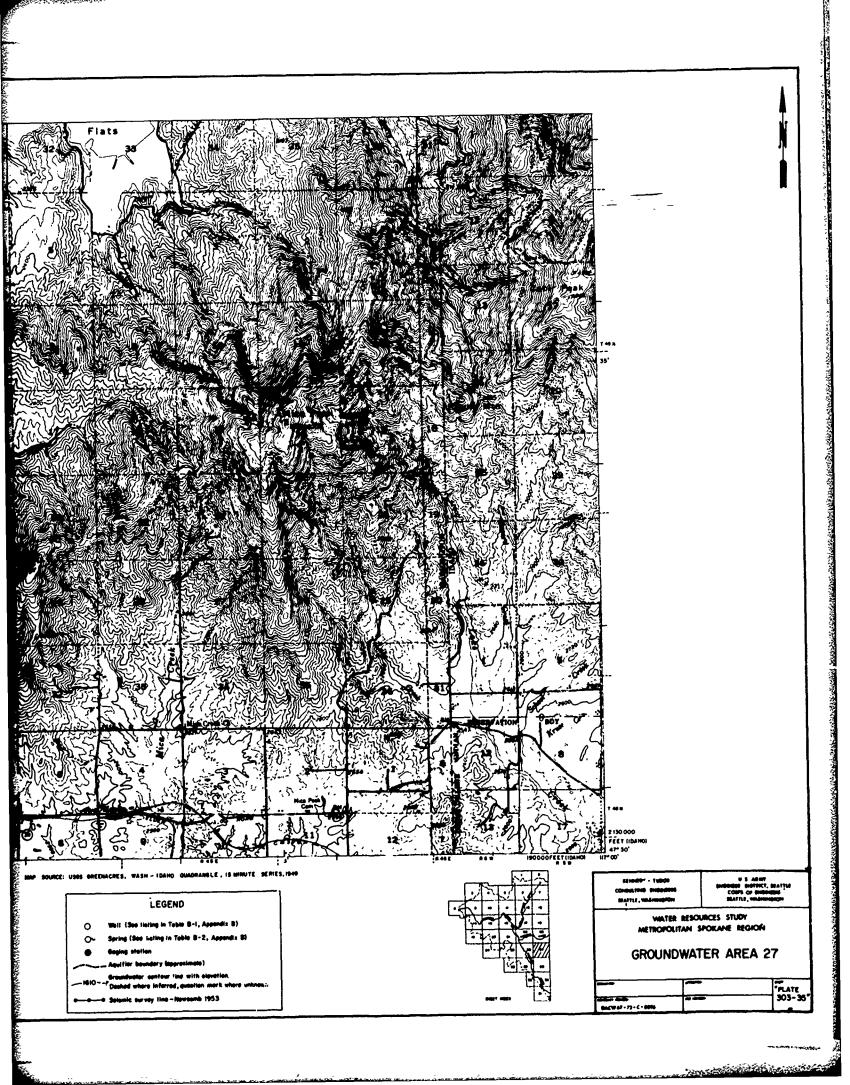


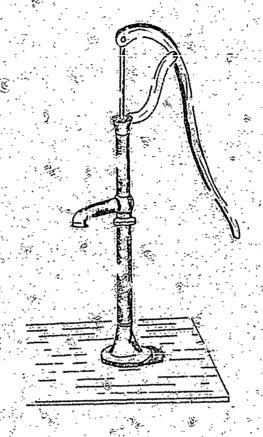












SECTION NOS

GROUNDWATER QUALITY DATA

WATER RESOURCES STUDY METROPOLITAN SPOKANE REGION

SECTION 405

GROUNDWATER QUALITY DATA

15 November 1974

Department of the Army, Seattle District Corps of Engineers Kennedy-Tudor Consulting Engineers

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 $[\]star$ All plates are large drawings bound at the end of this section.

SECTION 405

GROUNDWATER QUALITY DATA

Introduction

This section has four objectives:

- (1) To summarize prior existing groundwater quality data
- (2) Identify data gaps in the prior existing record
- (3) Describe the selection of a supplemental sampling program to fill data gaps
- (4) Report the results of the supplemental sampling program.

 Interpretation of the groundwater quality data is not an objective of this section. Interpretation and analysis are covered in Section 608.2.

There are two groundwater regimes of major areal extent, the primary Spokane Valley aquifer and the basalt aquifer, plus a number of other aquifers of smaller extent. The four objectives listed above are addressed in terms of data divided into aquifer categories designated "primary", "basalt" and "other".

An incidental objective of this section is the development of a groundwater quality file for insertion in the simulation model. The groundwater of the primary aquifer, most of which does not originate in the study area, forms an important constituent of the basin surface waters which must be accounted for in simulation of surface water quality.

Available Prior Existing Data

General. The general availability of groundwater quality data is

developed in Section 307-9. It is shown in 307-9 that there is a large amount of general chemical quality data available in the raw data files of the Department of Social and Health Services (DSHS) which are not categorized or systemitized except by County and which are not in STORET. These DSHS files are explored as part of this section to fill specific data gaps prior to development of the supplemental sampling program. Their usefullness is greatest in the areas of the basalt and other aquifers which are not as well represented in STORET and were not included in the USGS-EPA special program.

Primary Aquifer. Due to the great interest in this aquifer which provides, with one exception*, the entire municipal, industrial and agricultural supply of the Urban Planning Area, it is possible to assemble for this aquifer water quality data from thirty wells, all based on samples in the period 1970-74 and all covering a significant range of quality parameters. These data are all available from either STORET or the recent USGS-EPA program. These thirty wells and their quality data are shown in Appendix I. An overall evaluation of typical primary aquifer quality is presented in Table 1 based on statistical summary of twenty-five of the thirty wells. The following five wells are omitted from the summary because they are untypical in some respect as indicated below:

Number	Name	Anomalous Data				
25/42-13B1	Spokane Cold Storage	High manganese				
25/43-24G1	East Spokane #1	High solids				
25/44-201	Kaiser Eactgate	High solids, chlorides, chrome, zinc				
26/45-36N1 26/45-36Q1	G. N. Siverson Borden	fligh iron and zinc High zinc				

^{*} The one exception is the Kaiser Trentwood industrial cocling water diversion from the Spokane River.

It is evident that these five wells are being affected by influences not common to the rest of the primary aquifer wells and are not representative of typical conditions.

The locations of the wells listed in Appendix I are shown on Plate 608-1.

Basalt Aquifer. Groundwater quality data are available for 46 wells in the Basalt Aquifer by extending the search to the DSHS source and utilizing data back to 1942. The data are listed in Appendix II. Sources utilized are as follows:

Source	Number of Wells
DSHS	31
Van Denburgh and Santos	13
USGS (Unpublished)	1
Weigle and Mundorf	1_
	46

Locations of the wells listed in Appendix II are shown in Plate 608-2.

A summary of quality data for the Basalt Aquifer based on statistical

analysis of the data in Appendix II is presented in Table 2.

Other Aquifers. Of the other aquifers, the alluvium of the Little Spokane River valley is the most important from both an areal and utilization standpoint. Therefore, this area is treated separately and the data for the remaining areas is lumped together. Data for 16 wells in the Little Spokane basin and for 19 wells in other locations are available entirely from DSHS records for the period 1970-73. These data are compiled in Appendices III and IV respectively and locations are shown on Plate 608-2. Summarized data are shown in Table 3.

Summary Comparison of All Aquifers. Mean values of quality parameters for all aquifers are summarized in Table 4.

Identification of Data Gaps and Selection of Supplemental Sampling Program

Primary Aquifer. The data shown in Table 1, Appendix I and Plate 608-1 when evaluated for parameter and areal coverage indicates no need for expansion of the parameter list beyond the USGS-EPA selection but does indicate the need for areas not previously sampled.

Fifteen general areas of the primary aquifer are found to be inadequately represented. Selected wells to represent twelve of the fourteen areas are shown on Plate 608-1. There are no available wells in or even sufficiently near to areas marked 2, 8 and 9 on Plate 608-1 to fulfill requirements. There appears to be no present groundwater development at these sites, which therefore must remain gaps.

In addition to the wells selected for additional area coverage, two wells previously sampled by USGS-EPA are selected for an additional sampling to confirm what appear to be changing conditions revealed by the four periodic samples under USGS-EPA. The two wells selected for repeat sampling are as follows with the apparent anomalies in data which caused them to be selected:

Number	Name	Anomalous Data
25/42-13B1	Spokane Cold Storage	Solids moved steadily up from 179 to 231 mg/l and manganese jumped from nil to 140 ug/l
25/44 - 2Q1	Kaiser East Gate	Large increase in solids from 175 to 329, jump in chlorides from less than 10 mg/l to 60 mg/l and chromium from nil to 30 ug/l

The well at Kaiser Mead, number 26/43-16F, although previously sampled, but not in the USGS-EPA full spectrum of parameters, is selected for additional analysis to fill out the data on the downstream end of the aquifer.

Table 5 lists the wells selected for sampling and analysis in the primary aquifer. Those for additional areal coverage inside the study area are P-1, 3, 4, 5, 6, 7, 10, 11, 12, 13 and 14. One of the newly completed wells by the Bureau of Reclamation east of the Idaho border and outside the study area is selected for the opportunity it affords of getting a sample representative of the water entering from Idaho. This well is designated P-W, 5/51-28N. Wells designated sample P-X and P-Y are the repeat samples of USGS-EPA points, Kaiser Eastgate and Spokane Cold Storage respectively. Sample P-Z is the expanded parameter sample of Kaiser Mead.

The analysis program shown for all wells in the primary aquifer typically includes all the USGS-EPA chemical parameters. Bacteriological and hexane extractables are added for those in and downstream of

development. Aluminum is added for those near the aluminum industries.

Basalt Aquifer. Both areal and parameter coverage available are adequate for an overview of Basalt Aquifer water quality except that there are no checks available on possible contamination at three communities. Sampling and analysis for wells near Airways Heights, Spangle, and Latah are selected as shown on Table 5. Since the interest in this coverage is primarily possible pollution from the adjoining development, the analysis selected is bacteriological and detergents. Basalt aquifer samples are identified by the letter B.

Other Aquifers. The data gap situation for other aquifers is similar to that stated above for the basalt. Data are adequate in general but there are three areas of development for which there is no check on potential pollution. Three locations as indicated in Table 5 are sampled and analyzed for bacterial indicators and detergent. Samples in other aquifers are identified by the letter O.

Sampling Program and Results. Samples were collected in the period June 4, 1974 to June 10, 1974. Analyses were performed in accordance with the schedule shown on Table 5 by Pacific Environmental Laboratory of San Francisco. All analyses are according to "Standard Method for the Examination of Water and Wastewater", Current Edition, APHA, except as noted below. Fluoride was analyzed by specific ion eletrode. Mercury was analyzed by flameless atomic absorption spectrophotometry. Cadmium, chromium, copper, iron, lead, manganese, potassium, sodium, and zinc were

analyzed by atomic absorption spectrophotometry. The conductivity, pH, and water temperature were analyzed in the field. Coliform tests were performed in the field laboratory by the membrane filter method.

It should be noted that the Kjeldhal nitrogen determination reported here are by the manual micro Kjeldhal apparatus in accordance with standard methods and are not of the same degree of sensitivity as those reported by USGS-EPA which are done by an automated apparatus. In both cases the amounts reported are at the minimum range of the respective methods.

Reports of the analyses are summarized in Table 6.

Summary Groundwater Quality

The water quality data gained in the supplementary sampling and analysis program reported above are combined with the prior existing data to arrive at mean values representative of each aquifer. These results are shown in Table 7.

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TABLE 1
SUMMARY OF PRIOR AVAILABLE GROUNDWATER
QUALITY DATA FOR THE PRIMARY AQUIFER

Parameter	<u>Units</u>	Average Value	Standard Deviation	# Samples
Conductivity at 25°C	umhos	285	. 56	201
Residue (Total)	mg/l	177	46	150
Residue (Diss-180°)	mg/l	171	26	49
Residue	TON/AFT	0.23	0.04	48
pН	-	7.7	0.6	198
Temp.	°C	10.7	1.8	69
D. O.	mg/1	8.3	1.4	5
Hardness (As CaCO ₃)	mg/1	157	40	202
NH ₃ -N	mg/1	0.015	0.008	49
NO ₂ -N	mg/l	0.002	0.002	49
NO3-N	mg/l	1.521	1.258	50
Kjel-Nitrogen (N)	mg/1	0.111	0.146	49
PO ₄ -P (Total)	mg/l	0.014	0.012	50
PO ₄ -P(Ortho)	mg/l	0.010	0.012	49
Cl	mg/l	4.7	3.6	198
As (Diss)	ug/1	5	10	49
Cd (Diss)	ug/l	0	0	49
Cr (Diss)	ug/l	0	0	49
Cu (Diss)	ug/l	6	11	49
Fe (Diss)	ug/1	28	32	49
Fe (Total)	ug/l	169	356	146
Pb (Diss)	ug/l	2	3	43
Pb (Total)	ug/l	19	8	20
Mn	ug/l	7	10	201
Hg (Total)	ug/1	2	4	61
Zn (Diss)	ug/1	26	32	50
MBAS	mg/1	0.02	0.03	45

TABLE 2
SUMMARY OF AVAILABLE GROUNDWATER
QUALITY FOR THE BASALT AQUIFER

Parameter	<u>Units</u>	Average Value	Standard Deviation	# Samples
Conductivity at 25°	umhos	263	90	95
Dissolved Solids Calculated * Residue (180°C)	mg/1 mg/1	181 177	33 34	52 64
рН	-	7.7	0.3	95
Temp.	°C	14.0	3.3	56
Color	Color Units	5	7	90
Hardness, as CaCO3	mg/l	108	38	97
ALK. as CaCO3	mg/l	120	46	31
NO ₃ -N	mg/l	1.686	2.884	96
Total PO ₄ -P	mg/l	0.088	0.121	31
Ortho PO ₄ -P	mg/l	0.070	0.099	2
C1	mg/l	5.1	7.0	96
so ₄	mg/l	14.5	11.2	96
F	mg/l	0.308	0.215	96
Fe	ug/l	140	178	88

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^{*} Based on the sum of the concentrations of the individual major constituents.

TABLE 3

SUMMARY OF PRIOR AVAILABLE GROUNDWATER QUALITY
DATA FOR THE LITTLE SPOKANE BASIN
AND OTHER AQUIFERS

		Little S	pokane Basi	n Aquifer	Ot1	her Aquifers	
Pa:ameter	Units	Average Value	Standard Deviation	No. of Samples	Average Value	Standard Deviation	No.of Samples
рН	-	7.7	0.4	16	7.4	0.4	19
Conductivity	umhos/cm	292	106	16	285	145	19
Color	Color Units	4	2	16	7	6	19
Hardness (CaCO ₃)	mg/l	151	64	16	136	72	19
Alkalinity (CaCO ₃)	mg/1	139	54	16	114	58	19
Fe	ug/1	100	100	16	161	167	19
so ₄	mg/1	17.7	9.4	16	16.7	9.6	19
Cl	mg/l	3.4	4.5	16	8.6	10.3	19
F	mg/l	0.166	0.093	16	0.247	0.301	19
NO ₃ -N	mg/1	1.418	1.624	16	1.046	1.504	19
P04-P	mg/1	0.095	0.082	16	0.115	0.272	19

TABLE 4
SUMMARY OF AVAILABLE GROUNDWATER
QUALITY FOR ALL AQUIFERS

教育ない

		Mean Values by Aquifer						
Parameter	Units	Primary	Basalt	Little Spokane	Other			
Conductivity	umhos/CM	285	263	292	285			
Residue (Total)	mg/l	177	_	_	-			
Residue (180°C-Diss)	mg/l	171	177	-	***			
Residue	TON/AFT	0.23	-	-	_			
pН	-	7.7	7.7	7.7	7.4			
Temp.	°C .	10.7	14.0	-	-			
D. O.	mg/l	8.3	~	-	-			
Color	Color Units	-	5	4	7			
Hardness (CaCO3)	mg/l	157	108	151	136			
ALK (CaCO ₃)	mg/l	-	120	139	114			
ин ₃ -и	mg/l	0.015	~	-	-			
NO2-N	mg/l	0.002	-	-	-			
No ₃ -N	mg/1	1,521	1,686	1.418	1.046			
Kjel-Nitrogen (N)	mg/l	0.111		-	1.040			
PO ₄ -P (Total)	mg/l	0.014	0.088	0.095	0.115			
PO ₄ -P (Ortho)	mg/l	0.010	0.070	_	- 0.113			
C1	mg/1	4.7	5.1	3.4	8.6			
so ₄	mg/1	-	14.5	17.7				
F	mg/l	_	0.308	0.166	16.7			
AS (Diss)	ug/1	5	0.506	0.100	0.247			
Cd (Diss)	ug/1	ő	•	_	_			
Cr (Diss)	ug/1	Ö	•••	-	_			
Cu (Diss)	ug/l	6		-	_			
Fe (Diss)	ug/l	28						
Fe (Total	ug/1	169	140 *	100 *	161 *			
Pb (Diss)	ug/l	2	-	-	_			
Pb (Total)	ug/l	19		_	_			
Mn	ug/1	7	-	-	_			
Hg (Total)	ug/l	2	-	-	-			
Zn (Diss)	ug/l	26	~	_	_			
MBAS	mg/1	0.02	~	-	~			

^{*} Not known if these represent total or dissolved iron.

TABLE 5
SUPPLEMENTAL GROUNDWATER SAMPLING PROGRAM

Sample		USGS	Parameters
Number	Well Name	Well Number	Sampled
P-1	City, Hoffman #1 & #2	25/43-4B1	A-B-C
P-3	City, Ray St. #1 & #2	25/43-22F	A-B
P-4	Modern #8	25/44-17A1	A-B
P-5	Model #1	25/44-21L	A-B
	Vera #5	25/44-26D1	A-B
P-7	CID #2A, 2B & 2C	25/45-18R	A
P-10	WWP #3-6	26/43-3Q	A-B-C
P-11	Whitworth #2A	26/43-20D	A-B-C
P-12	N. Spokane # 1 & #2	26/43-28H	A-B-C
P-13	CID #6A, 6B & 6C	25/45-4C	A
P-14	MOAB #1	26/45-24P	A
P-W	USBR, Post Falls #1	5/51-28N	A
P-X	Kaiser Eastgate	25/44-2Q1	A-B-E
P-Y	Spokane Cold Storage	25/42-13B1	A
P-Z	Kaiser Mead	26/43-16F2	A-B-C-E
B-1	Airway Heights #3	25/41-26H	B-D
B-2	Spangle #2	22/43-4F	B-D
B-3	Latah #1	21/45-30B	B-D
0-1	Chattaroy Valley		
	Mobile Estates	28/43	B-D
0-2	Diamond Lake		
	(Degestrom Well)		B-D
0-3	Rivilla #1	26/43-6G1	B-D

Identification of Parameters Sampled

- A Full spectrum of chemical parameters per USGS-EPA program
- B Total Fecal Coliform
- C Hexane Extractables
- D Detergents (where A is not run, A includes detergents)
- E Aluminum

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TABLE 6 RESULTS OF GROUNDWATER SAMPLING PROGRAM

Minumocalum	-		ু কাক বিশ্বস	gelanggalder vilet, i en dere e	الم المحمد الما المحمد الما المحمد		Systematics of Statement of	Dige veganikaniyanin seriçi	y . yr. nagen agylagydd yr	- 144 28	er er Virtgementinge	THE PARTY OF THE P
												SENECES SERVICES SERVICES
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										LE 6	50005	
									· · · · · · · · · · · · · · · · · · ·	TER SAMPLING	Well Ide	tificatio
Parameter	Units	P-1 25/43-4B1*	P-3 25/43-22 F* 348	P-4 25/44-17A1* 270	P-5 25/44-21L*	P-6 25/44-26D1*	P-7 25/45-18R* 170		P-11 26/43-20D*	P-12 26/43-28H* 252	P-13 25/45-4C* 259	P-14 26/45-24
nductivity A sidue (180°C) sidue sperature	mg/1 TON:/AFT	176 .24 7.6 11.5	226 .31 7.5 12.0	539 .73 8.1 9.5	259 179 .24 7.75 12.0	235 160 .22 7.9 11.5	170 121 .16 7.85 11.0	265 184 .25 7.8 12.5	235 149 .20 7.3 11.5	252 146 .20 8.2 11.0	259 165 .22 7.9 9.0	167 113 .15 7.7 9.0
rdness (CaCO ₃) 3-X 2-X 3-X	mg/1 mg/1 mg/1 mg/1	154 <.056 <.002 1.1	178 <.056 <.002 2.8	152 <.056 <.002 1.2	132 <.056 <.002 2.2	132 <.056 <.002	88 <.056 <.002 1.2	148 <.056 <.002 1.5	128 <.056 <.002 1.3	136 <.056 <.002 1.5	140 <.056 <.002 1.1	116 <.056 <.002 0.17
eldahl Nitrogen 4-P (Total) 4-P (Ortho) Lorine uminum	mg/1 mg/1 mg/1 rg/1	<.28 .028 .028 2.0	<.28 .037 .037 7.0	<.28 .032 .032 0.5	<.28 .028 .022 2.0	<.28 .030 .024 1.0	<.28 .026 .022 <0.5	<.28 .040 .040 3.0	<.28 .086 .084 2.5	<.28 .032 .026 4.0	<.28 .032 .024 <0.5	<.28 .03: .03: 0.5
senic dmium romium pper on	ив/1 ив/1 ив/1 ив/1 ив/1	<6 <5 <5 200 10	<6 <5 <5 <5 <10	<6 <5 <5 <5 <10	<6 <5 <5 10 <10	< 6 < 5 < 5 < 5	.<6 <5 <5 330 <10	< 6 < 5 < 5 < 5 < 10	<6 <5 <5 <5 <10	<6 <5 <5 <5	< 6 < 5 < 5 50 10	<6 <5 <5 <5 <10
ad nganese rcury nc AS	AUS/1 AUS/1 AUS/1 AUS/1 mg/1	<10 <10 <.2 <5 <.05	< 10 < 10 < .2 30 < .05	<10 <10 <.2 <5 <.05	<10 <10 <.2 20 <.05	<10 <10 <.2 <5 <.05	30 <10 <.2 540 <.05	<10 <10 <.2 10 <.05	< 10. < 10 < .2 120 < .05	<10 <10 <.2 32 <.05	<10 <10 <.2 23 <.05	< 10 < 10 < .2 6 < .05
1 & Grease stal Coliform scal Coliform	mg/l #/100ml #/100ml	6.1 <1 <1	<1 <1	√1 <1	-4 <1	<1 <1		5.0 <1 <1	5.5 <1 <1	6.3 4 <1	 	<1 <1
SGS number.					•							
						~						<pre> <pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre></pre>
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GROUNDWATER SAMPLING PROGRAM

.20 .5 .5 .056 .002 .3	252 146 .20 8.2 11.0 136 <.056 <.002 1.5 <.28 .032 .026 4.0	P-13 25/45-4C* 259 165 .22 7.9 9.0 140 <.056 <.002 1.1 <.28 .032 .024 <0.5	116 	P-W 5/51-28%* 268 170 .23 7.95 10.0 144 <.056 <.002 1.3 <.28 .032	767 537 .73 7.75 11.0 228 <.056 .039 <.01	P-Y 25/42-1381* 353 233 .32 7.48 12.5 164 <.056 <.002 1.1	P-Z 26/43-16F* 300 165 .22 7.6 12.0 146 <.056 <.002	B-1 25/41-26H*		8-3 21/45-308*	0-1 28/43*	0-2	0-3 26/43-6C1*
. 20 . 5 . 5 . 056 . 002 . 3	252 146 .20 8.2 11.0 136 <.056 <.002 1.5 <.28 .032 .026 4.0	259 165 .22 7.9 9.0 140 <.056 <.002 1.1 <.28 .032 .024 <0.5	167 113 .15 7.7 9.0 116 <.056 <.002 0.17 <.28 .032	170 .23 7.95 10.0 144 <.056 <.002 1.3 <.28 .032	767 537 .73 7.75 11.0 228 <.056 .039 <.01	353 233 .32 7.48 12.5	300 165 .22 7.6 12.0			, <u>=</u>			
.056 .002 .3	146 .20 8.2 11.0 136 <.056 <.002 1.5 <.28 .032 .026 4.0 <6 <5	165 .22 7.9 9.0 140 <.056 <.002 1.1 <.28 .032 .024 <0.5	113 .15 7.7 9.0 116 <.056 <.002 0.17 <.28 .032	170 .23 7.95 10.0 144 <.056 <.002 1.3 <.28 .032	.73 7.75 11.0 228 <.056 .039 <.01	233 .32 7.48 12.5 164 <.056 <.002	165 .22 7.6 12.0			′ = • _			
	8.2 11.0 136 <.056 <.002 1.5 <.28 .032 .026 4.0	7.9 9.0 40 <.056 <.002 1.1 <.28 .032 .024 <0.5	7.7 9.0 116 <.056 <.002 0.17 <.28 .032 .032	7.95 10.0 144 <.056 <.002 1.3 <.28	7.75 11.0 228 <.056 .039 <.01	7.48 12.5 164 <.056 <.002	7.6 12.0 146 <.056			´ = ● _			
.056 .002 .3	11.0 136 <.056 <.002 1.5 <.28 .032 .026 4.0 <6 <.5 <.5	9.0 140 <.056 <.002 1.1 <.28 .032 .024 <0.5	9.0 116 <.056 <.002 0.17 <.28 .032 .032	10.0 144 <.056 <.002 1.3 <.28 .032	228 <.056 .039 <.01	12.5 164 <.056 <.002	12.0 146 <.056			•			
.056 .002 .3	<.056 <.002 1.5 <.28 .032 .026 4.0 <6 <55	<.056 <.002 1.1 <.28 .032 .024 <0.5	<.056 <.002 0.17 <.28 .032 .032	<.002 1.3 <.28 .032	<.019	<.002	146 <.056 <.002			•			
.056 .002 .3	<.056 <.002 1.5 <.28 .032 .026 4.0 <6 <55	<.056 <.002 1.1 <.28 .032 .024 <0.5	<.056 <.002 0.17 <.28 .032 .032	<.002 1.3 <.28 .032	<.019	<.002	146 <.056 <.002						
.002 .3 .28	<.002 1.5 <.28 .032 .026 4.0 <6 <5	<.002 1.1 <.28 .032 .024 <0.5	<.002 0.17 <.28 .032 .032	<.002 1.3 <.28 .032	<.019	<.002	<.030						
.3	1.5 <.28 .032 .026 4.0 <6 <5	1.1 <.28 .032 .024 <0.5	<.28 .032 .032	<.28 .032	<.01		~ · UU2						
.28 .086 .084 .5	.032 .026 4.0 <6 <5 <5	.032 .024 <0.5	.032 .032	.032			1.7						
.086 .084 .5	.026 4.0 <6 <5 <5	.024 <0.5	.032		1.12	<.28	<.28						
.5	4.0 <6 <5 <5	<0.5 		.030	.032 .024	.054 .038	.026 .024						
	<6 <5 <5			0.5	130	22.5	9.5						
	<5 <5	11			< 20		<20	~~					
	<5	< 6	<6	<6	<6	<6	<6				7-		
	· ·	< 5 < 5	<5 <5	<5 <5	< 5 10	<5 <5	<5 <5						
	< 5	50	< 5	<5	120	< 5	10	**					
	10	10	<10	<10	20	<10	< 10		***				
	< 10	<10	< 10	< 10 < 10	20 < 10	<10 .	< 10	,					
. 2	<10 <.2	<10 <.2	<10 <.2	<.2	<.2	<10 <.2	<10 <.2						
.05	32 <.05	23 <.05	6 <.05	<5 <.05	96 <.05	< 5 <.05	<5 <.05	₹.05	.07	<.0s	₹.05	<.os	₹.05
-		1.07				03		رن.	.07	7.03	\. 03	7.03	×.05
.5	6.3 4		<1		<u> </u>		5.5 <1	<u></u>	<1	<u></u>	2	<u></u>	<u></u>
	<i< td=""><td></td><td><1</td><td></td><td><1 <1</td><td>***</td><td><1 <1</td><td><1 <1</td><td><1 <1</td><td><1 <1</td><td><12</td><td><1 <1</td><td><1 <1</td></i<>		<1		<1 <1	***	<1 <1	<1 <1	<1 <1	<1 <1	<12	<1 <1	<1 <1
				,							TABLE GROUNDWATE	r sompling	PROGRAM
Sector 20 pt - market and		· ere obsamrativose Adiobliga	· · · Jt. productypopus space de			Salaran and an analysis and an	endenski Fidelija (S. Zagle Lasy, Propins	_				المارة والمارة	



TABLE 7

SUPPLARY OF GROUND WATER QUALITY IN THE STUDY AREA

		Ā	Primery Aqui	ier.	2		.	Little	bokane Basi	n Aquifer		ther Aquife	
Parameter	Units	Avg.	Std. Dev.	Samples	Avg.	Std. Dav.	Samples	Avg.	Std. Dev. Samples	Samples	Avg.	Std. Dev. S.	Samples
Conductivity	umpos (cm)	283	56	213	263		95	292	106	16	285	145	
Residue (180°C)	mg/1	176	54	5	177		99	•	ı	ı	•	ı	,
Residue	TON/AFT	. 24	.07	09	0.24		99	t	1	1	1	1	
нd		7.7	9.0	210	7.7		95	7.7	4.0	16	7.4	7.0	
Temp.	ပ္	10.7	1.7	18	14.0		26	1		, !	1	,	
D.0.	mg/1	8.1	1.1	ø	ı		1	1	1	1	ı	1	
Mardness (CaCO ₃)	mg/1	156	39	214	108		97	151	64	16	. 96	72	
NH1-N	mg/1	<.023	0.018	19	•		1	•	•	1	ŧ		
NO2-N	mg/1	. 602		19	1		1	•		. 1.	1	ı	
. N-EON	mg/1	1.506	1,161	62	1.686		96	1.418	1.624	16	1.04	1.504	
Kjeldahl Nitrogen(N)	mg/1	< 0.144		19	ı		1	ı	•	1	•		
POG-P (Total)	ng,'1	0.018		. 62	0.088		31	0.095	.0.082	16	0.11	0.272	
PO4-P (Ortho)	mg/1	0.014		19	0.070		7	ı	1	ı		,	
ប	mg/1	4.6		210	5.1		96	3.4	4.5	16	8.6	10.3	
A1	ug/1	ı	•	1	١,		1	1		1		1	
As		ĸη	O	19	ı,		1	•	ı	t	1	1	
છ		<1	7	19	1		ı	ı	1	ŧ	ı	,	
č.		~1	7	19	ı		1	1	1	ı	1	,	
		<15	67	19	1		1	ı	ı	t	1	•	
(Diss)		24	29	19	•		,	,	ı	1		1	•
Fe (Total)		169	356	146	1404		88	1004	100	16	19	167	19
(Diss)		47	•	55	1		1		ı	1		•	1
		19	œ	20	1		1		ı	ŧ	٠,	1	1
	u g /1	7	10	213	ı		1	ı	:	1	:		1
Hg (Total)	u8/1	н	•	73	ı		1	ı	ı		1	1	ı
	ug/1	34	73	62	ı			1	ı	1	t	1	1
	mg/1	0.03	0.03	57	~0.0		ო	< .05	0	e		,	1
	m8/1	5.7	9.0	4	•		1	ı	ŧ	•	•	•	1
Total Coliform	#/100 ml	7	н	o,	1		m	41	H	m	1	1	ı
	1/100 ml	41	0	o,	د 1		m	~1	0	m	•	1	t

* Data do not permit definition as to whether dissolved or total

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WATER QUALITY DATA PRIMARY AQUIFER

	_	We	11 Identific	ation by USG	S Number	
Parameter	Units	25/42-311-		25/42-138/		
Conductivity	umhos/cm	296	275	336	335	388
Residue (Total)	mg/l	179			0.16	029
Residue (180°C)	mg/1			179	188	210
Residue, Calculate						
Residue Loading	TON/AFT			0.34		
pН		8.2		7.4	7.7	7.9
Temperature	*C	8.2	11.5	12.4	12.0	12.0
Dissolved Oxygen	mg/1		5.9			
Hardness (CaCO3)	mg/l	145	134	160	150	170
Alkalinity (CaCO ₃)	mg/l					
NH3 -N	mg/1			0.01	0.01	0.01
NO ₂ -N	mg/1			0,002	0 001	0.002
NO3 -N	mg/1			1.7	1.4	
Kjeldahl-N	mg/1			0,12	0.16	007
PO ₄ -P (Total)	mg/1			0.06	0.06	0,10
PO ₄ -P (Ortho)	mg/1			0.05	0.06	.0.09
Chloride	mg/1	4	ļ	4	/2.6	24.0
Sulfate	mg/l	<u> </u>				
Fluoride	mg/l					
Aluminum	μg/1				•	
Arsenic	μg/1			0	6.0	2,0
Cadmium	ug/1			1,0	00	0
Chromium	g/1			0	00	0
Copper	ng/1)			2.0	4.0	3,0
Iron (Dis)	1/gبر			30	110	30
Iroa (Tot)	ug/1					
Lead (Dis)	1/gدر			0	40	0
Lead (Tot)	μg/1					
Manganese	μg/1	ļ	 	0	0	٥
Mercury	μg/1			0	O	0
Zinc	ر (gug/1			/0	20	30
MBAS	mg/1			660	0.03	0 02
011 & Grease	mg/1					
Total Coliform	#100 m1		1			
Fecal Coliform	#100 m1			-		
Other (Specify)						
				}		
Source		Storet -	s	4545- EPA .		
Sample Date		70-10-14	72-08-08	73-00-27	72. 19-16	クター 12 ー 17
Owner		city of		Spokane		
		Spokune Baxter#1		cold Storage		1.
		L	<u></u>	<u> </u>	<u> </u>	

PRIMARY AQUIFER

	-		11 Identific	ation by USG	S Number	
Parameter	Units	25/42-1381	25/43-116-1	<i>₽</i>	25/43-11-64	15,45-12 H
Conductivity	umhos/cm	410	315			296
Residue (Total)	mg/1	0,3/				164
Residue (180°C)	mg/1	23/				
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					-
рН		7.5				7.6
Temperature	°C	13.3	9.0	8,9		5-6
Dissolved Oxygen	mg/l		8,2	7.2		
Hardness (CaCO3)	mg/1	180	156			154
Alkalinity (CaCO ₃)		<u> </u>				
NH3 -N	mg/1	002				
NO ₂ -N	mg/1	0.003				
NO ₃ -N Kjeldahl-N	mg/1	1.6				
vlergaur-M	mg/l	0.25				
PO4 -P (Total)	mg/1	0.007				
PO4 -P (Ortho) Chloride	mg/1	0.006				2
Sulfate	mg/l mg/i					
Fluoride	mg/1					
tinorine	_					
Aluminum	րւց/1	<u></u>				
Arsenic	μg/1	//_				
Cadmium	/ug/1					
Chromium	1/guر	0				
Copper	μg/1	8	<u> </u>		<u> </u>	1
Iron (Dis)	1/وبر	15				
Iron (Tot)	μg/l		5-	20	20	0
Lead (Dis)	1/ ₈ גע	1_				
Lead (Tot)	μg/1			30	25	
Manganese	1/ ₈ بر	140		40	2.0	0
Mercury	ug/1	0	1.3	0.2 K	Q.2K	
Zinc	ير/gن <i>ر</i>	30				
MBAS	mg/l	<u> </u>				<u> </u>
Oil & Grease	mg/l	 	· · · · · ·			
Total Coliform	#100 ml					
Fecal Coliform	#100 m1					
Other (Specify)						
Source	·····	USGS-EFA	Storet -			
Sample Date		74-03-20	72-02 2	13-11-15	73-01-15	71-04-01
Owner		SMOKHNI CO D	1 - 10 50 80 -		57 78 811	ORCHALD AUG
		STUPP 13	121	F	No = 2	+/
		<u></u>		<u> </u>	<u> </u>	<u> </u>

WATER QUALITY DATA

<u> </u>	CINIAKY A	<u> </u>	HFER	,				
	Wei	11	Identif	ication	by	USGS	Number	
_								_

Residue (Total)	Units	25/43-1241	<u>-</u>			
Residue (Total)		-3/73 (27/				
Residue (Total)	mhos/cm	360	300	300	264	292
*	mg/1	221	145	306	140	172
Residue (180°C)	mg/1					
Residue, Calculated	mg/l					
Residue Loading T	on/aft					
		7.6	7.6	7./	-, -	. 76
pH Temperature	•c	7.6			7.5	75
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	204	148	3/6	1/2	164
Alkalinity (CaCO3)	mg/l		•			······································
NII3 -N	mg/1					
NO ₂ -N	mg/1					
NO3 -N	mg/1					
Kjeldahl-N	mg/1					
PO4 -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	4	.3	9	2	a
Sulfate	mg/1			•		
Fluoride	mg/l					*****************
	. 14					
Aluminum	μg/1	}				~~···
Arsenic Cadmium	иg/1					·
Cadmium Chromium	ug/1 ug/1					
Copper	1/وبر 1/وبر					
cobber	/4B/ 1					
Iron (Dis)	ng/1					·
Iron (Tot)	$\mu g/1$	200	60	20	40	180
Lead (Dis)	1/g <i>در</i>				,	
Lead (Tot)	μg/1					
Manganese	ug/1	9	9	9	0	0
Mercury	μg/1					
Zinc	ug/1					
MBAS	mg/1					
Oil & Grease	mg/l					
Total Coliform #	100 ml					
	100 ml					
Other (Specify)						
Source		Storet				
Sample Date		71-09-17	71-10-14	71-11-15	71-12-13	72-01-18
Owner		ORCHED AVE #				

PRIMARY AQUIFER

			Il Identific	ation by USG	S Number	
Parameter	Units	25/43-1241				
Conductivity	umhos/cm	3/2	300	326	310	308
Residue (Total)	mg/1	179	172	211	194	304
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pH	-	8.1	7.7	8.0	8.3	8.4
Temperature	°C				10.0	
Dissolved Oxygen	mg/1				, , , , , , , , , , , , , , , , , , ,	•
Hardness (CaCO3)	mg/1	176	148	188	154	194
Alkalinity (CaCO3)	mg/l					
NII3 -N	mg/1					
NO ₂ -N	mg/1					
$NO_3^ N$	mg/l	·				
Kjeldahl-N	mg/1					
PO ₄ -P (Total)	mg/1					
PO ₄ -P (Ortho)	mg/1					•
Chloride	mg/l	3	3	6	á	71
Sulfate	mg/l					
Fluoride	mg/l					
Aluminum	μg/1					
Arsenic	g/1					
Cadmium	ug/1					
Chromium	ົມg/1					
Copper	μg/1					
Iron (Dis)	յսց/1					
Iron (Tot)	μg/1	80	160	150	120	90
Lead (Dis)	1/gد <i>ر</i>					
Lead (Tot)	μg/1					
Manganese	/ug/1	3	6	6	0	300
Mercury	μg/1					
Zinc	μg/1					
MBAS	mg/1					
Oil & Grease	mg/l					
Total Coliform	#100 m1					
Fecal Coliform	#100 m1					
Other (Specify)						
Source		Storet				-
Sample Date		 	72-03-21	70 - 215	72-05-22	75-06-12
Owner		ORCHARD FIUE 21 ,			72 03 44	>

APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

11711		
Well Identificat	ion by HS	GS Number

		We.	11 Identific	ation by USG	S Number	
Parameter	Units	25/43-1211				
Conductivity	umhos/cm	330	300	320	3/0	
Residue (Total)	mg/1	(83	158	248	174	
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pН	-	7.7	7.8	7.9	7.4	
Temperature	*C					····
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	2.04	156	324	156	
Alkalinity (CaCO ₃)	_				··	
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO3 -N Kjeldahl-N	mg/1	ļi				
vlergaur-w	mg/l					
PO4 -P (Total)	mg/J.					
PO ₄ -P (Ortho)	mg/1					•
Chloride	mg/1	10	4	4	2	
Sulfate	mg/l					
Fluoride	mg/l					· · · · · · · · · · · · · · · · · · ·
Aluminum	μg/1					
Arsenic	µg/1					···
Cadmium	μg/1					
Chromium	ر 1/عبر					
Copper	µg/1					
Iron (Dis)	μg/1					
Iron (Tot)	/Jug/1	300	20	80	LOOK	
Lead (Dis)	/1g/1				30	30
Lead (Tot)	μg/1					
Manganese	μg/1	6	18	6	2	2
Mercu v	μg/1				18,8	0,2
Zinc	ug/1					
MBAS	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 m1	ł				
Fecal Coliform	#100 ml					
Other (Specify)						
6		6100 +		y pina gair lings to the state of the state		
Source Sample Date		Storet -				
		72-07-17	72-08-14	72-09-14	72-09-14	13-01-15
Owner		ORCHARD				
		HUE	l	ļ		
				l		l ,
		#,		1		
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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

MALINAN AUG				
We11	Identification	hv	uses	Humber

		we,	11 Identific	ation by USG	SHumber	
Parameter	Units	25/43-13NI				
Conductivity	umhos/cm	270	265	261	268	
Residue (Total)	mg/1	0.21	0,23	6.19	0.2	
Residue (180°C)	mg/1	154	170	140	149	
Residue, Calculate	$1 \frac{\sqrt{1}}{mg/1}$	1.2.7				
Residue Loading	TON/AFT					
		,,,,,		24	~ 0	
pH	•c	7.7	8,2 170	8,2 /8.5	7.9	
Temperature Dissolved Oxygen	n.2/1	10.0	770	/0.5	70'3	•
Hardness (CaCO3)	mg/1	140	/30	130	140	· · · · · · · · · · · · · · · · · · ·
Alkalinity (CaCO3)	mg/1					
NH3 -N	mg/1	0.02	0.02	0.01	0.03	
NO2 -N	mg/1	0	0.003	0.003	0.001	
NO ₃ -N	mg/1	0.84	1.00	1000	0.90	
Kjeldahl-N	mg/l	C.05	0.04	0.03	0,84	
PO4 -P (Total)	mg/1	0,011	0.006	0.031	0.007	
PO4 -P (Ortho)	mg/1	0.007		0.025	0.005	•
Chloride	mg/1	1.7	2.1	1.4	2.2	
Sulfate	mg/1					·
Fluoride	mg/l					
Aluminum	<u>1/عر</u>					
Arsenic	μg/1	29	5	2	2	·····
Cadmium	μg/1	1.0	Ü	0	0	
Chromium	μg/1	0	0	0	0	
Copper	μg/1	20	30	50	50	
7 (D/-)	/1	130	10	150	60	
Iron (Dis) Iron (Tot)	μg/1 μg/1	130	70			~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
Lead (Dis)	1/عمر 1/عدر	3	3	4	17	
Lead (Tot)	/1g/1					***************************************
Manganese	μg/1	0	10	30	. 20	
	•					
Mercury	μg/1	0,1	0	0	0	
Zinc	/18/1	50	0.01	90	70	
MBAS Oil & Grease	mg/1 mg/1	0	3.01	0	ؿ	
	_					
Total Coliform	#100 ml					
Fecal Coliform	#100 m1					
Other (Specify)						
		}				
Source		USGS-EPA-				
Sample Date		73-06-29	73-09-25	73-12-17	74-0322	_
Owner		wwp				
		#1-3				

APPENDIX T WATER QUALITY DATA PRIMARY AQUIFER Wall Identifi

Parameter	Units	25/43-14KI				15/43-23A1
Conductivity	umhos/cm	235	236	238	235	284
Residue (Total)	mg/1	0.17	0.21	0.19	0.18	167
Residue (180°C)	mg/1	126	152	136	130	
Residue, Całculate	d mg/1					
Residue Loading	TON/AFT					
рH		7.6	8.0	8.1	7.8	8.3
l'emperature :	*C	11,6	11.5	110	10.2	7.3
Dissolvel Oxygen	mg/l					
Hardness (CaCO3)	mg/1	110	120	120	120	120
Alkalinity (CaCO ₃)		 				<u></u>
nh3 -n	mg/1	0.01	0.01	0.01	0.02	
NO2 -N	mg/1	0.001	0.002	0.002	0 002	
NO ₃ -N	mg/1	1./	1.1	09	1,1	
Kjeldahl-N	mg/l	0.05	0.04	0.02	0.12	
PO4 -P (Total)	mg/1	0.012	0.008	0.010	0.005	
PO ₄ -P (Ortho)	mg/1	0.003	0.005	0.004	0.003	
Chloride	mg/l	1.5	2.1	1.8	2,3	5
Sulfate	mg/1					
Fluoride	mg/1	l				
Aluminum	μg/1					
Arsenic	µg/1	0	0		<u>o</u>	
Cadmium	ug/1	1.0	0		0	
Chromium	μg/ <u>1</u>	0	0			
Copper	дg/1	6.0	4.0	3.0	<u> </u>	
Iron (Dis)	μg/1	50	30	0	20	
Iron (Tot)	ug/1					430
Lead (Dis)	/1g/1	4	3			
Lead (Tot)	лg/1				· · · · · · · · · · · · · · · · · · ·	
Manganese	μg/1		0	0	14	23
Mercury	Mg/1	0.1	0	0	٥	
Źinc	μg/1	10	40	0	20	2
MBAS Oil & Grease	mg/1 mg/1	0	0	6.03		
Total Coliform Fecal Coliform	#100 ml #100 ml					
Other (Specify)						
Source		USGS-EPA-			•	Storet
Sample Date			73-09-25	73-12-17	74-03-20	1
Owner		ACME -				wwP
		CUNCRETE				#1-5A
		-				7 1-57
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Michigan St. Aithe

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WATER QUALITY DATA

PRIMARY AQUIFER

Vall Identifi

25/45.25.46 3 3 3 3 3 3 3 3 3	arameter	Units		ll Identific	ation by USG	S Number	
Idua (Total) mg/1 Idv							
Idue (180°C) mg/1			234	3"0			300
Automate Color C	ue (180°C)	mg/1	167	i'-4		225	175
erature of colved Oxygen mg/1 mess (CaCO3) mg/1 mess (CaCO3) mg/1 mess (CaCO3) mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1							
Dived Oxygen mg/1 mess (CaCO3) mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1			7.5	7,9	7.7	7.6	6.9
-N mg/1 -N mg/1 -N mg/1 -N mg/1 -N mg/1 -N mg/1 -P (Total) mg/	ess (CaCO3)	mg/l	156	144	160	220	154
-N mg/1 dahl-N mg/1 -P (Total) mg/1 -P (Ortho) mg/1 ride mg/1 stee mg/1 ride mg/1 ride mg/1 ride mg/1 minum	7	mg/l					
Ash	1 1						
Property	ahl-N						
ride mg/1							
ride mg/1 inum	lde		5	6	4	5	ク
inum							
1	rae	mg/l					
ium							
(Dis)							· .
(Dis)							<u> </u>
(Tot)							
(Tot)		μg/1					
(Tot)		ug/1	140	160	60	140	320
		J1g/1					
Grease mg/1 Coliform #100 ml Coliform #100 ml (Specify) Storet 71-07-21 71-09-16 71-10-13 71-11-15 71-15 WWP #1-54			15	15	2	6	9
Grease mg/1 Coliform #100 ml Coliform #100 ml (Specify) ce	ry.	μg/1					
Grease mg/1		Jug/l					
(Specify) Stored	Grease	mg/1					
Stored							٠.
1e Date 71-07-21 71-09-16 71-10-13 71-11-15 71-12 x wwP #1-5A	(Specify)						
1e Date 71-07-21 71-09-16 71-10-13 71-11-15 71-12 r	,,		Staret -				
# 1-5A	Date			71-09-16	71-10-13	71-11-15	71-12-13
#1-5A	······································		WWP				1
				·			
405-22			,				
405-22							
· ·				405-22			
				`			
					-AP-With Company		

APPENDIX Z WATER QUALITY DATA

我们就是我们的说话,我就说我了我说话,我们就是我们是我的人的,我们就是我们的人的,我们也没有一个一个,我们也没有一个一个,我们也会一个一个,我们也是我们的人,也是

PRIMARY	AQU	IF ER				
	Vol 1	Identiff	cation	hv	11969	Mumba

and the second s

	Well Identification by USGS Number						
Parameter Units	25/43-2314						
Conductivity umhos/cm	320	330	306	3/4	310		
Residue (Total) mg/l	249	169	179	210	181		
Residue (180°C) mg/1							
Residue, Calculated mg/l			,				
Residue Loading TON/AFT							
,	7.8	5.0	8.0	8.1	7.9		
pH C	7.5	8.0	8.0	· · · · ·	12.2		
Temperature °C Dissolved Oxygen mg/1			·		1212		
Hardness (CaCO3) mg/1	200	128	176	208	166		
Alkalinity (CaCO ₃) mg/l							
NH3 -N mg/1							
$NO_2 - N mg/1$							
$NO_3 - N$ mg/1							
Kjeldahl-N mg/l							
PO ₄ -P (Total) mg/1							
PO ₄ -P (Ortho) mg/1							
Chloride mg/1	5	7	5				
Sulfate mg/1							
Fluoride mg/l							
Aluminum µg/1							
Arsenic µg/1							
Cadmium Aug/1							
Chronium ug/1							
Copper µg/1							
Iron (Dis) µg/1							
Iron (Tot) µg/1	440	160	160	0	40		
Lead (Dis) µg/1							
Lead (Tot) ug/1	//		0		6		
Manganese µg/1	/2	6		<u>o</u> _			
Mercury µg/1							
Zinc ug/1							
MBAS mg/1							
Oil & Grease mg/1							
Total Coliform #100 ml	1	}			٠,		
Fecal Coliform #100 ml							
Other (Specify)							
e .	. 1						
Source Sample Date	Stovet -						
	72-01-18	72-02-14	72-03-03	72-04-18	72-05-11		
Owner	WWP						
	#1-54						
	, , , , ,				Í		
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APPENDIX I WATER QUALITY DATA

/	PRIMARY AQUIFER			
	U-11 Y1	L	71000	31

	Well Identification by USGS Number							
Parameter	Unite	25/43-33A1				 ₽.		
Conductivity	umhos/cm	300	300	198	280	310		
Residue (Total)	mg/l	179	173	202	191	218		
Residue (180°C)	mg/1							
Residue, Calculate	d mg/1			,				
Residue Loading	TON/AFT							
pH		8,3	7.6	8.0	8.0	7.5		
Temperature :	°C							
Dissolved Oxygen	mg/1							
Hardness (CaCO3)	mg/1	160	164	188	152	158		
Alkalinity (CaCO ₃)								
NH3 -N	mg/1					· ····································		
NO ₂ -N	mg/1							
NO ₃ -N	mg/1							
Kjeldahl-N	ng/1	 				······································		
_	_					**************************************		
PO4 -P (Total)	mg/1							
PO4 -P (Ortho)	mg/l							
Chloride	mg/1	7	11	5	4	4		
Sulfate	mg/l							
Fluoride	mg/1							
Aluminum	աց/1							
Arsenic	μg/1							
Cadmium	ug/1							
Chromium	μg/1							
Copper	\ng'.⊤							
						!		
Iron (Dis)	μg/1				******	-		
Iron (Tot)	//g/1	210	740	120	700	100K		
Lead (Dis)	1/g <i>ار</i>					40		
Lead (Tot)	Jug/1							
Manganese	Jug/1	3	6	12	15	3		
Mercury	μg/1					6.3		
Źinc	ug/1							
MBAS	mg/1					^		
Oil & Grease	mg/1							
Total Coliform	#100 ml					٠.		
Fecal Coliform	#100 ml							
Other (Specify)						and the same of th		
6		1 e . L				, Non		
Source Sample Date								
		72.06-16	72-07-24	72-08-25	72-05-14	72.09-14		
Owner		wwp						
		# 1-519			معاسب سيدريندو وتحادثان سيدرين سيدرين			
		1-514						
			: !					
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APPENDIX T WATER QUALITY DATA PRIMARY AQUIFER Well Identi

		We	11 Identifica	ation by USGS	Number	
Parameter	Unita	25/43-23A1	25/43-23A2		•	
Conductivity	umhos/cm		292	400	345	310
Residue (Total)	mg/1		176	178	179	206
Residue (180°C)	mg/1					
Residue, Calculate	$\frac{mg}{1}$					
Residue Loading	TON/AFT					
	2011, 122					
pH			8.1	7.7	8.0	7.2
Temperature	•G					
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1		152	148	180	208
Alkalinity (CaCO3)	mg/1					
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO3 -N	mg/1					
Kjeldahl-N	mg/1					
_	_					
PO4 ~P (Total)	mg/1					
PO4 -P (Ortho)	mg/1	<u> </u>				
Chloride	mg/1		8	7	5-	5
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	/1					
	μg/1	<u> </u>				
Arsenic	μg/1				····	
Cadmium	дg/1					
Chromium	лg/1					
Copper	μg/1					
Iron (Dis)	g/1					
Iron (Tot)	/ug/1.	30				
Lead (Dis)	μ g/1	15				
Lead (Tot)	/ug/1		0	0	. 300	10
Manganese	μg/1	2	6	6	6	6
nen2eneee	J-101 +	£				
Mercury	µg/1	0.2K				
Źinc	ug/1					
MBAS	mg/1		1			
Oil & Grease	mg/1		<u> </u>			
Total Coliform	#100 ml	-			,	٠,
Fecal Coliform	#100 ml					
Other (Specify)						
Source		Storet -				-
Sample Date			 			
oumbre nare		73-01-15	71-07-21	71-09-16	71-10-13	71-11-15
Owner		WWP	wwp			
		}	# 1-58	l		
		#1-54	1 -1-20			
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APPENDIX I WATER QUALITY DATA

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The same of the same	77 A A	Well Identification by USGS Number							
Parameter	Units	25/43-23A2		···					
Conductivity	umhos/cm	290	340	340	280	292			
Residue (Total)	mg/1	174	122	218	181	192			
Residue (180°C)	mg/1								
Residue, Calculated	1 mg/1								
Residue Loading	TON/AFT					······································			
рН		7.7	7.1	7,9	8.2	7.9			
Temperature	°C	· /· /	///		012				
Dissolved Oxygen	mg/1					•			
Hardness (CaCO3)	mg/1	156	200	2/2	148	156			
Alkalinity (CaCO ₃)	mg/1								
NH3 -N	mg/1								
NO2 -N	mg/1								
NO3 -N	mg/1								
Kjeldahl-N	mg/1								
PO4 -P (Total)	mg/1								
PO ₄ -P (Ortho)	mg/1	· · · · · · · · · · · · · · · · · · ·							
Chloride	mg/1	4	9	6	5	5			
Sulfate	mg/1								
Fluoride	mg/l								
Aluminum	μg/1								
Arsenic	μg/1								
Cadmium	ug/1								
Chromium	ر 1/gرر								
Copper	ug/1								
Iron (Dis)	/1								
Iron (Tot)	μg/1 μg/1	280	60	•	120				
Lead (Dis)		280	60	40	7.20	60			
Lead (Dis))1g/1								
Manganese	/1g/1 /1g/1	6	0	6	6	6			
	J481 -								
Mercury	$\mu g/1$								
Zinc	μg/1								
MBAS	mg/1					<u> </u>			
Oil & Grease	mg/l								
Total Coliform	#100 ml								
Fecal Coliform	#100 ml								
Other (Specify)				,					
Source		Stoyet -				i>			
Sample Date		71-12-13	72-01-18	72-02-14	72-03-31	72-04-18			
Owner	***************************************			 					
		WWP #1-5B							
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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

77777						
We	11 1	Aanti	fication	hv	HSGS	Number

		Well Identification by USGS Number						
Parameter	Units	25/43-23A2						
Conductivity	umhos/cm	310	280	300	288	300		
Residue (Total)	mg/1	181	165	190	218	229		
Residue (180°C)	mg/1							
Residue, Calculated	i mg/1							
Residue Loading	TON/AFT							
-11		7.9	7.9	7.9	7.9	8.2		
pH	•c	7.7	7,9	/, 7	7,7	8.2		
Temperature Dissolved Oxygen						•		
Hardness (CaCO3)	mg/1 mg/1		154	180	204	204		
Alkalinity (CaCO ₃)	mg/l	165	734	700	204	204		
•	_							
NH ₃ -N NO ₂ -N	mg/1 mg/1							
NO ₃ -N	mg/l							
Kjeldahl-N	mg/1				**			
PO ₄ -P (Total)	mg/1							
PO ₄ -P (Ortho)	mg/l					•		
Chloride	mg/l	, 6	8	//	7	8		
Sulfate	mg/l							
Fluoride	mg/l							
Aluminum	μg/1							
Arsenic	µg/1					•		
Cadmium	Jug/1							
Chromium	μg/1							
Copper	µg/1				-			
Iron (Dis) .	μg/ <u>1</u>							
Iron (Tot)	Jug/1	220	220	100	100	30		
Lead (Dis)	μg/1							
Lead (Tot)	μg/1							
Manganese	μg/1	4	0	6	16	6		
Mercury	ug/1							
Zinc	ug/i							
MBAS -	mg/1							
011 & Grease	mg/1							
m 1 . 0 . 1 . C	#1.001							
Total Coliform	#100 m1	<u> </u>	 					
Fecal Coliform	#100 m1							
Other (Specify)								
Source		Storet -						
Sample Date		72-05-11	72-06-16	72-07-24	72-08-23	72-08-14		
Owner		wwP						
		1		Ì				
		#1-5B				-		
			}					
		L	<u> </u>	<u> </u>	<u> </u>	L		

APPENDIX I WATER QUALITY DATA

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		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/43-2312	——→	25/43-2461		
Conductivity	umhos/cm	310		400	380	368
Residue (Total)	mg/1			226	222	181
Residue (180°C)	mg/1					
Residue, Calculate				181		
Residue Loading	TON/AFT					
pН		7.4		7.7	7.8	7.6
Temperature	*C			5.0		
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	156		190	196	200
Alkalinity (CaCO ₃)	mg/1		·			
NH3 -N	mg/1		·			
NO ₂ -N	mg/1					
NO ₃ -N Kjeldahl-N	mg/1	i				
vleidaui-N	mg/l					
PO ₄ -P (Total)	mg/1					
PO ₄ -P (Ortho)	mg/1					•
Chloride	mg/1	4		4	6	Z
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	μg/1					
Arsenic	$\mu g/1$					
Cadmium	/ug/l					
Chromium	μg/1					
Copper	дg/1					
Iron (Dis)	μg/1	JUUK	<u> </u>			
Iron (Tot)	$\mu g/1$	15		. 0	0	0
Lead (Dis)	1/g <i>در</i>					
Lead (Tot)	Mg/1					
Manganese	Jug/1	2 K		0	9	/2
Mercury	Mg/1	4.3	0.2 K			
Zinc	Jug/1		·····			
MBAS	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 m1	į į		į		
Fecal Coliform	#100 ml					
Other (Specify)						
Source		Stovet -			or - a party of the surgery .	
Sample Date		72-09-14	75-01-15	70-12-21	71-10-18	71-11-15
Owner		WWP		1		
			_	E. SPOKANE		
		#1.5B	-	1		•
				111		}
		[i				
		# 1-58		1 1		

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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

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		We:	11 Identific	ation by USG	S Number	
Parameter	Units	25/43-24GI				
Conductivity	umhos/cm	420	405	380	380	410
Residue (Total)	mg/1	2/2	219	248	235	248
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pH	-	7.8	7,8	8,0	8.0	8.0
Temperature	*C					
Dissolved Oxygen	mg/1					•
Hardness (CaCO3)	mg/l	224	240	536	2/2	240
Alkalinity (CaCO3)	mg/l					
NH3 -N	mg/1					
NO2 -N	mg/1					
NO3 -N	mg/1					
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1					•
Chloride	mg/l	13	9	6	7	
Sulfate	mg/1					
Fluoride	mg/1					
Aluminum ·	աց/1					
Arsenic	μg/1					
Cadmium	ug/1					
Chromium	ug/1					
Copper	µg/1					
Iron (Dis)	յսց/1					•
Iron (Tot)	µg/1	340	240	380	460	0
Lead (Dis))1g/1					
Lead (Tot)	ug/1	·				
Manganese	μg/1	9	٥	6	3	0
Mercury	ug/1	!				
Zinc	ug/1					
MBAS	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify)						
Source		Storet	والمحسودة والمحور الي		-	-
Sample Date		71-12-13	73-01-18	72-02-14	73-03-31	72-04-18
Owner		1, 12, 13			/	
		E. SPOKANE		1		
		1		 		
		# /		1		
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APPENDIX I WATER QUALITY DATA

PRIMART	400	JIPER			
	17. 11	T1464	 L	HOOC	31

Parameter Un:	its 15/43-14G1	11 Identific			
nductivity umhos		400	420	380	360
idue (Total) mg		285	217	240	246
sidue (180°C) mg	/1				
idue, Calculated mg					
rane mountile TON/VI				-	
	8.3	7.3	7.8	7.8	7.7
perature °(
solved Oxygen mg, dness (CaCO3) mg,		260	200	2/6	192
alinity (CaCO ₃) mg/		200		2/0	172
-N mg.	/1				
-N mg,	/1				
-N mg/ lahl-N mg/					
gur_n mg	*				
P (Total) mg					
P (Ortho) mg					
ride mg,		//		8	7
te mg/					
щу	-				
inum jug,					
nic "ug,	/1				
nium jug,	/1				
omium "ng, per "ng,	/1				
μ6/					
n (Dis) ug,					
n (Tot) jug	/1 40	360	. 0	300	140
l (Dis) jıg, l (Tot) jug,	/1				
ganese jug		0	3	6	9
·					
cury ug	/1				
s mg/	/1				
S mg, & Grease mg,					
-					
al Coliform #100 m al Coliform #100 m					
er (Specify)					
er (shecrry)					
	. ,				1
4456,	storet -				
ole Date	72-05-11	12.06-12	72-07-7	72-08-14	72-09-14
er					
	E. SPOKANE	<u>.</u>			
	# /				
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APPENDIX T WATER QUALITY DATA PRIMARY AQUIFER

Well	Identific	ation by	USGS	Number

Warran A	97 - / A	we	TT TGENTILIC	ation by USG	5 Number	
Parameter	Units	25/43-2461	25/44-131			
Conductivity	umhos/cm	410	284	197	325	293
Residue (Total)	mg/1	236				
Residue (180°C)	mg/1		160	160	177	160
Residue, Calculated	1 mg/1					
Residue Loading	TON/AFT		0,12	0.22	0.24	0.22
•						
рН		7.3	7.7	7.9	7.9	7.6
Temperature	*C		10,6	10.5	10.0	9.6
Dissolved Oxygen	mg/1				<u></u>	
Hardness (CaCO3)	mg/1	199	140	150	160	140
Alkalinity (CaCO ₃)	mg/l					
NH3 -N	mg/1		0.01	0.01	0.01	0.03
NO ₂ -N	mg/1		0.001	0.003	0.002	0.001
NO ₃ -N	mg/1		0.81	0,71	2.10	0.82
Kjeldahl-N	mg/l		0.05	0.03	C 08	0.40
PO ₄ -P (Total)	mg/1		0.010	0 008	0.003	0.005
PO4 -P (Orthe)	mg/1		0.002	0007	0.008	0.003
Chloride	mg/l	6	1.5	1.1	1.4	0.9
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	μg/1					
Arsenic	ug/1		3	4	4	3
Cadmium	ug/1		0	0	0	0
Chromium	<u>1/8لار</u>		0	0	0	0 5
Copper	μg/1		10	16	2	5
Iron (Dis)	μg/1		60			
Iron (Tot)	ug/1	100				
Lead (Dis)	1/gرر					
Lead (Tot)	ug/1	15				
Manganese	μg/1	3				
Mercury	/1g/1	6.5				
Zinc	1/g <i>بر</i>					
MBAS	mg/1		ن	0.06	0.06	0.01
Oil & Grease	mg/l					
Total Coliform	#100 m1					
Fecal Coliform	#100 ml					
Other (Specify)				•		
orner (spectry)						
Source		Storet	USGS- EPA-		and the second s	
Sample Date				H7-100 A-	M 9 . 1	711 4= 0
		72-09-14	73-06-27	13-07-25	73-12-17	74-03-20
Owner		E. SPOKANE	SPOKANE			
		1	INO.PK			
		# /	ł .			-
			#2	·		
		1				
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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

خسنداللمشك تدليليك					
	Well	Identification	bν	USGS	Number

Bananahan	II and the second	We:				
Parameter	Units	25/44-291			-	25/44-761
Conductivity	umhos/cm	336	320	357	590	309
Residue (Total)	mg/l					
Residue (180°C)	mg/1	175	174	197	329	169
Residue, Calculated						
Residue Loading	TON/AFT	024	2.24	027	0.45	0.23
рН		7.6	7.9	7.9	7.5	7.6
Temperature .	°C	10.2	9.5	9.5	9,50	9.6
Dissolved Oxygen	mg/1	10.2				1.6
Hardness (CaCO3)	mg/1	150	150	170	190	160
Alkalinity (CaCU3)	mg/1				170	
~ ~	_				~ · · · ·	0 12
ин3 -и	mg/1	0.01	0.01	0.02	0.69	
NO ₂ -N	mg/1		0002	0.003	0.013	
NO3 -N	mg/1	2.0	7.4	3, /	5.4	0.84
Kjeldahl-N	mg/l	0.08	0.05	0.06	0.91	0.04
PO4 -P (Total)	mg/l	0.008	0.004	0.070	0,004	0.009
PO ₄ -P (Ortho)	mg/l	0.003	0.004	0.018	0.003	0.007
Chloride	mg/1	4.4	3.0	7.3	60	
Sulfate	mg/1					
Fluoride	mg/1					
Aluminum	րg/1					
Arsenic		· · · · · · · · · · · · · · · · · · ·	6	3	2	
Cadmium	μg/1	4,	0		6	6
	/1g/1	' ^		<u> </u>		
Chromium	лg/1	0	0	0	30	5
Copper	μg/1	9			4.4	1
Iron (Dis)	μg/1	40	20	60	10	30
Iron (Tot)	ug/1					
Lead (Dis)	1/פונ	3	5-	0	/	0
Lead (Tot)	ug/1					
Manganes e	µg/1	0		U		0
Mercury	μg/1	0	0.1	0	0	0.1
Żinc	μg/1 μg/1	360	60	80	90	10
MBAS .	mg/1	0.03	0.05		0.10	10
Oil & Grease	mg/1	J. 53	0,00	<u> </u>	<u> </u>	
Maka 1 - 0 - 115 -	_					
	#100 m1 #100 m1	}				
	, 200 mg					
Other (Specify)						<u> </u>
Source		4545-EPA-				
Sample Date	*		13-00-25	73-12-18	73-03-20	73-06-27
Owner		KAISER		13 .7-16	.,,	ORCHARD
		(TRENTWOOD)	.		_	AUE
		EAST GATE				#2
		LAN GALL				1

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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER

Well	Identification	bv	USGS	Number

	ation by USC	on by USGS Number				
Parameter	Units	25/44-761			25/44-1802	
Conductivity Kesidue (Total)	umhos/cm	306	3/5	32/	282	278
Residue (180°C)	mg/l mg/l	206		170	7/	
Residue, Calculate	d mg/1	700	175	172	150	193
Residue Loading	TON/AFT	0.28	0.24	0.23	0.02	0.06
pH		8./	8.0	7.7	7.6	. 8.1
Temperature .	*C	9.0	8.0	9.6	10,6	10.5
Dissolved Oxygen	mg/1					- AZ-Z-
Hardness (CaCO3)	mg/l	150	160	160	140	140
Alkalinity (CaCO ₃)	mg/1					
NH3 -N	mg/1	0.02	0.01	0,01	0.01	0.01
NO2 -N	mg/1	0.002	0.003	0,001	0	0.002
NQ3 -N	mg/1	0176	. 1.00	/3	112	1.1
Kjeldahl-N	mg/l	0.02	0.10	0,04	0.03	0.07
PO ₄ -P (Total)	'mg/1	0.011	0.016	0.014	0.010	0.011
PO ₄ -P (Ortho)	mg/1	01011	0.012	0.008	0.005	0.006
Chloride	mg/l	2.1	1.9	2.2	210	2.5
Sulfate	mg/l					
Fluoride	mg/l					
Aluminum	μg/1				·	
Arsenic	ug/1	/	6	3	2	2
Cadmium	ug/1	0	/	0	0	0
Chromium .	μg/1	0		0	0	0
Copper	μg/1	2	3	3	6	
Iron (Dis)	_{1/gل}	10	30	20	40	20
Iron (Tot)	/ug/1					
Lead (Dis)	g/1,		0	/		
Lead (Tot)	ng/1					
Manganese	μg/1	/0	0		10	0
Mercury	μg/1	0	0	O.	0	0
Żinc	ug/1	20	0	30	10	40
MBAS ·	mg/1	0	0,03	0	0	0.08
Oil & Grease	mg/l					
Total Coliform	#100 m1		`			
Fecal Coliform	#100 m1					
Other (Specify)						
Source		USGS-EPA-				>
Sample Date			73-12-18	73-04-19	73- 04-27	73-09-15
Owner		ORCHARIS	, - , - , 0	70 04 7		
		AUE			wu	
		#2		-	#1-4	
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APPENDIX T WATER QUALITY DATA PRIMARY AQUIFER Well Identifi

	· · · · · · · · · · · · · · · · · · ·	We	11 Identific	ation by USG	S Number	
Parameter	Units	25/44-1802	25/44-1901	-		_
Conductivity	umhos/cm	270	369	390	373	394
Residue (Total)	mg/1					
Residue (180°C)	mg/1	147	2//	206	207	226
Residue, Calculate						
Residue Loading	TON/AFT	0.20	0.29	0.28	0.28	0,31
pН		8.0	7.5	7.7	7.8	7.7
Temperature .	°C	11.0	11.9	11.5	11.5	11.2
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	130	190	180	180	190
Alkalinity (CaCO3)	mg/l					
NH3 -N	mg/1	001	0.04	0.01	0.01	0.01
NO ₂ -N	mg/1	0.003	0	0.001	0.002	0,001
NO3 -N	mg/l	1.1	2.8	3.0	2.6	2.2
Kjeldahl-N	mg/l	0.06	0.07	0.23	0.07	0.13
PO4 -P (Total)	mg/l	0.010	0.023	0,025	0.026	0.026
PO4 -P (Ortho)	mg/1	0,007		0.009	0.023	0.018
Chloride	mg/1	1.8	6.0	11.0	6.0	7.1
Sulfate	mg/l		3			
Fluoride	mg/l					
Aluminum	<i>յ</i> սց/1					
Arsenic	μg/1 μg/1		- 41	6	6	4
Cadmium	/ug/1	4	7	0	0	0
Chromium	μg/1	0	0	0		0
Copper	μg/1	3	8	2	<u>0</u>	8
T (D4-)	/1		(0	30	10	10
Iron (Dis) Iron (Tot)	дg/1	10	10	30	70	70
Lead (Dis)	μg/1 μg/1	 				ļ
Lead (Dis)	μg/1	 	 			
Manganese	μg/1 μg/1	0	0	0	10	36
80	<i>7-61</i> ~	<u>~</u>			10	5.0
Mercury	ла/1	0	0,1	0	0	0
Žinc	μg/1	0	20	10	10	20
MBAS	mg/1	0,06		0.04	0.04	0.08
Oil & Grease	mg/1					
Total Coliform	#100 m1					
Fecal Coliform	#100 m1					
Other (Specify)						
Source Sample Pote		USGS-EPA.				
Sample Date					73-12-18	74-03-20
0wner		1831010	EDGECLIFF SANITORIUM			
			SANITORIUN			
		#1-4				•
		L	J <u></u>		<u> </u>	

PRIMARY AQUIFER

***************************************	·-		Vell Identif	cation by US	GS Number	
Parameter	Units	25/44-156			Now Manner	<u> </u>
Conductivity	umhos/cm		·	2		
Residue (Total)	mg/1	138				
Residue (180°C)	mg/1		108	167	148	144
Residue, Calculate				· · · · · · · · · · · · · · · · · · ·		·
Residue Loading	TON/AFT					
pH		8,3	7.6	7.5	7.8	7.7
Temperature	*C	10,				 '''
Dissolved Oxygen	mg/1				-	
Hardness (CaCO3)	mg/1	108	136	14	148	134
Alkalinity (CaCO3)	mg/l					1
NH3 -N	mg/1					
NO ₂ -N	mg/1				-	
NO ₃ -N Kjeldahl-N	mg/1					
vletgeut-N	mg/l					
PO4 -P (Total)	mg/1	L		Ì		
PO4 -P (Ortho)	mg/1				 	
Chloride	mg/1	0	14	4	. 4	8
Sulfate	mg/1				1	
Fluoride	mg/l					
Aluminum	μg/1	1	ļ			(
Arsenic	<u>1/وبر</u>			†		}
Cadmium	ug/1			 	 	
Chromium	₂ /1					
Copper	بر 8/ر					İ
Iron (Dis)	₀ 1عر		1			
Iron (Tot)	µg/1	1080	120	10	 	-
Lead (Dis)	, Jig/1		7.30	1	0	320
Lead (Tot)	ر g/1				 	
Manganese	Jug/1	29	0	6	6	9
Mercury	μg/1					
Źinc	/ug/1		 		<u> </u>	
MBAS .	mg/1				<u> </u>	
Oil & Grease	mg/1					
Total Coliform	#100 m1					
	#100 ml					
Other (Specify)						
Source_	i	Storet -				
Sample Date	T	70-05-12	71-09-14	71-10-14	71-11-16	71-12-14
)wner		MODERN				
	Ī	ELECTRIC				
		#,				Ĭ
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		RIMARY A		ation by USG	S Number	
Parameter	Units	25/44-15E2				
Conductivity	umhos/cm	240	240	260	140	280
Residue (Total)	mg/1	106	156	167	184	158
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1			·		
Residue Loading	TON/AFT					
рН	***	7.0	7.3	7.8	7.8	7.9
Temperature .	°C					10.0
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	. 172	136	156	168	156
Alkalinity (CaCO ₃)	mg/1					
NH3 -N	mg/1					
NO2 -N	mg/1	· · · · · · · · · · · · · · · · · · ·				
103 -N	mg/l					
Cjeldahl-N	mg/l					
PO4 -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	4		0	4	4
Sulfate	mg/1					
Fluoride	mg/1					
	_					
Aluminum	μg/1					
Arsenic	μg/1	ļ				
Cadmium	ug/1	ļ				
Chromium	лв/1					
Copper	1/8بر					
Iron (Dis)	µg/1					
Iron (Tot)	μg/1	180	320	140	200	140
Lead (Dis)	11g/1					
Lead (Tot)	µg/1					
Manganese	g/1	18	3	0	19	6
Mercury	بر عبر					
Żinc	g/1					
MBAS ·	mg/1					
Oil & Grease	mg/l					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify)						
_						
Source		Storet -				
Sample Date		72-01-17	72-02-16	72-03-28	72-04-19	72-05-11
Owner		MUDERN				
				1		
		ELECTRIC				
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ATTINGS OF STREET

APPENDIX I WATER QUALITY DATA INMARY AND INFER

AUIA A LUI ALCE K	
Well Identification by USC	: Number

Parameter Units Conductivity umhos/cm Residue (Total) mg/l Residue (180°C) mg/l Residue, Calculated mg/l Residue Loading TON/AFT pH —— Temperature : °C	254 522 300 174 7.4	288 135 7.5	252	360	260
Residue (Total) mg/1 Residue (180°C) mg/1 Residue, Calculated mg/1 Residue Loading TON/AFT pH	/99	136	232	340 151	
Residue (Total) mg/1 Residue (180°C) mg/1 Residue, Calculated mg/1 Residue Loading TON/AFT pH	/99	136	165	151	
Residue (180°C) mg/1 Residue, Calculated mg/1 Residue Loading TON/AFT pH					
Residue, Calculated mg/1 Residue Loading TON/AFT pH	7.4	7.5	1		
Residue Loading TON/AFT pH	7.4	7.5			
	7.4	7.5			
	7.4	7,51			•
Temperature . T.			7.9	., 6	75
Dissolved Oxygen mg/l					
Hardness (CaCO3) mg/1	160	116	156	140	135
Alkalinity (CaCO3) mg/l		····			
NH3 -N mg/1					
NO2 -N mg/1					
NO3 -N mg/1					
Kjeldahl-N mg/1					
PO4 -P (Total) mg/1					
PO4 -P (Ortho) mg/1					
Chloride mg/1	5		.3		7
Sulfate mg/1 Fluoride mg/1					
Fluoride mg/1					
Aluminum µg/1					
Arsenic µg/1					
Cadmium _ug/l					
Chromium µg/1					
Copper µg/1					
Iron (Dis) μg/1					
Iron (Tot) µg/1	280	20	200	٧'۵	1004
Lead (Dis) jig/1					
Lead (Tot) _ug/1					, ()
Manganese µg/1	3	6		116	2.
Mercury ug/1	•				
Žinc µg/1					
MBAS mg/1					
011 & Gresse mg/1					
Date 1 0 145					
Total Coliform #100 ml Fecal Coliform #100 ml					
recer collivia yiou mi					
Other (Specify)					
Source	Storet				-
Sample Date		19 011-11	114.64 12	19-09 14	12 (4 11
Owner					12 17 17
Anner	MUDI KIN			l	1
	ELECTIC				
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APPENDIX I WATER QUALITY DATA PRIMARY A QUIFER Well Identifi

Well Identification by USGS Number							
Parameter	Units	25/44-2601					
Conductivity	umhos/cm	340	340	370	340	346	
Residue (Total)	mg/1	2/2	188	260	196	186	
Residue (180°C)	mg/1						
Residue, Calculated							
Residue Loading	TON/AFT						
**		4,	M /	4 -	7.8	ת ני	
pH	•C	7.6	7.6	7,3	1,8	7.7	
Temperature	_	10.0					
Dissolved Oxygen Hardness (CaCO3)	mg/1	1.89.1	1,-,	477.1	172	.~~	
Alkalinity (CaCO ₃)	mg/1 mg/1	174	156	244		178	
NH3 -N NO2 -N	mg/l mg/l						
NO ₃ -N	mg/1						
Kjeldahl-N	mg/1	,					
	6/						
PO ₄ -P (Total)	mg/1						
PO4 -P (Ortho)	mg/1						
Chloride	mg/1	3	8	10	6	3	
Sulfate	mg/1				· · · · · · · · · · · · · · · · · · ·		
Fluoride	mg/1						
	_						
Aluminum	μg/1						
Arsenic	ug/1						
Cadmium	$\mu g/1$						
Chromium	$\mu g/1$						
Copper	μg/1				**		
Iron (Dis)	$\mu g/1$						
Iron (Tot)	/ug/1	0	200	. 40	20	220	
Lead (Dis))1g/1						
Lead (Tot)	/ug/1		0	3	19	6	
Manganese	μg/1	6			29	6	
Mercury	Jug/1						
Zinc	ug/1				···		
MBAS	ng/1						
Oil & Grease	mg/1						
Total Coliform	#100 m1						
Fecal Coliform	#100 m1						
Other (Specify)							
_		1 . 1					
Source		Storet -					
Sample Date		70-11-04	71-08-09	71-09-14	71-10-13	71-11-16	
Owner		VERA I.D					
		# 5			^	 	
		L			<u> </u>	L	

PRIMAR	$\mathcal{L}_{\mathcal{H}}$	DULEER			
	Well	Identificat:	ion by	USGS	Number

Wassama Arasa		Well Identification by USGS Number				
Parameter Un	its 25/44-2601					
Conductivity umhos	/cm 326	370	340	310	324	
Residue (Total) mg		207	254	181	208	
Residue (180°C) mg						
Residue, Calculated mg	/1					
Residue Loading TON/A						
pH ~	- 7.1	8,3	8,2	7.9	7,9	
	c Zii					
Dissolved Oxygen mg	The same of the sa				***************************************	
Hardness (CaCO3) mg		200	264	172	208	
Alkalinity (CaCO ₃) mg		. 200	467	1,12	208	
-						
NH3 -N mg					·····	
NO2 -N mg						
NO ₃ -N mg Kjeldahl-N mg	/ 1					
vlergettran mR	/ -					
PO ₄ -P (Total) mg					-	
PO ₄ -P (Ortho) mg					•	
Chloride mg		5	3	ર		
Sulfate mg					·	
Fluoride mg	/1					
Aluminum µg	/1					
Arsenic jug	/1	 				
Cadmium ,ug	/1		·			
Chromium ,ug						
Copper	/1					
cobbet Wa	/1					
Iron (Dis) µg						
Iron (Tot) ug		60	160	100	10	
Lead (Dis)	/1					
Lead (Tot) ug	/1	<u> </u>				
Manganese ,ug	/1 9	15	6		0	
Mercury Aug	/1					
Zinc jug		<u> </u>				
MBAS mg		 		· · · · · · · · · · · · · · · · · · ·		
Oil & Grease mg						
m						
Total Coliform #100 Fecal Coliform #100		 				
2000 001210100 11400						
Other (Specify)						
Souver Sample Date	Storet -					
	71-12-13	72-01-17	72-02-14	72-03-31	72-04-18	
Owner	VERA I.D					
	1					
	# 5					
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APPENDIX I WATER QUALITY DATA

Parameter	Units	We: 25/444-2601		ation by USG		
ductivity	umhos/cm	200	326	386	332	326
idue (Total)	mg/1	134	217	195	215	180
idue (180°C)	mg/1					
idue, Calculated idue Loading	mg/l TON/AFT					
rene nonering	2011/ 211 1					
		8,0	8.0	7.9	7.9	6.8
perature	°C	11.0				
solved Oxygen iness (CaCO3)	mg/1 mg/1	136	192	180	232	
linity (CaCO ₃)	mg/1 mg/1	126		780	- A2-A	182
-N	mg/1					
·N	mg/1					
·N	mg/1					
lah1-N	mg/l					
P (Total)	mg/1					
-P (Ortho)	mg/1					
ide	mg/1	3	4	11	3	4
ite :ide	mg/1					
tas	mg/l	 				
num	յոց/1	[
nic	$\mu g/1$					
um	μg/1					
ium r	2/gبر 1/2بر					
•	μg/1					· · · · · · · · · · · · · · · · · · ·
(Dis)	րց/1					
(Tot)	/ug/1	<i>ಫ 0</i>	200	140	60	60
(Dis)	/1g/1					
(Tot) nese	μg/1 μg/1	60	9	9	3	6
	\u2\ 1	60				6
ry	$\mu g/1$					
	Jug/1					
Cross	mg/1					
Grease	mg/1					
Coliform	#100 m1					
	#100 ml					
/a						
(Specify)		ļ				
rce		Storet -				
e Date		72-05-12	72-06-12	72-07-17	72-68-23	72-09-14
		VERA 10				
		VEKA 10				
		# 5				
		خ ا				
		<u> </u>			· · · · · · · · · · · · · · · · · · ·	
			405-40			
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KIMIAKT HOUIPER			
Well Identification	bv	HSGS	Number

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	Well Identification by USGS Number				
Parameter Units	25/44-2601	25/44-2781			
Conductivity umhos/cm	 	260	384	284	276
Residue (Total) mg/1	223	140	244	194	147
Residue (180°C) mg/l					
Residue, Calculated mg/1					************
Residue Loading TON/AFT					
residue rosging 100/Wil					
рН	7.1	7.6	7.8	7.8	6.5
Temperature °C		10.0			
Dissolved Oxygen mg/1					•
Hardness (CaCO3) mg/1	183	108	248	158	148
Alkalinity (CaCO ₃) mg/1	/02	100			
	<u> </u>			~	~·····································
$NH_3 - N \qquad mg/1$					
$m_2 - n \qquad m_2/1$	·				
$NO_3 - N mg/1$					
Kjeldahl-N mg/1					
PO ₄ -P (Total) mg/1		,			
					
PO4 -P (Ortho) mg/1	<u> </u>			8	
Chloride mg/1	2	3	99	8	4
Sulfate mg/1					
Fluoride mg/1					
Aluminum µg/1					
Arsenic µg/1					ļ ————
Cadmium ug/1				 	
					
			 	 	
Copper µg/1					
Iron (Dis) µg/1					
Iron (Tot) µg/1	100K	0	200	120	300
Lead (Dis) µg/1					
Lead (Tot) µg/1	15				1
	2	.3	6	9	9
Manganese µg/1				7	
Mercury µg/1	4,2				
Zinc µg/1					
MBAS mg/1					
Oil & Grease mg/1					
m-h-1 0-145-m- #100 -1		1			
Total Coliform #100 ml Fecal Coliform #100 ml			 	 	
recar collion #100 mr				 	
Other (Specify)					
6	St. 1				
Souvee Sample Date	Storet -				
manhar mass	72-09-14	70-09-29	71-09-14	71-10-14	71-11-16
Owner	VERA I.D	Modern			
	•	11.446	 		·· >
	# 3-	Elect. #9			1
	- 3		1		i
		1	1	1	1
	1		l	<u></u>	

APPENDIX T WATER QUALITY DATA PRIMABY A SOLIT

RIMAB	$\mathbf{y}_{\mathbf{A}}$	$\Omega U J E E$	T.K			
	Wall	Tdentif	fostion	har	HECE	Humbar

Well Identification by USGS Number						
Parameter	Units	25/44-2781			·	***************************************
Conductivity	umhos/cm	300	300	234	250	268
Residue (Total)	mg/1	178	182	/38	156	223
Residue (180°C)	mg/1					
Residue, Calculate	d mg/l					
Residue Loading	TON/AFT					
••	ļ	.		- 0		
pH	°C	8.0	8.3	7.9	7.8	7.2
Temperature Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	144	152	144	/52_	208
Alkalinity (CaCO ₃)	mg/1		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		196-	
NH3 -N	mg/1					
NO2 -N	mg/1					
NO ₃ -N	mg/î					
Kjeldahl-N	mg/1					
PO4 -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1		····			······································
Chloride	mg/1	28	20	2	2	0
Sulfate	mg/1					
Fluoride	mg/l				[
Aluminum	μg/1					
Arsenic	μg/1					
Cadmium	/ug/1				(
Chromium	μg/1					<u> </u>
Copper	μg/1			1		
Iron (Dis)	μg/1					
Iron (Tot)	Jug/1	0	100	100	220	120
Lead (Dis)	1/gدر .	ļ			 	
Lead (Tot)	Jug/1	ļ		 	ļ <u>-</u>	
Manganese	₀ /1	18	6	3	3	3
Mercury	աց/1					
Źinc	ug/1					
MBAS ·	mg/1	<u> </u>		<u> </u>	 	<u> </u>
0il & Grease	mg/1					
	_					
Total Coliform	#100 m1			ļ		
Fecal Coliform	#100 m1				 	
Other (Specify)						
(),					<u> </u>	
_			(į		
Source		Storet -				
Sample Date		71-12-14	12-01-17	12-02-16	72-03-28	72-04-19
Owner		Modern				
		Elect #9				···
		1 1000		ļ		1
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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER Well Identifi

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		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/44-2781-				
Conductivity	umhos/cm	284	260	304	276	300
Residue (Total)	mg/1	169	/73	185	172	161
Residue (180°C)	mg/1					
Residue, Calculated	d mg/1				<u> </u>	
Residue Loading	TON/AFT					
pH		8,1	8.0	7.3	7.8	7.5
Temperature	•c	12.8			7.0	
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/l	170	148	140	172	152
Alkalinity (CaCO ₃)	mg/l			1.1.		
NH3 -N	mg/1					
NO ₂ -N	mg/1			 		
NOj -N	mg/1			 		
Kjeldahl-N	mg/1			 		
-						
PO4 -P (Total)	mg/l					
PO4 -P (Ortho)	mg/1					·
Chloride	mg/l	5	5	12	5	5
Sulfate	mg/l			ļ		
Fluoride	mg/l			ļ		
Aluminum	μg/1					
Arsenic	µg/1					
Cadmium	ug/1					
Chromium	1/gu					
Copper	μg/1					
Iron (Dis)	μg/1					
Iron (Tot)	/ug/1	140	10	380	40	100
Lead (Dis)	μg/1					
Lead (Tot)	ر العربي 1/عبر	<u> </u>		····		
Manganese	ي 1/gر	9	0	6	9	6
_						
Mercury	μg/1	<u></u>			ļ <u></u>	
Zinc	ug/1					
mbas	mg/1				ļ	<u> </u>
Oil & Grease	mg/1	<u> </u>				ļ
Total Coliform	#100 m1					
Fecal Coliform	#100 ml					
Other (Specify)						
Sample Date		72-05-11	72-06-19	72-07-17	72-08-23	72-09-14
Owner		Modern Elect #9				

APPENDIX I WATER QUALITY DATA

PRIMARY AQUIFER			
Wall Tanadedaa	2	11000	31

	W(ell Identific	ation by USG	S Number	,
Parameter Uni	ts 25/44-27E1	25/44-29A1			
Conductivity umhos/	cm 300	125	384	340	350
Residue (Total) mg/		199	220	202	220
Residue (180°C) mg/	1				
Residue, Calculated mg/	1				
Residue Loading TON/AF	T				
рН	7.4	7.9	7.9	7.5	. 7.6
Temperature °C		10.0			
Dissolved Oxygen mg/	1				
Hardness (CaCO3) mg/		142	196	192	200
Alkalinity (CaCO3) mg/	1				
NH3 -N mg/					
NO ₂ -N mg/					
$MO_3 - N$ mg/	1				
Kjeldahl-N mg/	1				
PO4 -P (Total) mg/	1				
PO ₄ -P (Ortho) mg/					
Chloride mg/	1 3	3	7	7	5
Sulfate mg/	1				
Fluoride mg/	1	 			
Aluminum µg/	1				
Arsenic µg/					
Cadmium ug/					
Chromium µg/			<u> </u>		
Copper ug/	1				
Iron (Dis) µg/					
Iron (Tot) µg/		140	0	120	40
Lead (Dis) jig/		<u> </u>	<u> </u>		
Lead (Tot) ug/					
Manganese jug/	$\frac{1}{2\kappa}$	6	ĝ	0	3
Mercury jug/	1 3.8				
Zinc µg/	1	<u> </u>			
MBAS mg/	1				
Oil & Grease mg/	1	 	 		
Total Coliform #100 m	1				
Fecal Coliform #100 m					
Other (Specify)					
Sample Date	72-09-14	70-09-23	71-09-14	71-10-13	71-11-15
Owner	Modern	WWP			
	Modern Elect # 9	#2-4	to electronical designation of the second de		

Well	Ident	ificatio	n bv	USGS	Number

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Description.	77-14-	w(ell Identific	ation by usu	2 Number	
Parameter	Units	25/44-2941	 			
Conductivity	umhos/cm	330	330	340	300	300
Residue (Total)	mg/1	238	152	197	193	155
Residue (180°C)	mg/l		1			
Residue, Calculated	mg/l	~~ ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	†			
	TON/AFT				**********	
	2011, 222 2	, <u>, , , , , , , , , , , , , , , , , , </u>			····	
рН		7,8	8.3	8.2	7,8	8.0
Temperature	*C	•				
Dissolved Oxygen	mg/1					•
Hardness (CaCO3)	mg/1	232	188	200	172	172
Alkalinity (CaCO ₃)	mg/1					
NH3 -N	mg/1					****
NO ₂ -N	mg/1					
NO ₃ -N	mg/1		 			
Kjeldahl-N	mg/1		 			
vîernenr.u	· mR\T		-			
PO ₄ -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	9	6	.5-	4	. 3
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	$\mu g/1$					
Arsenic	μg/1		<u> </u>			
Cadmium	µg/1					
Chromium	μg/1					
Copper	µg/1					
Iron (Dis)	/1					
	μg/1	1.70	180		3 / 5	9
Iron (Tot)	ug/1	140	100	420	360	
Lead (Dis)	/1g/1	<u></u>	 	<u></u>		
Lead (Tot)	ug/1					
Manganese	µg/1	36	 	6	3	0
Mercury	μg/1			•		
Zinc	ug/1			1		T
MBAS	mg/1					
011 & Grease	mg/1					
	#100 m1			L		
Fecal Coliform	#100 ml					
Other (Specify)						
Source		Storet-				
Sample Date		71-12-13	72-01-18	72-02-14	72-03-31	72-04-18
Owner		W.W.P				
		#2-4				
		2-7				
		<u> </u>	1		<u> </u>	<u> </u>

APPENDIX T.
WATER QUALITY DATA
PRIMARY AQUIFER
Well Identification by USGS Number

Parameter Unit		11 Identific	ation by USG	S Number	
Parameter Unit	25/44-2941				
nductivity umhos/o		320	368	350	
sidue (Total) mg/:		192	192	2/9	
idue (180°C) mg/l idue, Calculated mg/l					
idue Loading TON/AF					
	8./	7.9	8.0	7.6	
perature °C	10.0	 	0,0	7.6	
solved Oxygen mg/					
dness (CaCO3) mg/3	178	176	188	173	
alinity (CaCO ₃) mg/1		 			
-N mg/:			·		
-N mg/:					
dahl-N mg/					
-P (Total) mg/:					
-P (Total) mg/: -P (Ortho) mg/:		 		·	
oride . mg/:	6	7	12	4	
fate mg/:					
oride mg/:	·	 		ļ	i
ıminum µg/i					
senic µg/:					
lmium ug/:					
comium ug/	<u> </u>			 	
pper ug/:	` 			 	
n (Dis) µg/					
on (Tot) µg/	60	180	260	100K	/5
d (Dis) µg/	!	 	<u></u>	420	20
id (Tot) /ug/: uganese /ug/:		 	 	720	20
•		1			
cury "ug/	<u> </u>	 		9,3	0,2 K
nc /ug/. AS mg/		 		 	
S mg/: . & Grease mg/:	<u> </u>	 		 	
tal Coliform #100 m		-	<u> </u>		-
al Coliform #100 m	L	 		 	
mer (Specify)					
,-,,,,					
	Storet-				<u> </u>
ou <i>rce</i> mple Date		 		1	
	72-05-11	72-06-16	72-07-17	12-09-14	73-01-15
ner	WWP				
	#2-4	-	 		
			1	, resident	
			1		
		<u> </u>	L	<u> </u>	L
		405-46			
		40 3-46			
					THE PERSON NAMED IN COLUMN TWO IS NOT THE OWNER.

APPENDIX I
WATER QUALITY DATA
PRINIARY AQUIFER
Well Identification by USGS Number

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Units whos/cm mg/1 mg/1 mg/1 DN/AFT	25/45-1501 236 163	270	260 196	228	250
mg/1 mg/1 mg/1 ON/AFT					
mg/1 mg/1 mg/1 ON/AFT			196	134	7.5
mg/1 mg/1 DN/AFT					159
mg/1 DN/AFT			ł		
ON/AFT		L			
A	7,3	8:1	7.4	7.7	7.7
•c	11.7	1	(18		
- 1	· · ·	 			
	156	/32	20.4	//2	/36
		1-2-			
					
		 	·		
mg/1		 	ļ		
mg/1					
- / 1					
		ļ	 		ļ
		 	 		<u> </u>
		7	5	4	5
		<u> </u>		<u> </u>	
mg/l			 		
μg/1					
µg/1				}	
ng/1					
$\mu g/1$					
μg/1					
119/1				ł	
	120		100	140	50
		 	1	 	
		 			
μg/1	4	15	5		3
na/1					
		 	 	 	
		 	 	 	
		 	 	 	
	Storet -				
	71-07-21	71-08-18	71-09-16	71-10-13	71-11-10
	Holiday_		Managar Managar up		
	,				
	mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1	mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1	mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1	mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1	mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1

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Wall Identification	hw	TISCS	Mumhar

Well Identification by USGS Number						
Parameter	Units	25/45-15171-		_	· ·	
Conductivity	umhos/cm	216	248	248	7	184
Residue (Total)	mg/1	127	.97	164	156	176
Residue (180°C)	mg/1					•
Residue, Calculate						
Residue Loading	TON/AFT					
pH .	***	7.2	7.5	8,2		7.3
Temperature	°C				·	
Dissolved Oxygen	mg/1	·			,	
Hardness (CaCO3)	mg/l	124	94	148	118	196
Alkalinity (CaCO3)	mg/1					
NH3 -N	mg/l					
NO2 -N	mg/l					
NO3 -N	mg/l					
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/l					
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	2	4	5	2	2
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	μg/1					
Arsenic	μg/1	<u> </u>				
Cadmium	Jug/1					
Chromium	1/grر					
Copper	µg/1					
Iron (Dis)	μg/1					
Iron (Tot)	µg/1	340	20	. 80	60	80
Lead (Dis)	1/81ر					
Lead (Tot)	ug/1					
Manganese	μg/1	0		9	6	0
Mercury	ug/1					
Zinc	/ug/1		·	1	1	
MBAS -	mg/1		 		·	
0il & Grease	mg/1					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify)						
_						
Source Sample Date		Storet-		 		
	•	71-12-13	72-01-18	72-626	72-03-28	72-04-18
Owner		Holiday Hills				
				Parkin digram. * * d fra		Parkettining and the second
		<u> </u>	<u></u>	<u> </u>		<u> </u>

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APPENDIX I WATER QUALITY DATA

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PRIMIARY ADVIFER			
Well Identification	h	HCCC	Mumbay

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AND THE PROPERTY.

		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/45-1501-				
Conductivity	umhos/cm	192	242	248	246	256
Residue (Total)	mg/1	124	138	141	149	144
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					<u> </u>
Residue Loading	TON/AFT					
рН		7,8	8.7	7.6	8.1	7.5
Temperature	•c	11.7				
Dissolved Oxygen	mg/1		<u> </u>			-
Hardness (CaCO3)	mg/1	120	124	/36	144	140
Alkalinity (CaCO3)	mg/l					
NH3 -N	mg/1		1			
NO ₂ -N	mg/1		ļ	 		
NO ₃ -N	mg/î		 	 		
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/1		1			
PO4 -P (Ortho)	mg/1		 	 	· · · · · · · · · · · · · · · · · · ·	
Chloride	mg/1	7	6	9	2	5
Sulfate	mg/l		 	 		
Fluoride	mg/l					<u> </u>
A4 4						
Aluminum	μg/1					
Arsenic	μg/1	<u> </u>				
Cadmium	ug/1					<u> </u>
Chromium	μg/1					
Copper	ug/1					·
Iron (Dis)	μg/1					
Iron (Tot)	µg/1					
Lead (Dis)	ر g/1					
Lead (Tot)	ug/1	300	4000	20	300	80
Manganese	μg/1	9	6	3 -	3	6
Mercur y	μg/1					
Zinc	ug/1					T
MBAS	mg/1					T
Oil & Grease	mg/l		<u> </u>			Ţ <u> </u>
Total Coliform	#100 ml					
Fecal Coliform	#100 m1					
Other (Specify)						
6						
Source Sample Date		Storet -	ļ			
•		72-05-10	72-06-19	72-07-24	72-08-14	72-09-14
Owner		Holiday				-

4.1.20			
Well	Identification	by USGS	Number

		11 Identific	ation by USG	S Number	
Parameter Units	25/45-1501				•
Conductivity umhos/cm	260		266	259	263
Residue (Total) mg/l	181				
Residue (180°C) mg/l			158	157	154.
Residue, Calculated mg/1					
Residue Loading TON/AFT			0,21	0.21	0.21
рн			7.6	8./	8.1
Temperature °C			12.0	12.0	12.0
Dissolved Oxygen mg/1					
Hardness (CaCO3) mg/l	125		120	120	120
Alkalinity (CaCO ₃) mg/l					
NH3 -N mg/1			0.03	0.01	0.01
$NO_2 - N$ mg/1			0,000	0.002	0.002
$NO_3 - N$ mg/1			218	1.5	2.0
Kjeldahl-N mg/1			0.06	0.05	0.04
PO ₄ -P (Total) mg/1			0.021	0,023	0.073
PO ₄ -P (Ortho) mg/1			0.021	0:022	0.073
Chloride mg/l	2	İ	2.3	2.3	2,5
Sulfate mg/1			<u> </u>		
Fluoride mg/l					
Aluminum µg/1					
Arsenic µg/1		 	5	6	5
Cadmium /ug/1			 	D	7
Chromium jug/1		·	0	0	
Copper µg/1			3	6	3
Iron (Dis) µg/1			10	10	0
Iron (Tot) µg/1	100 K	/5	 		
Lead (Dis) µg/1	100 8	1	 	2	6
Lead (Tot) µg/1	20	20	 		
Manganese µg/1	2	2	10	0	0
Managem	0- (
Mercury µg/1 Zinc µg/1	20,4	0.2K	0,1	0	0
Zinc µg/1 MBAS mg/1		 	0.00	50	0
Oil & Grease mg/l			0.00	0.04	002
m1 0.115 #100 1					
Total Coliform #100 ml Fecal Coliform #100 ml					
	70				42
Other (Specify)	72-09-14	12-01-15	73-66-28	73-09-25	13-12-18
Source	Storet	3	1154.5 500		
Sample Date	210.61		4565-EPA -		
-	<u> </u>				
Owner	Holiday Hills				
	Hills				
	<u></u>	<u> </u>	<u> </u>	L	<u> </u>

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是在这里的第三人称形式,在这个个是在这里的时间,不是是这个人,他们也是是一个人,也是是一个人,也是是一个人,也是是一个人,也是一个人,也是一个人,也是一个人,也是

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		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/45-1501	26/42-131(5)			-
Conductivity	umhos/cm	242	350		367	360
Residue (Total)	mg/l					
Residue (180°C)	mg/l	143			200	186.
Residue, Calculate						
Residue Loading	TON/AFT	'_19			0,27	0.25
рH		7.8	7.5	•	7,6	8.0
Temperature	*c	11.8	10.0	10.0	10.8	11.0
Dissolved Oxygen	mg/l		8.7	9.3		
Hardness (CaCO3)	mg/l	120	179		180	180
Alkalinity (CaCO ₃)	mg/1					
NH3 -N	mg/1	0.03			0.02	0.02
NO ₂ -N	mg/1	0.002			0,002	0.002
NO ₃ -N	mg/1	//3	2,2		2,5	2.5
Kjeldahl-N	mg/1	0.56	 		0.05	0.2/
PO ₄ -P (Total)	mg/1	0.023	0.01		0.009 .	0.048
PO4 -P (Ortho)	mg/l	0.02/			0.005	0.043
Chloride	mg/1	2.6			5.8	5,9
Sulfate	mg/1					
Fluoride	mg/l					
Aluminum	بر 1/عبر					
Arsenic Cadmium	μg/1	2			/	9
Chromium	1/عبر 1/عبر	0			0	0
Copper	1/عبر 1/عبر	5			0	0
copper	7081 ±			***		
Iron (Dis)	_{1/8} بر	20			10	50
Iron (Tot)	<u>/اوبر</u>			. 15		
Lead (Dis)	g/1				0	3
Lead (Tot)	μg/1		20	15		
Manganese	1/g <i>بر</i>				0	
Mercury	μg/1	0.4	15,4		0,/	0
Zinc	μg/1 μg/1	20	1917	0,2 K	10	10
MBAS	mg/1				0.00	0.00
011 & Grease	mg/1					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify)						
(0)						
						1
Source		USGS-EPA	Storet -		USGS- EPA -	
Sample Date		74-03-20	72-09-13	13-01-15	73-66-27	73-09-26
Owner						
		Holiday Hills	Hatchery - Springs			
		17116	springs			1
		<u></u>	<u> </u>]	<u> </u>

APPENDIX I WATER QUALITY DATA

PRIMARY	AQU	IFE	3		

		We]	ll Identific	ation by USGS	S Number	
Parameter	Units	26/42-115(5)		26/42 - 12AI(3)		<u> </u>
Conductivity Residue (Total)	umhos/cm mg/1	354	358	301	29.7	29/
Residue (180°C) Residue, Calculate	mg/l	198	195	166	165	157
Residue Loading	TON/AFT	0,27	0,27	0.23	0.22	0.2/
pH	 •c	8.0	7.7	7.7	8./	8,2
Temperature Dissolved Oxygen	mg/1	11.0	10.3	11.8	10.5	10.0
Hardness (CaCO ₃) Alkalinity (CaCO ₃)	mg/1 mg/1	180	170	150	150	140
NH3 -N	mg/1	0:01	0.01	0.02	0.01	0.01
NO ₂ -N NO ₃ -N	mg/1 mg/1	2,3	0:00 l 2:D	1.3	1.1	0,004
Kjeldahl-N	mg/l	0.04	0.26	0:05	0.15	0.15
PO4 -P (Total) PO4 -P (Ortho)	mg/1 mg/1	0.012	0,013	0.003	0:004	0.008
Chloride	mg/l	0.010 5.6	0.007 5.8	2.3	3.4	2.5
Sulfate Fluoride	mg/1 mg/1					
Aluminum	_{1/8} بر					
Arsenic	μg/1	4	3	/	0	<u>5</u>
Cadmium Chromium	/ນg/1 /ນg/1	0	0	0	0	0
Copper	дg/1	/	3	/	0	0
Iron (Dis)	μg/1	0	20	30	10	10
Iron (Tot) Lead (Dis)	μg/1 μg/1	0	2	· /	2	0
Lead (Tot) Manganese	μg/1 μg/1	40	36	0	Q	0
Mercury	ر 1/gu	<i>e</i>	0	0	0	
Zinc MBAS	ug/1 mg/1	0.02	20	30	10	20
Oil & Grease	mg/1					
Total Coliform Fecal Coliform	#100 ml #100 ml					
Other (Specify)						
Source		USGS-EPA				-
Sample Date		73-12-17	74-03-19	73-06-29	73-09-26	73-12-17
Owner		Hatchery Springs	>	Spokane Country -		

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APPENDIX I WATER QUALITY DATA PRIMARY AQUIFER Wall Identifi

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Parameter	Units	26/42-12AI(3)	11 Identific	acton by esc	2 Mamber		
onductivity	umhos/cm	287	150	210	172	180	┪ .
esidue (Total)	mg/1		92	131	105	120	İ
esidue (180°C)	mg/1	156]
esidue, Calculated esidue Loading	mg/l TON/AFT	0.2/					1
		8.0	7.8	7:1	7.5	7:6	1
mperature ·	•c	9.5	8.7		· · · · · · · · · · · · · · · · · · ·	1:8	1
ssolved Oxygen	mg/1						1
rdness (CaCO3)	mg/l	140	60	88	88	/28]
kalinity (CaCO3)	mg/l						4
3 -N	mg/1	0.01					4
2 -N 3 -N	mg/1 mg/1	0.002					-
ldahl-N	mg/1	0.12					1
-P (Total)	mg/1	0.013					
4 -P (Ortho)	mg/1	0.003]
lorida	mg/1	2,2	D		2	3	1
lfate uoride	mg/1 mg/1						1
uminum	μg/1						1
senic	дв/1 дв/1	2				!	1
dmium	μg/1	6					1
romium	μg/1	0]
pper	μg/1	0					1
n (Dis)	μg/1	10					1
on (Tot)	/Jg/1		0	.0	20	140	4
ad (Dis) ad (Tot)	ມg/1 ມg/1	2	ļ	 	 	 	4
nganese	μg/1 μg/1	14	6	18	15	. 9	1
ccury	μg/1	0					
nc	1/وبر	10					1
AS	mg/1	0.05					4
1 & Grease	mg/1						1
	#100 ml #100 ml	<u> </u>			ļ	 	4
	ATON WT						†
her (Specify)							1
our <i>ce</i>		USGS-EPA	Storet -				
mple Date	·····	74-03-19	70-05-14	71-09-14	71-10-13	71-11-17	1
ner		Spokane	C.1.D	1	13		1
		Country	#2A				1
		club					
]
			405-53				
			,00				
THE PERSON NAMED IN COLUMN TWO IS NOT THE OWNER OF THE OWNER OF THE OWNER.	entervisent septiminate, silitate	వి. మెలికోగా ఇచ్చే ప్రాథమేకు సమాజుపుల్లు ప్రాథాపుల్లు ప్రాథాపుల్లు ఉన్నాయి. అయినికి ఇచ్చారు. అయినికి అయినికి మ	BANGARAM ARABAMAN TANAM MENANTER SECTION SECTI	randers hilly of a market or otherwise of the 19 46 049644	and the second s	ra Silisik Kalilli	Torras.

APPENDIX T WATER QUALITY DATA PRIMARY AQUIFER Wall Ident

		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/45-18R1	·			
Conductivity	umhos/cm	170	180	/72	150	150
Residue (Total)	mg/1	104	118	104	111	/36
Residue (180°C)	mg/1					•
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT				•	
_1 7	900 000	7.5	7.7		7.7	7.7
pH	°C	(<u>.</u> 3	· · · · · · · · · · · · · · · · · · ·	8.2		
Temperature	-					
Dissolved Oxygen	mg/1			84	G (132
Hardness (CaCO3)	mg/1	100	82		84	134
Alkalinity (CaCO ₃)						
NH3 -N	mg/1					
NO ₂ -N	mg/1			 		
NO3 -N	mg/1	ļ	ļ — —————	ļ		
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1	ļ. <u></u>		<u> </u>		
Chloride	mg/1	6	7	2	0	3
Sulfate	mg/1					
Fluoride	mg/l			ļ		
Aluminum	μg/1					
Arsenic	μg/1					
Cadmium	/ug/1					
Chromium	μg/1					
Copper	ug/1					
Iron (Dis)	μg/1					ļ
Iron (Tot)	ug/1	0	120	140	140	140
Lead (Dis)	g/1ر					
Lead (Tot)	ug/1					
Manganese	μg/1	0	0	3	6	3
Mercury	μg/1.	ļ				
Zinc	ду/1 ду/1			 	 	
MBAS -	JUB/ 1 ma/1			·	 	
Oil & Grease	mg/1		 			
OII & Grease	mg/l					
Total Coliform	#100 ml			<u> </u>		
Fecal Coliform	#100 ml					
Other (Specify)				<u> </u>		
C						
Source		Storet-				
Sample Date		71-12-14	72-01-17	72-02-16	72-03-28	72-04-19
Owner		C.1.D. #2A				
						

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וומש	1/4007171	cation by	Here	Tree of the co
4677	TACHETET	CALTON DA	0303	MUMDER

nos/cm mg/1	25/45-18R1 164 98	159	/68	,,,,	-
mg/1		159	148	44.5	
				168	180
	/ 0	122	126	100	367
mg/1					
mg/1					
/aft					
	7:8	7:7	7.4	714	7.2
*c [
mg/l [1
mg/1	90	98	112	100	88
mg/1 [
mg/1					
mg/1	•				
mg/1					
mg/l					
mg/1					
mg/1				,	
mg/1 [7	10	2	/
mg/1 [
mg/l					
μg/1					
ug/l					
ug/l					
ug/1 [
ug/1					
ug/1					
ug/1	20	140	680	40	100 K
ug/1 [
ug/l					10
ug/l		0	6	6	2.K
ug/1					4.6
ug/1 [1				
mg/1					
mg/1	···				
0 ml					
0 m1					
	Storet -				
		02 24 10	72.02.21	A A A A	
		172-06-19	12-07-24	72-08-14	72-09-14
	_	I	1	1	i
	C.I.D			ن کرپیش در پنو پن مشکده مستخصصت شده مست	
	18/1 18/1 18/1 18/1 18/1 18/1 18/1 18/1	19/1 19/1	18/1 18/1	18/1 18/1	18/1

William William Strate and his Strate and British and the Strategy of the Stra

	/ <u></u>	We	ll Identific	ation by USG	S Number	
Parameter	Units					24/43-5LI(S)
Conductivity	umhos/cm	293	290	292	291	393
Residue (Total)	mg/1					
Residue (180°C)	mg/1	157	164	159	162	22/
Residue, Calculate Residue Loading	d mg/1 TON/AFT	0.21	0.22	0.22	0.22	zi 2 o
Kestage rosatus	ION/AFI	0,2	UILL	0.02	0,24	0.30
pН		7.6	81/	8-3	7.9	7.6
Temperature	°C	11.8	12.0	11.0	10.8	11.6
Dissolved Oxygen	mg/1				,	
Hardness (CaCO3)	mg/1	140	140	140	140	140
Alkalinity (CaCO ₃)	mg/l	2	<u>.</u>			0.004
NH3 -N NO2 -N	mg/1 mg/1	0.03	0.01	0,003	0.01	0.000
NO ₃ -N	mg/1	1.20	0.006	0,97	1:3	1/3
Kjeldahl-N	mg/l	0.03	0.12	0103	0,22	0.05
	_					
PO4 -P (Total)	mg/1	0.008	0,022	0,014	0.032	0.006
PO4 -P (Ortho)	mg/1	0.007	0,001	0,013	0.006	0.002
Chloride Sulfate	mg/1	4.0	3,5	3.7	4,0	15,0
Fluoride	mg/1 mg/1					-
11401146	mg/ ±				}	
Aluminum	µg/1	ļ	į			
Arsenic	$\mu g/1$	5	7	4	7-	10
Cadmium	$\mu g/1$	/	0	0	0	
Chromium	μg/1	0	0	0		ن
Copper	μg/1	2	0	22	4	0
Iron (Dis)	1/ <u>و</u> بر	10	20	20	0 ز	40
Iron (Tot)	$\mu g/1$					1-70
Lead (Dis)	/1g/1	0	3	0	/	0
Lead (Tot)	Jug/1					
Manganese	1/guر		0	0	43	10
Mercury	/1		0	0	0	0,1
Zinc	дуд/1 Дуд/1	40	90	150	160	10
MBAS -	mg/1	0.00	0.04	0.02	0.00	0.00
Oil & Grease	mg/1					
	_					
Total Coliform	#100 ml	<u></u>				
Fecal Coliform	#100 ml		ļ		 	
Other (Specify)			!			
-				1		
Source		USGS- EPA-				- Vin Ar Physiology are a second
Sample Date					1	
		73-06-29	73-09-26	73-12-17	74-03-19	73-06-29
Owner		,3.W		-		Wandermera
		Livergood			,	
					j	

APPENDIX I WATER QUALITY DATA

PRIMARY AQUIFER

Parameter	Units		11 Identific	ation by USO		<u> </u>
		26/43-5LI(5)			26/43-781(3)	
Conductivity	umhos/cm	387	394	39/	305	30 ¥
Residue (Total)	mg/l					
Residue (180°C)	mg/1	214	222	2/7	168	166.
Residue, Calculate Residue Loading	ed mg/1 TON/AFT	0,29	030	A 2 D	ļ	0.23
vestane rogating	ION/ AF I	V,2/	030	0,30	0,23	0123
pH ·	***	8.0	81	7.7	7.6	8.0
Temperature	*C	12.0	11.5	10,8	11.2	11.0
Dissolved Oxygen Hardness (CaCO3)	mg/1 mg/1	172	180	/22		
Alkalinity (CaCO ₃)		170	780	170	150	150
NH3 -N	mg/1	0.01	0.02	0.01	0.02	0.02
NO ₂ -N	mg/1	0.002	0.007	0,001	0.001	0.002
NO3 -N	mg/1	1.7	0.95	4/	1.3	11/
Kjeldahl-N	mg/1	0.11	0.09	0.05	0.05	0.15
PO4 -P (Total)	mg/l	0.005	0.554	2010		1 4 4 3
PO4 -P (Ortho)	mg/1	0,004	0,005	0.010	0.001	0.003
Chloride	mg/1	140	14.0	1410	2.3	2,3
Sulfate	mg/1			1		
Fluoride	mg/1					
Aluminum	μg/1					
Arsenic	μg/1	3	,	2	- 4	64
Cadmium	μg/1	0	0	0	 	٥
Chromium	μg/1	0	0	0	0	O
Copper	μg/1	Ö	0	2	0	0
Iron (Dis)	μg/1	30	10	90	10	10
Iron (Tot)	/ug/1				1	
Lead (Dis)	ر 1/gبر	.3	0	2	0	/
Lead (Tot)	/ug/1					
Manganese	μg/1	0		29	0	0
Mercury	بر عربر	0	0	0	0	0./
Zinc	ير 1/ <u>ع</u> بر	20	0	20	10	10
MBAS	mg/l	0.00	0.05	0,08	0,00	
Oil & Grease	mg/l					
Total Coliform	#100 m1			٠.		
Fecal Coliform	#100 ml					
Other (Specify)						
Source		USGS-EPA-				
Sample Date		73-09-26	73-12-17	74-03-19	13-06-29	73-09-26
Owner		Wandermerc			Dept. of	

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PATITIFICA AUDITER			
Woll Identification	2.	TICCO	Mumban

				ation by USG		
Parameter	Units	26/43-781(5)		26/45-35FI-		26/45-36N
Conductivity	umhos/cm	294	311	276	277	296
Residue (Total)	mg/1					
Residue (180°C)	mg/1	155	162	149	151	139·
Residue, Calculate	ed mg/l					
Residue Loading	TON/AFT	0.2/	0,22	0,20	0.2/	0,19
pΗ		8.2	810	7.6	8.0	714
Cemperature ·	°C	10.0	9.6	8,6	8,5	9,6
Dissolved Oxygen	mg/1	·				
iardness (CaCO3)	mg/1	150	150	140	140	150
Alkalinity (CaCO ₃)) mg/1					
NH3 -N	mg/1	0.02	0.01	0.01	0.01	0.09
102 -N	mg/1	0:002	0.005	0,000	0,001	0.001
103 -N	mg/1	0.85	111	0,47	0.43	0.62
(jeldahl-N	mg/l	0.06	0104	0.05	0.04	0,09
204 -P (Total)	mg/1	0,008	0.007	0,003	0,006	0,014
204 -P (Ortho)	mg/1	0.005	01003	0,002	0.006	0.001
Chloride	mg/1	2.3	2,4	0.8	10	1.3
Sulfate	mg/l					
luoride	mg/l					
luminum	µg/1					
rsenic	$\mu g/1$	3	2		2	0
Cadmium	$\mu g/1$	0	0		! 0	3
Chromium	1/هبر	0	0	10	0	0
Copper	μg/1		66			70
ron (Dis)	μg/1	10		110	30	2400
Iron (Tot)	/\g/1			<u> </u>		<u> </u>
Lead (Dis)	1/פונ	0		0	11	45
Lead (Tot)	ug/1			 		
langanese	μg/1		36	D	10	30
lercury	ug/1	0	0,1	Out	0	0,1
line	ug/1	0	20	30	30	560
IBAS 011 & Grease	mg/1 mg/1	0.02	0,00	0,00	0.04	0.00
	_					
Cotal Coliform Secal Coliform	#100 ml #100 ml					
Other (Specify)						
_						
Sample Date	- 170, - 1-2-10,	USGS-EPA-				
•			74-03-19		73-09-25	73-06-28
wner		Dept. of		C.1.D.		G. N. Siverson
		CIAME		= 10A		JACKS BA

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		ne ne	TI Identifit	ation by USG	2 Number	
Parameter	Units	26/45-36N/		26/45-3601-		
Conductivity	umhos/cm	30/	865	274	274	279
Residue (Total)	mg/1					
Résidue (180°C)	mg/1	158	168	148	155	176 .
Residue, Calculate	ed mg/1					
Residue Loading	TON/AFT	0,21	0.23	0.20	0.2/	0.24
рH		7.9	8.2	7.8	7.9	8.0
Temperature ·	*C	9.0	9,5	11.0	11.0	11.0
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	150	160	140	140	140
Alkalinity (CaCO3)) mg/1					
NH3 -N	mg/1	0.03	0,06	0.04	0.02	0.01
NO2 -N	mg/1	0.001	0.006	0.000	0,004	0.002
NO3 -N	mg/1	0.87	0.77	1.0	1.0	0.24
Kjeldahl-N	mg/l	0,09	0,04	0,05	0.06	0.03
PO4 -P (Total)	mg/l	0,005	0.023	0,004	0.008	0.009
PO4 -P (Ortho)	mg/1	0.004	0.006	0,002	0.005	0.009
Chloride	mg/l	1,0	1.2	15	118	1.6
Sulfate	mg/l					
Fluoride	mg/l					
Aluminum	<u>1/gبر</u>					
Arsenic	µg/1	5	ク	4	3/	6
Cadmium	ug/1	0		0	0	0
Chromium	<u>1/وىر</u>	0	0	0	0	0
Copper	μg/l	29				4
Iron (Dis)	g/1	550	1500	50	60	60
Iron (Tot)	µg/1					
Lead (Dis)	μg/1			2	4	0
Lead (Tot)	/ug/1			<u> </u>		
Manganese	بر 1/8سر		80	0	0	0
Mercury	ug/1	0	0	0	0	٥
Zine	ug/1	250	460	120	120	160
MBAS	mg/1	0.00	0,00	0.00	0.03	0.00
Oil & Grease	mg/1					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify)						
Source						
Sample Date			02 15 15		-	1
		73-09-26	73-12-18		73-09-26	73-12-18
Owner		G. N		Borden -	 	
		Siverson				

APPENDIX T WATER QUALITY DATA

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T.J.o.	1 74.	~ - 1	. L	TICOC	Much

		We	11 Identific	ation by USO	S Number	<u> </u>
Parameter	Units	26/46-31MI				
Conductivity	umhos/cm	232	208	232	260	247
Residue (Total)	mg/1	102	165	137	162	/33
Residue (180°C)	mg/1					•
Residue, Calculated						
Residue Loading	TON/AFT	<u> </u>				
рĦ	*****	8.4	7.7	7,6	7.8	7.6
Temperature	°C	8,4				
Dissolved Oxygen	mg/1	•				
Hardness (CaCO3)	mg/l	100	160	122	162	120
Alkalinity (CaCO ₃)	mg/l					
NH3 -N	mg/1					
NO2 -N	mg/1	·				
NO ₃ -N	mg/1					
Kjeldahl-N	mg/l					
PO ₄ -P (Total)	mg/l					
PO4 -P (Ortho)	mg/l					
Chloride	mg/1	3	3	3	3	5
Sulfate	mg/l					
Fluoride	mg/1					
Aluminum	µg/1					
Arsenic	μg/1					
Cadmium	/ug/1					
Chromium	µg/1					
Copper	дg/1					
Iron (Dis)	بر 1/gبر					
Iron (Tot)	μg/l	420	80	. 20	D	400
Lead (Dis))1g/1					
Lead (Tot)	/ug/1					<u> </u>
Manganese	дıg/1	60	/5	9	3	3
Mercury	μg/1			ļ		
Zinc	ug/1					
MBAS	mg/1					
011 & Grease	mg/1					
Total Coliform	#100 m1					
Fecal Coliform	#100 ml					
Other (Specify)						
Source		Storet -				
Sample Date			71-09-14	71-10-13	71-11-16	71-12-14
Owner		C.1.D. #1/A				

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APPENDIX I WATER QUALITY DATA

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	Wall	Tdone464	401400	L	HEAG	Mumbas

Towns to the	77	we we	11 Identific	ation by USC	S Number	
Parameter	Units	26/46-31M1-				
Conductivity	umhos/cm	246	254	240	220	240
Residue (Total)	mg/1	2/5	151	155	177	138
Residue (180°C)	mg/l					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pH		7.5	8.3	7.8	7.8	8.1
Temperature	*C					
Dissolved Oxygen	7. 1		 			10.6
Hardness (CaCO3)	mg/1 mg/1	IG.	1:1:1		160	/34
Alkalinity (GaCO3)		196	144	/32	7.00	137
						
NH3 -N	mg/1		 			
NO ₂ -N NO ₃ -N	mg/1 mg/1		 	 		
Kjeldahl-N	mg/1			 		†
	_					
PO ₄ -P (Total)	mg/1					
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	3	6	0	3	2
Sulfate	mg/1					<u></u>
Fluoride	mg/l	**************************************				
Aluminum	µg/1					
Arsenic	بروبر 1/عبر		<u> </u>			
Cadmium	μg/1		 			
Chromium	بر 1/عر		1		·	
Copper	7.87.1 1/8لر		<u> </u>	 		
-						
Iron (Dis)	μg/1		 	 	· · · · · · · · · · · · · · · · · · ·	<u> </u>
Iron (Tot)	µg/1	0	80	.180	120	220
Lead (Dis)	/1g/1					
Lead (Tot)	μg/1	 		ļ		
Manganese	μ g/1	66	6	0	3	<u> </u>
Mercury	1/وبر					
Zinc	ر 1/ <u>ق</u> بر					
MBAS	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 m1				Ì	}
Fecal Coliform	#100 m1					
Other (Specify)						
Source		Storet -				-
Sample Date		72-01-17	72-02-16	72-03-28	72-04-19	72-05-12
Owner		C.1.D.				1
	•	#//4			<u> </u>	
	i	1/7				
			<u> </u>	1	<u> </u>	1

APPENDIX I WATE: QUALITY DATA

KIMAKT HO	DIFER			
Well	Identification	hv	TISGS	Mumber

			11 Identific	ation by USG	S Number	
Parameter	Units	26/46-31M1-				
		 				
	mhos/cm	224	244	220	240	
Résidue (Total)	mg/1	140	150	410	158	
Residue (180°C)	mg/1					
Residue, Calculated	mg/1					
Residue Loading T	on/aft					
. **		~ .		7.7	77	
pH	••	7.6	7.4		7.2	
Temperature	*C	<u> </u>				
Dissolved Oxygen	mg/1		ļ	200		
Hardness (CaCO3)	mg/1	/32	136	372-	121	
Alkalinity (CaCO ₃)	mg/l					
ин3 -и	mg/1		<u></u>			
NO ₂ -N	mg/1					
NO ₃ -N Kjeldahl-N	mg/1		 			******
vlergaur-v	mg/l					
PO ₄ -P (Total)	mg/l]				
PO4 -P (Ortho)	mg/1	 	 	 		
Chloride	mg/1	3	3	 		
Sulfate	mg/1	a	 	 		
Fluoride	mg/l		 		 	
Limorage	mB) 7	<u></u>		 		
Aluminum	րց/1		}		}	
Arsenic	μg/1					
Cadmium	μg/1					
Chromium	μg/1		 		<u> </u>	
Copper	μg/1		1	 	 	
Copper	/-B/ -		1		i	
Iron (Dis)	μg/1					
Iron (Tot)	µg/1	60	240	.60	100K	15
Lead (Dis)	/1g/1					
Lead (Tot)	ug/1				20	30
Manganese	ug/1	3	6	9	0	2_
_	•					
Mercury	$\mu g/1$				8.2	0,2 K
Zinc	$\mu g/1$					
MBAS	mg/1					
011 & Grease	mg/1					
_					1	
	100 m1					
Fecal Coliform #	100 ml			<u> </u>		ļ
_			•	1		
Other (Specify)						ļ
		}				
6		1-4.4				_
Source		Storet -	 		+	-
Sample Date		72-06-19	72-07-24	72-09-14	72-09-14	73-01-15
Owner		 	1	 	 	1.00.10
AMITEL		C.1. D.]		
		=11A				
		1		1	}	
		1				
				!		1
		<u> </u>		<u></u>		<u> </u>

APPENDIX II WATER QUALITY DATA

BASALT AQUIFER Well Identification by USGS Number Units Parameter 21/45-308 22/45-190 23/41-12N 22/42-48 280 232 690 216 256 umhos/cm Conductivity Residue (Total) mg/1 mg/l mg/l Residue (180°C) Residue, Calculated Residue Loading TON/AFT 7.6 7.3 7.3 7.05 7.56 pН *C Temperature mg/1 Dissolved Oxygen 156 104 88 Hardness (CaCO3) mg/1106 328 146 250 Alkalinity (CaCO3) mg/1110 NH3 -N mg/1 NO₂ -N NO₃ -N mg/1 mg/1 0.52 14.0 3,60 1.54 0.46 mg/1 Kjeldahl-N 0.075 PO₄ -P (Total) mg/10.140 0.010 0.13 0.068 PO4 -P (Ortho) mg/133,0 5,0 2.25 Chloride mg/112.0 Sulfate 16.9 mg/124.8 50.4 18.0 0.98 Fluoride 0.29 0.36 0,37 mg/10.198 0.35 Aluminum 1/وبر Arsenic µg/1 Cadmium μg/1 Chromium μg/1 Copper μg/1 Iron (Dis) 2/عبر μg/1 520 100 320 Ó 860 Iron (Tot) μg/1 Lead (Dis) Lead (Tot) ug/1 Manganese μg/1 Mercury **Jug/1** μg/1 Żinc **MBAS** mg/1 Oil & Grease mg/1#100 m1 Total Coliform Fecal Coliform #100 ml 5 7 23 5. 11 Other (Specify) Color DSHS -Source Sample Date 72-01-04 71-04-22 71-12-21 10-07-24 71-01-22 #/ Owner # / #1 #2 #3 Waverly Hts City of Latah City of Fairfield Cheney Spangle

APPENDIX II WATER QUALITY DATA BASALT AQUIFER

	and the same of th			
	Identification	•	***	•• •
Wali	Idontitioation	P-31	110110	Clinmbay.
MCTT	THEILETTTCALTON	υv	0000	MUMDEr

Well Identification by USGS Number Parameter Units 22/11/22 22/11/23 22/11/23							
rarameter	UIILLS	23/41-130	23/41-130	23/41-13E	23/41-136	23/41-130	
Conductivity	umhos/cm	263	256	336	284	200	
Residue (Total)	'mg/1						
Residue (180°C)	mg/1					<u> </u>	
Residue, Calculate			·		 		
Residue Loading	TON/AFT						
ρĤ		8,25	8:03	8.15	7:6	7.3	
lemperature .	°C						
Dissolved Oxygen	mg/1	٠					
Hardness (CaCO3)	mg/1	94	88	118	104	72	
Alkalinity (CaCO ₃)	mg/l	124	116	124	122	108	
NH3 -N	mg/1						
NO2 -N	mg/1						
NO3 -N	mg/l	0,97	0.93	1.18	2.14	3.05	
Kjeldahl-N	mg/l			<u> </u>	 		
PO4 -P (Total)	mg/1	0.013	0	. 0	0.313	0,551	
PO4 -P (Ortho)	mg/1		-				
Chloride	mg/l	2,5	2.25	8,0	4.0	1,5	
Sulfate	mg/1	10.92	8.25	33.25	16.8	5.4	
Fluoride	mg/l	0.35	0,36	0.36	0,32	0,21	
Aluminum	μg/1						
rsenic	ug/l						
Cadmium	ug/1						
Chromium	μg/1						
Copper	µg/1			<u> </u>		ļ	
Iron (Dis)	μg/1						
Iron (Tot)	ug/1	660	280	0	40	240	
Lead (Dis)	μg/1						
Lead (Tot)	ر 1/ <u>و</u> س			<u> </u>			
Manganese	μg/1					 	
Mercury	յսց/1						
Žinc	μg/1						
MBAS ·	mg/1						
Oil & Grease	mg/1			-			
Potal Coliform	#100 ml						
Pecal Coliform	#100 m1						
Other (Specify) Cold	or	20	15	2	10	8	
<u>.</u>			! }	1			
Source Comple Date		DSHS -		i			
Sample Date				70-07-20	70-10-16	70-10-16	
Owner		# l	#2 Chenej	-4	#/	*2	
		Cheney	Cheney	Chancy	Eustern W. State College		
		/		1	State College		

APPENDIX II WATER QUALITY DATA BASALT AQUIFER

Well	Identification	bу	USGS	Number

Parameter Uni Conductivity umhos/ Residue (Total) mg/ Residue (180°C) mg/ Residue, Calculated mg/ Residue Loading TON/AI pH Temperature (Cacos) mg/ Hardness (Cacos) mg/	71	229	24/41-23 240	25/41-25E 332	25/41-26H 150
Residue (Total) mg/ Residue (180°C) mg/ Residue, Calculated mg/ Residue Loading TON/AI pH Temperature consolved Oxygen mg/	71		240	332	150
Residue (180°C) mg/ Residue, Calculated mg/ Residue Loading TON/AI pH Temperature consistency of the physical physi	7.72				
Residue (180°C) mg/ Residue, Calculated mg/ Residue Loading TON/AI pH Temperature consistency of the physical physi	7.72				
Residue, Calculated mg/ Residue Loading TON/Al PH Temperature *(Dissolved Oxygen mg/	7.72	7.4.5		·	•
Residue Loading TON/AI PH Temperature *(Dissolved Oxygen mg/	7.72	7.4.5			
pH Temperature °(Dissolved Oxygen mg/	7.72	7.4.5			
Temperature . °(Dissolved Oxygen mg/	;	1 1		 	
Temperature . °(Dissolved Oxygen mg/	;	7.95	7.2	7.65	7.35
Dissolved Oxygen mg/			· · · · · · · · · · · · · · · · · · ·	·	
	/7 i	l			
			124	96	64
		80			
Alkalinity (CaCO ₃) mg/		100	126	53	50
NH3 -N mg/				<u> </u>	<u> </u>
NO ₂ -N mg/				<u> </u>	<u> </u>
NO3 -N mg/	1 0,2	0.9	5.2	10.3	412
Kjeldahl-N mg/	/1			ļ	ļ
PO4 -P (Total) mg/	1 0.013	0	0	0.153	0./37
PO4 -P (Ortho) mg/		<u> </u>		1	1
Chloride mg/		1,25	6.0	2.5	5,25
Sulfate mg/				17.6	23.0
		9.35	16.8		
Fluoride mg/	0,39	0.36	0,23	0.16	0,146
Aluminum µg/	/1			·	
Arsenic µg/	/1				
Cadmium /ug/					
Chromium jug/					
Copper µg/			**************************************	 	
2-6'				 	1
Iron (Dis) µg/	/1			1	
Iron (Tot) ug/		300	D	340	240
Lead (Dis) jug/		200		1 7 7 0	1 270
		 		 	
Lead (Tot) ug/				 	
Manganese µg/	1				
	,,				1
Mercury /ug/		 		 	
Žinc "už/				ļ	
MBAS mg/		<u> </u>			
Oil & Grease mg/	/1			 	
Wahal Californ 8100 -		1		1	1
Total Coliform #100 m		 	 	 	
Fecal Coliform #100 s	M	 	ļ	ļ	<u> </u>
Orbon (8-0045-1)/-1	4	15		5	55
Other (Specify)Colov	 	1		 	1 22
Source	DSHS -				
Sample Date	70-12 - 21	7 -07 21	71-08-23	71-08-17	71-08-17
Owner	#2	#1	-/	* 3	=2
	D 15	r. J. 14	E. 1.Va.	A. Auta-I	Δ
	Rockford	Eastern W.	Four Lakes	Airway	AITHOY.
		State Hospital	I	Heights	Meights
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APPENDIX II. WATER QUALITY DATA BASALT AQVIFER.

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		We	11 Identific	ation by US	GS Number	
Parameter	Units	25/42 - 29	22/45-19	23/42-5	24/40-10	24/40-114
Conductivity	umhos/cm	220	3/2	240	226	660
esidue (Total)	mg/1					
ēsidue (180°C) ēšidue, Calculatéd	mg/1 mg/1				 	
	TON/AFT	-				-,
	·					
oH	•c	7.95	7.35	7,45	7,78	8.1
emperature issõlved Oxygen	mg/l	<u></u>			 	
ardness (CaCO3)	mg/1	99	116	130	80	104
Ikalinity (CaCO ₃)	mg/l	104	106	106	108	74
H3 -N	mg/1					
02 -N 0N	mg/1 mg/1	1.9	9 / 3	/33	1.75	10
03 -N jēldāhl-N	mg/1		9.63	1.22	1/3	1.8
, 						
04 -P (Total)	mg/1	0.016	.0.065	0.052	0.026	0.085
04 -P (Ortho)	mg/1	335	2210	3,75	210	2/0
hloride Sulfate	mg/1 mg/1	2.25 5.4	36.4	40.8	10:48	36.0
Luoride	mg/1	0.28	0,24	0,145	0.38	0.14
	_					
luminum	ug/1			 		
rsenic Cadmium	μg/1 μg/1	<u> </u>		-		+
hromium	ug/1				1	
opper	μg/1					
i (D.)	, In					
ron (Dis) ron (Tot)	μg/1 μg/1	10	0	160	340	280
eād (Dis)	بر الهمر 1/عدر .	<u>'</u>	-	1	1	
ead (Tot)	ug/1			-d x a		
anganese	μg/1			<u> </u>		
ercury	ug/1					
inc	ug/1			-		
BAS ·	mg/1					
il & Grease	mg/1			}	ļ	
otal Coliform	#100 m1	1			1	
	#1.00 ml		1			+
		3	13	1	//2	C
ther (Specify) Colo	r		12	14	/3	8
ource		DSHS -				
ample Date		70-12-10	70-08-11	70-08-26	72-07-21	73-07-10
wner		<u> </u>	10-08-11			# 14
		Spokunc Int. Airport	Phangeist Res in Fairfield	Fish Lake	Eastern W.	14
		Int. Airport	Res in	1613	State Hosp	
			Laning	1	1 - in the state of	
	· · · · · · · · · · · · · · · · · · ·					
			405-66)		
or dination ness state graft in a since o				-		
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APPENDIX II WATER QUALITY DATA BASALT AQUIFER

				
Well	Identification	Ъу	USGS	Number

Well Identification by USGS Number						
Parameter	Units	24/42-12J	24/42 -16 M	24/42 - 22	24/43-28L	24/43 - 28Q
Conductivity	umhos/cm	292	620	2/2	340	380
Residue (Total)	mg/1					
Residue (180°C)	mg/1					
Residue, Calculate				<u> </u>		
Residue Loading	TON/AFT					
_11		- 1	_ ,		7.0	
pH Temperature :	°C	7.6	7.6	7.7	7.5	7,4
Temperature . Dissolved Oxygen	mg/1			 		, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
Hardness (CaCO3)	mg/1	/38	274	74	160	158
Alkalinity (CaCO ₃)	mg/1	142	254	97	168	72
NH3 -N	mg/1	1.7.	1-27-	1		
NO2 -N	mg/1			 		
NO ₃ -N	mg/1	1.05	4.9	0.75	1:09	2.0
Kjeldahl-N	mg/1	1103	7.7	1	7107	
d						
PO ₄ -P (Total)	mg/1	0.028	0,235	0.003	0.020	0.029
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	2.5	48.0	1.25	5.0	9.0
Sulfate	mg/1	7,7	58.0	7.7	16.8	22.0
Fluoride	mg/1	0.20	0.182	0.34	0.21	0.3/
	4.0		1		}	
Aluminum	ير 1/عر			ļ	ļ	
Arsenic	μg/1		 	ļ	ļ	
Cadmium	/1g/1		 	 		ļ
Chromium	μg/1	}		 		ļ
Copper	μg/1					
Iron (Dis)	1/عبر					
Iron (Tot)	μg/1	0	200	230	20	160
Lead (Dis)	ון 1 און				1	
Lead (Tot)	ug/1					
Manganese	μg/1					
Mercury	$\mu g/1$					
Žinc	1/وب ر					
MBAS	mg/l			 		ļ
Oil & Grease	mg/1	ļ		 	ļ	
Total Coliform	#100 m1	ļ				1
Fecal Coliform	#100 m1		 	-		
recar ourtion	ATOO MT	 	 	 	 	
Other (Specify) Colo	or	10	Ь	8	13	11
(-,-,-,-,-,-,-,-,-,-,-,-,-,-,-,-,-,-,-,		<u> </u>		1	 	
				}		
Source		D5145-		 		
Sample Date		71-07-21	72-06-05	70-06-03	71-06-07	71-06-07
Owner		# /	# 3	Marshall	72	724]
		2	· ·		Hangman	}
		Hallman?	PICNIL	Comm. W.S.		-
		Runchos	Pines	1	Valley Golf	1
		}	1			1
* 		L		<u> </u>	1	L

APPENDIX IL WATER QUALITY DATA BASALT AQUIFER

Well	Identification	by USGS	Number

Well Identification by USGS Number						
Parameter	Units	24/44-28N	24/44 -33 F	25/41-26	25/42-18F	25/42
Conductivity	umhos/cm	240	192	360	172	208
Residue (Total)	mg/1					
Residue (180°C)	mg/l					
Residue, Calculate	d mg/1			•		
Residue Loading	TON/AFT					
рĦ	***	7.4	7.6	7,8	7,3	8.0
Temperature	°c					
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	116	128	122	72	112
Alkalinity (CaCO ₃)		138	138	144	56	148
NH3 -N	mg/1	<u> </u>				
NO ₂ -N	mg/1		 			
NO ₃ -N	mg/1	0.029	0.6	11.2	3,4	/,2
Kjeldahl-N	mg/1	0.027		11.2		
KJerudir	mg/ x		<u> </u>			
PO ₄ -P (Total)	mg/l	0.085	0.261	0.196	0.023	0.029
PO ₄ -P (Ortho)	mg/l			<u> </u>		
Chloride	mg/1	3,25		9.0	1.0	7.0
Sulfate	mg/1	14.5	9.5	15.8	17.7	13.8
Fluoride	mg/l	0.325	0.28	0.10	0:13	0.38
Aluminum	μg/1					
Arsenic	μg/1					
Cadmium	ug/1				<u> </u>	
Chromium	ug/1					
Copper	/ug/1					
Iron (Dis)	μg/ <u>1</u>		ļ. ———			
Iron (Tot)	μg/1	140	620	180	100	40
Lead (Dis)	1/g <i>ار</i>		ļ	<u> </u>	 	
Lead (Tot)	μg/1		 	 		
Manganese	дg/1	<u> </u>	 			
Mercury	μg/l					
Żinc	µg/1					
MBAS .	mg/1					
Oil & Grease	mg/l					
Total Coliform	#100 m1					
Fecal Coliform	#100 ml					
Other (Specify) Co	lor	6	10	4.	20	2
Source		DSHS -				
Sample Date		72-10-10	71-07-29	70-09- ?	71-05-06	72-08-30
Owner		ļ		* 4	- 1	Spokane
		Tom Hodges	County Park	1 '		1 1 Amend
	Ź		Tought with	Airway	Balmer	Int. Airport
				Heights	Gardens	
						1
			<u> </u>		<u> </u>	

APPENDIX II WATER QUALITY DATA BASALT AQUIFER

Well Identification by USGS Number						
Parameter	Units	22/43-32L			24/40-2261-	-
Conductivity	umhos/cm	359	255	265	284	277
Residue (Total)	mg/1					
Residue (180°C)	mg/1	285	178		197 191	195
Residue, Calculate		266	185		191	201
Residue Loading	TON/AFT				<u> </u>	
pH .	gas-da	7,3	7.8	7.9	7.9	. 716
Temperature :	*C	9,4		14.4	13.3	20.6
Dissolved Oxygen	mg/1	,				
Hardness (CaCO3)	mg/1	118	97	100	118	118
Alkalinity (CaCO ₃)						
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO ₃ -N	mg/l	12.6	0:00		0.090	0.00
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/1					`
PO4 -P (Ortho)	mg/1	0.140	0.00			
Chloride	mg/1	13.0	3.0		5,5	3.8
Sulfate	mg/1	24.0	11.0		20	21
Fluoride	mg/1	0.3	0.5		0,2	0.3
Aluminum	1/وبر					
Arsenic	μg/1					
Cadmium	ug/1					
Chromium	μg/1		,			
Copper	µg/1					
Iron (Dis)	_{9/1} وبر	30	200.		120	
Iron (Tot)	μg/1					
Lead (Dis)	<u>1/g</u> ير					
Lead (Tot)	ر/gu <u>ر</u>					
Manganese	ду g/1			····		
Mercury	μg/1					
Żinc	1/وبر					
MBAS _	mg/1					۴. ۰
Oil & Grease	mg/1		ļ			
Total Coliform	#100 m1					٠,
Fecal Coliform	#100 ml		·			
Other (Specify) Colo	r	5	0		5	٥
Source		Van D \$ 5	VanDIS -		Van D 45	
Sample Date			7	1		F6 . 5 . 5 .
		61-05-02	59-12-01	60-05-16	57-11-06	58-07-22
Owner		W. Hendrixson	E. State -	-	45C -	
		<u> </u>	1		<u> </u>	

APPENDIX II WATER QUALITY DATA BASALT AQUIFER

		<i>JSASALT AG</i> We	11 Identific	ation by USG	S Number	
Parameter	Units	24/40-2241-			24/41-3N	
Conductivity	umhos/cm	294	296	291	220	220
Residue (Total)	mg/l					
Residue (180°C)	mg/1	204	205	. 197	163	167
Residue, Calculate		206	201	. 203		
Residue Loading	TON/AFT					
pH		8.3	8.2	8.0		. 7.9
Temperature :	°C	15.6		12,2	15.0	16:1
Dissolved Oxygen	mg/1	1.5				
Hardness (CaCO3)	mg/1	//7	118	116	90	ق ع
Alkalinity (CaCO ₃)						
NH3 -N	mg/1	ļ				
NO2 -N	mg/1			A A.C.	2 : 2 2	
NO ₃ -N Kjeldahl-N	mg/1	0.00	0.045	0.045	0,023	0.00
vlergur-M	mg/l					
PO ₄ -P (Total)	mg/1			***************************************	·	····
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	5,5	5.0	6.2	2.8	2,4
Sulfate	mg/l	21	19	19	//	
Fluoride	mg/l	0,4	0.30	0.30	0,20	0.40
Aluminum	յւց/1					
Arsenic	/1/g/1			and the same of the language of the same of		
Cadmium	ug/1					
Chromium	ر 1/gu					
Copper	µg/1					
Iron (Dis)	յսg/1	100	30	70	20	. 20
Iron (Tot)	/1g/1					
Lead (Dis)	ر 1/ورر					
Lead (Tot)	Jug/1				· · · · · · · · · · · · · · · · · · ·	
Manganese	μg/1					
Maraum	/1		l e			
Mercury Żinc	յոց/1 յոց/1					
MBAS		<u> </u>				
0il & Grease	mg/l mg/l		l			
OII & OICASC	ω 8 / τ					· · · · · · · · · · · · · · · · · · ·
Total Coliform	#100 m1					٠,
Fecal Coliform	#100 m1	ļ				
Other (Specify) Cold	or-	5	5	5		
			{			
Source		Van D. \$5 -			Vun D. 45	
Sample Date		59-09-23	60-09-12	60-11-08	47-02-26	47-08-05
Owner	····	U.S. Gov't			U.S. Gort	
		450			Forchild	
		1	}		#2	
		İ	*			
		1				
		<u></u>	<u></u>		J	

APPENDIX II WATER QUALITY DATA

BASALT AQUIFER			
Wall Tlandfilmedam	١	TICCC	37

Well Identification by USGS Number							
Parameter	Units	24/41-3N-			•	•	
Conductivity	umhos/cm	225	218.	203	215	212	
Residue (Total)	mg/1						
Residue (180°C)	mg/l	164	168	155	164	164	
Residue, Calculate					171	170	
Residue Loading	TON/AFT					· 	
рН	tion (tra)	7,6	7:7	7.6	7.7	7.7	
Temperature :	*Ç		. 1510	13.9	13.3 .	15.0	
Dissolved Oxygen	mg/1						
Hardness (CaCO3)	mg/1	89	87	80	88	85	
Alkalinity (CaCO3)					ļ	··	
NH3 -N	mg/1						
NO2 -N	mg/1						
NO3 -N	mg/1	0.00	0.203	0.790	0.068	0.045	
Kjeldahl-N	mg/1						
PO ₄ -P (Total)	mg/1	-					
PO ₄ -P (Ortho)	mg/1						
Chloride	mg/1	3.8	2.6	2.1	2.2	2.1	
Sulfate	mg/l	1		12	- 11	10	
Fluoride	mg/l	0.3	0.2	0.2	0,2	0,4	
Aluminum	μg/1						
Arsenic	μg/1						
Cadmium	ug/1						
Chromium	ير) 1/gu						
Copper	$\mu g/1$						
Iron (Dis)	րց/1	40	60	130	50	10	
Iron (Tot)	ug/1						
Lead (Dis)	<u>/1</u> g/1						
Lead (Tot)	1/ <u>g</u> بر				•		
Manganese	μg/1						
Mercury	ug/1						
Żinc	ug/1						
MBAS	mg/1						
Oil & Grease	mg/1					· .	
Total Coliform	#100 m1						
Fecal Coliform	#100 m1						
Other (Specify) Col	lor				'2	5	
.		V. 046 -					
Sample Date		Van. D. \$5 -				-1 1/ 1/	
		48-01-01	48-08-11	49-07-19	50-12-06	51-X-X	
Owner		U.S.Govt. Fairchild -					
					<u></u>		

APPENDIX II WATER QUALITY DATA

SAJALT AL	2011	EK			
	Wa11	Identification	har	TICCC	M.

	~	We	11 Identific	ation by USG	S Number	
Parameter	Units	24/41 - 3N -				
Conductivity Residue (Total)	umhos/cm mg/l	220	183	219	224	219
Residue (180°C)	mg/1	166	143	164	166	160
Residue, Calculate	d mg/1	167	143	.168	170	164
Residue Loading	TON/AFT					
_						
pН		7.5	7,5	7.9	7,2	7.7
Temperature ,	°C	16.1	15,0	16.1	•	11.1
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	89	68	86	86	86
Alkalinity (CaCO ₃)						
NH3 -N	mg/1					
NO ₂ -N	mg/1	·.				
$NO_3 - N$	mg/l	0.023	1.806	0.090	0.045	0.068
Kjeldahl-N	mg/l					
PO ₄ -P (Total)	mg/1				-	
PO ₄ -P (Ortho)	mg/1					
Chloride	mg/1	2./	2.3	2.6	2.8	2.5
Sulfale	mg/l	11	9.7	- 11	10	11
Fluoride	mg/l	0.3	0,3	0.3	0.3	0.4
A.9 2	_					
Aluminum	дg/1					
Arsenic	/1g/1		 			
Cadmium	/ug/1					
Chromium	μg/1	<u> </u>	 			
Copper	дg/1					
Iron (Dis)	μg/1	40	30	80	100	. 460
Iron (Tot)	ug/1					
Lead (Dis)	1/g <i>ار</i>			-		
Lead (Tot)	/ug/1			1		
Manganese	μg/1					
W	/1					
Mercury Žinc	/ug/1	<u> </u>) 	
MBAS	ug/1 mg/1					
Oil & Grease	mg/1 mg/1		 	<u> </u>		<u> </u>
Total Coliform	#100 m1					٠,
Fecal Coliform	#100 m1					
Other (Specify) Co	lov	5'	2	4	5	٥
Source		Van D \$5 -		•	,	~•
Sample Date		53-01-14	53-12-15	54-10-06	55-D6-16	56-06-05
Owner		}	1		75 70 16	20.00
OMITEL		U.S.Govt				I
		Fairchild -	1		-	
		= 2	!			
			į			
					1	
			J.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	I	<u> </u>	<u> </u>

Well Identification by USGS Number							
Parameter	Units	24/41-3N -					
Conductivity	umhos/cm	215	184.	214	218	208	
Residue (Total)	mg/1						
Residue (180°C)	mg/l	155	139	154	164	156	
Residue, Calculate	ed mg/1	155 158			166	157	
Residue Loading	TON/AFT						
рĦ		7,6	7,4	7.6	7.8	7.8	
Temperature :	•c	12.2	12.2	1414	20.6	15.6	
Dissolved Oxygen	mg/l	121	12.2	1717		13/0	
Hardness (CaCO3)	mg/1	85	71	85	88	82	
Alkalinity (CaCO ₃)							
NH3 -N	mg/1						
	mg/1						
NO ₂ -N NO ₃ -N	mg/1 mg/1	0,226	2:100	0.050	0.136	0.519	
Kjeldahl-N	mg/1	VILLE	E100	4.474			
v) eragur.u							
PO ₄ -P (Total)	mg/1	1		_			
PO ₄ -P (Ortho)	mg/l						
Chloride	mg/1	1.8	2.0	2,2	1.8	2.2	
Sulfate	mg/l	. 14	//	<u> </u>	10	12 .	
Fluoride	mg/l	0.3	0.2	0.3	0.4	0.4	
Aluminum	- 1/gu <u>ر</u>				•		
Arsenic	μg/1						
Cadmium	ug/1						
Chromium	μg/1						
Copper	μg/1						
••	, .						
Iron (Dis)	μg/1	120	30	170		290	
Iron (Tot)	µg/1						
Lead (Dis)	//giر						
Lead (Tot)	_ug/1						
Manganese	μg/1						
•	/-						
Mercury	μg/1						
Žinc	$\mu g/1$						
MBAS	mg/1				<u></u>		
Oil & Grease	mg/1	 	ļ				
Total Coliform	#100 m1		}			٠.	
Fecal Coliform	#100 m1		 	 			
				_	_		
Other (Specify) Co	lor	0	0	5	5	٥	
		1					
Source		Van D. \$5 -					
Sample Date					re		
		56-10-30	57-07-30	57-11-06	58-07-22	59-09-22	
Owner		U.S. 6004				_	
		Fairchild -		 	<u> </u>	† -	
		#2	1	!	ł	1	
		1	1	1	}	1	
		1	I		1		
		1	l	l	<u> </u>	1	

DAJACI MOUT	<u> </u>			
Well	Identification	hv	HSGS	Number

Well Identification by USGS Number						
Parameter	Units	24/41-3N-		24/41 - 11NI	•	
Conductivity	umhos/cm	275	262	129	121	136
Residue (Total)	mg/1					
Residue (180°C)	mg/1	168	162	. 110	110	112
Residue, Calculate	d mg/1	173	160	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	109	112
Residue Loading	TON/AFT					
_	·					- ^
pΗ	-	8.0	7.8	7,5	7,1	. 7.8
Temperature .	°C	12,2	15.6	12.8	15,0	12.6
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/l	133	125	52	51	52
Alkalinity (CaCO ₃)						
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO ₃ -N	mg/1	1.016	0.723	1.355	1,400	1.603
Kjeldahl-N	mg/1					
PO4 -P (Total)	mg/l			1		
PO4 -P (Ortho)	mg/1					
Chloride	mg/1 mg/1	5.2	3.5	1.0	1,2	2.8
Sulfate	mg/1	14	14	5.8	6.3	7.0
Fluoride	mg/1	0.3	0.2	0.2	0,2	0.5
FIUOLIGE	mg/ T		0,2	1	0,2	013
Aluminum	յսց/1					
Arsenic	ر ₈ رر					
Cadmium	/ug/l					
Chromium	1/g <i>ير</i>					
Copper	μg/1					
4	•	263			2.	, .
Iron (Dis)	$\mu g/1$	280	630	80	30	. 10
Iron (Tot)	ng/1			ļ		
Lead (Dis)	1/g/ر					
Lead (Tot)	Jug/1			ļ		
Manganese	μg/1			ļ		
Mercury	μg/1					
Žinc	ug/1					
MBAS	mg/l					~
Oil & Grease	mg/1					<u>.</u>
	M100 1		}			
Total Coliform Fecal Coliform	#100 m1 #100 m1		-	-		<u> </u>
recal Collinim	ATOO IIIT			-		
Other (Specify) Col	lor	0	5	5	0	0
			3			
Source		Van D. 25 -			ervance materials results	
Sample Date		60-11-08	61-10-10	157-11-05	58-07-22	59-09-23
Owner				14.5.60vt		31 01 23
OMITEL		U.S.Govt.		· -		
		Fairchild -	1	N. 11 3761L		1
		= 2	1	•		
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			1	ļ		
		<u> </u>	<u> </u>	1	l	!

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Well	Identification	by USG	S Number

		We	11 Identific	ation by USG	S Number	
Parameter	Units	24/41-11N1	24/41-23KI	25/40 -14RI -		
Conductivity	umhos/cm	135	•	229	240	235
Residue (Total)	mg/1					
Residue (180°C)	mg/1	107	188 187	155	167	172
Residue, Calculate	d mg/l	105	187		172	164
Residue Loading	TON/AFT					····
pH	***	7.5		7.8	7,5	8,3
Temperature ·	*C	10.0		14,4	20.0	16.1
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/l	50	135	93	100	98
Alkalinity (CaCO ₃)	_ mg/1					
NH3 -N	mg/1					
NO2 -N	mg/1					
no <mark>3</mark> -n	mg/1	1.400	5.42	0.158	0.226	0.113
Kjeldahl-N	mg/1	ļ				
PO ₄ -P (Total)	mg/1				-	
PO4 -P (Ortho)	mg/1			<u></u>		
Chloride	mg/1	2.2	1:7	3.0	5.2	3.8
Sulfate	mg/1	6.6	10	9.6	8.1	8.2
Fluoride	mg/l	0,2	0:1	.0.3	0.7	0.6
Aluminum	_{1/وبر}					
Arsenic	2/gu <u>ر</u>					
Cadmium	ر /gug/1					
Chromium	1/g <i>بر</i>		<u> </u>			<u> </u>
Copper	J18/1			ļ		
Iron (Dis)	1/ <u>و</u> بر	60	/20	40		20
Iron (Tot)	ug/1			<u> </u>		
Lead (Dis)	<i>J</i> 18/1					
Lead (Tot)	µg/1					
Manganese	дg/1		 	ļ		}
Mercury	ر 1/ویر					
Żinc	ر _{(عبر}			1	1	
mbas	mg/1					<u> </u>
Oil & Grease	mg/1			 		
Total Coliform	#100 ml					٠.
Fecal Coliform	#100 m1		ļ			
Other (Specify) Col	or	5	0	5	0	0
			4545			
Source		Van. D. \$ 5	unpublished	Van D. + 5 -		
Sample Date		60-11-08	70-11-10	57-11-06	58-07-22	59-09-23
Owner		U.S.Gov't Well 37 C+L	Four Lakes Water Dist	4.5.60vt. 87L		-
		<u></u>				

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APPENDIX III WATER QUALITY DATA BASALT AQUIFER Wall Iden

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Well Identification by USGS Number								
Parameter	Units	25/40-14RI	25/40-34NE#		+ · · ·			
Conductivity	umhos/cm	247	183	281	282	288		
Residue (Total)	mg/1					 		
Residue (180°C)	mg/l	165	203	. 206	205	198		
Residue, Calculate	d mg/1	174	 ~~	,203	209	206		
Residue Loading	TON/AFT							
_11		0						
pH		8.0	7.8	7,2	7,7	7.7		
Temperature	°C	11.1	10:6	2/.1	15.6	12.2		
Dissolved Oxygen	mg/1		 					
Hardness (CaCO3)	mg/1	97	108	112	111	[1]		
Alkalinity (CaCO ₃)								
NH3 -N	mg/1							
NO ₂ -N	mg/1							
NO3 -N	mg/1	0.090	2.71	2.7/	2.94	2.94		
Kjeldahl-N	mg/1							
PO4 -P (Total)	mg/l			İ				
PO4 -P (Ortho)			 	ļ	 	<u> </u>		
Chloride	mg/1		+	 	 			
Sulfate	mg/1	5,5	5.0	4,5	5.2	6,8		
Fluoride	mg/1	7.8	4.5	4.2	5.4	5.0		
riuoride	mg/l	0,4	0,3	0:4	0.8	0,4		
Aluminum	_{1/عبر}		Ì					
Arsenic	رور 1/وير			<u> </u>	 	}		
Cadmium	μg/1	 	 	 	 	}		
Chromium	μg/1		 	 	 	}		
Copper	μg/1		 	 	 			
- The state of the	<i>7</i> 467 ±		 	 	 			
Iron (Dis)	$\mu g/1$	40	40		0	. 30		
Iron (Tot)	/ug/l					<u> </u>		
Lead (Dis)	1/g <i>در</i>					·		
Lead (Tot)	μg/1				1	 		
Manganese	ug/1							
V	. 19	ļ						
Mercury Żinc	/ug/1				-			
MBAS	/1/gu		 	ļ	ļ	ļ <u> </u>		
Oil & Grease	mg/1	 	 	ļ	·			
Ull a Grease	mg/l	 	 			<u> </u>		
Total Coliform	#100 m1		1	ì				
Fecal Coliform	#100 ml	}	 		 	ļ		
	,		_	· · · · · · · · · · · · · · · · · · ·	 			
Other (Specify) Cold	σ Υ	5	5	_0	0	5		
-								
6						-		
Source Sample Date		Van D \$5 -						
nambre pare		60-11-08	57-11-05	58-07-12	59-09-23	60-11-08		
Owner		u.s.goit	U.S.Govt		<u> </u>	 		
		1	1			-		
		87L	876#2			1		
			1					
			<u></u>		1			

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		We	11 Identific	ation by USC	S Number	
Parameter	Units	25/41-IRI-			25/41-1061-	
Conductivity	umhos/cm	270	283.	291	373	334
Residue (Total)	mg/1					
Residue (180°C)	mg/1	188	190 199	. 193	257	239
Residue, Calculate	d mg/1	151	155	.203	ч	244
Residue Loading	TON/AFT					
рН		7.2	8.0	7.8.	7.9	7.4
<u>.</u>	°C	20.0	16.7	13.9	16.1	18.5
Temperature : Dissolved Oxygen	mg/1	20.0	18//	13,7	1011	
Hardness (CaCO3)	mg/1	124	126	124	160	150
Alkalinity (CaCO ₃)				<u>'</u>	1	
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO3 -N	mg/1	0.0	0.0	0.068	11.97	10.39
Kjeldahl-N	mg/1					
PO4 -P (Total)	mg/1		•	L		
PO4 -P (Ortho)	mg/1				1	
Chloride	mg/1	2.5	3.0	6.2	3.0	3.0
Sulfate	mg/1	6.4	9.3	11	22	20
Fluoride	mg/1	4.3	0.5	0.4	6.1	0.2
Aluminum	μg/ <u>1</u>				<u> </u>	
Arsenic	μg/1			ļ	}	
Cadmium	μg/1					
Chromium	ير 1/عبر			 	 	
Copper	μg/1			 		-
Iron (Dis)	₂ /1		560	150	30	
Iron (Tot)	μg/1					
Lead (Dis)	/1 g/1					
Lead (Tot)	μg/1					
Manganese	μg/1			<u> </u>		
Mercury	μg/1				!	İ
zinc	ug/1	 		1		
MBAS	mg/1			 	†	~
Oil & Grease	mg/1	<u> </u>		1		
Total Coliform	#100 ml	<u></u>				'.
Fecal Coliform	#100 ml			<u> </u>	ļ	ļ
Other (Specify) Col	or	0	0	5	5	0
Source		Van D. \$ 5 -				-
Sample Date		58-07-23	59-09-23	60-11-08	57-11-06	58-07-22
Owner		U.S.Govit		1	u.s. Govt	_
		07C#2		-	07L	-
		0,0 2	}			
				1		
		I	t	I	4	

DADAGE AGUER	<u> </u>			
Well	Identification	hv	USGS	Number

Parameter Units 25/4/ 106 25/4/ - 28 25/4/ - 34	330 256 237 . 7,4 15,6 128
Residue (Total) mg/l Residue (180°C) mg/l 2/9 203 /45 2/9 2/9 Residue (Calculated mg/l 223) 2/2 /445 2/9 224 Residue Loading TON/AFT PH	256 237 . 7,4 15,6 . 128
Residue (Total) mg/1 Residue (180°C) mg/1 Residue, Calculated mg/1 Residue Loading TON/AFT pH	237 . 7,4 15,6 . 128
Residue (180°C) mg/1	237 . 7,4 15,6 . 128
Residue, Calculated mg/1 Residue Loading TON/AFT pH	. 7,4 15,6 128
Residue Loading TON/AFT	15,6 128 :
PH	15,6 128 :
Temperature Dissolved Oxygen Mg/1 Hardness (CaCO3) Mg/1 Alkalinity (CaCO3) Mg/1 NO2 -N Mg/1 NO3 -N Mg/1 PO4 -P (Total) Mg/1 Chloride Mg/1 Fluoride Mg/1 Arsenic Aluminum Ag/1 Arsenic Ag/1 Cadmium Ag/1 Chromium Ag/1 Chromium Ag/1 Chromium Ag/1 Copper Ag/1 Iron (Dis) Iron (Tot) Lead (Die) Aug/1 Manganese Ag/1 Mercury Mg/1 Mercury Mg/1 Markanic Ag/1	15,6 128 :
Dissolved Oxygen mg/1 Hardness (CaCO3) mg/1 Alkalinity (CaCO3) mg/1 NH3 -N mg/1 NO2 -N mg/1 NO3 -N mg/1 Fieldahl-N mg/1 PO4 -P (Tota1) mg/1 Chloride mg/1 Fluoride mg/1 Fluoride mg/1 Aluminum	128
Dissolved Oxygen mg/1 Hardness (CaCO3) mg/1 Alkalinity (CaCO3) mg/1 NN3 -N mg/1 NO2 -N mg/1 NO3 -N mg/1 V 58 7.69 0.337 8.81 PO4 -P (Total) mg/1 Chloride mg/1 Fluoride mg/1 Fluoride mg/1 Aluminum	11.06
Hardness (CaCO3) mg/1 Alkalinity (CaCO3) mg/1 NH3 -N mg/1 NO2 -N mg/1 NO3 -N mg/1 Kjeldahl-N mg/1 PO4 -P (Total) mg/1 Chloride mg/1 Fluoride mg/1 Fluoride mg/1 Arsenic	11.06
Alkalinity (CaCO ₃) mg/1 NH3 -N mg/1 NO2 -N mg/1 NO3 -N mg/1 PO4 -P (Total) mg/1 PO4 -P (Ortho) mg/1 Chloride mg/1 Sulfate mg/1 Fluoride mg/1 Arsenic \(\nu_{1}\) Cadmium \(\nu_{1}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chromium \(\nu_{2}\) Chrom (Dis) \(\nu_{2}\) Iron (Tot) \(\nu_{2}\) Lead (Tot) \(\nu_{2}\) Manganese \(\nu_{2}\) Mercury \(\nu_{2}\) Mercury \(\nu_{2}\) mg/1 mg/1 \[\text{Ng/1} \\ N	11.06
NH3 -N mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1 mg/1	
NO2 -N mg/1	
NO3 -N	
Rjeldahl-N mg/1 m	
PO4 -P (Total) mg/1 PO4 -P (Ortho) mg/1 Chloride mg/1 Sulfate mg/1 Fluoride mg/1 Fluoride mg/1 Arsenic \(\mu_g/1 \) Chromium \(\mu_g/1 \) Chromium \(\mu_g/1 \) Copper \(\mu_g/1 \) Iron (Dis) \(\mu_g/1 \) Lead (Dis) \(\mu_g/1 \) Lead (Tot) \(\mu_g/1 \) Mercury - \(\mu_g/1 \) Mercury - \(\mu_g/1 \) Mercury - \(\mu_g/1 \)	9,5
PO4 -P (Ortho) mg/1	9,5
PO4 -P (Ortho) mg/1	9,5
Chloride mg/1 2.2 4.2 2.8 7.8 Sulfate mg/1 18 17 10 13 Fluoride mg/1 0.4 0.3 0.3 0.3 Aluminum	9,5
Sulfate mg/1 18 17 10 13 Fluoride mg/1 0.4 0.3 0.3 0.3 Aluminum µg/1	1
Fluoride mg/1 0.4 0.3 0.3 0.3 Aluminum	14
Aluminum	0.3
Arsenic µg/1	
Arsenic µg/1	
Cadmium	
Chromium	
Copper	
Iron (Dis)	
Iron (Tot)	
Iron (Tot)	
Lead (Dis)	
Lead (Tot) µg/1 Manganese µg/1 Mercury µg/1	
Manganese µg/1 Mercury - µg/1	
Mercury - µg/1	
Man Man Man Man Man Man Man Man Man Man	
MBAS mg/1	~
Oil & Grease mg/l	
W5/1	
Total Coliform #100 ml	••
Fecal Coliform #100 ml	
Other (Specify) Color 0 5 2 0	O
(opcosz), co.o.	
Source Van D. \$5	
Sample Date 59-09-23 60-11-8 53-12-16 56-10-30	58-07-23
	20-01-23
Owner U.S. Goit Fairchild U.S. Gov't	
07L #3	
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· · · · · · · · · · · · · · · · · · ·		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/41-34		25/42-25 SN4		25/42-2981
Conductivity	umhos/cm	302	327.	284	305	211
Residue (Total)	mg/1					
Residue (180°C)	mg/1	231	229	173	177	166
Residue, Calculated		226	230	ر71 <i>ر</i>	185	162
Residue Loading	TON/AFT					
рН		7.6	8.0	8.2	7.7	7.5
Temperature :	*C	13.3	12,2	14.4	10.0	
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	123	130	140	159	91
Alkalinity (CaCO ₃)	_mg/l					
NH3 -N	mg/l					
NO2 -N	mg/1	:				
NO ₃ -N	mg/1	9,03	9,71	1,152	1,264	0,452
Kjeldahl-N	mg/l					ļ
PO ₄ -P (Total)	mg/l					i,
PO4 -P (Ortho)	mg/1		, , , , , , , , , , , , , , , , , , ,			
Chloride	mg/l	7.0	9.5	3.5	3,5	1.8
Sulfate	mg/l	13	15	14	16	6.7
Fluoride	mg/l	0,2	0.3	0.1	0,0	0,2
Aluminum	րց/1					
Arsenic	μg/1 μg/1			 		
Cadmium	ug/1			<u> </u>		
Chromium	رومر 1/عر					
Copper	μg/1					
					^ -	
Iron (Dis)	лв/1 /1	130		0	90	40
Iron (Tot)	µg/1		620			·
Lead (Dis) Lead (Tot)	ا/giر 1/1					
Manganese	1/وبر 1/وبر				<u> </u>	
manganese	748/1					
Mercury	برو رر					
Žinc	$\mu g/1$					
MBAS	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 m1					٠.
Fecal Coliform	#100 ml			<u> </u>		
Other (Specify) Color	~	5	5	0	5	3
Source	•	Van D. \$5 -				
Sample Date	<u> </u>	59-09-22	/4 11 40	CC 00 0=	1-11-40	62 42 111
0	·	ļ	60-11-08	59-09-22	60-11-08	52-02-14
Owner		4.s.Govt -	-	us Gort -		Spokane Int. Airput
			Į.			TAL ULANI
		1	1			
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Well Identification	hv	2021	Numbe

	Well Identification by USGS Number						
Paramete::	Units	25/42-29RI					
Conductivity	umhos/cm	253	213	211	2/3	211	
Residue (Total)	mg/1						
Residue (180°C)	mg/1	188	155	160	153	157	
Residue, Cakculate	ed mg/1	182	163	158	152	152	
Residue Loading	TON/AFT						
•					٠.		
pH		8,0	8,1	7.8	8.1	8.0	
Temperature .	°C	12.8			11:1	11:7	
Dissolved Oxygen	mg/1	<u> </u>	<u> </u>				
Hardness (CaCO3)	mg/1		88	88	86	88	
Alkalinity ($CaCO_3$)		ļ	ļ <u>.</u>				
NH3 -N	mg/1	<u> </u>	<u> </u>			·	
NO2 -N	mg/1	<u> </u>					
NO ₃ -N	mg/1	2,71	0,723	0.587	0,948	0.948	
Kjeldahl-N	mg/l						
PO4 -P (Total)	mg/1						
PO4 -P (Ortho)	mg/1		1				
Chloride	mg/1	4,5	1.6	215	118	1.8	
Sulfate	mg/1	15	6	7.2-	6.8	6.5	
Sullace Fluoride	mg/1	0.1	0,3	0,2	0.3	0.1	
LIUOLIGE	mg/ T		1 - 5,5	1	0.5	0,1	
Aluminum	րջ/1						
Arsenic	μg/1						
Cadmium	ug/1						
Chromium	μg/1						
Copper	μg/1						
Iron (Dis)	μg/1	80	50	60	0	. 20	
Iron (Tot)	μg/1 μg/1		 	1	 		
Lead (Dis)	μg/1 μg/1	ļ		 	 		
Lead (Dis)	ر الهدر 1/gu			 		 	
			 	 	ļ		
Manganese	ng/1		 		ļ		
Mercury	μg/1	[
Żinc	μg/1						
MBAS	mg/1		·			~	
Oil & Grease	mg/1						
Total Coliform	#100 ml						
Fecal Coliform	#100 ml		<u> </u>	<u> </u>		 	
		3	2	5	. 0	0	
Other (Specify) (of	ior	 	 				
Source		Van D \$5-	<u> </u>				
Sample Date		52-10-15	53-10-27	155-01-07	5-12-22	56-12-18	
Owner					 	-	
Owiter		Spokune _ Int. Airport					
					<u> </u>		

		Ţ.	Vell Identif:	cation by U	SGS Number	
Parameter	Units	25/42-29R			<u> </u>	
Conductivity	umhos/cm	214	258.	220	71/	
Residue (Total)	mg/1		1 200:	220	216	209
Residue (180°C)	me/1	153	195	.161		
Residue, Calculate	ed mg/1		190	164	158	158
Residue Loading	TON/AFT			1	/38	162
pH	Ca 14	7.8	7.8	7.8	8.0	8.0
Temperature	•c	5.0	4.4		14.4	10.0
Dissolved Oxygen	mg/1		·		1717	1010
Hardness (CaCO3)	mg/1	91	118	93	88	90
Alkalinity (CaCO ₃)) mg/1					
NH3 -N	mg/1					
NO ₂ -N	mg/1					
NO ₃ -N	mg/1	1:42	2:19	1.3/	1,20	1.06
Kjeldahl-N	mg/l					
PO4 -P (Total)	mg/1					
PO ₄ -P (Ortho)	mg/1					-
Chloride	mg/1	2.0	3.0	2.2	2.0	2,5
Sulfate	mg/1	6.7	14	7.8	6,2	7.4
Fluoride	mg/l	0.2	0.3	0.4	0,3	0,3
			1		- 0/3	013
Aluminum	μg/1					Į
Arsenic	μg/1					
Cadmium	μg/1				 	·
Chromium	ng/1					
Copper	μg/1					
Iron (Dis)	μg/1	70	30	20	10	0
Iron (Tot)	ug/1					
Lead (Dis)	1/g <i>ונ</i>			1		
Lead (Tot)	ug/1					
Manganese	μg/1					
Mercury	μg/1					
Żinc	ug/1		 	 		
MBAS	mg/1			 		
011 & Grease	mg/1					
Total Coliform	#100 m1					
Fecal Coliform	#100 m1			 	 	
Other (Specify) Cold	m	5	0	0	5	0
Source		Van D \$ 5 -		<u> </u>		
Sample Date		57-11-06	58-09-26	59-09-29	60-09-21	61-10-03
Owner		Spokane Int Airport				

		We	11 Identific	ation by USG	S Number	
Parameter	Units	25/42-31] [25/42-27K		·	
Conductivity	umhos/cm		328	ł		
Residue (Total)	mg/1	121				
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1			,		
Residue Loading	TON/AFT					
рH		7:1	7.0			
Temperature ,	°C			~~~~	•	
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	55	144	,		
Alkalinity (CaCO ₃)	mg/1		74			
инз -и	mg/1					
NO ₂ -N	mg/1	·				
NO3 -N	mg/1	1,49	918			
Kjeldahl-N	mg/l					
		į				
PO4 -P (Total)	mg/1		0			
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	1,2	11.0			
Sulfate	mg/1	4.9	50.5			
Fluoride	mg/1	0,2	0.083			
44						
Aluminum	μg/1 /1	ļ				
Arsenic	ug/1	<u> </u>			<u> </u>	
Cadmium	μg/1				 	
Chromium	յս _ե , 1 յսց/1					
Copper	WR/ I					·
Iron (Dis)	րց/1	80	80			
Iron (Tot)	μg/1	i	1			
Lead (Dis)	ارودر 1/ودر					·
Lead (Tot)	/ug/1				·	
Manganese	رومر 1/8لا		 			
	<i>)</i> 0,					
Mercury	μg/1		1		1	1
Žinc	ug/1					
MBAS	mg/1		1			~ .
Oil & Grease	mg/1					- ,
Total Coliform	#100 m1			<u> </u>		•,
Fecal Coliform	#100 m1			[
			4			
Other (Specify) Colo	~		ļ		<u> </u>	
		1		}		1
_		111 4 6 6	Della		1	1
Source		W+M	DSHS		 	
Sample Date		42-07-29	71-11-09	•		
Owner		Geiger	120			1
		Field	? Spring			}
		1	School			
						[
		1				1
				<u> </u>	<u> </u>	<u> </u>
						,

APPENDIX III

WATER QUALITY DATA

LITTLE SPOKANE AQUIFER

Well Identification by USGS Number

Well Identification by USGS Number						
Parameter [Inits	27/43-22M	27/43-32K	27/43-33B	27/43 - 34H	28/42-02M
Conductivity umbo	s/cm	400	440	200	370	180
	ng/1					
	ng/1					
	ng/1			•		
Residue Loading TON/	AFT					
pH ,		7.9	7.7	7.0	8.0	7,8
Temperature :	°C				,	
	ng/1					
	ng/1	188	224	88	206	86
•	ng/1	. 168	189	96	200	82
	ng/1					
	ng/1	·,				
	ng/1	11,2	1.82	4,2	0.87	25,8
Kjeldahl-N n	ng/1					
	1/g	0.300	0,094	0.094	.104	.17.3
	ng/1					
	ng/1	3,0	410	5.75	2,5	1,5
	ng/1	17.4	23.9	14,4	18.3	8.6
Fluoride	ng/1	0,20	0,22	0,48	0:10	0.108
	1g/1					
	1g/1					
	1g/1					
	1g/1	<u></u>		! 		
Copper	1g/1					
	1g/1					
	1g/1	O	80	0	180	0
	1g/1					
	1g/1			ļ		
Manganese	1g/1					
	ug/1					
	ug/1					
	ng/1					
Oil & Grease	ng/1	ļ			ļ	
Total Coliform #100) m1					
) m1					
Other (Specify) Color		8	Ч	o o	5	3
		Down				
Source		DSHS-		-		
Sample Date		70-08-03	71-08-11	70-09-30	72-10-10	70-10-20
Owner	 	+ 9	+ 1	# 8	# 1	* 3
		Rivernew Hills	Whitworth	Whitworth	Kellogg	Deer Park

APPENDIX III WATER QUALITY DATA LITTLE SPOKANE AQUIFER

		We	11 Identific	ation by USG	S Number	
Parameter	Units			28/43-23M	·	27/43 -10 J
Conductivity	umhos/cm	180	184	280	180	410
Residue (Total)	mg/1					
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1			•		
Residue Loading	TON/AFT					
pН		1.8	7.7	7.9	7.35	8,3
Temperature .	°c					<u>*</u> -
Dissolved Oxygen	mg/1		 			
Hardness (CaCO3)	mg/1	92	94	184	88	1/2
Alkalinity (CaCO ₃)	mg/1	108	92	120	70	220
		100		120	10	
NH3 -N	mg/1			ļ		
NO ₂ -N	mg/1	77.7	12.6	15.05	/2 6	
NO ₃ -N Kjeldahl-N	mg/1	13,4	13.5	5,08	/2.8	1.75
Kleidaur-N	mg/l					
PO4 -P (Total)	mg/l	0.111	0.104	0.186	0.147	0,006
PO4 -P (Ortho)	mg/l		<u> </u>			
Chloride	mg/1	.75	115	0.5	0	5.9
Sulfate	mg/1	9,0	10.6	42,2	9.7	26,3
Fluoride	mg/l	0.11	0.115	0.09	0.12	0.17
Aluminum	μg/l				d permitte de la constante de	
Arsenic	ug/1			<u> </u>		
Cadmium	/ug/1				<u> </u>	
Chromium	ug/1			 	<u> </u>	
Copper	μg/1				 	<u> </u>
Обррег	J. 61 -				İ	
Iron (Dis)	μg/1	<u> </u>			<u> </u>	
Iron (Tot)	$\mu g/1$	60	40	0	0	240
Lead (Dis)	, 1g/1					
Lead (Tot)	ug/1					
Manganese	μg/1					
Mercury	μg/1					
Žinc	μg/1 μg/1	 		1		
MBAS .	mg/1		 			
Oil & Grease	mg/1		 	 	 	
OTT A ALEASE	m2/ T		<u> </u>		 	
Total Coliform	#100 m1		<u> </u>	<u> </u>		
Fecal Coliform	#100 m1	<u></u>	<u> </u>	 	 	
Other (Specify) Col	or	5	4	5	3	0
Source		DSHS-				
Sample Date		70-10-	70-10-20	70-08-03	70-10-20	73-06-21
Owner		21	<i>"</i> 2	710	1 +4	= 1
		David Politi)	2	1	Wahoo
		Deer Park	Deer Park	hattaray	Deer Park	•
				H. is		Add
		<u> </u>	J		<u> </u>	J

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APPENDIX III WATER QUALITY DATA LITTLE SPOKANE AQUIFER

Parameter	Units	17/43-26M	27/43-29M	27/43 -345	27/43 - 34K	28/42-36A
Conductivity	umhos/cm	370	360	420	340	200
Residue (Total)	mg/1					
Residue (180°C)	mg/1			,		
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pH		7.9	8,3	7.8	7.7	. 7.0
l'emperature	°C					
Dissolved Oxygen	mg/1					
lardness (CaCO3)		202	196	252	232	82
Alkalinity (CaCO ₃)	mg/l	. 196	176	186	170	81
NH3 -N	mg/1					
NO ₂ -N	mg/1			 		
NO ₃ -N	mg/1	0.038	5,2	3.0	1.68	2.01
Kjeldahl-N	mg/1	0,038	3,2	3.0	7168	C.61
PO4 -P (Total)	mg/1	0.127	.042	0,036		0
PO4 -P (Ortho)	mg/1	0.727	1072	0,036		<u>-</u>
Chloride	mg/1	2.5	1,5	19.0	2	
Sulfate					3.5	110
	mg/1	26.8	18.3	22,6	20.6	7,3
Fluoride	mg/1	0.124	0,18	0,20	0,143	0.168
Aluminum	µg/1					
Arsenic	μg/1		·			
Cadmium	ug/1					
Chromium	μg/1		!			1
Copper	ug/1					
Iron (Dis)	μg/1					
Iron (Tot)	ug/l	260	140	200	200	20
Lead (Dis)	1/gر .					
Lead (Tot)	/ug/1					{
langanese	μg/1					
Mercury	ug/1					1
Žinc	ug/1		1	1		†
MBAS .	mg/1	 				
Oil & Grease	mg/1					
Total Coliform	#100 m1					
Fecal Coliform	#100 ml					
recal collidim	ATOO III	ļ		 		
Other (Specify) Col	or	7	2	5	8	6
.		Dene -				
Source		DSHS -	ļ			
Sample Date		72-10-10	72-06-03	72-10-10	72-10-10	12-08-29
Owner		" ,	E. 0	M+ Spokane		Wash State
		Colbert Elem. School	Pleeger	Motel	Park Cafe	Dragoon Creal2

APPENDIX III WATER QUALITY DATA LITILE SPOKANE AQUIFER

The state of the s

	2771	S SPOKANE We		ation by USG	S Number	
Parameter	Units	29/43-4				=
						
Conductivity	umhos/cm	752				
Residue (Total)	mg/1					
Residue (180°C)	mg/1					
Residue, Calculate						
Residue Loading	TON/AFT					
pН		6.9				
Temperature :	°c					
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/l	88				
Alkalinity (CaCO ₃)		. 69				
NH3 -N	mg/1					
NO ₂ -N	mg/1	· · · · · · · · · · · · · · · · · · ·				
NO3 -N	mg/1	1,4				
Kjeldahl-N	mg/1					
-	<u>.</u>					
PO ₄ -P (Total)	mg/1	0.01				
PO ₄ -P (Ortho)	mg/1					
Chloride	mg/1	1,5				
Sulfate	mg/l	7.2		Ì		
Fluoride	mg/l	0,126		 		
Aluminum	μg/1		ĺ			
Arsenic	/ug/1		<u> </u>	 		
Cadmium	/ug/1					
Chromium	ug/1				İ	
Copper	дв/1 лg/1			 		
oopper	7-67					
Iron (Dis)	μg/1					}
Iron (Tot)	/ug/1	180				
Lead (Dis)	,)1g/1	100		 		
Lead (Tot)	ug/1			 		
Manganese	μg/1		 			
	7-18-7		 	 	 	i
Mercury	/ug/1				}	!
Żinc	jug/1			-		
MBAS	mg/1				1	
Oil & Grease	mg/1			1		
			<u> </u>			
Total Coliform	#100 m1			[1
Fecal Coliform	#100 m1			1		
Other (Specify) Col	or	4			j	1
				}		
			1	•	1	ļ
Source		DSHS	1	1		j
Sample Date			 	1	1	
		73-05-09				
Owner		Carmil		-		
		Estates		1	}	
		C>14(T)		}		1
			1	i		1
			1	}		
			!	<u> </u>		

THE SECTION OF THE PROPERTY OF

APPENDIX IV. WATER QUALITY DATA

DIMER AQUIFE.	<u> </u>			
Wall Ta	antification	h	TICCC	Marsh

Parameter Unit Conductivity umhos/c	B 25/45-14N	25/45-2301	Newman Lake	action lill	l .
Conductivity umhos/c			,	26/42-144	26/42 -16
	n 120	124	200	220	294
Residue (Total) mg/1					
Residue (180°C) mg/1					
Residue, Calculated mg/1		 			
Residue Loading TON/AFT			<u> </u>		
worder noderne touth					
pH	7.3	7.1	6.4	7,2	7.5
Temperature . *C					
Dissolved Oxygen mg/1					
Hardness (CaCO3) mg/1	88	56	56	120	188
Alkalinity (CaCO ₃) mg/1	58	56	42	108	164
	1	†	72	108	167
NH3 -N mg/1		ļ			
NO ₂ -N mg/1		<u> </u>			L
$NO_3 - N$ mg/1	DISS	0.92	7.80	2,7	1,43
Kjeldahl-N mg/1	ļ	 			
PO4 -P (Total) . mg/1	0.016	0.062	0,036	0.114	0
PO4 -P (Ortho) mg/1	Kiii K			**************************************	
Chloride mg/1	4.5	8.5	34,0	6,5	6.0
Sulfate mg/1	18,0	4.9	8.9		
Fluoride mg/1				7,4	23.7
ridoride m8/r	0,06	0,082	0.08	0118	0,193
Aluminum / / / / / / / / / / / / / / / / / / /	1	l			1
Arsenic µg/1		 	 		
Cadmium µg/1		 	 	····	
, — · · · · · · · · · · · · · · · · · ·	ļ	 	ļ		
Chromium µg/1		 	ļ		
Copper µg/1	 	 			
Iron (Dis) µg/1					ŀ
Iron (Tot) µg/1	0	40	170	70	180
Lead (Dis) /1g/1		T			180
Lead (Tot) µg/1		 	 		
· · · · · · · · · · · · · · · · · · ·	 	 			
Manganese µg/1		 -			
Mercury jug/1					
Žinc µg/1			<u> </u>		
MBAS mg/1		 			
Oil & Grease mg/l		 	 		
orr a grease mg/r					
Total Coliform #100 ml	(
Fecal Coliform #100 ml					
	17		-	2	
Other (Specify) Color	17	8	5	3	5
Sources	DSHS -				-
Sample Date	71-04-20	71-04-13	69-04-28	71-09-20	73-04-23
Owner		 	Newman		Fairview
	Liberty Lake	Liberty			1 -
	Lake	Lake	Lake		Additions
	1				
		1			
				}	

A STATE OF THE PROPERTY OF THE

APPENDIX TV WATER QUALITY DATA OTHER AQUIFERS

Well	Iden	tificat:	ion by	USGS	Number

		We	11 Identific	ation by USO	S Number	·
Parameter	Units	26/44-290	26/44-32R	26/44 - 32 Q	26/45-256	27/41-276
Conductivity	umhos/cm	528	400	500	200	460
Residue (Total)	mg/1					
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
••		— ,		٠.		
pH	°C	7.1	7.8	7.3	7,2	7.5
Temperature	7.		<u> </u>	<u> </u>		
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	246	184	256	102	128
Alkalinity (CaCO ₃)	mg/1	130	178	206	86	184
NH3 -N	mg/1				<u> </u>	
NO2 -N	mg/1	0.0		11 5	ļ	
NO ₃ -N Kjeldahl-N	mg/1 mg/1	24.8	5,75	11.3	1.04	9,75
vlergaur-N	mg/r		 	 		
PO ₄ -P (Total)	mg/1	.068	.052	.082	,072	1.226
PO4 -P (Ortho)	mg/1					1.46
Chloride	mg/l	10.25	3.25	8.5	0	2.25
Sulfate	mg/l	37.1	2218	25.5	12.7	11.9
Fluoride	mg/l	0.148	0.17	0.07	0.13	0.114
Aluminum	μg/1		l			
Arsenic	ر g/1					
Cadmium	/ug/l					
Chromium	ر _{ير} 1/وير					
Copper	μg/1					
Iron (Dis)	μg/1					
Iron (Tot)	$\mu g/1$	460	120	180	140	0
Lead (Dis)	1/gi <i>ر</i>		<u> </u>	<u> </u>		
Lead (Tot)	μg/1			}		<u> </u>
Manganese	1/8بر				ļ	<u> </u>
Vanauur	/1					
Mercury Żinc	/ug/1			-	 	<u> </u>
MBAS .	μg/1 mg/1					
Oil & Grease	mg/l		 -	 		
OII a diease	шВ/ т			 		
Total Coliform	#100 m1			1		
Fecal Coliform	#100 m1			1		
				1		
Other (Specify) Cold	or	7	20	7	/3	9
, , , , , , , , , , , , , , , , , , , ,				1	1	
				1		
Sources		D5H3-		!		
Sample Date		71-04-14	71-05-05	71-09-27	71-04-01	70-08-27
Owner				 	 	
		Marvin	settlement	Fleasant	Moab	Cutler
	ļ	Bartel	settiement	Trees	!	Esenbarth
			j 1	į	1	
				j		
	Ì			1		

APPENDIX TV WATER QUALITY DATA

OTHER AQUIFERS

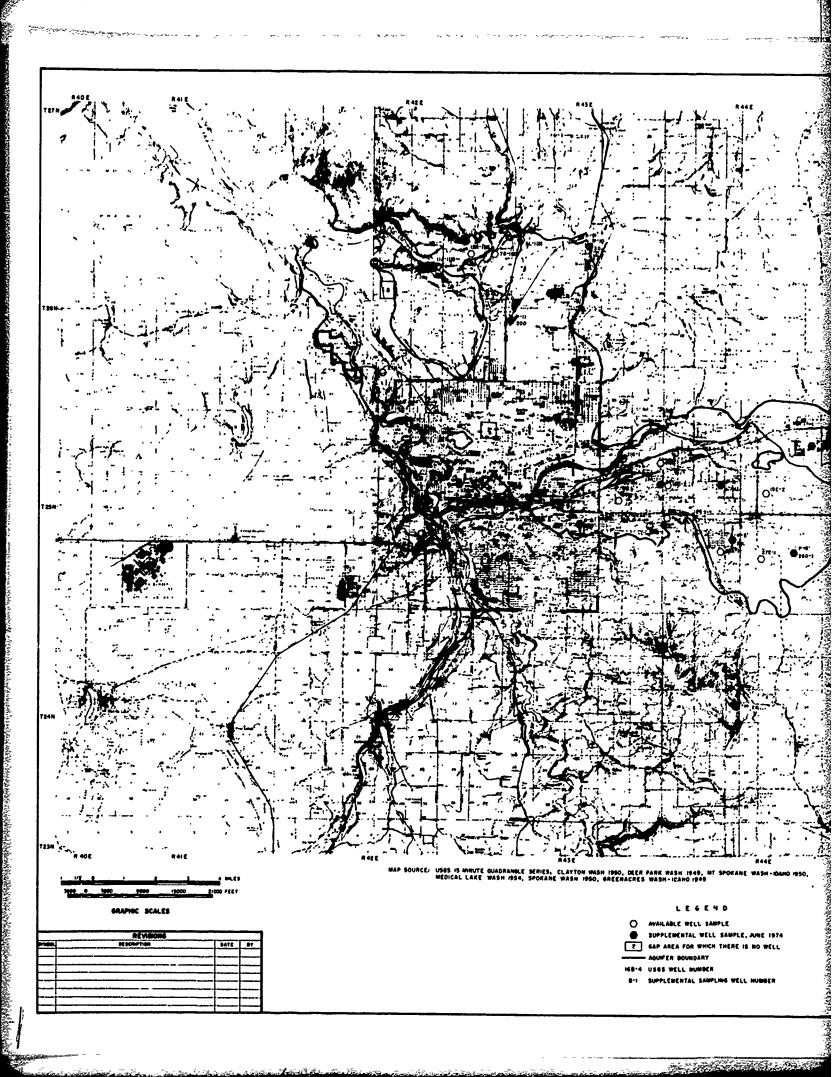
Well Identification by USGS Number						
Parameter	Units	25/45-25F	25/45-256	26/44 - 28 K	26/45-246	27/42 - 21R
Conductivity	umhos/cm	220	214	340	246	240
Residue (Total)	mg/1					
Residue (180°C)	mg/1					
Residue, Calculate	d mg/1					
Residue Loading	TON/AFT					
pH .		7.4	7.8	7.4	7.3	. 7.6
Temperature .	°C				·	
Dissolved Oxygen	mg/1					
Hardness (CaCO3)	mg/1	88	172	177	144	196
Alkalinity (CaCO ₃)	mg/1	92	153	102	112	168
NH ₃ -N	mg/1			•		
NO ₂ -N	mg/1	:				
NO_3^-N	mg/1	0.58	<.01	15.9	1.4	0.21
Kjeldahl-N	mg/1					
PO ₄ -P (Total)	mg/1		.098	.127	1059	.104
PO ₄ -P (Ortho)	mg/l					<u></u>
Chloride	mg/1	3,5	3,5	8.5	39.5	3.5
Sulfate	mg/1	26.3	16.8	24,8	13.3	21.3
Fluoride	mg/l	1.08	1.07	0,20	0.09	0.11
Aluminum	μg/1				•	
Arsenic	μg/1					
Cadmium	ug/1					-
Chromium	μg/1					
Copper	μg/1					
Iron (Dis)	μg/1					
Iron (Tot)	/ug/1	240	10	60	0	60
Lead (Dis)	1/g <i>در</i>		<u> </u>	<u> </u>		<u> </u>
Lead (Tot)	ug/1					
Manganese	µg/1	,	ļ			
Mercury	μg/1					
Żinc	ر 1/ویر					
MBAS .	mg/1					
Oil & Grease	mg/1					
Total Coliform	#100 m1					
Fecal Coliform	#100 m1					
Other (Specify) Cold	pΥ	18	4	4.	6	4
Sources		DSH5			 	-
Sample Date		71-06-07	72-09-15	71	71-08-02	71-09-20
Owner		Liberty Lake Park	Liberty Lake Park		Moab	Spokune Lake Park
		<u> </u>	<u>L</u>	<u> </u>	<u> </u>	<u>L</u>

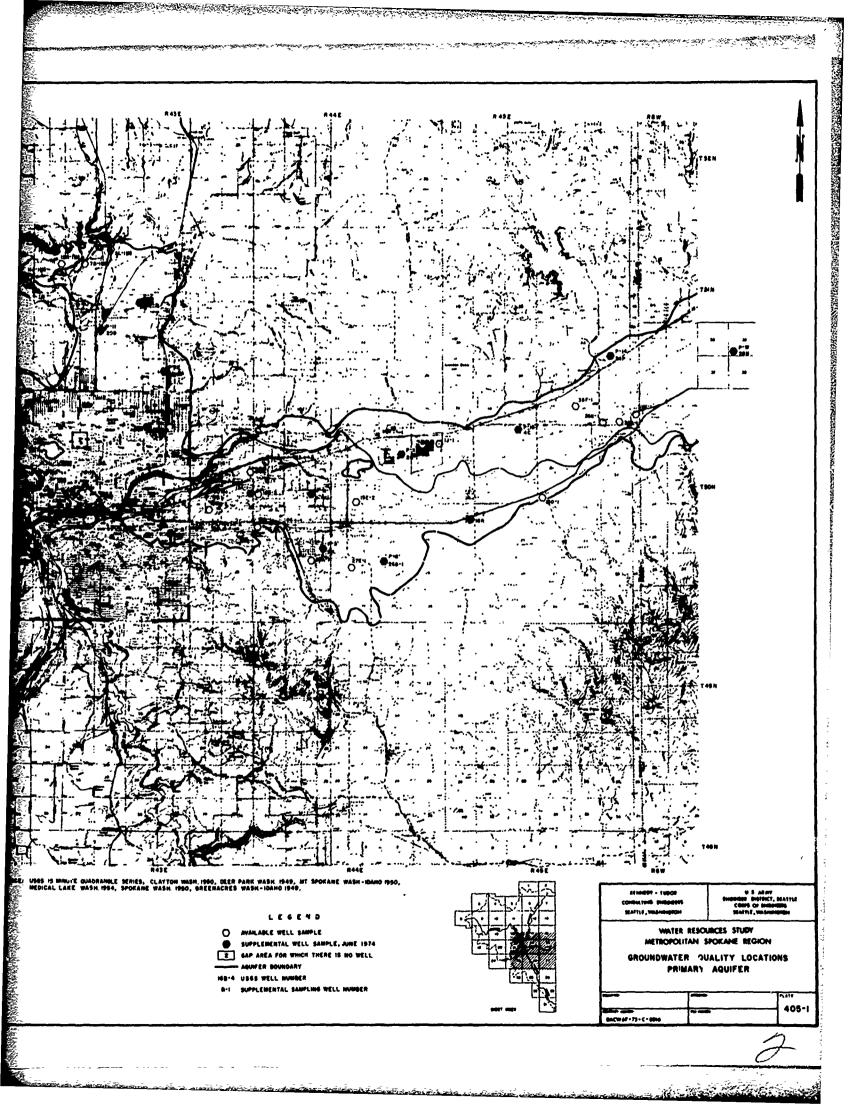
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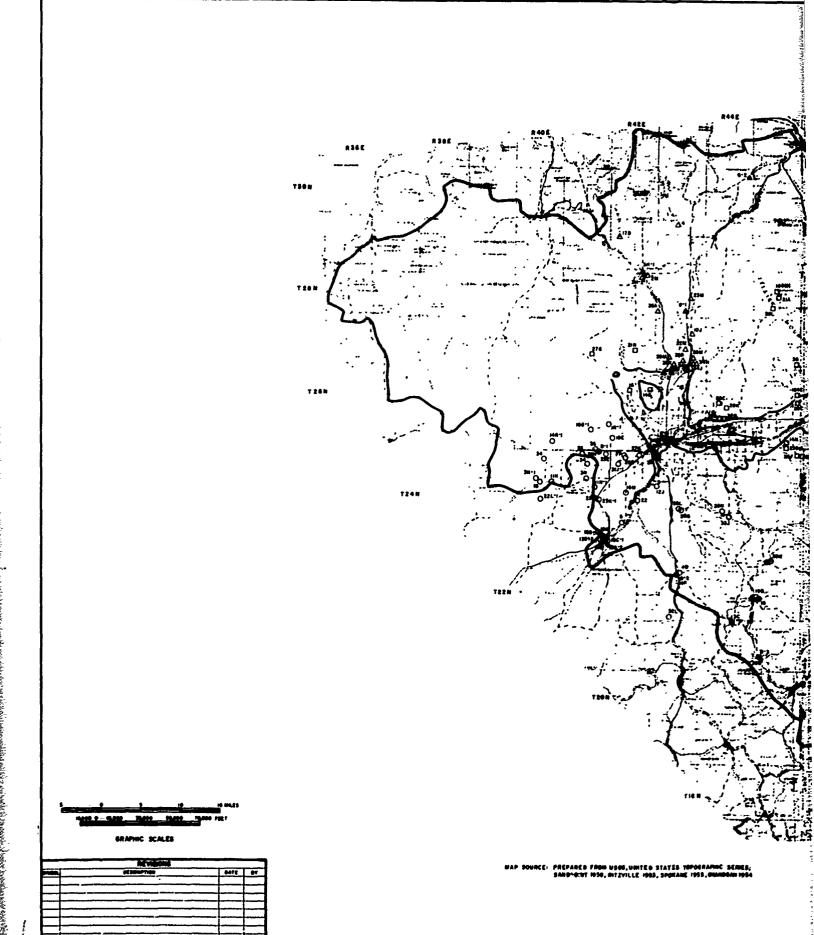
APPENDIX IV WATER QUALITY DATA

OTHER AQUIFERS

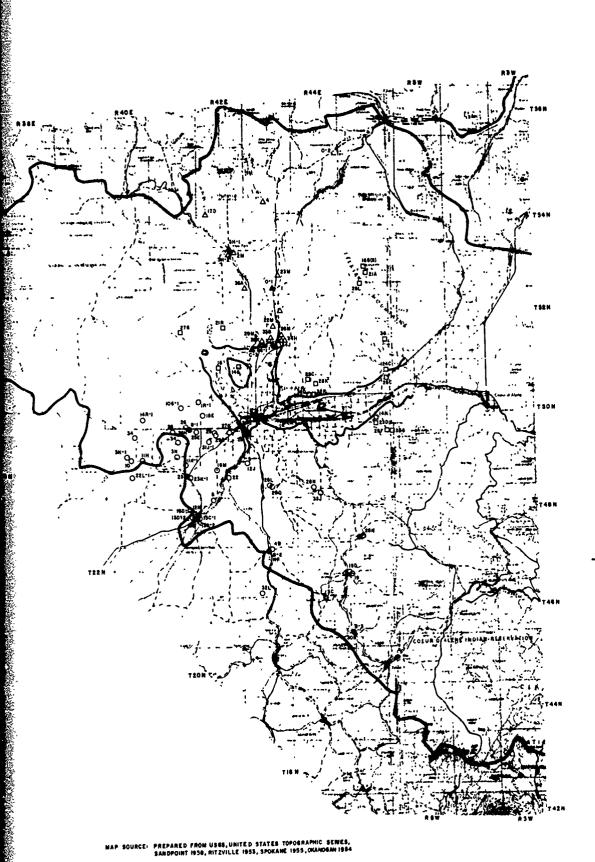
		We We		ation by USG	S Number	
Parameter	Units	27/45-36	28/45-164 spring	28/45 - 21A	28/45-28L	-
Conductivity	umhos/cm	400	22	18.4	50.8	
Residue (Total)	mg/1					
Residue (180°C)	mg/l					
Residue, Calculate				•		
Residue Loading	TON/AFT					
рН		7.9	7.1	7./	7.3	
Temperature	°C					
Dissolved Oxygen	mg/l					
Hardness (CaCO3)	mg/1	240	68	32	68	
Alkalinity (CaCO ₃)	mg/l	200	36	34	66	
NH3 -N	mg/l					
NO ₂ -N	mg/1					
NO ₃ -N	mg/1	2.8	0.61	0,17	0.32	
Kjeldahl-N	mg/1					
PO4 -P (Total)	mg/1	1023	0	-0/0	,033_	
PO4 -P (Ortho)	mg/1					
Chloride	mg/1	8.5	5,5	2.5	5.0	
Sulfate	mg/1	2510	1.9	14,2	0,3	
Fluoride	mg/l	0.35	0.13	0.30	0.13	
Aluminum	րg/1					
Arsenic	μg/1					
Cadmium	μg/1					
Chromium	дg/1 лg/1			 		
Copper	дg/1		-			
Coppen	7-61-					
Iron (Dis)	μg/1					
Iron (Tot)	/ug/1	80	260	600	380	
Lead (Dis)	/1g/1		L			
Lead (Tot)	μg/1					
Manganese	лg/1					
Mercury	μg/1					İ
Źinc	ug/1					
MBAS	mg/1				l -	
Oil & Grease	mg/l					
Total Coliform	#100 ml					
Fecal Coliform	#100 ml					
Other (Specify) Col	or	0	0	5	٥	
		-				
Sources		D5H5-		<u> </u>		
Sample Date			71-04-19	73-06-19	72-01-19	
Owner		, , 00-21	00 77	1 12 00 17	17-06-17	
2 2 . — -					Salar Artista	
				İ		
					<u> </u>	<u>. </u>







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PRIMARY AQUIFER BOUNDARY-FOR DETAILS SEE PLATE 405-F

AVAILABLE WELL SAMPLE, 6/SALT AQUIFER

AVAILABLE WELL SAMPLE, LIFFE SPOKANE AQUIFCR AVAILABLE WELL SAMPLE, OTHER SQUIFERS SUPPLEMENTAL WELL SAMPLE, JUNE 974, BASALT AQUIFER

SUPPLEMENTAL WELL SAMPLE, JUNE 974, BASALT AQUIFER SUPPLEMENTAL WELL SAMPLE, JUNE 1974, LITTLE SPOKANE AQUIFER

the second of th

USSS WELL NUMBER
SUPPLEMENTAL SAMPLING WELL NUMBER
WELLS 22/45-18 & 100: 24/41-23 & 23K-1, "3/41-28 & 28K-1
25/42-28 & 23K-1 BIGHT REFER TO ONE WE. L EACH, BUT
THIS DIFFERENCIATION CARNOT BE PRECISELY DETERMINED
FROM THE DATA SOURCES.

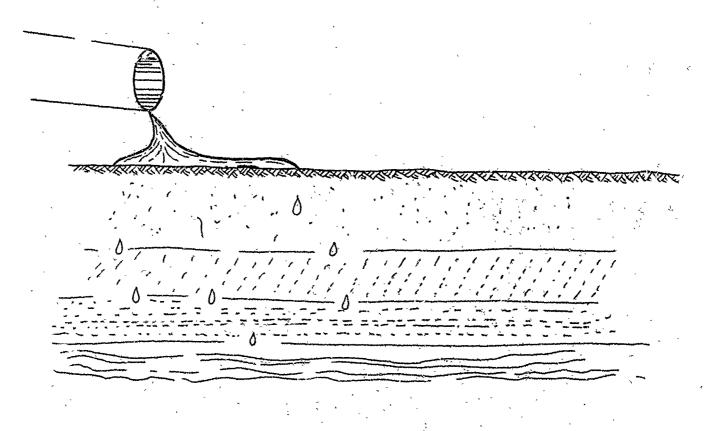
CONSULTING ENGINEERS MATTLE, WASH

U S ABRIT M DISTRICT, SEATTLE MS OF SHOOMERS

WATER RESOURCES STUDY METROPOLITAN SPOKANE REGION

GROUNDWATER QUALITY LOCATIONS BASALT AND OTHER AQUIFERS

405-2 BACW 67 - 73 - C - 6000



OCCUPA 608.1

THE EFFECT OF SURFACE APPLIED WATERS ON GROUNDWATER
QUALITY IN SPOKANE VALLEY

WATER RESOURCES STUDY METROPOLITAN SPOKANE REGION

SECTION 608.1

THE EFFECT OF APPLIED SURFACE WATERS ON GROUNDWATER QUALITY IN THE SPOKANE VALLEY

18 April 1975

By David K. Todd, Consulting Engineer, Berkeley, California in cooperation with Kennedy-Tudor

Department of the Army, Seattle District Corps of Engineers Kennedy-Tudor Consulting Engineers

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SECTION 608

THE EFFECT OF APPLIED SURFACE WATERS ON GROUNDWATER QUALITY IN THE SPOKANE VALLEY

Introduction

A significant body of groundwater quality data on the Spokane Valley aquifer existed prior to the inception of this study. Concurrent with the study, the USGS-EPA groundwater quality monitoring program added extensively to these data. Additional groundwater quality sampling for this study further supplemented the areal coverage of the USGS-EPA program. Inspection of these accumulated data, unsupported by other considerations, does not permit a conclusive answer to the question: "Are the individual disposal systems (septic tanks and drain fields), which serve approximately 55,000 persons in the Spokane Valley, significantly affecting groundwater quality?"

The investigations of Crosby, et al (6,7,8)* reported a moisture deficit beneath drainfields in summer, suggesting that the septic tank moisture addition was being totally removed by evapotranspiration. On the other hand these same studies reported no significant salt build-up in the soil, which suggests that at some time of the year any salt accumulations are flushed downward. This evidence appears to be contradictory regarding the ultimate fate of the drainfield effluent and its associated dissolved salts.

The known facts about the hydrology of the Spokane Valley aquifer indicate that a flow of approximately 1,000 cubic feet per second (cfs) enters the study area at the Idaho boundary, largely unaffected by man's surface activities up to that point, except for some irrigation in the vicinity of Post Falls. This large flow passes westward under the Spokane Valley, first beneath an area of irrigated agriculture and then under an extensive suburban area serviced entirely by on-site sewage disposal systems. The flow turns northward under the City of Spokane, which is essentially all served by a sewage collection system, and continues northward, again under unsewered suburbs in North Spokane, to discharge into the lower reaches of the Little Spokane River. Throughout the Spokane Valley, the aquifer materials of high permeability extend from the ground surface down to the water table with no known extensive layers of low permeability materials to act as barriers to vertical migration of moisture. There is minor recharge of the aquifer by the Spokane River in the area east of Greenacres. There is a major discharge of some 500 to 600 cfs from the aquifer to the Spokane River along most of the reach from Greenacres to Spokane Falls. The aquifer is extensively penetrated by wells from the Idaho line to the

^{*}See List of References for identification of citations.

Spokane city limits, but there are few wells inside the City and only a moderate number north of the City.

These hydraulic conditions, most significantly the large aquifer flow and the complex interchange with the river, indicate that if septic tank drainfield percolation does reach groundwater its probable effects on chemical quality of the groundwater would be difficult to identify unless supported by such gross indicators as bacterial contamination or detergents. These gross indicators are subject to a high degree of removal by the deep soil layer above the water table, and have not been detected in routine well water examinations. Therefore, it was decided to study analytically the drainfield percolation mechanism to determine if this effluent could be reaching the groundwater; and if so, to estimate the order of magnitude of the changes in groundwater chemical quality that could take place, with emphasis on the total dissolved mineral content, which is known to be largely unaffected by percolation through soil.

The purpose of this portion of the total Water Resources Study is first to determine from calculations of the evapotranspiration mechanism whether moisture is available for percolation under suburban development conditions after evapotranspiration needs are satisfied. If net percolation quantities are determined to exist, a second objective is to estimate the total dissolved solids load which would be carried by this leachate to the water table and its probable effect on the quality of groundwater. A third objective is then to reexamine the existing water quality data and previous related investigations in the light of this knowledge for possible confirmation or to suggest the kind of field investigation needed for confirmation.

It is not the purpose of this portion of the total study to draw conclusions from the above described analyses in either the public health or wastewater management fields. This portion of the study primarily is concerned with the analytical determination of whether the liquid component of septic tank effluent and its accompanying dissolved salts are or will be reaching the water table in the Spokane Valley.

Summary of Conclusions

An analysis of the evapotranspiration mechanism for urban and suburban land use conditions in the Spokane Valley indicates that a significant proportion of the leachate from septic tank drainfields is available for percolation to the water table of the groundwater. The analysis of soil moisture behavior is based on a conservative interpretation of data and a conservative application of soil moisture transport technology. Notwithstanding the conservative approach, the analytical results indicate a net surplus of leachate available for percolation to groundwater.

An evaluation of the physical transport mechanism under pre-

vailing soil and moisture conditions indicates that the dissolved mineral salt content of septic tank effluent is the most reliable indicator for identifying the arrival of leachate at the water table. Since the water supply for the Spokane Valley is drawn from groundwater, the total dissolved solids content of the septic tank effluent is the sum of the salt* content of the water supply and that added by domestic use, which in this case makes the salt content of septic tank effluent more than double that of the water supply. The estimated quantity of leachate reaching the groundwater is sufficient to transport all of the dissolved salts downward with no accumulation in the soil. Therefore, the entire dissolved solids content of the septic tank effluents should be reaching the groundwater, except for minor dissolved constituents such as phosphates, which are known to react with soil particles, or nitrates, which may be partially taken up by plant roots.

This analysis concludes that there is, as a result of the indicated percolation of septic tank effluent and other applied surface waters, an accumulation of these leached flows joining the surface of the native groundwater as it progresses westerly through the valley. The depth and dissolved solids concentration of a layer of leachate accumulated along a groundwater flow line, which for simplification of representation is assumed to be unmixed, is evaluated for both present and forecast year 2020 conditions. Because these leached flows have a significantly higher dissolved solids content than the native groundwaters, their presence should be in evidence as a layer or zone of significantly higher dissolved solids content. This analysis does not address the mechanism of mixing below the water table, because of the lack of depth specific groundwater quality data, but assumes that the depth and degree of mixing are limited and not sufficient to eliminate a distinctive graduation of dissolved solids concentration. The calculated results at the downstream end of the groundwater flow line indicate that the maximum range of solids concentration expected at present is up to 93 milligrams per liter (mg/l) above the mean natural groundwater background of 155 mg/1. The forecast year 2020 maximum incremental concentration is calculated to be 101 mg/l above the natural background.

Confirmation of these concentration differentials from existing water quality data is difficult and uncertain due to lack of information about the penetration of the groundwater by the wells from which samples were drawn. Depending upon depth of penetration of the groundwater and construction details which determine the levels from which water may enter the casing, a well may be pumping native water from beneath the leachate layer, water from the leachate layer itself, or a mixture from both. Despite this limitation, existing water quality data appear to confirm the calculated trend which indicates an increase in total dissolved solids as the groundwater flows westward under the areas served

^{*}Salt is used herein in the general sense to mean any dissolved inorganic compound and is not limited to sodium chloride.

by septic tanks. The available data, however, contain isolated anomolous results which cannot be interpreted for lack of correlative information, and a specific sampling program addressed to meeting the inadequacies of the existing data is recommended. The most critical data need is for depth specific groundwater quality data which suggests that additional groundwater sampling be performed with strict attention given to depth of sample and that simultaneous measurements be made at a number of locations along the groundwater flow path through the valley. Additional depth specific data would provide a firm basis for further verification of analytical results and additional information about the specific chemical composition of the total salt content.

The soil salinity and moisture measurements made in 1967 by the Washington State University investigators appear to verify the analytical result that there should be no salt accumulation above the water table due to the continuous flushing action of the percolate. The reported low soil moisture conditions are likewise compatible with the analytical results which indicate that vertical transport of percolate requires only very small increments of soil moisture above field capacity. (Field capacity is the maximum soil moisture content that can be held by capillar forces and which will not drain under the action of gravity.)

The analytical results of the forecast impact at year 2020 when compared with the present impact, both measured in terms of volume of leachate and dissolved solids concentration, indicate that the present impact on groundwater quality is already a significant proportion of the ultimate level.

Subsurface Water Movement Under Natural Conditions

General. Analysis of groundwater quality within Spokane Vallay requires a determination of how and when water, if any, migrates from the land surface to the water table. Before introducing the complexity of septic tanks and suburban development, it is useful to begin with natural conditions—that is, raw land that is neither occupied nor irrigated by man. Subsequent analyses modify these results to reflect suburban conditions.

The approach followed here is to evaluate the monthly water balance of the surface soil layer for natural conditions. Combined for an entire year, these balances indicate whether surplus water is available for percolation to the water table.

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Water Balance for Natural Conditions. A water balance for a surface soil layer, such as shown in Figure A, can be determined by evaluating each of the flow components. This is done on a monthly basis for representative conditions in the Spokane Valley. Precipitation is determined from local climatological records. Evapotranspiration can be computed for given climatic, locational, and soil conditions. Surface runoff and percolation are outflows when the moisture holding capacity of the soil is exceeded. For the highly permeable soils in the Spokane Valley, surface runoff is judged to be negligibly

small; therefore, it is assumed that all excess moisture in the soil layer percolates downward.

Location. Soil and climatic conditions representative of Spokane Valley are selected for that portion most subject to septic tank installations; this is assumed to be centered near Opportunity, Washington, at an elevation of 2,000 feet MSL.

Soils. The primary soil in the Spokane Valley is designated a Garrison gravelly loam (GgA) by the Soil Conservation Service. This is an excessively drained soil formed in gravelly glacial outwash material. Its permeability is described as moderate to very rapid. In the top 5 feet it has a water holding capacity (field capacity) of about 5 inches.

Temperature. A 13 year record of air temperature in Spokane (Lat. 47°40'N., Long. 117°25'W.: elev. 1,875 ft.) is available (1) and has a mean value of 48.3°F. This compares with means of 47.8°F. at Spokane Airport and 47.8°F. at Coeur d'Alene. The Spokane (city) record is adopted as most representative of the Valley area. Monthly values are listed in Table 1 and plotted in Figure B.

Precipitation. An isohyetal map prepared in another section of this report shows that the mean annual precipitation in the vicinity of Opportunity is 20 inches. The monthly pattern of precipitation is obtained by correlation with the records of the Spokane (city) weather station, located approximately 9 miles westward. Values are listed in Table 1 and are plotted in Figure C.

Potential Evapotranspiration. Given the above data, mean monthly potential evapotranspiration is computed for the Valley using the method of Thornthwaite (2). Potential evapotranspiration is defined as the maximum amount of water which if available could be removed from the soil by the combined processes of evaporation and transpiration. The values are listed in Table 1 and plotted in Figure C. Note that values range from zero for months with mean temperatures at or below freezing to a maximum in mid-summer. The annual total is 25.51 inches. A hypothetical moisture surplus exists in months when precipitation exceeds potential evapotranspiration (October to March), and a hypothetical moisture deficiency occurs when the reverse is true (April to September), as shown in Figure C.

Actual Evapotranspiration. Because of the moisture deficiency in the Valley during the summer months, the actual evapotranspiration will be something less than the potential. Actual evapotranspiration is defined as the computed amount of water lost considering the limitation of moisture availability. Moisture released to the atmosphere comes from any available precipitation plus soil moisture. For the given climatic and soil conditions of the Valley, the actual evapo-

transpiration can be computed by the Thornthwaite method (2). Monthly values are listed in Table 1 and plotted in Figure D. The annual total is 13.73 inches, which is only 54% of the potential evapotranspiration.

Moisture Deficit. Moisture deficit is a measure of the difference between the potential and actual evapotranspiration. Deficit values are listed in Table 1 and plotted in Figure D. Note that the deficit goes from zero in March to large values in July and August and then drops to zero again in October.

Soil Moisture Storage. Soil moisture exists in three forms which react differently to external forces. Hygroscopic water is that which is so tightly bound to the soil particles that it is not available to plants nor can it be moved by gravity; it can only be removed by heat. Capillary water is bound less tightly to the soil and is available to plant roots but does not move downward under the action of gravity. The sum of the hygroscopic moisture and the capillary water is referred to as the field capacity of the soil. The hygroscopic portion is so small that it can be neglected for the purpose of this study and the entire field capacity is assumed to act as capillary water and be subject to uptake by plants. Moisture above the field capacity is free to move under the force of gravity and would also be available to plant roots within the root zone.

The field capacity of typical Spokane Valley soils is 1 inch of water per foot or approximately 8 percent of the total volume.* The total voids are approximately 30 percent of total volume so that 8/30 of the voids can be occupied by moisture that does not move under the force of gravity.

The amount of moisture stored in the soil will vary seasonally depending upon variations in precipitation and evapotranspiration. Using the Thornthwaite method (2), monthly values of moisture content for the representative Spokane Valley GgA soil (60 inches thick) are computed. Values in inches of water within the soil layer are listed in Table 1 and plotted in Figure E. The maximum moisture holding capacity of 5 inches occurs only in late winter when moisture from snow melt and precipitation greatly exceed evapotranspiration. Note in Figure E that the soil moisture begins to decrease in April, falls rapidly during the spring months, and reaches values of less than one inch for the entire July to October period. Finally, it recovers rapidly during November and December and reaches its maximum again in February.

Snow Pack Moisture Storage. In addition to soil moisture storage, there is a temporary supplemental storage in winter due to the snow pack on the ground surface. Applying the Thornthwaite method (2) to Spokane Valley conditions yields a snow pack moisture storage of 3.15 in January only (see Table 1 and Figure E).

^{*}Source: Soil Survey, Spokane County, Washington, U.S. Dept. of Agriculture, Soil Conservation Service.

Percolation to Groundwater. The above water balance analysis reveals that surplus water over and above potential evapotranspiration is available only during late winter. The snow pack moisture storage of 3.15 inches melts and becomes available at this time, and in addition, there is excess February and March precipitation. Thus, total mean annual available surplus water for a year amounts to 6.01 inches (see Table 1). This percolates downward through the soil layer and to the water table only in late winter. In summer the combination of low precipitation and high evapotranspiration leads to a dessication of the surface layer. Based on this finding, there is a net annual downward transport of water under natural conditions for average and above average years of precipitation. In years of significantly below average precipitation it is possible there would be no net surplus for downward movement.

Subsurface Water Movement Under Suburban Conditions

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General. Before proceeding to an analysis of subsurface moisture movement under present and future conditions, involving full recognition of the spatial distribution of development, it is useful to analyze a generalized suburban area with simplifying assumptions. These simplifying assumptions favor maximum evapotranspiration and minimize the potential for creation of surplus soil moisture that would percolate downward. A calculation of this kind should demonstrate whether downward percolation is to be expected under actual suburban conditions.

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Determination of a monthly water balance for the same Spokane Valley surface soil layer under suburban conditions requires evaluation of moisture inputs from (a) precipitation, (b) septic tank effluent, and (c) garden irrigation as shown in Figure A for developed conditions.

<u>Precipitation</u>. Representative precipitation for Spokane Valley is defined above and is shown in Table 1.

Septic Tank Effluent. To estimate the effluent from septic tanks in a generalized suburban condition an assumption regarding density of development is required. For extensive tracts of land zoned for residential development, an average lot size of about 14,000 square feet can be adopted. This is equivalent to approximately 3 lots per acre. If the average family consists of 4 persons and the typical effluent production rate is 90 gpcpd*, then the average flow rate per month would be:

(4 persons) (3 lots/Acre) (90 gpcpd) (30 days) = 32,400 gallons/month/Acre **

Converting this to a flow depth over the gross area yields:

^{*}gpcpd - gallons per capita per day.

^{**}Note that this is not intended to represent typical average conditions in Spokane Valley but rather the more extreme condition that exists in areas developed as subdivisions.

$\frac{32,400 \text{ gal/mo/A} \times 12 \text{ in/ft}}{7.48 \text{ gal/ft}^3 \times 43,560 \text{ ft}^2/\text{A}} = 1.19 \text{ in/mo}.$

The actual area occupied by the leach lines from three septic tanks on an acre of land would amount to only about 10% of the gross area. Normally such effluent would tend to be distributed slightly laterally by capillary action but mostly downward due to gravity drainage, as indicated in Figure F. The lateral flow component can become significant where the sub-soil layers are highly stratified and where impermeable layers tend to perch and to spread the effluent. In Spokane Valley there appears to be little physical evidence which shows that either of these situations occur; however, vertical permeabilities are undoubtedly lower than horizontal ones.

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If septic tank effluent moves essentially downward, it follows that the volume subject to evapotranspiration loss will be minimal because the area of excessively moistened soil will be minimal. On the other hand the opposite extreme would be to assume a lateral spreading of the effluent over the entire area. Although physically highly improbable, this assumption can be useful because it represents the situation tending to maximize the effect of evapotranspiration. Note that this assumption also makes no deduction for the areas covered by streets and houses, which, in reality, are not capable of contributing to evapotranspiration loss. Thus, two assumptions are made, both highly favorable to evapotranspiration and unfavorable to leaving excess moisture available for percolation. With these assumptions, the above computation giving a uniform effluent rate of 1.19 in/mo. can be adopted; this totals 14.28 inches per year. The sum of precipitation plus septic effluent expressed in inches per month over the gross generalized suburban area is listed in Table 2; this amounts to a total annual moisture input of 34.28.

Garden Irrigation. Irrigation of gardens and lawns is another moisture source in a suburban area. This will occur during the growing season and will be at a maximum during the driest and hottest months (July and August) when the moisture deficit is greatest.

Again a simplifying assumption can be made favoring evapotranspiration loss of the septic tank effluent by neglecting garden irrigation which would supplement the effluent in supplying a part of the total evapotranspiration demand. It should be noted that Spokane Valley garden irrigation use is known to be very high from water use records.

Water Balance. Values of temperature, precipitation, and potential evapotranspiration are identical to those shown in Figures B and C in Table 1.

Actual evapotranspiration under generalized suburban conditions

will differ from that of natural conditions because of the increased moisture available from septic tank effluent. Monthly values of actual evapotranspiration computed by the Thornthwaite method are listed in Table 2 and plotted in Figure G. The annual total is 20.35 inches, which is 80% of the potential evapotranspiration.

Moisture deficit, which is the difference between the potential and actual evapotranspiration, is listed in Table 2 and plotted in Figure G. Note that the annual moisture deficit has decreased to less than half of that found under natural conditions, and most of this now occurs in July and August.

Soil moisture storage values within the 60-inch thick GgA soil layer are listed in Table 2 and plotted in Figure H. Under suburban conditions the maximum moisture holding capacity of 5 inches occurs from November to April; minimum values occur only in the late summer months of August and September. A 60-inch layer is selected as typical of the depth of influence of the average non-orchard crop as suggested by Thornthwiate (2).

Snow pack moisture storage is listed in Table 2 and shown in Figure H. According to the Thornthwaite method, this amounts to the total January precipitation.

Percolation to Groundwater. The above water balance indicates that surplus water over and above potential evapotranspiration is available from November through April (see Table 2). The total annual surplus water is 14.22 inches; this is more than double that determined for natural conditions (see Table 1) and is 41% of the total precipitation plus septic tank effluent. Thus, a significant net downward transport of water and associated dissolved solids could occur under generalized suburban development conditions.

Having found that water is available for percolation, it is useful to determine how much moisture in excess of field capacity would be required to account for continuing vertical transport of the surplus moisture under steady state flow condition. For this calculation, a conservative assumption would consider a minimum area and a maximum amount of moisture. Therefore, a calculation is made to determine the steady state moisture to transport the entire septic tank effluent, undiminished by evapotranspiration through an area limited to the actual size of the drainfield. The typical drainfield required by County regulations for a three-bedroom home is approximately 1,800 square feet in area and would be receiving water at the rate of 0.32 inch per day, say 0.50 inch per day. Vertical unsaturated flow at the rate of 0.50 inch per day, or 0.31 gallons per square foot per day, will take place under steady state conditions at a soil moisture of 9.0 percent by volume or 8 percent above the field capacity level of 8.3 percent by volume. This calculation is made on the assumption that the vertical

permeability is only one tenth of the estimated horizontal permeability of the Spokane Valley aquifer material, again a conservative assumption.

What this result indicates is that the observed soil moisture content above the water table should not deviate materially from that of field capacity because only a very small increment is all that is required to sustain percolation rates that could account for transport of all surplus moisture. Therefore, even during the percolation season, soil moistures at depth should remain close to field capacity.

Estimation of Percolation Under Present-Day Conditions

General. The previously described water balance methodology can be applied to present day specific development patterns in the Spokane Valley. Quantities of percolation to the water table from precipitation, septic tank effluent, lawn irrigation, and agricultural irrigation can thus be calculated. Knowing this downward flow rate, the effect on groundwater quality can then be estimated by use of an appropriate quality index parameter.

Dissolved inorganic salts are added to water by domestic use. The laboratory test for dissolved inorganic salts is identified as "total dissolved solids" (TDS). A very large percentage of the materials identified by the TDS test pass undiminished through typical treatment plants including septic tanks and through soil. Only phosphates and certain heavy metals are removed by attachment to or reaction with soil particles. Phosphates usually represent less than fifteen percent of TDS and heavy metals, when present, are a fraction of one percent.

The observed TDS concentration in the City of Spokane wastewater effluent is of the order 440 mg/l. The City water supply which is from Spokane Valley groundwater has an average TDS content of approximately 170 mg/l. This indicates an incremental TDS addition of 270 mg/l. Normal ranges of increases in various salts and TDS due to domestic use are shown in Table 3. Note that the City experience falls within the normal range.

Since the Spokane Valley suburbs have the same water supply, a similar relationship would be expected between the TDS of Spokane Valley wastewater and the TDS of its supply. Thus, septic tank leachate at depth should have a TDS concentration more than double that of the natural groundwater. Therefore, total dissolved solids is selected as an appropriate identifier of the moisture from septic tank effluent as it moves downward to the water table. Since the City experiences some dilution by significant infiltration flows, a value of 300 mg/l is selected as the domestic TDS increment in Spokane Valley use for computational purposes.

Groundwater Flow Direction. It is expected that the largest change in quality of groundwater will occur at the western end of Spokane Valley because of the general east-to-west flow of groundwater. The western boundary of septic tank installations coincides with the

western limit of Planning Unit SV-3 and with the eastern city limit of Spokane. Refer to Figure K. Therefore, groundwater quality in terms of TDS is computed for a hypothetical well located at T25N,R43E,14E1* as an index of the impact of septic tank percolation.

Percolation reaching the groundwater at this location must come from along the flow line of groundwater upstream within Spokane Valley and extending to the Idaho State Line. The required flow line is traced as a dashed line on the water table contour map in Figure K. The water table contours are developed from the following data sources:

(1) Well water levels from Appendix I of Task Report Section 303.

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- (2) USGS Water-Supply Paper 889-B.
- (3) Aerial photographs of May 1, 1973.
- (4) Consolidated Irrigation District monthly well water levels.
- (5) Rating curves of Spokane River gaging stations.
- (6) Bed profile of Spokane River.

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All water levels are corrected to a September datum to eliminate seasonal variations.

The flow line above well T25N,R43E,14El meanders up the southern half of Spokane Valley. It crosses to the north side of Spokane River at RM 91.6, crosses back to the south side at RM 94.1, and follows close to the river up to the Idaho State Line. The total flow length is 79,800 feet; the distance within each Planning Unit intersected is listed in Table 4.

Applied Water. Mean annual precipitation for each Planning Unit in Spokane Valley is estimated, taking into account the increasing precipitation east of the City of Spokane. Values are listed in Table 4.

Septic tank effluent is computed from estimates of domestic sewage for each Planning Unit for present-day conditions. Annual values in inches mean depth over the area of each Planning Unit are listed in Table 4.

^{*}USGS well identification system: Township 25 North, Range 43 East, Southwest quarter of the northwest quarter of Section 14.

Lawn irrigation* is computed as the difference between water demand and domestic sewage. Estimates in inches mean depth for each Planning Unit for 1975 conditions are shown in Table 4.

Agricultural irrigation is computed from estimates of irrigated acreage for each Planning Unit. It is assumed that 3.0 feet of water are applied annually to all irrigated areas. Annual values in inches mean depth over the area of each Planning Unit are listed in Table 4.

Total applied water is the sum of precipitation, septic tank effluent, lawn irrigation, and agricultural irrigation. Values for each Planning Unit appear in Table 4. The largest quantities of applied water occur in the more heavily developed western portions of the Spokane Valuey (Planning Units SV-3 and SV-5). Also, the relatively small contributions of septic tank effluent to the total applied water should be noted: 12 percent in Planning Unit SV-3, about 2 percent in Planning Units SV-5 and SV-6, and less than 1 percent in Planning Units SV-7 and SV-8.

Percolation. By application of the Thornthwaite water balance method (2) on a monthly basis, the percolation to groundwater for each Planning Unit is obtained. Potential evapotranspiration rates are reduced for estimated current impervious areas. The following values are used:

Planning Unit	Impervious Area, %
sv-3	13.0
SV- 5	2.6
SV-6	2.6
3V~7	0.5
sv-8	1.0

The resulting annual values of percolation are listed in Table 4.

In order to compute the quantity and quality of the percolate from the sum of contributions along the flow line, expressed in inches depth per year, it is first necessary to estimate the flow velocity of the groundwater with which the percolate is assumed to move.

Groundwater Movement

A representative groundwater velocity can be computed using Darcy's law: $v_a = \frac{Ri}{CC}$

^{*}Lawn irrigation is defined as non-commercial irrigation and for Spokane Valley includes many private pastures and other extensive areas.

where v_a is the average actual (tracer) velocity, K is permeability, i is the water table gradient, and \propto is porosity. Based upon given geologic data for Spokane Valley, permeability can be estimated at 75,000 gpd/ft, i = 10 ft/mi, and \propto is 0.30. Substituting these values in the above equation yields 63 ft/day.

Applying this calculated groundwater flow rate, the total depth of water percolating within each Planning Unit along the flow line is computed. Values are shown in Table 4. A sample calculation for Planning Unit SV-3 is

$$\frac{37,500 \text{ ft. x } 11.11 \text{ in/yr.}}{63 \text{ ft/day x } 365 \text{ day/yr.}} = 18.1 \text{ inches}$$

This states that the percolation from SV-3 arriving at the surface of the groundwater at a rate of 11.11 inches per year and traveling with the groundwater at a rate of 63 feet per day will have an accumulated depth of 18.1 inches at the downstream end of 37,500 feet of exposure. The values shown in Table 4 for flow line percolation for each unit have the same interpretation. Since each exposure is successive, the indicated depths of flow line percolation are cumulative so that the percolated waters have a depth of 34.1 inches at the downstream end of the flow line at the hypothetical well T25N,R43E,14E1. Note that this depth is the net water quantity and not the vertical space that this volume would occupy when filling the void spaces of the aquifer.

Percolation Quality. To determine the TDS concentration of the percolating water for 1975 conditions, the following TDS values are assumed for the various sources:

Percolation Source	Salinity, TDS, mg/1
Precipitation	10
Septic Tanks	45 5
Lawn Irrigation	155
Agric. Irrigation	155

The above values are based on an average measured salinity of 155 mg/l for groundwater entering the Spokane Valley near the Idaho State Line. Assuming that salt does not accumulate within the zone of aeration above the water table because of the net downward movement of water, the salinity of the percolation can be computed on a mass balance basis. Values for each Planning Unit are listed in Table 4. A sample calculation for Planning Unit SV-3 follows

$$\frac{19.50(10)+3.95(455)+8.05(155)+0}{11.11} = 302 \text{ mg/1 TDS}$$

With the quantities and qualities of percolation given, the

mean weighted concentration of percolation, Cpm, at the end of the flow line can be computed using an equation of the form

$$Cpm = \frac{\sum Qp \times Cp}{\sum Qp}$$

where Qp is percolation depth for one Planning Unit and Cp is the TDS of percolation for the Planning Unit. Inserting values from Table 4, the mean salinity becomes 248 mg/1 TDS.

This result indicates that the 34.1 inches of accumulated depth of percolated waters at the downstream end of the flow line, if mixed among themselves but unmixed with the underlying groundwater body, would have a mean TDS of 248 mg/1.

Groundwater Quality. Having determined the quantity and quality of the percolate which is reaching the surface of the groundwater, it is necessary to assume how the percolate mixes with the natural groundwater before an evaluation can be made of the impact upon water withdrawn by wells. Undoubtedly some mixing of the percolate and the groundwater occurs, but due to the nature of flow in a porous medium, it is probable that the majority of the percolate remains unmixed in a layer at the top of the groundwater body.

The implication of a substantially unmixed layer of percolate traveling on the surface of the native groundwater body is that wells with varying depths of penetration and levels of openings will yield waters of varying quality. A well penetrating deep into the native body and without casing perforations into the percolate layer could, if no mixing were induced in rhe aquifer by a very high pumping rate, yield water having a quality equal to that of the native water. On the other hand, a well which penetrates only the percolate layer could produce water with a quality of the undiluted percolate. In terms of the TDS concentration, the concentration could range from 155 to 248 mg/1. Wells which penetrate and have openings into both layers would be expected to have qualities intermediate between these extremes.

The following calculation is made to illustrate the quality in terms of TDS which could appear from a well which penetrates and withdraws water from both the percolate and native water layers. The 34.1 inches of accumulated percolate would occupy a layer 9.5 feet in the typical aquifer materials with 30 percent voids.

A hypothetical well at T25N,R43E,14El in the western portion of the Spokane Valley is assumed to penetrate 20 feet below the water table, as shown on Figure I. It is further assumed that water is drawn into the well casing in proportion to the length exposed to each layer. The top 9.5 feet of casing would then be exposed to waters of 248 mg/l TDS while the lower 10.5 feet would be exposed to waters of

155 mg/l TDS. Mixing these two waters within the well yields pumped groundwater having a concentration as follows:

$$\frac{9.5(248) + 10.5(155)}{20.0} = 199 \text{ mg/1 TDS}$$

Large diameter open bottom dug wells which do not penetrate deeply would induce similar mixed qualities by turbulent mixing caused by high pumping rates.

Estimation of Percolation for the Year 2020

In order to evaluate trends in groundwater quality for the Spokane Valley, estimates of percolation are made for the year 2020. Calculations are based on projections of population, water use, domestic sewage production, lawn irrigation, and agricultural irrigation for each contributing Planning Unit in the Valley. Comparative population figures are shown in Table 5.

These data show that the population along the flow line assumed for demonstration purposes will increase by 54 percent over the 1975 level. Furthermore, most of the population growth will be concentrated in Planning Unit SV-3, which, in 1975, was already the primary source of septic tank effluent (see Table 4).

Using the projections described above and the following estimates of impervious areas for the year 2020, calculations of the annual water balance and percolation quality values are made for the year 2020.

Planning Unit	1mpervious Area for Year 2020, %
sv-3	18
SV-5	4
sv-6	4
SV-7	1
SV8	2

The year 2020 results are summarized in Table 6 for each Planning Unit. The format of Table 6 is identical to that of Table 4 so changes in any of the variables between 1975 and 2020 can readily be seen. A key assumption in the 2020 analysis is that all sewage continues to be disposed of by means of septic tanks; introduction of a sewerage system in part or all of the Valley would affect quantities of septic tank effluent and, consequently, groundwater quality.

Comparing 2020 conditions in Table 6 with 1975 conditions in Table 4, the following points can be noted:

- (1) Septic tank effluent increases in all areas and Planning Unit SV-3 continues as the predominant contributor.
- (2) Lawn irrigation increases moderately in all areas except Planning Unit SV-5.
- (3) Agricultural irrigation decreases substantially in all Planning Units.
- (4) Total applied water remains essentially unchanged except for a small increase in Planning Unit SV-3.

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- (5) Percolation increases moderately in Planning Unit SV-3 but changes insignificantly in other areas.
- (6) TDS of percolation remains remarkably constant in all areas.

The mean salinity and accumulated volume of the percolate at the end of the flow line will be 256 mg/l TDS and 41.0 inches, respectively, in 2020; this compares with 248 mg/l and 34.1 inches, respectively, for 1975. Similarly, the salt concentration of mixed waters pumped from a hypothetical well penetrating 20 feet below the water table at the western portion of the Spokane Valley in 2020 will be 213 mg/l TDS; the comparable 1975 value is 199 mg/l.

Based on a native groundwater quality of 155 mg/l TDS, the present "hypothetical well" quality at 199 mg/l represents an increase of 28 percent. Similarly, the forecast "hypothetical well" quality in year 2020 at 213 mg/l represents an increase over native condition of 37 percent. These comparisons suggest that groundwater quality in Spokane Valley will change relatively slowly during the next 45 years (1975 to 2020) and that the present day impact of septic tank effluent is a significant proportion of the anticipated future impact. Thus, with a future population increase of 54 percent in Spokane Valley, there will be an estimated increase of 32 percent in well water salinity based on the background level of 155 mg/l in the native groundwater.

Field Verification of Calculated Quality

The foregoing calculations indicate that a substantial amount of percolation from septic tank effluent and other sources should be reaching the surface of the native groundwater and that there should be evidence of this percolate in the form of increased total dissolved solids above that of the native groundwater. The apparent groundwater quality should vary with distance westward from the Idaho State Line and with depth of well penetration below the water table. The scarcity of water quality data, and particularly those associated with well penetration, makes verification of the analytical results difficult.

There are two sources of recent and synoptic data for water quality expressed as TDS. One source is the USGS-EPA program for the period June 1973 to March 1974. For comparative purposes and correlation with the water table contours developed for September, the

sampling of September 1973 is selected. Table 7A lists the wells, the penetration below the September water table, the TDS (Residue 180°C), and the distance west from the Idaho State Line. The other source is the Spokane County Health District survey covering the period September 1971 through September 1972. The mean TDS for the two September samplings is selected for analysis. The quality data, well penetrations and distance from the State Line are listed in Table 7B. Well locations from both sources are shown on Figure K.

The data from Tables 7A and 7B are plotted in Figure Jl and J2 respectively. Note that in both cases, a trend of increasing TDS with distance westward is apparent. This is in basic agreement with the foregoing analytical results. A total of four points from Figure J-1 and one point from Figure J-2 suggest a groundwater TDS of approximately 155 mg/l near the Idaho State Line. Two points from Figure J-2 near the State Line have very low TDS values, approaching that of Spokane River water. There is no apparent explanation for these anomalous values other than extremely deep penetration below the water table. In the area between 7 miles west of the State Line to the City limits (approximately 14.3 miles west of the State Line) there are four points from Figure J-l and eight points from Figure J-2 that strongly support the trend of increasing TDS, suggesting a mean value of 21.5 mg/l at the City limit. In Figure J-1 there are two wells in the western area with moderate penetrations of the aquifer that do not follow the trend but are substantially at State Line TDS values. Again, there is no apparent explanation for these anomalous salinities.

The well penetration data listed in Table 7-A and 7-B are plotted beside the well points in Figures J-1 and J-2. Except for a tendency for wells with low penetrations to lie above the trend line and those with high penetrations to lie below the line, no verification of the effect of well depth can be obtained from these limited field data.

Independent Field Investigations

General. A group of investigators at Washington State University studied moisture and pollutant movement at selected sites in Spokane Valley during the summer of 1967. Their findings have been published (6,7,8) and are of interest because they provide independent evidence generally supporting the previously described analytic studies. The following paragraphs briefly summarize the field investigations and then interpret their results in terms of the analytic work.

Site 1. This site (T25N,R44E,20J) was a nursing home and was selected as an example of extreme loading of pollutants in the Valley. Six test holes were drilled in the main sanitary drainfield and three in a separate laundry drainfield; in addition, two test holes for control were drilled nearby. Hole depths ranged from 41 to 66 feet, while the water table was at a depth of 125 feet. Soil samples were collected at 5 to 10 foot intervals and were analyzed for total coliforms, fecal coliforms, enterococci, moisture, chloride, nitrate, detergents, and particle size distribution.

The sample results indicated that bacteria were rapidly

removed in the upper soil layers and that low and relatively uniform values of chloride, nitrate, and detergents were present. Moisture contents were also generally low.

Of particular interest to this study are the salinity measurements reported under the sanitary drain field. In all test holes the chloride ion concentrations found below the top ten feet, that is below the zone of high moisture, are consistently low, never exceeding 10 $\mu g/g^*$ of soil, with most being less than $5 \mu g/g$. The fact that the highest concentrations are associated with the highest moisture contents indicates that salts are in the dissolved form rather than as precipitated deposits. The low salt concentrations must result because they are being constantly flushed out rather than being left behind by evapotranspiration.

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The following calculations are made to demonstrate these interpretations:

(1) A typical value for the per capita chloride increment to sanitary sewage is 0.035 pounds per day. For the 100 occupants of the nursing home this totals 3.5 pounds per day or 1,278 pounds per year. If this quantity of chloride for one year were held by evapotranspiration in a soil prism 100 feet square, centered on the 40 foot square drainfield, and extending to the depth of the borings at 60 feet, it would appear as a concentration of approximately 17.5 µg/g. In 1967 the drain field had been operating at least 10 years, by which time total chloride would have a concentration of 175 µg/g. These values are 10 and 100 times, respectively, more than the measured values.

It is of interest to calculate the zone of soil influence that would have to be involved to lower the mean concentration of 10 years deposition to 2.5 $\mu g/g$, which is the order of magnitude of the mean observed chloride concentration. The prism of soil would have to weigh 5,110 million pounds and have dimensions 842 feet square by 60 feet deep, or 584 feet square and 125 feet deep, extending to the water table. The potential for flows from a 40 foot square drainfield to permeate such extensive volumes is very small.

(2) Similarly, if a typical value for chloride ion increment due to domestic use of 40 mg/l is selected and the native groundwater is assumed to have 1.5 mg/l, then the total drainfield effluent quality would be 41.5 mg/l. Taking,

^{*}µg/g - micrograms per gram.

by weight at 40 foot depth (in Hole No. CN-5), the computed chloride content of the soil, assuming this moisture to be drainfield percolate at 41.5 mg/l, is 1.04 µg/g. This compares with a measured value of 2 µg/g. Similarly, the mean for all observed moistures is 4.06% in Hole No. CN-1; the computed chloride content is 1.68 µg/g and the mean of the observed chlorides is 2.6 µg/g. These results indicate that the observed chloride concentrations in the soil moisture are about twice that expected of drainfield effluent. This suggests that about half of the drainfield moisture is traveling with all of its salt content and that there is no significant salt accumulation in the soil.

Site 2. This site (T25N,R44E,26N) was a dairy and was selected on the basis that it should have the highest pollution of this type in the Valley. Two test holes were drilled in the loafing area and one at a waste disposal site; in addition, a test hole was drilled in a nearby area for control. Hole depths ranged from 51 to 71 feet, while the water table was at a depth of 90 feet. Soil sampling and analysis procedures were identical to those at Site 1.

Sample results showed that bacteria were again rapidly removed in the upper soil layers. Chloride and nitrate were higher than at Site 1, but were still low and relatively uniform. Detergent analyses were negative; however, the tests were not considered conclusive.

Site 3. This site (T25N,R45E,7A) was a trailer park servicing six mobile homes; the pollution loading here was relatively light. Three test holes were drilled in the drainfield, which was adjacent to the Spokane River. Hole depths ranged from 31 to 43 feet, while the water table was at a depth of 45 feet. Soil sampling and analysis procedures were as before.

Sample results showed no bacteria below 6 feet; low and relatively uniform values of chloride, nitrate, detergents, and moisture were again observed, showing no evidence of salt accumulation.

Site 4. This site (T25N,R43E,14E) was located on the west side of a gravel quarry. Three test holes were drilled here as control holes for the drainfield study of the Valley. Hole depths pend trated to the water table at 67 feet. Soil sampling and analysis procedures were as before.

Sample results showed only one positive bacterial determination, negligible nitrate, and relatively uniform chloride values comparable to those found under drainfields in the Valley. Moisture contents again were low and relatively uniform.

Control borings had been made previously at both the nursing home and the dairy. In both cases, the chloride concentrations were found to be negligibly small. At the quarry site control borings, however, significant chlorides were found, the mean value for the two holes below 10 feet being 2.75 µg/g. This is the same magnitude as found under the two drainfields. The variability was extreme, for example, going from 5.51 µg/g at 51 feet to 0.63 µg/g at 61 feet. These results appear to be anomolous and are judged to not represent control conditions; past quarry operations which recycle wash waters into the quarry may have caused this condition.

Summary. Review of the studies by the Washington State University investigators indicates that their observed results are in basic agreement with the previously described analytic work. Specifically, chloride and nitrate concentrations in soils below septic tank drainfields were found to be generally low. With a net annual downward transport of septic tank effluent, precipitation, and irrigation waters, as computed herein, salt concentrations should remain low because of the leaching action of the percolating water moving to the water table.

Recommendations for Further Investigations

The analytic studies on groundwater quality are verified to a limited extent by available data as shown in Figure J and also by the earlier work of the Washington State University Investigators (6,7,8). It would be desirable, however, to confirm and to extend these findings by additional field investigations. The purpose of the additional work would be to:

- (1) Define more comprehensively the down-valley changes in groundwater quality.
- (2) Obtain information on the variation in groundwater quality as a function of depth of penetration of wells.
- (3) Provide a firmer foundation for future projections of groundwater quality and for wastewater management planning.

A field program should as a minimum consist of a series of well and water quality measurements along the flow line sketched in Figure K. Weils should be selected where it is possible to ascertain the following facts:

- (1) Depth of water table at time of sampling.
- (2) Depth of well and its penetration into the groundwater at the time of sampling.
- (3) Location and size of openings in the well casing through

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which groundwater enters the well.

- (4) Elevation of the pump suction relative to the water table at time of sampling.
- (5) Pumping rate at the time of sampling.

Where large diameter dug wells are available, it would be most useful to collect samples under quiescent conditions at various depths in the aquifer without pumping.

For each well sampled a complete physical description should be obtained in accordance with the above listing. The water quality sampling should, as a minimum, cover the following parameters:

- (1) Temperature
- (2) Conductivity, or total dissolved solids
- (3) Chlorides
- (4) Nitrate nitrogen
- (5) Detergents
- (6) Fecal coliform

It is recommended that the above described field investigational program be undertaken as soon as possible so that a more complete verification of the analytical results presented in this report can be achieved.

Temperature, °F 26.7 32.0 40.3 48.1 55.8 Precipitation, 3.15 2.04 1.70 1.10 1.83 potential Evation, in. 0 0.61 1.71 3.14 transpiration, in. 0 0 0.61 1.71 3.14 hoisture Storage, in. 4.73 5.00 5.00 4.41 3.38 Storage, in. 3.15 0 0 0 0.00 0.00 0 0 0 0 0 0 0 0 0 0 0	Evapotranspiration, and Soil	Soil Moisture Data	Data for	: Spokane Valley	Valley			
Temperature, °F 26.7 32.0 40.3 48.1 55 Precipitation, 3.15 2.04 1.70 1.10 1. Potential Eva- potranspira- tion, in. Moisture Deficit, in. Soil Moisture Storage, in.	1 J	J	A	S	0	Z	Q	YR.
Precipitation, 3.15 2.04 1.70 1.10 1. Potential Eva- potranspira- tion, in. Noisture Deficit, in. Soil Moisture Storage, in. Storage, in. Total Perco- Total Perco- Intion, in. Solution, in. Solution Storage, in. Total Perco- Intion, in. Solution Storage, in. Total Perco- Intion, in. Solution	5.8 61.8	6.69	67.0	60.1	49.5	37.5	31.1	48.3
Potential Eva- potranspira- tion, in. Actual Evapo- transpira- tion, in. Moisture Deficit, in. Soil Moisture Storage, in. Snow Pack Moisture Snow Pack Moisture Sn	.83 1.44	0.52	0.65	0.91	1.74	2.40	2.52	20.00
Actual Evapo- transpira- tion, in. Moisture Deficit, in. Soil Moisture Storage, in. Show Pack Moisture Storage, in. Storage, in. Total Perco- lation, in. 1.69 2. 0.061 0.061 0.002	.14 4.38	5.61	4.78	3.15	1.67	0.46	0	25.51
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Snow Pack Moisture Storage, in. 3.15 0 0 0 Total Perco- lation, in. 3.15 1.77 1.09 0	.38 1.86	99.0	0.29	0.20	0.27	2.21	4.73	1
Total Perco- lation, in. 3.15 1.77 1.09 0	0	0	0	0	0	0	0	3.15
	0	0	0	0	0	0	0	6.01

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	YR.	34.28	25.51	20.35	5.16	ı	3.15	14.22	
	Q	3.71	0	0	0	5.00	0	3.71	
suc	z	3.59	0.46	0.46	0	5.00	0	0.08	
Conditi	0	2.93	1.67	1.67	0	1.95	0	0	
Suburban	တ	2.10	3.15	2.26	0.89	0.69	0	0	
TABLE 2 for Spokane Valley under Generalized Suburban Conditions	A	1.84	4.78	2.85	1.93	0.85	0	0	
der Gene	Ъ	1.71	5.61	3.55	2.06	1.55	0	0	
TABLE 2: Valley un	Ŋ	2.63	4.38	4.10	0.28	3.41	0	0	
TAI pokane V.	E	3.02	3.14	3.14	0	4.88	0	0	
1	А	2.29	1.71	1.71	0	5.00	0	0.58	
Monthly Water Balance Data	×	2.89	0.61	0.61	0	5.00	0	2.28	
Water Ba	ξъ	3.23	o	0	0	5.00	0	3.23	
fonthly (b	4.34	0	0	0	5.00	3.15	4.34	
,24 -		Precipitation and Septic Tank Effluent, in.	Potential Evapo- transpiration, in.	Actual Evapo- transpiration, in.	Moisture Deficit, in.	Soil Moisture Storage, in.	Snow Pack Moisture Storage, in.	Total Percola- tion, in.	

TABLE 3
Normal Ranges of Increases in Inorganic Salts in Domestic Sewage (3)

Mineral	Mineral range (mg/1)
Dissolved solids	100 - 300
Boron (B)	0.1 - 0.4
Sodium (Na)	40 - 70
Potassium (K)	7 - 15
Magnesium (Mg)	3 - 6
Calcium (Ca)	6 - 16
Total Nitrogen (NO ₃)	20 - 40
Phosphate (PO ₄)	20 - 40
Sulfate (SO ₄)	15 - 30
Chloride (C1)	20 - 50
Alkalinity (as CaCO ₃)	100 - 150

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TABLE 4
Summary of Annual Water Balance and Percolation Quality Values for Spokane Valley Planning Units at Present (1975)

Planning Unit sv3 SV7 sv8 SV5 SV6 Mean Annual Precipitation, 19.5 21.5 21.0 22.0 22.0 in. Septic Tank Effluent, in. 3.95 0.74 0.70 0.16 0.16 8.05 6.06 2.26 0.98 1.67 Lawn Irrigation, in. Agricultural Irrigation, 6.44 0.72 7.27 5.67 5.19 in. 30.40 28.81 29.09 Total Applied Water, in. 32.22 33.07 Percolation, in./yr. 11.11 10.91 8.47 8.16 8.24 37,500 7,700 4,000 18,000 12,600 Flow Length, ft. 3.66 1.48 6.38 4.51 Flow Line Percolation, in. 18.1. 302 240 222 162 165 TDS of Percolation, mg/1

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TABLE 5
Population Trends in Spokane Valley

Planning Unit	1975 Population	2020 Population
SV-3	30,000	43,800
SV-5	2,000	3,267
SV-6	1,000	1,575
SV-7	1,900	3,560
SV-8	2,300	5,190
Tota	37,200	57,392

TABLE 6
Summary of Annual Water Balance and Percolation Quality Values for Spokane Valley Planning Units at Year 2020

	·		Plannin	g Unit	
	sv3	SV5	sv6	SV7	sv8
Mean Annual Precipitation, in.	19.5	21.0	21.0	22.0	22.0
Septic Tank Effluent, in.	6.34	1.31	1.07	0.36	0.48
Lawn Irrigation, in.	10.09	4.60	.2.95	1.72	2.54
Agricultural Irrigation, in.	0.21	4.71	2.99	4.75	3.33
Total Applied Water, in.	36.14	31.62	28.01	28.83	28.35
Percolation, in./yr.	15.39	9.42	8.09	8.68	8.31
Flow Length, ft.	37,500	7,700	4,000	18,000	12,600
Flow Line Percolation, in.	25.1	3.16	1.41	6.79	4.55
TDS of Percolation, mg/l	304	239	200	160	162

TABLE 7A

SELECTED GROUNDWAFER QUALITY AND
WELL DATA FOR SPOKANE VALLEY
SEPTEMBER 1973

Well Location	Well Penetra- tion below Water Table, ft.	TDS, mg/1*	Distance from Idaho State Line, mi.
26/45, 36Q1	-3	155	0.7
26/45, 36N1	23	158	1.1
26/45, 35F1	117	151	1.8
25/45, 15D1	75	157	3 1
25/44, 1J1	77	160	6.9
25/44, 2Q1	45	174	7.6
25/44, 7C1	31	206	11.9
25/44, 18D1	42	193	12.1
25/44, 19D1	8	206	12.1
25/43, 13A1	41	170	12.3
25/43, 14K1	31	152	13.6

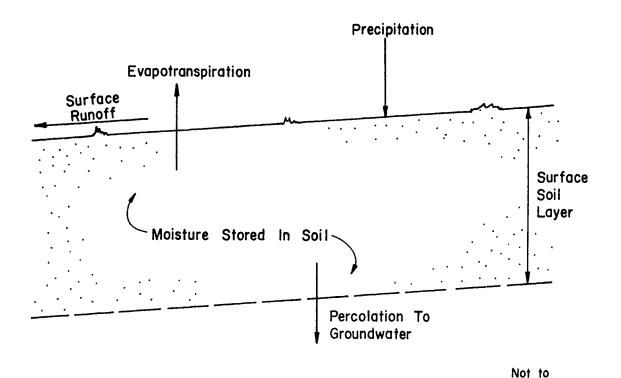
^{*}Source USGS-EPA Program measured on September 25-26, 1973.

TABLE 7B
SELECTED GROUNLWATER QUALITY AND
WELL DATA FOR SPOKANE VALLEY
SEPTEMBER 1971-72

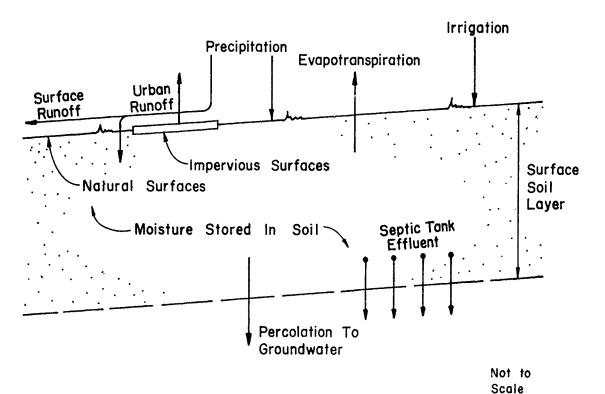
Well Location	Well Penetra- tior below Water Table, ft.	TDS, mg/1*	Distance from Idaho State Line, mi.
26/46, 31M	110	120	0.2
25/45, 15D1	75	155	3.1
25/45, 18R	130	105	5.5
25/44, 26L1·	0-32**	195	7.8
25/44, 27E1	63	164	9.3
25/44, 16E1	28	174	10.2
25/44, 29A1	5-26**	206	10.5
25/43, 12H1	35	210	12.3
25/43, 24L	9	216	12.8
25/43, 23A1	80	182	13.3
25/43, 23A2	74	196	13.3

*Source: Spokane County Health District
Mean values for September 1971 and 1972 computed from r
conductivity using a factor of 0.56.

**Perforated casing.



NATURAL CONDITIONS



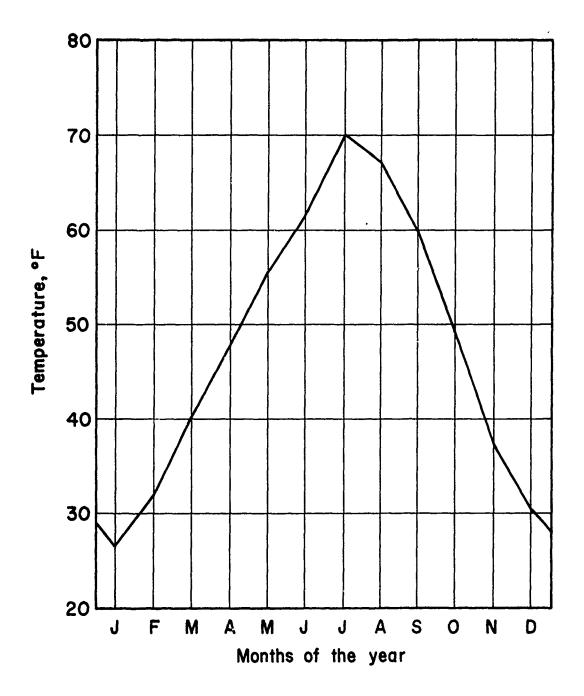
DEVELOPED CONDITIONS

WATER RESOURCES STUDY
METROPOLITAN SPOKANE REGION
Dept. of the Army, Seattle District
Corps of Engineers
Kennedy - Tudor Consulting Enquineers

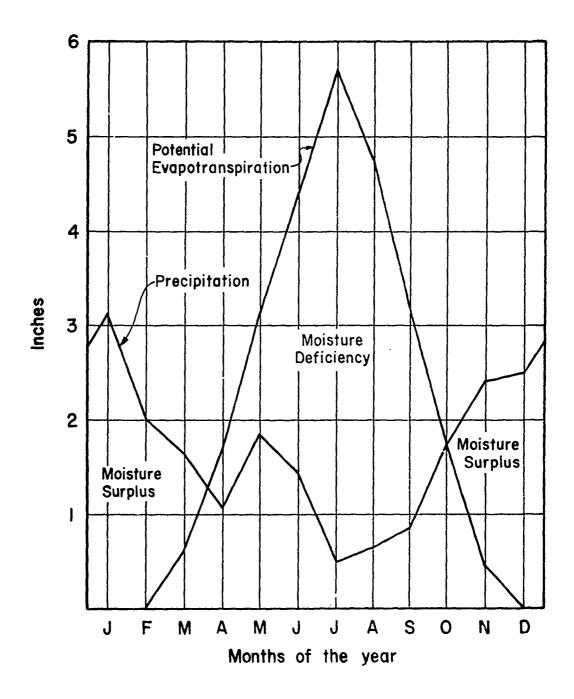
Schematic Diagram of Water Balance for a Surface Soil Layer under Natural and Developed Conditions

Figure A the book of the constitution of the constitution of the second of the constitution of

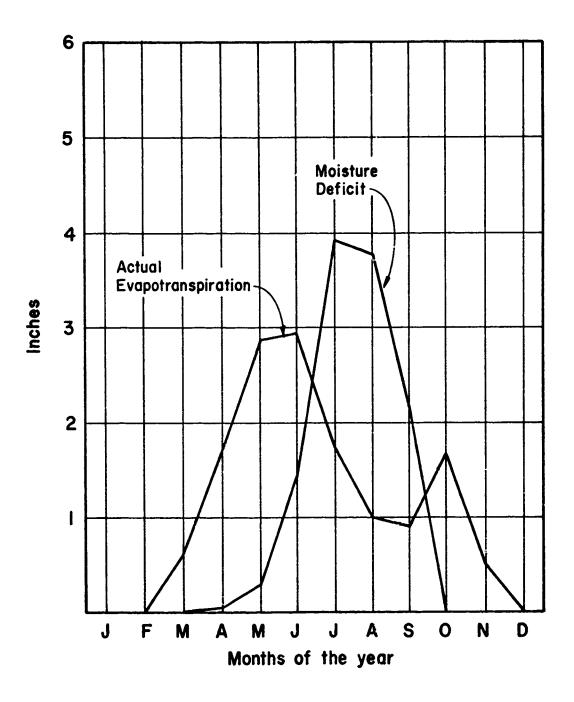
Scale



Monthly Mean Air Temperatures for Spokane Valley

Figure B 

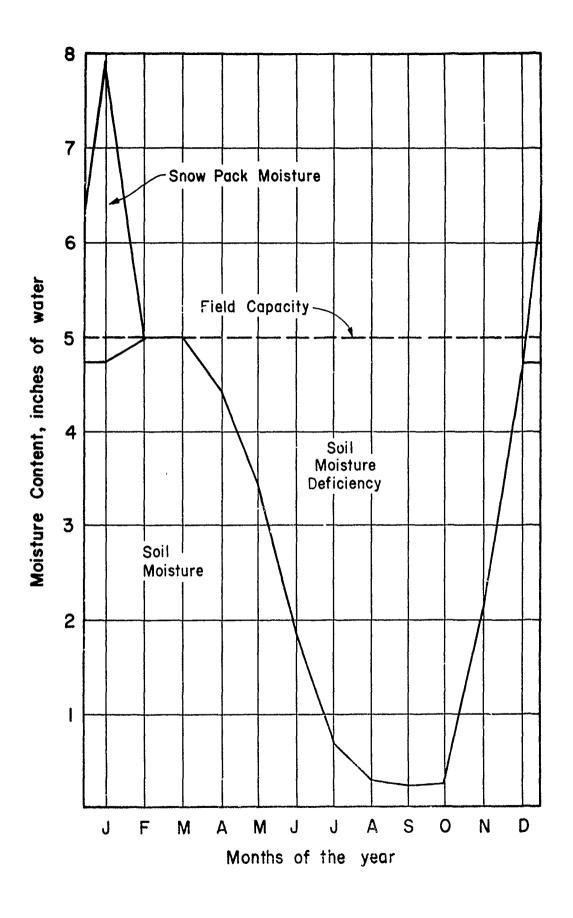
Monthly Mean Precipitation and Potential Evapotranspiration for Spokane Valley

Figure C 

Monthly Actual Evapotranspiration and Moisture Deficit for Natural Conditions in Spokane Valley

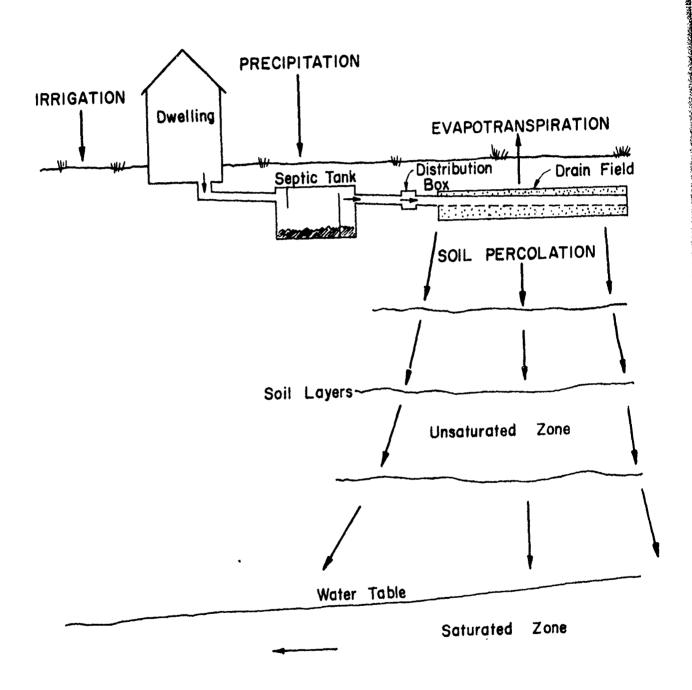
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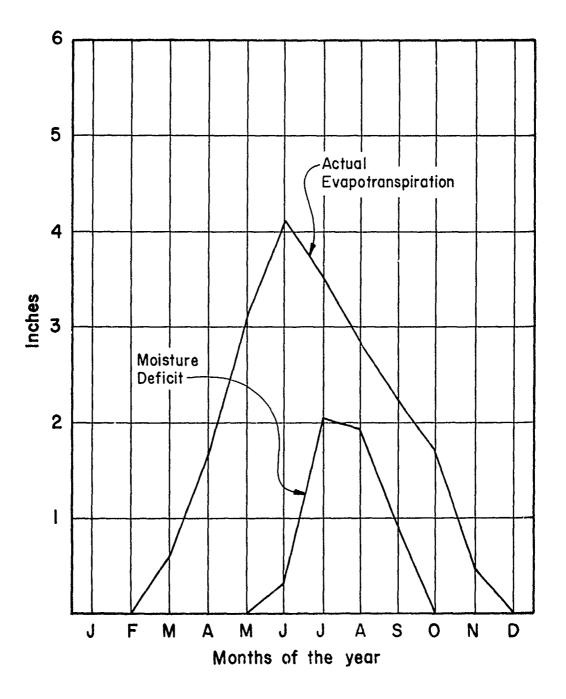
Monthly Soil and Snow Pack Moisture Storage for Natural Conditions in Spokane Valley

Figure E



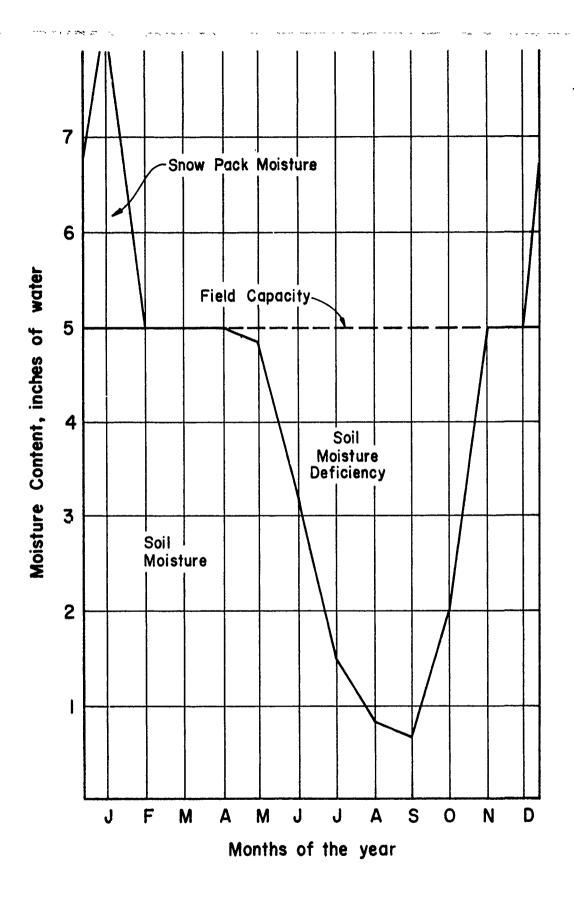
Disposal of Household Wastewater through a Conventional Septic Tank - Drainfield System

Figure F



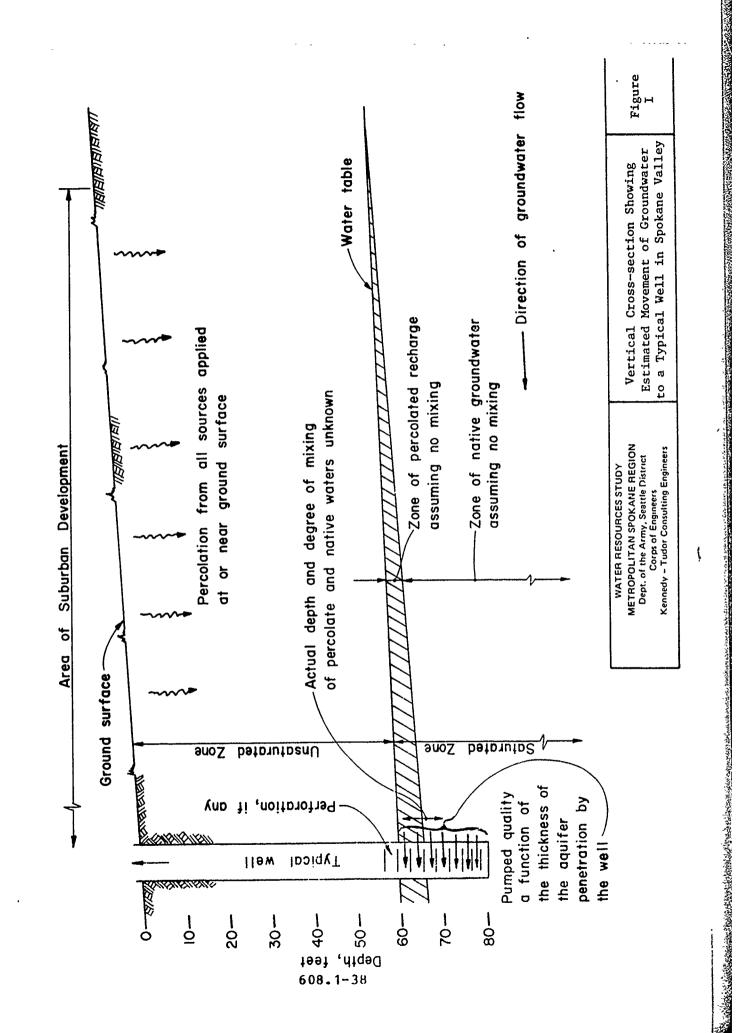
Monthly Actual Evapotranspiration and Moisture Deficit for Spokane Valley under Generalized Suburban Conditions

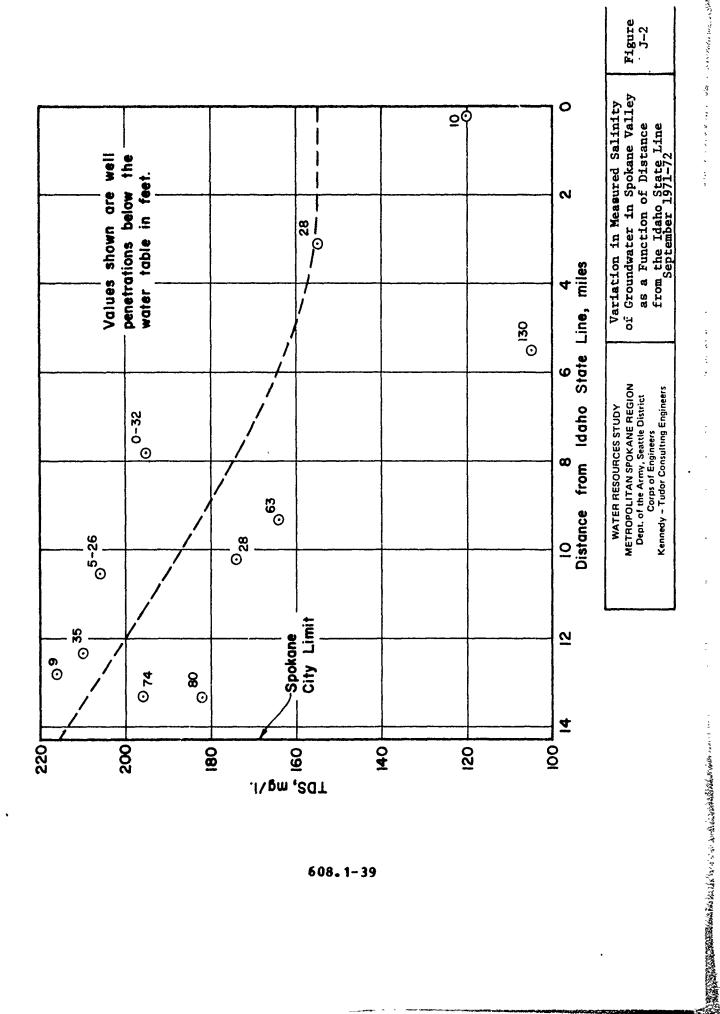
Figure G



Monthly Soil and Snow Pack Moisture Storage for Spokane Valley under Generalized Suburban Conditions

Figure H

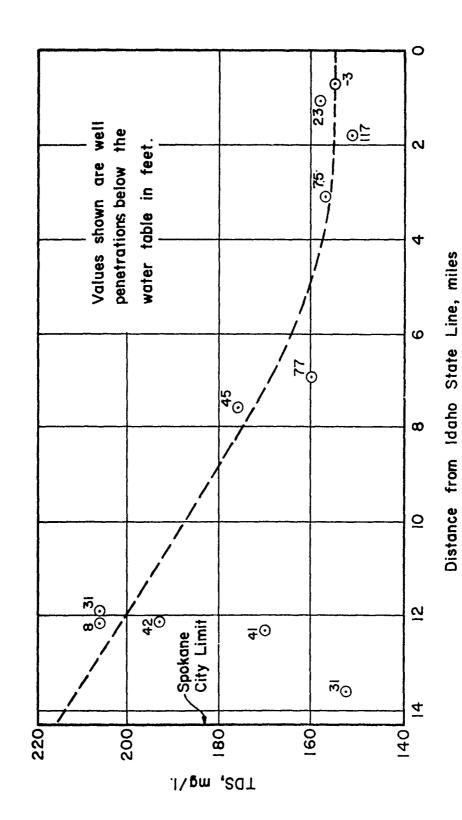




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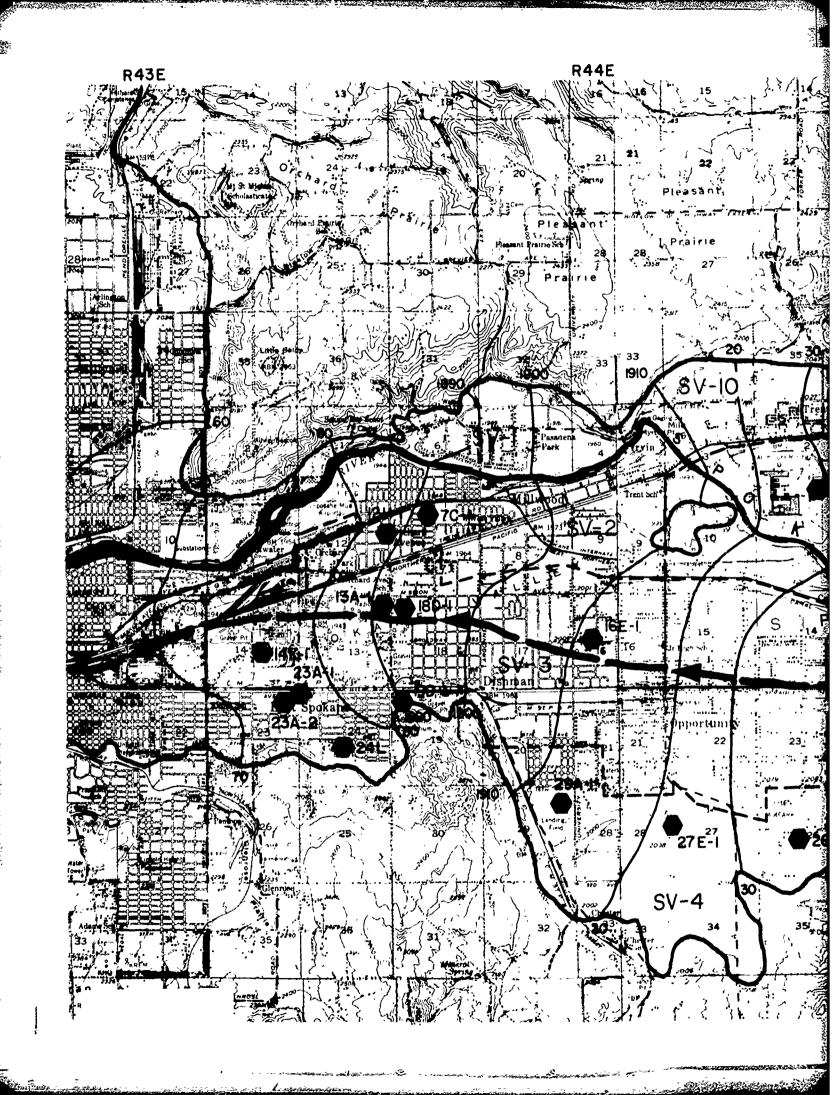
Figure J-1

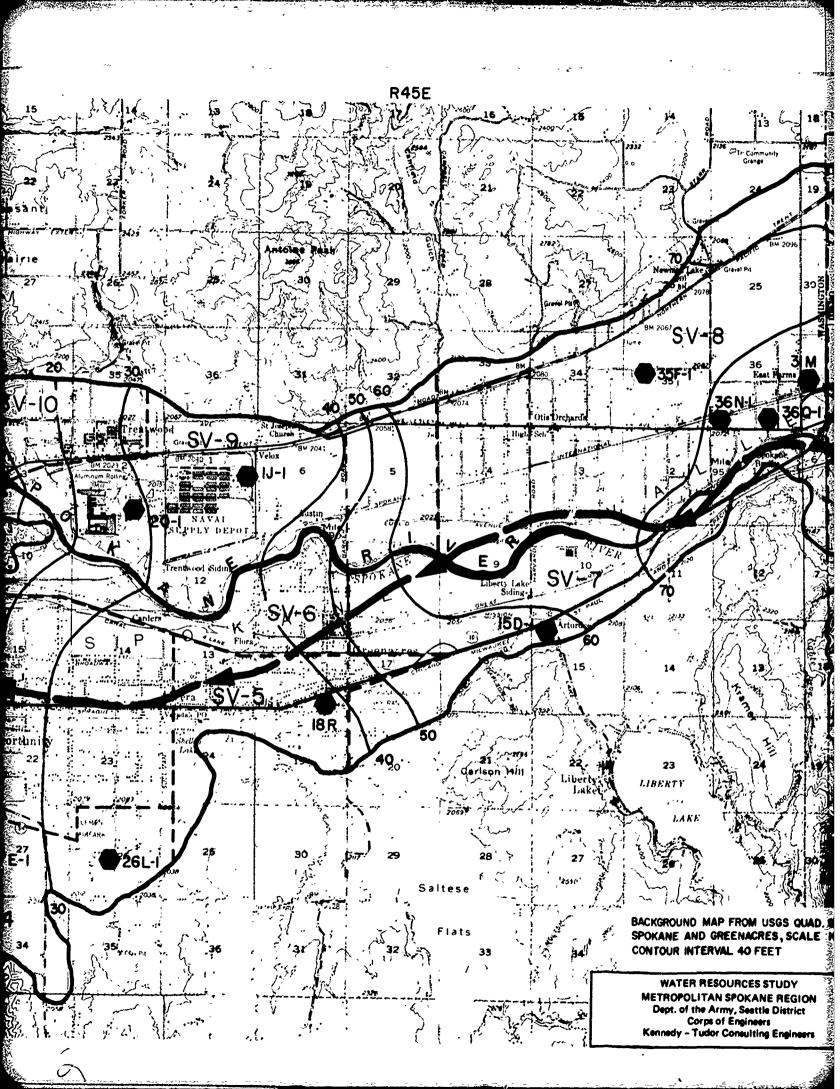
Variation in Measured Salinity of Groundwater in Spokane Valley

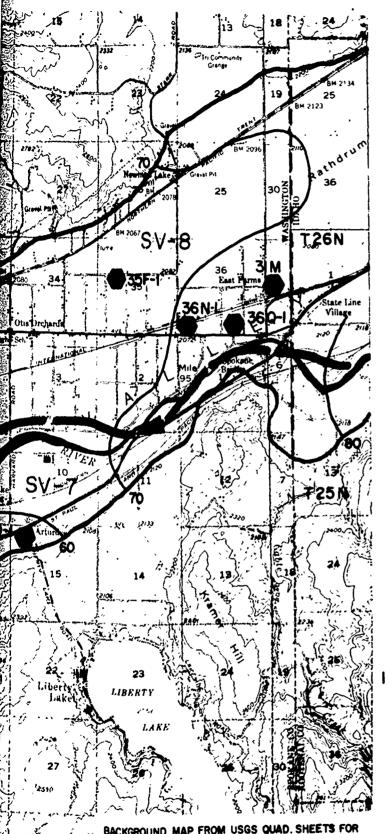
as a Function of Distance from the Idaho State Line September 1973

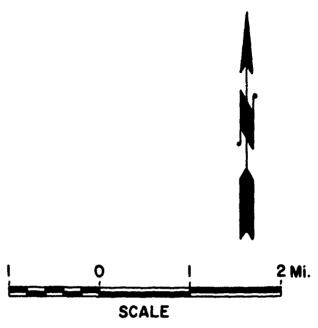
METROPOLITAN SPOKANE REGION Bept. of the Army, Seattle District Corps of Engineers Kennedy - Tudor Consulting Engineers

WATER RESOURCES STUDY









LEGEND

SPOKANE VALLEY
AQUIFER BOUNDARY

CONTOURS OF WATER
TABLE, FEET ELEVATION,
SEPTEMBER CONDITION

SELECTED GROUNDWATER
FLOW LINE FOR WATER
QUALITY ANALYSIS

-- PLANNING UNIT BOUNDARY

SV-3 PLANNING UNIT IDENTIFICATION

LOCATION OF SELECTED WELLS & USGS NO.

BACKGROUND MAP FROM USGS QUAD. SHEETS FOR SPOKANE AND GREENACRES, SCALE 1:62,500 CONTOUR INTERVAL 40 FEET

WATER RESOURCES STUDY
METROPOLITAN SPOKANE REGION
Dept. of the Army, Seettle District
Corps of Engineers
Kennedy – Tudor Consulting Engineers

SPOKANE VALLEY WATER TABLE CONTOURS AND GROUNDWATER FLOW LINE

FIG. K

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